

THE
MERCK
INDEX
★
THIRTEENTH EDITION

THE MERCK INDEX

AN ENCYCLOPEDIA OF
CHEMICALS, DRUGS, AND BIOLOGICALS

THIRTEENTH EDITION

Editorial Staff

Maryadele J. O'Neil, *Senior Editor*

Ann Smith, *Senior Associate Editor*

Patricia E. Heckelman, *Associate Editor*

John R. Obenchain Jr., *Editorial Assistant*

Jo Ann R. Gallipeau, *Technical Assistant*

Mary Ann D'Arecca, *Administrative Associate*

Susan Budavari, *Editor Emeritus*

*Published by
Merck Research Laboratories
Division of*

MERCK & CO., INC.
Whitehouse Station, NJ

2001

MERCK & CO., INC.
Whitehouse Station, NJ
USA

1st Edition—1889
2nd Edition—1896
3rd Edition—1907
4th Edition—1930
5th Edition—1940
6th Edition—1952
7th Edition—1960
8th Edition—1968
9th Edition—1976
10th Edition—1983
11th Edition—1989
12th Edition—1996

Library of Congress Catalog
Card Number 89-60001
ISBN Number 0911910-13-1

Copyright © 2001 by MERCK & CO., INC.
All rights reserved. Copyright under the Universal Copyright Convention
and the International Copyright Convention.
Copyright reserved under the Pan-American Copyright Convention.

Printed in the USA

Xylidine

Brown crystals from alcohol.

USE: Indicator, used in 0.02% soln: pH 1.2 red, 2.8 yellow, 9.6 blue.

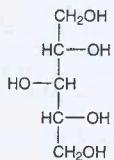
10139. Xylidine. [1300-73-8] *α,α-Dimethylbenzeneamine; dimethylaniline; aminodimethylbenzene.* C₈H₁₁N; mol wt 121.18. C 79.29%, H 9.15%, N 11.56%. (CH₃)₂C₆H₃NH₂. There are six isomeric xylidines. Prep'd by reduction of corresponding nitro-compounds: Allchin, US 1867962 (to I.C.I.); physical properties of all six isomers also given: Birch et al., J. Am. Chem. Soc. 71, 1362 (1949); van Loon et al., Rec. Trav. Chim. 79, 977 (1960); Bergmann, Berkovic, J. Org. Chem. 26, 919 (1961).

All except *o*-4-xylidine are liquids above 20°. d 0.97-0.99, bp 213-226°. They are sparingly sol in water, sol in alcohol and form more or less sol salts with the strong mineral acids.

Caution: Potential symptoms of overexposure are anoxia, cyanosis, methemoglobinemia; lung, liver and kidney damage. See NIOSH Pocket Guide to Chemical Hazards (DHHS/NIOSH 97-140, 1997) p 336.

USE: Chiefly in the manuf of dyes.

10140. Xylitol. [87-99-0] *xylo-Pentane-1,2,3,4,5-pentol; xylite; Eutrit; Kannit; Klinit; Kylit; Newtol; Torch; Xyliton.* C₅H₁₂O₅; mol wt 152.15. C 39.47%, H 7.95%, O 52.58%. Intermediate in metabolism of D-glucose through glucuronate cycle in livers. Prep'd by reduction of xylose: G. Bertrand, Bull. Soc. Chim. France [3] 5, 555 (1891); E. Fischer, R. Stahel, Ber. 24, 538 (1891). Prepn of metastable crystals: M. L. Wolfrom, E. J. Kohn, J. Am. Chem. Soc. 64, 1739 (1942); of stable form: J. F. Carson et al., ibid. 65, 1777 (1943). Crystal structure: H. S. Kim, G. A. Jeffrey, Acta Crystallogr. 25B, 2607 (1969). Use in prevention of dental caries: E. Grunberg et al., Int. J. Vit. Nutr. Res. 43, 227 (1973); A. Scheinin, K. K. Makinen, DE 2606533 (1976 to Hoffmann-La Roche), C.A. 85, 149140h (1976). Acute toxicity: S. Salminen et al., Toxicol. Lett. 18, Suppl. 1, 37 (1983). Reviews of toxicity, metabolism and use as dietary additive: International Symposium on Metabolism, Physiology and Clinical Uses of Pentoses and Pentitols, B. L. Horecker et al., Eds. (Springer-Verlag, New York, 1969) 408 pp; Sugars in Nutrition, H. L. Sipple, K. W. McNutt, Eds. (Academic Press, New York, 1974) *passim*; G. E. Demetrakopoulos, H. Amos, World Rev. Nutr. Diet 32, 96-122 (1978); R. Ylikahri, Adv. Food Res. 25, 159-180 (1979). Book: Xylitol, J. N. Counsel, Ed. (Applied Science, London, 1978) 191 pp.

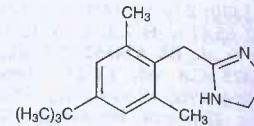


Stable form: orthorhombic needles from THF, prisms from ethanol; mp 93-94.5°; d 1.52. Metastable form: colorless, monoclinic, lath-shaped crystals from anhydrous methanol; hygroscopic; mp 61-61.5°. Solv of stable form (g/100 g soln): abs methanol 6.0; abs ethanol 1.2; water 64.2. Relative sweetness equal to sucrose. LD₅₀ orally in mice: approx 22 g/kg (Salminen).

USE: As oral and intravenous nutrient; in anticaries preparations.

10141. Xylometazoline. [526-36-3] *2-[[4-(1,1-Dimethyl-ethyl)-2,6-dimethylphenyl]methyl]-4,5-dihydro-1*H*-imidazole; 2-(4-*tert*-butyl-2,6-dimethylbenzyl)-2-imidazoline.* C₁₆H₂₄N₂; mol wt 244.37. C 78.64%, H 9.90%, N 11.46%. α -Adrenergic agonist; topical vasoconstrictor. Prepn: Hüni, US 2868802 (1959 to Ciba). Pharmacology: S. Morimoto, H. Tanaka, Osaka Shiritsu Daigaku Igaku Zasshi 18, 211 (1969), C.A. 72, 20437n (1970). Comprehensive description: Y. Golander, W. J. DeWitte, Anal. Prof. Drug. Subs. 14, 135-156 (1985). GC determin in plasma and urine: A. Sioufi et al., J. Chromatog. 487, 81 (1989). Clinical trial in allergic rhinitis: M. Fradis et al., J.

Laryngol. Otol. 101, 666 (1987); in nasal surgery: J. P. Campbell et al., Otolaryng. Head Neck Surg. 107, 697 (1992).

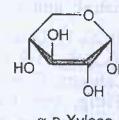


mp 131-133°.

Hydrochloride. [1218-35-5] Neo-Rinoleina; Novorin; Olynd; Otriven; Otrivin; Xymelin. C₁₆H₂₄N₂·HCl; mol wt 280.84. Solv in water: up to 3%; also sol in methanol, ethanol. Practically insol in ether, benzene.

THERAP CAT: Decongestant.

10142. Xylose. [58-86-6] *D-Xylose; wood sugar; Xylo-med; Xylo-Pfan.* C₅H₁₀O₅; mol wt 150.13. C 40.00%, H 6.71%, O 53.29%. Widely distributed in plant materials, especially in wood (maple, cherry), in straw, in hulls. Not found in free state, but in form of xylan, a polysaccharide built from D-xylose units and occurring in association with cellulose. Xylose occurs also as part of glycosides. Isoln from corn cobs by boiling with 8% H₂SO₄: Monroe, J. Am. Chem. Soc. 41, 1002 (1919). Peanut shells and cottonseed hulls also are practical sources of xylose. Ling, Nanji, J. Chem. Soc. 1923, 620. Configuration: Hudson, Yanovsky, J. Am. Chem. Soc. 39, 1029 (1917); Haworth, Nature 116, 430 (1925). Review on history, constitution and prep: Harding, Sugar 24, 14 (1922).



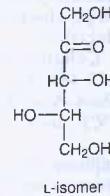
α -D-Xylose

Monoclinic needles or prisms. Very sweet taste. mp 144-145° (Wheeler, Tollens, Ann. 254, 309); mp 153-154° (Heben, Compt. Rend. 110, 970). d₄²⁰ 1.525. Shows mutarotation: [α]_D²⁰ +92° → +18.6° (16 hrs c = 10). One gram dissolves in 0.8 ml water. Sol in pyridine, hot alcohol. pKa (18°): 12.14. Reduces warm Fehling's soln. Upon heating with water in closed tube to 140° or by boiling with dil H₂SO₄, furfural is formed.

USE: In tanning, dyeing, and as a diabetic food.

THERAP CAT: Diagnostic aid (intestinal function).

10143. Xylulose. [5962-29-8] *threo-Pentulose.* C₅H₁₀O₅; mol wt 150.13. C 40.00%, H 6.71%, O 53.29%. L-Form has been found in the urine of humans with pentosuria. Prep of DL-form: Gascoigne, Chem. & Ind. (London) 1959, 402; of D-form: Mendicino, J. Am. Chem. Soc. 82, 4975 (1960); of L-form: Wolfrom, Bennett, J. Org. Chem. 30, 458 (1965). Isol of DL-form from the acid hydrolysate of bagasse hemicellulose: Banerjee et al., Sci. Cult. (Calcutta) 27, 498 (1961), CA 56, 11682d (1962). Enzymic prep of L-form: Hough, Jours. Chem. & Ind. (London) 1952, 907; eidem, J. Chem. Soc. 1952, 4047. Formation of L-form in normal humans and guinea pigs and its utilization by guinea-pig liver preps: Touster et al., J. Am. Chem. Soc. 76, 5005 (1954). Reviews: The Carbohydrates, W. Pigman, Ed. (Academic Press, New York, 1957) pp 80-87, 759, 795; Methods in Carbohydrate Chemistry vol. 1, R. L. Whistler, M. L. Wolfrom, Eds. (Academic Press, New York, 1962) pp 94-101.



L-isomer

p-Isomer.
p-Isomer p-
yellow crystals
+24° (15 min)
l-Isomer. S
L-Isomer p-
dil alc, mp 128°
in ethanol).

**10144. 1-Xylylphenylazo]-2-
xylylazo)-2-napl
& C Red no. 14
78.24%. H 5.84%
xylylazo deriv. j
2-naphthol: J. N
Smyth, G. G. Mc
Ginn, J. L. Radc**

Red needles, mp
benzene.

10145. Xylyl
51.92%, H 4.90%,
Org. Chem. 17, 522
Soc. 32, 403 (1955)
Soc. 1960, 3340.