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10/551,205	11/14/2006	Nicholas S. Bodor	0056192-000024	4092
	7590 04/04/200 INGERSOLL & ROOI		EXAM	INER
POST OFFICE			LAU, JON	ATHAN S
ALEAANDKIA	A, VA 22515-1404		ART UNIT	PAPER NUMBER
		1623		
			NOTIFICATION DATE	DELIVERY MODE
			04/04/2008	ELECTRONIC

# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

ADIPFDD@bipc.com

		Ар	plication No.	Appl	icant(s)	
	Office Action Comment	10	/551,205	BOD	OR ET AL.	
	Office Action Summary	Exa	aminer	Art U	Jnit	
			athan S. Lau	1623		
Period fo	The MAILING DATE of this communica or Reply	ation appears	on the cover sheet	with the corresp	oondence ad	ldress
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Status						
1) 🛛	Responsive to communication(s) filed	on <i>04 Janua</i>	ry 2008.			
•	•		on is non-final.			
′—	Since this application is in condition fo	r allowance e	except for formal ma	atters, prosecut	ion as to the	e merits is
7—	closed in accordance with the practice		•	-		
Dispositi	on of Claims	·	•			
•	Claim(s) <u>1-35 and 56-98</u> is/are pending	a in the annli	action			
•—	4a) Of the above claim(s) <u>13-35 and 63</u>	•		idoration		
	Claim(s) is/are allowed.	<u>1-01</u> 15/416 WI	thurawit from cons	deradori.		
· —	Claim(s) <u>1-12,56-66 and 82-98</u> is/are r	raiactad				
·	Claim(s) is/are objected to.	ejected.				
•	Claim(s) are subject to restriction	on and/or ele	ction requirement			
0)[]	ciaiii(s) are subject to restricte	on and/or ele	cuon requirement.			
Applicati	on Papers					
9)🛛	The specification is objected to by the I	Examiner.				
10)🛛	The drawing(s) filed on <u>28 Se<i>ptember</i> .</u>	<u>2005</u> is/are:	a)⊠ accepted or b	) ☐ objected to	by the Exar	miner.
	Applicant may not request that any objection	on to the draw	ing(s) be held in abey	ance. See 37 Cl	FR 1.85(a).	
	Replacement drawing sheet(s) including the	ne correction is	required if the drawi	ng(s) is objected	to. See 37 CI	FR 1.121(d).
11) 🔲	The oath or declaration is objected to b	y the Examir	ner. Note the attach	ed Office Action	n or form P7	ГО-152.
Priority u	ınder 35 U.S.C. § 119					
12)	Acknowledgment is made of a claim fo	r foreign prio	rity under 35 U.S.C	. § 119(a)-(d) o	r (f).	
a)[	☐ All b)☐ Some * c)☐ None of:	_	-			
	1. Certified copies of the priority do	ocuments hav	ve been received.			
	2. Certified copies of the priority do	ocuments hav	e been received in	Application No	)	
	3. Copies of the certified copies of	the priority d	ocuments have be	en received in tl	his National	Stage
	application from the Internationa	al Bureau (PC	T Rule 17.2(a)).			
* S	See the attached detailed Office action	for a list of th	e certified copies n	ot received.		
Attachmen						
	e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTC	J_Q/8\		w Summary (PTO-4 lo(s)/Mail Date		
	nation Disclosure Statement(s) (PTO/SB/08)	<i>-</i> 070 <i>)</i>	5) 🔲 Notice o	of Informal Patent A		
Pape	Paper No(s)/Mail Date <u>See Continuation Sheet.</u> 6) Other:					

 $Continuation \ of \ Attachment(s)\ 3).\ Information \ Disclosure \ Statement(s)\ (PTO/SB/08),\ Paper\ No(s)/Mail\ Date : 11\ pgs/14Nov2006,\ 10Aug2007,\ 8Nov2007,\ 4Jan2008.$ 

## **DETAILED ACTION**

This application is the national stage entry of PCT/US04/09387, filed 26 Mar 2004; and claims benefit of provisional application 60/458,922, filed 28 Mar 2003; and claims benefit of provisional application 60/484,756, filed 02 July 2003; and claims benefit of provisional application 60/541,247, filed 04 Feb 2004.

Claims 1-35 and 56-98 are pending in the current application. Claims 13-35 and 67-81, drawn to non-elected inventions, are withdrawn. Claims 1-12, 56-66, and 82-98 are examined on the merits herein.

However, the parent applications provisional application 60/458,922, filed 28 Mar 2003; provisional application 60/484,756, filed 02 July 2003; and provisional application 60/541,247, filed 04 Feb 2004; upon which priority is claimed fail to provide adequate support under 35 U.S.C. 112 for the instant claims 1-12, 56-66, and 82-98 of this application since parent applications 60/458,922, filed 28 Mar 2003; and 60/484,756, filed 02 July 2003 are not seen to disclose the amorphous cladribine-cyclodextrin complex of in the independent claims 1, 56 and 82. Written description for claims 1-11 and 56-65 may be found in provisional application 60/541,247, filed 04 Feb 2004, however no support is found for the percent by weight present in the inclusion complex and the non-inclusion complex of instant claims 12 and 66, the temperature range of about 40 to about 80 °C of claims 82 and 83, the temperature range of about 45 to about 50 °C of claim 85, or the temperature range of about -40 to about -80 °C of claim 89. Thus, the filing date of the instant claims 12, 66, 82, 83, 85, 88 and 89 are deemed to be the filing date of the instant application, 14 Nov 2006. The filing date of instant

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claims 1-11, 56-65, 84, 86 and 87 are deemed to be the filing date of provisional application 60/541,247, filed 04 Feb 2004. If applicant disagrees, applicant should present a detailed analysis as to why the claimed subject matter has clear support in the earlier priority applications. Applicant is reminded that such priority for the instant limitations requires written description and enablement under 35 U.S.C. § 112, first paragraph.

#### Election/Restrictions

Applicant's election with traverse of the invention of Group I, claims 1-12, 56-66, and 82-98, in the reply filed on 04 Jan 2008 is acknowledged. The traversal is on the ground(s) that the amorphous nature of the complex is part of the special technical feature of the cladribine-cyclodextrin complex in a solid oral dosage form. This is not found persuasive because Schultz et al. (US Patent 6,194,395, of record) explicitly discloses a cladribine-cyclodextrin complex in a solid oral dosage form (column 5, lines 50-52). Further, Schultz et al. references the method of making said solid oral dosage form disclosed in WIPO Publication WO97/18839 (cited in PTO-892), which is drawn to the embodiment wherein the melt-extruded forms consist essentially of amorphous material (page 8, lines 14-15). Therefore WIPO Publication WO97/18839 provides evidence that it was recognized in the prior art that the product disclosed by Schultz et al. inherently includes amorphous cladribine-cyclodextrin complex in a solid oral dosage form. While the International Search Report is factually correct in stating that Schultz et al. is silent about specific ratios of cladribine to cyclodextrin and amorphous forms, a

patent need not teach, and preferably omits, what is well known in the art. By referencing WIPO Publication WO97/18839 Schultz et al. demonstrates that the <u>amorphous</u> cladribine-cyclodextrin complex produced by the melt-extrusion method is well known in the prior art. Finally, to address the scientific issue regarding the equilibrium presence of both the inclusion and non-inclusion complex, while the mathematical equation for the equilibrium of the cladribine-cyclodextrin inclusion complex and the cladribine-cyclodextrin non-inclusion complex would be different for cladribine and cyclodextrin in a solvent versus cladribine and cyclodextrin in a molten state, ie. a liquid mixture absent solvent, the equilibrium and thus equilibrium products would still be present.

The requirement is still deemed proper and is therefore made FINAL.

Claims 13-35 and 67-81 are withdrawn from further consideration pursuant to 37 CFR 1.142(b), as being drawn to a nonelected inventions, there being no allowable generic or linking claim. Applicant timely traversed the restriction requirement in the reply filed on 04 Jan 2008.

# Specification

The disclosure is objected to because of the following informalities:

- a) The blanks identifying the provisional patent application numbers on page 23, lines 25 and 27 must be replaced with the application numbers.
  - b) The minor typographical error "comples" on page 22, line 12.

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Appropriate correction is required.

# Claim Rejections - 35 USC § 112

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

Claims 2, 11 and 57 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

The term "saturated" in claims 2, 11 and 57 is a relative term which renders the claim indefinite. The term "saturated" is not defined by the claim, the specification does not provide a standard for ascertaining the requisite degree, and one of ordinary skill in the art would not be reasonably apprised of the scope of the invention. The following definition for "saturated" is provided in the specification, page 10 lines 1-13:

The term "saturated" when used in conjunction with a complex of cladribine in amorphous cyclodextrin means that the complex is saturated with cladribine, that is, the complex contains the maximum amount of cladribine which can be complexed (by means of both inclusion and non-inclusion complexes) with a given amount of cyclodextrin under the conditions of complexation used. A phase solubility study can be used to provide this information, as described in more detail hereinafter. (Conditions for the complexation are also described in more detail below.) Alternatively, a saturated complex may be arrived at empirically by simply adding cladribine to an aqueous solution of the selected cyclodextrin until no more cladribine goes into solution; ultimately, excess cladribine, if any, is removed (by filtration or centrifugation) and the solution lyophilized to provide the dry saturated complex.

The saturated complex is defined in relation to a maximum amount of cladribine which can be complexed under the conditions of complexation used. However, this amount is defined only empirically. A saturated aqueous solution is invoked with regard to this empirically defined maximum amount of cladribine which can be complexed, but the

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claims are drawn to a saturated complex in a solid oral dosage form, not a saturated solution. Therefore one of ordinary skill in the art would not be reasonably apprised of the scope of the invention because the maximum amount would have to be determined empirically for each composition. For the purpose of furthering prosecution, Examiner has interpreted the "maximum amount of cladribine which can be complexed" to be the weight ratio of 1:10 for the cladribine:cyclodextrin complex, based on guidance given on page 31, lines 18-20.

Claim 2 recites the limitation "the complex" in line 2. There is insufficient antecedent basis for this limitation in the claim. It is unclear which complex is referred to by the term "the complex," the inclusion complex (a), the non-inclusion complex (b), both complexes or the complex cladribin-cyclodextrin complex.

Similarly, claim 11 recites the limitation "saturated complexes" in line 3. There is insufficient antecedent basis for this limitation in the claim. It is unclear which complex is referred to by the term "complexes," the inclusion complex (a), the non-inclusion complex (b), both complexes or the complex cladribin-cyclodextrin complex.

The term "a point located on a phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin" in claim 11 is a relative term which renders the claim indefinite. The term "a point located on a phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin" is not defined by the claim, the specification does not provide a standard for ascertaining the requisite degree, and one of ordinary skill in the art would not be reasonably apprised of the scope of the invention. The broadness of the term "a point

located on a phase solubility diagram" does not necessarily render the term indefinite. However, no standard is given for what "phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin" is referred to in the claim, such as what temperature, pressure, or solvent this phase solubility diagram describes. Therefore one of ordinary skill in the art would not be reasonably apprised of the scope of the invention from the term "a point located on a phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin".

# Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1-5, 11, 56-60, 82-90 and 94-98 are rejected under 35 U.S.C. 102(b) as being anticipated by Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) as evidenced by Baert et al. (WIPO Publication WO97/18839, published 29 May 1997, cited in PTO-892).

Schultz et al. discloses a solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin (column 2, lines 31-39), meeting the limitations of instant claims 1 and 56. The disclosed product is substantially identical to a product-by-process meeting the limitations of instant claims 82-90 and 94-96. Schultz et al. discloses the use of  $\beta$ - and  $\gamma$ -cyclodextrins (column 2, lines 56-58) and derivatives

wherein one or more cyclodextrin hydroxy groups are replaced with groups such as methyl, hydroxypropyl, carboxymethyl (column 3, lines 26-27) or sulfobutylcyclodextrins (column 4, lines 22-24), meeting the limitations of instant claims 3-5 and 58-60. The phrase "one or more cyclodextrin hydroxy groups" combined with the absence of specific structural details of which hydroxyl group is substituted with a methyl group meets the limitation of "randomly methylated β-cyclodextrins" of instant claims 3 and 58. Schultz et al. discloses the solid oral dosage form in the form of a tablet (column 5, lines 37-38) including the excipients sorbitol and magnesium stearate (column 6, lines 2-7), disclosing a product that is substantially identical to a product-by-process meeting the limitations of instant claims 97 and 98. Schultz et al. discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, which would lead one of skill in the art to instantly envision a cladribine to cyclodextrin ratio ranging from 15 mg:100 mg to 15mg:500 mg, or 1:6.67 to 1:33.3 by weight (column 6, lines 23-31). These values molar ratios that correspond to "a point located on a phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin," meeting the limitations of instant claim 11. The language of instant claim 11 as disclosed requires only that the point be located on a phase solubility diagram for said complexes, not that the point be located on the curve defining a saturated complex such as the curve disclosed in the Figure, meaning that any composition according to claim 1 necessarily meets the limitations of instant claim 11 as disclosed. The instant specification suggests that maximum amount of cladribine which can be complexed gives a weight ratio of 1:10 for the cladribine:cyclodextrin complex. Therefore a

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composition comprising the cladribine:cyclodextrin complex that contains a cladribine to cyclodextrin ratio of 1:6.67 describes a composition that comprises a "saturated" complex and meets the limitations of instant claims 2 and 57. The open language of "comprising" allows for the presence of uncomplexed cladribine in the composition.

Schultz et al. incorporates-by-reference the method of making said solid oral dosage form (Schultz et al.column 5, lines 50-52) disclosed in WIPO Publication WO97/18839, Baert et al., which provides evidence in the embodiment wherein the melt-extruded forms consist essentially of amorphous material (Baert et al. page 8, lines 14-15). Therefore Baert et al. provides evidence that it was recognized in the prior art that the product disclosed by Schultz et al. inherently includes amorphous cladribine-cyclodextrin complex in a solid oral dosage form.

To address the scientific issue regarding the equilibrium presence of both the inclusion and non-inclusion complex, while the equation for the equilibrium of the cladribine-cyclodextrin inclusion complex and the cladribine-cyclodextrin non-inclusion complex would be different for cladribine and cyclodextrin in a solvent versus cladribine and cyclodextrin in a molten state due to the lack of a solvent, the equilibrium and thus equilibrium products, the cladribine-cyclodextrin inclusion complex and the cladribine-cyclodextrin non-inclusion complex, would still be inherent in the product disclosed by Schultz et al.

Claims 82-90 and 94-98 are drawn to a product-by-process. The disclosed product is substantially identical to the instantly claimed product-by-process, an amorphous solid pharmaceutical oral dosage form comprising cladribine and

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cyclodextrin. "[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted) (Claim was directed to a novolac color developer. The process of making the developer was allowed. The difference between the inventive process and the prior art was the addition of metal oxide and carboxylic acid as separate ingredients instead of adding the more expensive pre-reacted metal carboxylate. The product-by-process claim was rejected because the end product, in both the prior art and the allowed process, ends up containing metal carboxylate. The fact that the metal carboxylate is not directly added, but is instead produced in-situ does not change the end product.). See MPEP 2113.

# Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of

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the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Claims 1, 6-10, 12, 56, 61-66, 82 and 91-93 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Baert et al. (WIPO Publication WO97/18839, published 29 May 1997, cited in PTO-892).

Schultz et al. as evidenced by Baert et al. discloses as above. Schultz et al. implicitly discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, or a cladribine to cyclodextrin ratio ranging from 1:6.67 to 1:33.3 by weight (column 6, lines 23-31).

Schultz et al. does not specifically disclose the composition comprising a cladribine to cyclodextrin ratio from about 1:10 to about 1:16 (instant claims 6, 7, 10, 61, 62 and 65), or a ratio of about 1:14 (instant claims 8 and 63) or about 1:11 (instant claims 9 and 64). Schultz et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b) (instant claims 12 and 66). Schultz et al. does not specifically disclose the product-by-process wherein 12.00 parts by weight of cladribine and 172.50 parts by

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weight of hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process (instant claim 91 and 93), to give a cladribine to cyclodextrin ratio of 1:14.38.

Schultz et al. does not specifically disclose the product-by-process wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl- $\beta$ -cyclodextrin are introduced in step (i) of the process (instant claim 92), to give a cladribine to cyclodextrin ratio of 1:10.55.

Baert et al. discloses a solid mixture comprising one or more cyclodextrins and an insoluble active ingredient embedded into the cyclodextrin carrier (abstract), and teaches ratios of active ingredient to cyclodextrin of from about 1:100 to 100:1, from about 1:5 and 5:1 and from about 1:3 to 3:1 (page 11, lines 1-5). These ratios are interpreted as mole ratios because Baert et al. teaches the use of different active ingredients with different molecular weights. A mole ratio of active ingredient to cyclodextrin of about 1:3 for cladribine (MW 285.7 g/mol) and  $\beta$ -cyclodextrin (MW 1135 g/mol) gives a ratio by weight of approximately 1:11.9. The ratio of 1:11.9 meets the limitation of both a ratio of about 1:11 and a ratio of about 1:14 according to the non-limiting definition of "about" as a variance of 20% provided in the instant specification page 9, lines 6-11.

It would have been obvious to one of ordinary skill in the art at the time of the invention to practice the solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin disclosed by Schultz et al. in the ratios of cladribine and cyclodextrin taught by Baert et al. One of ordinary skill in the art would be motivated to combine the Schultz et al. and Baert et al. because Schultz et al. incorporates-by-

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reference Baert et al. and because Baert et al. suggests that improving a similar product according to the teachings of Baert et al. has beneficial properties such as high bioavailability and dissolution rate (Baert et al. page 7, lines 25-27). One of ordinary skill in the art would have an expectation of success because the ratios taught by Baert et al. fall within the range of ratios that is implicitly disclosed by Schultz et al. Schultz et al. in view of Baert et al. does not teach the specific cladribine to cyclodextrin ratios of 1:14.38 or 1:10.55, however these ratios are encompassed by the prior art and Baert et al. suggests optimization of the ratio (Baert et al. page 11, lines 1-5). See also MPEP 2144.05 II.A, "Generally, differences in concentration or temperature will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical." Schultz et al. in view of Baert et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b). However, it is well known in the art that the formation of an inclusion complex from a non-inclusion complex is an equilibrium process, and the position of this equilibrium is dependent on the concentrations of the cladribine and cyclodextrin. This molecular inclusion equilibrium is a process inherent in the formation of the inclusion complex in both aqueous solutions and hot melt liquid mixtures, and Baert et al. teaches variation of the ratio of cladribine to cyclodextrin and hence their relative concentration.

It is noted that In re Best (195 USPQ 430) and In re Fitzgerald (205 USPQ 594) discuss the support of rejections wherein the prior art discloses subject matter which

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there is reason to believe inherently includes functions that are newly cited or is identical to a product instantly claimed, namely the position of the equilibrium process governing formation of an inclusion complex and a non-inclusion complex. In such a situation the burden is shifted to the applicants to "prove that subject matter shown to be in the prior art does not possess characteristic relied on" (205 USPQ 594, second column, first full paragraph).

#### Conclusion

No claim is found to be allowable.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan S. Lau whose telephone number is 571-270-3531. The examiner can normally be reached on Monday - Thursday, 9 am - 4 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Shaojia Anna Jiang can be reached on 571-272-0627. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jonathan Lau Patent Examiner Art Unit 1623

/Shaojia Anna Jiang, Ph.D./ Supervisory Patent Examiner, Art Unit 1623

				Application/Control No.		Applicant(s)/Patent Under		
Notice of References Cited			10/551,205	Reexamination BODOR ET AL.				
			Examiner	Art Unit				
				Jonathan S. Lau	1623		Page 1 of 1	
				U.S. P	ATENT DOCUMENTS			
*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Name Classificat		Classification		
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# FOREIGN PATENT DOCUMENTS

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#### **NON-PATENT DOCUMENTS**

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\*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).)

Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

**Notice of References Cited** 

Part of Paper No. 20080325

# **PCT**

# WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 6: (11) International Publication Number: WO 97/18839 A61K 47/48 A1 (43) International Publication Date: 29 May 1997 (29.05.97) (81) Designated States: AL, AM, AU, BB, BG, BR, CA, CN, CZ, PCT/EP96/05118 (21) International Application Number: EE, GE, HU, IL, IS, JP, KG, KR, LK, LR, LT, LV, MD, MG, MK, MN, MX, NO, NZ, PL, RO, SG, SI, SK, TR, TT, 20 November 1996 (20.11.96) (22) International Filing Date: UA, US, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, (30) Priority Data: 95203219.1 23 November 1995 (23.11.95) EP GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). (34) Countries for which the regional or international application was filed: AT et al. Published (71) Applicant (for all designated States except US): JANSSEN With international search report. PHARMACEUTICA N.V. [BE/BE]; Turnhoutseweg 30, B-Before the expiration of the time limit for amending the 2340 Beerse (BE). claims and to be republished in the event of the receipt of amendments (72) Inventors; and (75) Inventors/Applicants (for US only): BAERT, Lieven, Elvire, Colette [BE/BE]; Gouden Boomstraat 52/3, B-8000 Brugge (BE). PEETERS, Jozef [BE/BE]; Sint Corneliusstraat 64, B-2430 Beerse (BE). VERRECK, Geert [BE/BE]; Salvialaan 5, B-2980 Zoersel (BE), (74) Agent: DE CORTE, Filip; Janssen Pharmaceutica N.V., Patent Dept., Turnhoutseweg 30, B-2340 Beerse (BE).

#### (54) Title: SOLID MIXTURES OF CYCLODEXTRINS PREPARED VIA MELT-EXTRUSION

#### (57) Abstract

Process for preparing a solid mixture comprising one or more cyclodextrins and an insoluble active ingredient characterized in that said process comprises a melt-extrusion step, wherein the active ingredient is embedded into the cyclodextrin carrier.

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## SOLID MIXTURES OF CYCLODEXTRINS PREPARED VIA MELT-EXTRUSION

- The present invention involves a process for preparing solid mixtures by melt-extrusion 5 comprising one or more active ingredients, preferably one or more practically insoluble active ingredients and one or more cyclodextrins. The invention further concerns pharmaceutical compositions comprising the above mixture.
- WO 94/11031, published on May 5, 1994, discloses a method of manufacturing a high-10 quality enclosure compound using extrusion techniques. In this document the extrusion of cyclodextrins together with an active ingredient is mentioned. However, the document discloses the use of a wet mixture (i.e. including water or another solvent) to feed into the extruder.

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French patent application 2,705,677 published on December 2, 1994 describes microgranules obtained by extrusion-spheronisation containing a cyclodextrin. The extrusion-spheronisation technique is the combination of an agglomeration technique, i.e. extrusion, and a shaping technique, i.e. the spheronisation. Said patent application actually teaches the formation of microgranulates containing  $\beta$ -cyclodextrin (Kleptose®) and microcrystalline cellulose (Avicel®) and as active ingredients ketoprofen and paracetamol. The extrusion technique used in the above-mentioned patent application consists in preforming a humid mass by forcing said human mass through a nozzle thus forming long strands of extruded material. The document does not mention melt-extrusion at all.

EP 0,665,009, published as international application on April 24, 1994, discloses a method of dislocating the crystalline condition of crystalline medicine by extruding said crystalline material as such, i.e. without any excipient such as cyclodextrins.

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- In J. Pharm. Pharmacolog., vol 44, No 2, pages 73-8, Uekama et al show how amorphous nifedipine powders were prepared by spray-drying with hydroxypropyl-\u00b1cyclodextrins. The document does not mention melt-extrusion.
- 35 In Pharm. Weekbl. Sci. Ed., 1988, vol 10, No 2, page(s) 80-85, Van Doorne et al, the complex formation between β-cyclodextrins and six antimicotic imidazole derivatives was studied. In said study gels and creams comprising antimicotics were prepared

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whereby a 1.8 % solution of  $\beta$ -cyclodextrin was added instead of purified water. There is no mentioning of extrusion at all.

- In J. Antimicrob. Chemother., 1993, vol 32, No 3, pages 459-463, Hostetler *et al* describe the effect of hydroxypropyl-β-cyclodextrin on the efficacy of oral itraconazole in disseminated murine cryptococcosis. In said document the authors describe how itraconazole is solubilized in hydroxypropyl-β-cyclodextrin resulting in a 100 ml solution. There is no mentioning at all of an extrusion process.
- In Jpn. J. Med. Mycol., 1994, vol 35, No 3, page 263-267, Mikami *et al* describe the effect of carrier solvents on the efficacy of oral itraconazole therapy in aspergilossis in mice. Again this document discloses itraconazole being solubilized in hydroxypropyl-β-cyclodextrin. There is no mentioning of extrusion techniques.
- In "Effect of 2-Hydroxypropyl-β-cyclodextrin on Crystallization and Polymorphic Transition of Nifedipine in Solid State", Pharmaceutical research, vol 11, No 12, 1994, Uekama et al. describe a glassy mixture of 2-hydroxypropyl-β-cyclodextrin obtained by heating said mixture and immediately cooling said mixture to 0 degrees Celsius. There is no teaching that this mixture can be extruded.

US 5,009,900 describe glassy matrices that are useful for introducing and/or retaining and/or stabilizing the volatile and/or labile components in cooked and uncooked food products. These glassy matrices comprise chemically modified starch having a dextrose equivalent not greater than about 2; maltodextrin, corn syrup solids or a polydextrose, and a mono- or disaccharide. The document does disclose extrusion to form glassy matrices. However, there is no specific mentioning of cyclodextrins and of therapeutically or pharmaceutically active ingredients.

None of the above mentioned documents disclose the present invention.

Although WO 94/11031 and French patent application 2,705,677 disclose extrusion of mixtures of cyclodextrins and actives ingredients, said documents do not mention the use of meltextrusion. The technique described in WO 94/11031 and French patent application 2,705,677 has a main disadvantage, that a humid mass needs to be prepared which requires adding to the cyclodextrin and the active ingredient a certain amount of water and in most cases others solvents such as ethanol or methanol. Removing the water and/or other solvents is often a troublesome production step, which often leads to

irreproducibility because not all of the solvent can be removed. Moreover, with practically insoluble active ingredients the amounts of water and/or adjuvant solvents needed make the above technique unpractical on a production scale. Another disadvantage of the technique described in the prior art is that the drying step can induce unwanted crystallization of the active ingredient.

These problems are solved in the present invention by the use of a melt-extrusion process to form solid mixtures comprising one or more cyclodextrins and insoluble active ingredients.

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The present process is advantageously applicable when said active ingredient is sensitive to a solvent such as water or an organic solvent, because it does not require any solvent. The term "sensitive" used herein means that the active ingredient is readily (e.g. within about one hour) influenced by a solvent to such an extent that its physical, chemical and/or biological properties are substantially modified or changed.

The present process is further advantageous because it does not require a drying step, during which insoluble active ingredients often tend to crystallize.

The term "insoluble" hereinabove and hereinunder refers to three categories of compounds, i.e. the "very slightly soluble", "practically insoluble" and "insoluble".

The terms "very slightly soluble", "practically insoluble" or "insoluble" are to be understood as defined in the United States Pharmacopeia 23, NF 18 (1995) page 7, i.e. a "very slightly soluble" compound requires from 1000 to 10,000 parts of solvent for 1 part of solute; a "practically insoluble" or "insoluble" compound requires more than 10,000 parts of solvent for 1 part of solute. The solute referred to in these cases are water or aqueous solutions.

- Three examples of this type of insoluble compounds are: itraconazole, loviride and (±)-ethyl (R\*,R\*)-4-[5-[1-[1-[(4-chlorophenyl)hydroxymethyl]propyl]-1,5-dihydro-5-oxo-4<u>H</u>-1,2,4-triazol-4-yl]-2-pyridinyl]-1-piperazinecarboxylate (hereinafter referred to as compound 1).
- Itraconazole is an art-known antifungal. Loviride is an art-known anti-retrovirally active compound, particularly useful in treating HIV-infected patients.

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- (±)-Ethyl (R\*,R\*)-4-[5-[1-[1-[(4-chlorophenyl)hydroxymethyl]propyl]-1,5-dihydro-5- $0x0-4\underline{H}-1,2,4$ -triazol-4-yl]-2-pyridinyl]-1-piperazinecarboxylate is described as compound No. 3, in WO 95/27704 published on October 19, 1995.
- The compounds that are suitable to be used in this technique are compounds that show no appreciable decomposition at the temperatures needed to melt and extrude the mixture of said one or more active ingredients with the cyclodextrin or cyclodextrins.
- The term "active ingredient" further refers to compounds or mixtures of compounds which are pharmaceutically or therapeutically or cosmetically active for treating humans or animals.
  - The present invention provides a process for preparing a solid mixture comprising one or more cyclodextrins and an (insoluble) active ingredient, comprising a melt-extrusion step wherein one or more cyclodextrins are combined with the one or more active ingredients.
  - Melt-extrusion is a polymer extrusion technique which involves embedding an active ingredient in one or more carriers. In this technique the active ingredient and excipients are molten in the extruder and hence embedded in thermoplastic and thermomelting polymers. The resulting molten mass is then forced through one or more nozzles resulting in a thermoplastic strand or strands.
- An extruder comprises an inlet structure, a cylindrical structure called "barrel", a die and a screw or screws. A schematic overview is shown in Figure 1.
  - The inlet structure mostly is funnel shaped.
- The barrel may comprise one or more barrel units and the screw or screws extend through them.
  - Extruders are available in two general types, namely a single-screw extruder comprising one screw and a multi-screw extruder comprising two or more screws. While this invention can be carried out using either type of extruder, the use of a multi-screw extruder, particularly a twin-screw extruder is preferred. A twin-screw extruder (and a multi-screw extruder) is more efficient in that the plural screws interferring with

each other precludes follow-up movement of the active ingredient and, moreover, the intermeshing of the screws provides a high energy output physically.

An interesting mode of operating the screws is to operate them in a corotating mode.

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The screw or screws may have different shapes such as, for example, a trapezoidal screw, a trapezoidal cut screw, trapezoidal reverse cut screw, ball screw, kneading paddle and these may be used in the desired combination.

- The load fed into the extruder via the inlet structure is forced by the screw or screws to advance, shorn and blended by the screw within the barrel and extruded from the orifice or orifices of the die. The temperature of the barrel or of the barrel units can be controlled via a heating element or even if necessary by a cooling element.
- The rotational speed of the screw can be set within the allowable range of the extruder used.

A person skilled in the art is able to select the screw geometry and combination of unit screws. The principal function of the screw is to transport, crush and knead the material that is being extruded.

The orifice configuration may be circular, elliptical, rectangular or hexagonal.

Hence, said melt-extrusion step comprises the following substeps:

- a) mixing one or more cyclodextrins with the active ingredient or active ingredients,
  - b) optionally mixing additives,
  - c) heating the thus obtained mixture until melting of one of the components,
  - d) forcing the thus obtained mixture through one or more nozzles;
  - e) cooling the mixture till it solidifies.

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If desired, as mentioned above, the thermomelting mixture comprising one or more cyclodextrins and active ingredient(s) may comprise any suitable additive. When, for instance, the cyclodextrin(s) or the active ingredient(s) or one of the other possible additives is apt to be oxidized, an anti-oxidizing agent may be incorporated, preferably in small amounts, such as, for instance 100 to 5000 ppm when compared to the total weight of the mixture. Furthermore, conventional auxiliary additives such as pigments, flavors, stabilizers, preservatives and buffers may be added.

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If necessary conventional pharmacologically acceptable plasticizers, such as long chain alcohols, ethylene glycol, propylene glycol, thriethylene glycol, butanediols, pentanols, hexanols, polyethylene glycols, aromatic carboxylates (e.g. dialkyl phthalates,

trimellitates, benzoates or terephthalates), aliphatic dicarboxylates or fatty acid esters can also be added. Preferably however, a plasticizer is not needed.

The term "melting" should be broadly interpreted. "Melting" can also refer to the fact that some transition is made to a glassy state, in which it is possible for one component of the mixture to get embedded into the other. In particular cases, one component will melt and the other component(s) will dissolve in the melt thus forming solid solutions, which show advantageous dissolution properties.

The possible formation of these solid solutions is one of the further advantages of the present invention. It will be appreciated by a person skilled in the art that mixing two or more solids, i.e. one or more cyclodextrins and the active ingredient or ingredients, and subsequently melting these solids together will give rise to different products than when the said solids are first brought into contact with water or another solvent and then extruded.

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A characteristic of the melt extruded mixtures of the present invention is the fact that they contain substantially less water or any other solvent than mixtures being extruded in an other way.

25 Preferably the present melt extruded mixtures contain no water or solvent, apart from the water or solvent that eventually is contained in the crystal structure of the active ingredient.

It will be appreciated that the temperature inside the extruder is an important parameter.

When different barrel units are present, different temperatures can be applied. A person skilled in the art is able to establish the required temperatures by taking the desired type of cyclodextrin or cyclodextrins or even the complete mixture that is going to be extruded and observing the behaviour as a function of temperature with the aid of a melting point measuring instrument, such as a Kofler hot bench, a microscope hot stage type or a differential scanning calorimeter, e.g. type DSC 7 Series - Perkin Elmer.

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The cooling can be performed without using any auxiliary means, i.e. it most often suffices to let the thermoplastic strand coming out of the extruder cool down to the ambient temperature of the production site. Of course, cooling aids may be used.

Once these thermoplastic strands are cooled down these strands can be milled to obtain a powdered form of the mixture of cyclodextrin or cyclodextrins and the active ingredient.

A person skilled in the art will appreciate that the milling can influence the physical characteristics of the extrudate. During milling the temperature of the material can rise because of the friction and also high shear forces are exerted on the material that is to be milled. Both temperature and mechanical or shear forces can result in a transition of the physical state of the material that is to be milled. A person skilled in the art has sufficient means at his disposal to control temperature and shear forces and thus to control the milling process.

The two processes, i.e. melt extrusion and milling can be combined into one configuration as is shown in Figure 1. The mixture of one or more cyclodextrins and one or more active ingredients in combination with possible other additives is feed via a funnel like inlet. The mixture is then melt-extruded and the mixture is forced through a nozzle onto a conveyor belt. While being transported on the conveyor belt the extrudate cools down. The cooled melt extrudate is fed into a chopper which forms pellets. These pellets may be further milled if required.

This powdered material still has the beneficial properties (high bioavailability, dissolution rate, etc.) and it can be used in the conventional way to prepare pharmaceutical, therapeutical or cosmetical solid dosage forms.

An additional advantage of the present invention is that the active ingredient as well as the cyclodextrins may be transformed in a amorphous form or even that a solid solution is formed. A person skilled in the art will appreciate that this modification of physical state from crystalline to amorphous or to solid solutions is highly advantageous for the dissolution.

35 The fact whether the melt extruded mixture contains amorphous material or contains a solid solution or consists essentially of amorphous material or a solid solution can be measured or checked using differential scanning calorimetry. When there is crystalline

material present in the melt extruded mixture a differential scanning calorimeter will show an endothermic melting peak. When amorphous material or a solid solution is mainly present in the melt extruded mixture a differential scanning calorimeter will not show an endothermic melting peak. Visual inspection of the melt extrudate allows for a distinction between amorphous material and a solid solution. In case the melt extrudate is opaque then both the cyclodextrin(s) and the active ingredient are in a amorphous form. In case melt extrudate is clear then a solid solution was formed.

Curves of differential scanning calorimetry are shown in Figures 2 to 7.

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Interesting embodiments of the present invention are these melt extruded mixtures which consist mainly of amorphous material.

More interesting embodiments of the present invention are these melt extruded mixtures which consist essentially of amorphous material.

Even more interesting embodiments of the present invention are those melt extruded mixtures which consist mainly of a solid solution of the active ingredient or active ingredients in the cyclodextrin or the cyclodextrins.

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Preferred embodiments of the present invention are those melt extruded mixtures which consist essentially of a solid solution of the active ingredient or active ingredients in the cyclodextrin or the cyclodextrins.

- Another advantage of the present invention is that the granulation step in forming pharmaceutical, therapeutical or cosmetical compositions can be omitted, because the powdered material can simply be mixed with other excipients and compressed into, for instance, tablets or another solid pharmaceutical, therapeutical or cosmetical form.
- Depending upon the characteristics of the melt extruded mixture, the size of the pellets of said melt extruded mixture or the mesh of the powder of said melt extruded mixture and, of course, dependent upon the other auxiliaries that are added to the unit dosage forms the unit dosage form may give immediate release or sustained release.
- If desired, said solid pharmaceutical form may also be provided with a conventional coating to improve the appearance and/or the flavor (coated tablets) or additionally to target the release of the active ingredient.

Suitable tablets may have the following compositions and may be prepared in a conventional way. The amounts given are of course dependent upon the dose required for the pharmaceutical, therapeutic or cosmetic activity.

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## Composition A

	milled melt extrudate	100 - 500 mg
	microcrystalline cellulose	100 - 300 mg
10	crospovidone	10 - 200 mg
	colloidal silicon dioxide	1 - 5 mg
	sterotex	2 - 10 mg

# Composition B

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milled melt extrudate	1 <b>0</b> 0 - 500 mg
Microcelac (TM) (1)	200 - 300 mg
crospovidone	70 - 200 mg
talc	20 - 50 mg
sterotex	7 - 10 mg
colloidal silicon dioxide	1 - 5 mg
magnesium stearate	2 - 10 mg

The cyclodextrin to be used in the aforementioned compositions include the pharmaceutically acceptable unsubstituted and substituted cyclodextrins known in the art, more particularly  $\alpha$ ,  $\beta$  or  $\gamma$  cyclodextrins or the pharmaceutically acceptable derivatives thereof.

Substituted cyclodextrins which can be used in the invention include polyethers
described in U.S. Patent 3,459,731. In general, unsubstituted cyclodextrins are reacted with an alkylene oxide, preferably under superatmospheric pressure and at an elevated temperature, in the presence of an alkaline catalyst.

Since a hydroxy moiety of the cyclodextrin can be substituted by an alkylene oxide which itself can react with yet another molecule of alkylene oxide, the average molar substitution (MS) is used as a measure of the average number of moles of the substituting agent per glucose unit. The MS can be greater than 3 and theoretically has no limit.

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Further substituted cyclodextrins are ethers wherein the hydrogen of one or more cyclodextrin hydroxy groups is replaced by C<sub>1-6</sub>alkyl, hydroxyC<sub>1-6</sub>alkyl, carboxy-C<sub>1-6</sub>alkyl or C<sub>1-6</sub>alkyloxycarbonylC<sub>1-6</sub>alkyl or mixed ethers thereof. In particular such substituted cyclodextrins are ethers wherein the hydrogen of one or more cyclodextrin hydroxy groups is replaced by C<sub>1-3</sub>alkyl, hydroxyC<sub>2-4</sub>alkyl or carboxyC<sub>1-2</sub>alkyl or more in particular by methyl, ethyl, hydroxyethyl, hydroxypropyl, hydroxybutyl, carboxy-methyl or carboxyethyl.

In the foregoing definitions the term "C<sub>1-6</sub>alkyl" is meant to include straight and branched saturated hydrocarbon radicals, having from 1 to 6 carbon atoms, such as, methyl, ethyl, 1-methylethyl, 1,1-dimethylethyl, propyl, 2-methylpropyl, butyl, pentyl, hexyl and the like.

Such ethers can be prepared by reacting the starting cyclodextrin with an appropriate O-alkylating agent or a mixture of such agents in a concentration being selected so that the desired cyclodextrin ether is obtained. The said reaction is preferably conducted in a suitable solvent in the presence of an appropriate base. With such ethers, the degree of substitution (DS) is the average number of substituted hydroxy functions per glucose unit, the DS being thus 3 or less.

In the cyclodextrin derivatives for use in the compositions according to the present invention, the DS preferably is in the range of 0.125 to 3, in particular 0.3 to 2, more in

particular 0.3 to 1 and the MS is in the range of 0.125 to 10, in particular of 0.3 to 3 and more in particular 0.3 to 1.5.

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Of particular utility in the invention are the  $\beta$ -cyclodextrin ethers, e.g. dimethyl- $\beta$ -cyclodextrin as described in Drugs of the Future, Vol. 9, No. 8, p. 577-578 by M. Nogradi (1984) and polyethers, e.g. hydroxypropyl  $\beta$ -cyclodextrin and hydroxyethyl  $\beta$ -cyclodextrin, being examples. Such an alkyl ether may be a methyl ether with a degree of substitution of about 0.125 to 3, e.g. about 0.3 to 2. Such a hydroxypropyl cyclodextrin may for example be formed from the reaction between  $\beta$ -cyclodextrin an propylene oxide and may have a MS value of about 0.125 to 10, e.g. about 0.3 to 3.

A more novel type of substituted cyclodextrins is sulfobutylcyclodextrines. These type are also envisaged in the present invention.

The ratio of active ingredient over cyclodextrin may vary widely. For example ratios of 1/100 to 100/1 may be applied. Interesting ratios of active ingredient over cyclodextrin range from about 1/10 to 10/1. More interesting ratios of active ingredient over cyclodextrin range from about 1/5 to 5/1. Most interesting ratios range from about 1/3 to 3/1. Preferred ratio is aabout 1/1.

The use of a mixture of cyclodextrins, either different types  $(\alpha, \beta, \gamma)$  or different substitution (2-hydropropyl or methyl) or different substitution grades in sometimes recommendable to decrease the melting point.

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## Description of the drawings

Figure 1 is a schematic representation of a configuration for carrying out the present invention.

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- Figure 2 is a differential scanning calorimetry curve (DSC curve) of non-milled Batch No 1 material. (see Example 1)
- Figure 3 is a differential scanning calorimetry curve of milled Batch No 1 material (see Example 1)
  - Figure 4 is a differential scanning calorimetry curve of Batch No 2 material (see Example 1)
- Figure 5 is a differential scanning calorimetry curve of of Batch No 3 material (see Example 1)
  - Figure 6 is a differential scanning calorimetry curve of of Batch No 4 material (see Example 1)

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Figure 7 is a differential scanning calorimetry curve of of Batch No 5 material (see Example 1)

## 35 Example 1

Extruded samples of active ingredient with hydroxypropyl-β-cyclodextrin (HP-β-CD) were obtained using a Twin Screw Extruder type MP19 APV Baker (commercially available from the company APV Baker) with a die having a orifice of 3 mm. The process parameters for each individual experiment are shown in the table 1. This type of extruder has a L/D ratio of 15 and a screw pattern: 4D FS - 4x30 FP - 4x60 FP - 4x90 P - 4x60 RP - 2.5D FS - 2x30 FP - 2x60 FP - 2x90 P - 3x60 RP - 3 DFS. (4D refers to a transportelement having a length of 4 times the screw diameter of the feed screw type; 4x30 FP refers to 4 forward paddles positioned with mutual angle of 30

degrees, 4x60 RP refers to a working zone having reverse paddles positioned with a mutual angle of 60 degrees)

- In this type of extruder the mixture is fed by a feeding screw turning a constant feeding speed (v1) ( A feeding speed of 10 revolutions per minute amounts to a feeding speed of 1,5 kg per hour) onto the twin transporter screws having a diameter of 18 mm turning at a transporter speed (v2). These speeds are rotational speeds (revolutions per minute).
- The mixture is then transported into a first heating zone (t1). Here the rate of transport diminished by a difference of the configuration of the twin transporter screws i.e. the rotational transporter speed v2 remains the same but the material does not progress as quickly.
- Subsequently, the molten mass is transported by again normal configuration twin transporter screws to a second heating zone (t2) where the rate of transport is again diminished by a difference of configuration of the twin transporter screws.

After this second heating the thermomelting mixture is transported to the nozzle of the apparatus.

Table 1

mixture	Batch. No	t <sub>1</sub> (°C)	t2 (°C)	t <sub>p</sub> (°C)	v1 (rpm)*	v2 (rpm)*
$\frac{\text{compound 1}}{\text{HP-}\beta\text{-CD}} \cdot \frac{1}{3}$	1	256	283	280	10	100
$\frac{\text{itraconazole}}{\text{HP-}\beta\text{-CD}} : \frac{1}{1}$	2	263	265	279	10	20
$\frac{\text{itraconazole}}{\text{HP-}\beta\text{-CD}} : \frac{1}{3}$	3	264	265	280	10	20
$\frac{\text{loviride}}{\text{HP-}\beta\text{-CD}}:\frac{1}{1}$	4	274	285	292	10	80
$\frac{\text{loviride}}{\text{HP-}\beta\text{-CD}}: \frac{1}{3}$	5	258	265	274	10	20

\* rpm = revolutions per minute

- t<sub>1</sub>: temperature of the first heating zone

- t2: temperature of the second heating zone

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- tp: temperature inside the barrel

- v1: rate of feeding screw

- v2 : twin transporter screws speed(rotational).

5 In every case the mixture of active ingredient and 2-hydroxypropyl-β-CD gave a solid solution.

# Example 2

Extruded samples of active ingredient with dimethyl-β-cyclodextrin (DM-β-CD) were obtained using extruder type MP19 - APV Baker with the process parameters as shown in the table 2.

## Table 2

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mixture	Batch. No.	t1 (°C)	t2 (°C)	t <sub>p</sub> (°C)	v1 (1) (rpm)*	v <sub>2</sub> (rpm)*
$\frac{\text{compound 1}}{\text{DM-}\beta\text{-CD}} : \frac{1}{1}$	6	241	245	254	0	20
$\frac{\text{itraconazole}}{\text{DM-}\beta\text{-CD}}:\frac{1}{1}$	7	239	240	253	0	20
$\frac{\frac{\text{loviride}}{\text{DM-}\beta\text{-CD}} : \frac{1}{l}$	8	248	250	263	0	20

<sup>\*</sup> rpm = revolutions per minute

(1) The apparatus was fed manually, without using the feeding screw. In every case the mixture of active ingredient and DM- $\beta$ -CD.

20 - t1: temperature of the first heating zone

- t2: temperature of the second heating zone

- t<sub>D</sub>: temperature inside the barrel

- v<sub>1</sub>: feeding screw speed (rotational)

- v<sub>2</sub>: twin transporter screw speed (rotational).

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# Example 3

The dissolution of the melt extrudate of Batch No 1 was compared with the dissolution of the "physical mixture" (i.e. the mixture of the two component in the ratio as shown for Batch No. 1, but not melt extruded).

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An amount of 100 mg of milled melt extrudate of Batch No 1 was added to a volume of 900 ml of artificial gastric juice at a temperature of 37 degrees Celsius. The stirring method used was the peddle method with a peddle moving at 100 rotations per minute. Using UV spectrometry the relative amount of dissolved extrudate was measured during 1 hour.

The same procedure for the "physical mixture" was followed.

The results of this dissolution process are shown in Table 3

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Table 3

time mixture (minutes)	milled extrudate Batch No 1 (% of total amount dissolved)	corresponding physical (% of total amount dissolved)
0	0.00	0.00
5	62.10	1.71
15	70.20	14.67
30	72.63	21.06
45	74.07	26.10
60	74.25	28.35

## Example 4

The melting behaviour was measured by using differential scanning calorimetry. The calorimeter used is the Perkin-Elmer 7 Series Thermal Analysis System. In all cases the rate of heating was set at 20 degrees Celsius per minute.

Figure 2 shows the DSC curve of melt extrudate of Batch No 1 before milling. The curve shows no endothermic or exothermic peaks and it was established by visual inspection that the molten material was a clear solution, thus indicating that the non-milled melt extrudate of Batch No 1 is a solid solution.

Figure 3 shows the DSC curve of melt extrudate of Batch No 1 after milling. The curve shows no endothermic or exothermic peaks and it was established by visual inspection that the molten material was a clear solution, thus indicating that the milled melt extrudate of Batch No 1 is a solid solution.

Figure 4 shows the DSC curve of meltextrudate of Batch No 2 before milling. The curve shows no endothermic or exothermic peaks and it was established by visual inspection that the molten material was <u>not</u> a clear solution, thus indicating that the non-milled melt extrudate of Batch No 2 is a mixture of amorphous material.

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Figure 5 shows the DSC curve of meltextrudate of Batch No 3 before milling. The curve shows a small endothermic peak. The data on said small peak are as follows: X1 = 117.600 degrees Celsius, X2 = 143.200 degrees Celsius, Peak at 132.695 degrees Celsius, Area is 38.126 mJ,  $\Delta H$  is 3.768 J/g, Height is 1.520 mW and the onset is at 125.816 degrees Celsius. Said small peak is very probably due to an impurity in the cyclodextrins. It was established that the non-milled melt extrudate of Batch No 3 is a mixture of amorphous material.

Figure 6 shows the DSC curve of meltextrudate of Batch No 4 before milling. The curve shows a few small endothermic peaks. Hence, it was established that the non-milled melt extrudate of Batch No 4 is a mixture of amorphous material containing small amounts of crystalline material

Figure 7 shows the DSC curve of meltextrudate of Batch No 5 before milling. The curve shows no endothermic or exothermic peaks and it was established by visual inspection that the molten material was <u>not</u> a clear solution, thus indicating that the non-milled melt extrudate of Batch No 5 is a mixture of amorphous material.

## Example 5

The melt extrudate of Batch No. 1 was milled and sieved. By mixing the appropriate amounts a tablet having the following composition was prepared in an art-known way:

	milled extrudate batch No 1	480 mg
	microcrystalline cellulose	218 mg
30	Aerosil	3 mg
	magnesium stearate	5 mg
	crospovidone	144 mg

## **Claims**

- Process for preparing a solid mixture comprising one or more cyclodextrins and
   one or more active ingredients characterized in that said process encompasses a melt-extrusion step, wherein the active ingredient is embedded into the cyclodextrin carrier.
- 2. A process as claimed in claim 1, wherein the melt-extrusion process comprises the following substeps:
  - a) mixing one or more cyclodextrins with one or more active ingredients, and
  - b) optionally mixing additives;
  - c) heating the thus obtained mixture until melting of one of the components;
  - d) forcing the thus obtained mixture through one or more nozzles;
- e) cooling the mixture till it solidifies.
  - 3. A solid mixture obtainable by the process as described in any of claims 1 or 2, with the proviso that nifedipine in combination with 2-hydroxypropyl-β-cyclodextrin is excluded.

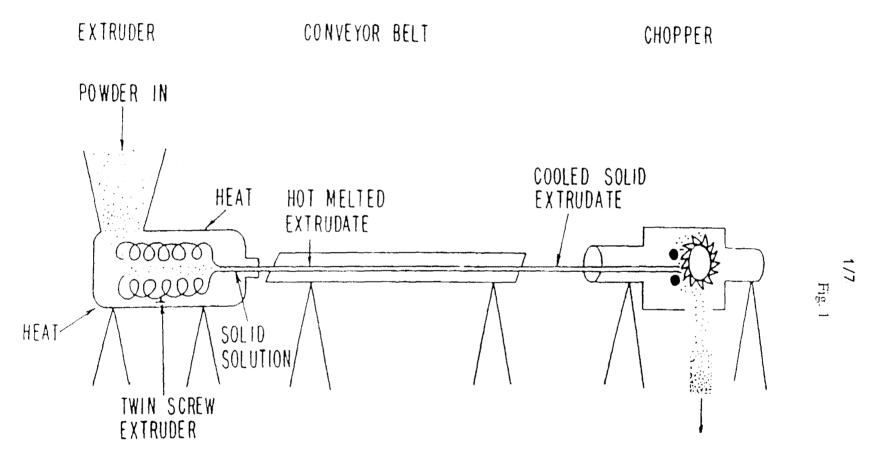
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- 4. A solid mixture as claimed in claim 3 characterized in that the active ingredient or active ingredients are insoluble according to the definition of US Pharmacopeia.
- 5. A solid mixture as claimed in claims 3 or 4, wherein substantially only one type of cyclodextrin is present.
  - A solid mixture as claimed in any of claims 3 to 5 wherein a cyclodextrin is hydroxypropyl-β-cyclodextrin.
- A solid mixture as claimed in any of claim 3 to 5 wherein a cyclodextrin is dimethyl-β-cyclodextrin.
  - 8. A solid mixture as claimed in any of claims 3 to 7, wherein the active ingredient is itraconazole.

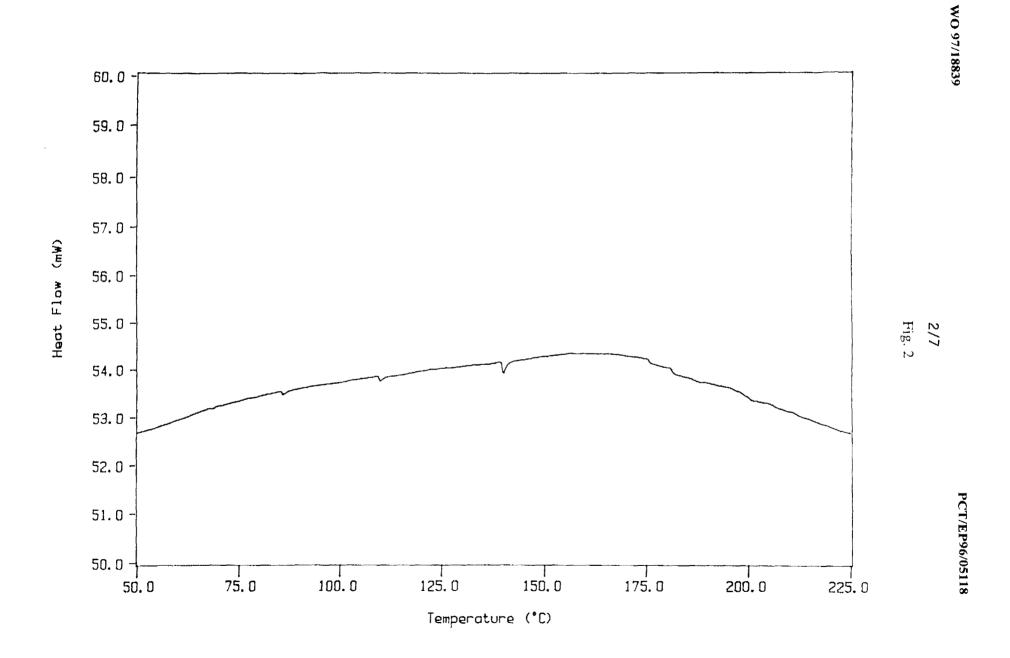
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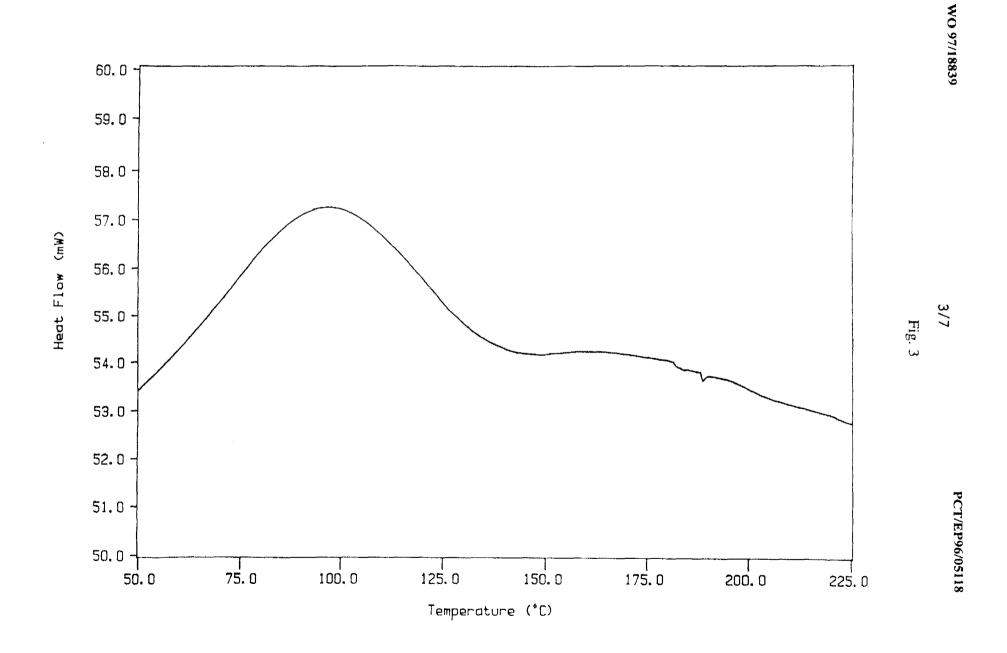
9. A solid mixture as claimed in any of claims 3 to 7 wherein the active ingredient is loviride.

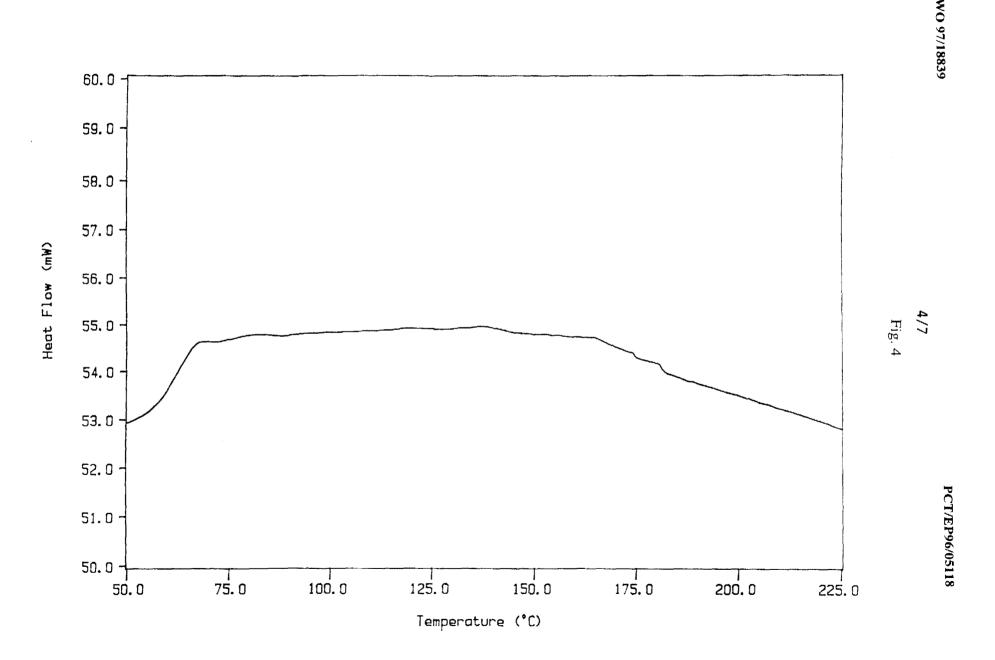
- 10. A solid mixture as claimed in any of claims 3 to 7 wherein the active ingredient is (±)-ethyl (R\*,R\*)-4-[5-[1-[1-[(4-chlorophenyl)hydroxymethyl]propyl]-1,5-dihydro-5-oxo-4H-1,2,4-triazol-4-yl]-2-pyridinyl]-1-piperazinecarboxylate.
- 5 11. A pharmaceutical composition comprising milled melt extrudate and other excipients.
- 12. A process for preparing a pharmaceutical composition as claimed in claim 11 characterized by milling appropriately the solid mixture as claimed in any of
   10 claims 4 to 10, intimately mixing the thus obtained powdered material with other pharmaceutically acceptable excipients and further processing into pharmaceutical dosage forms.

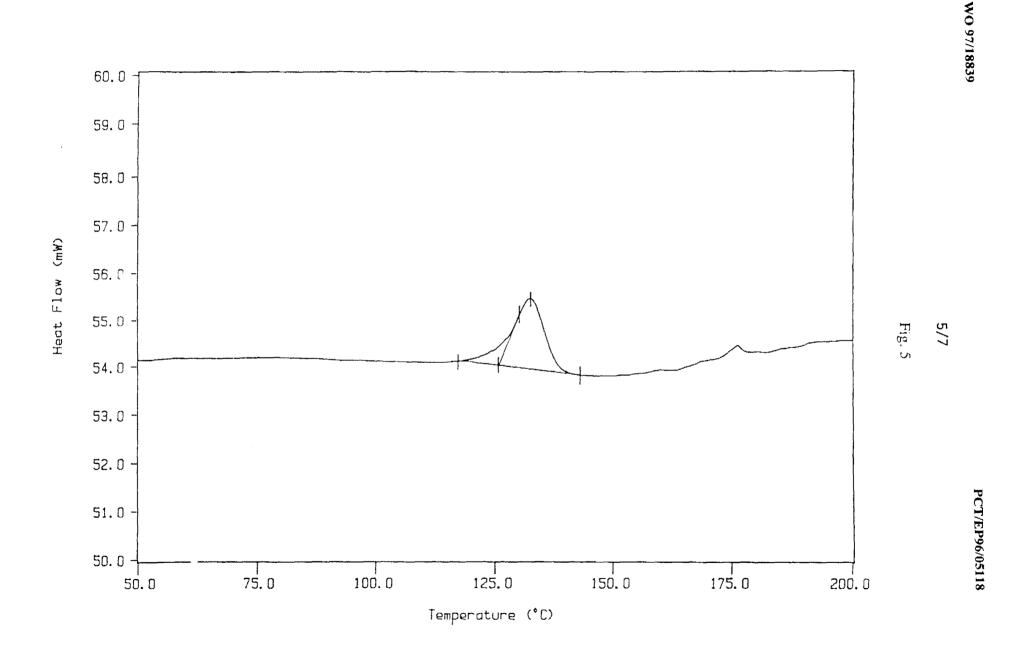


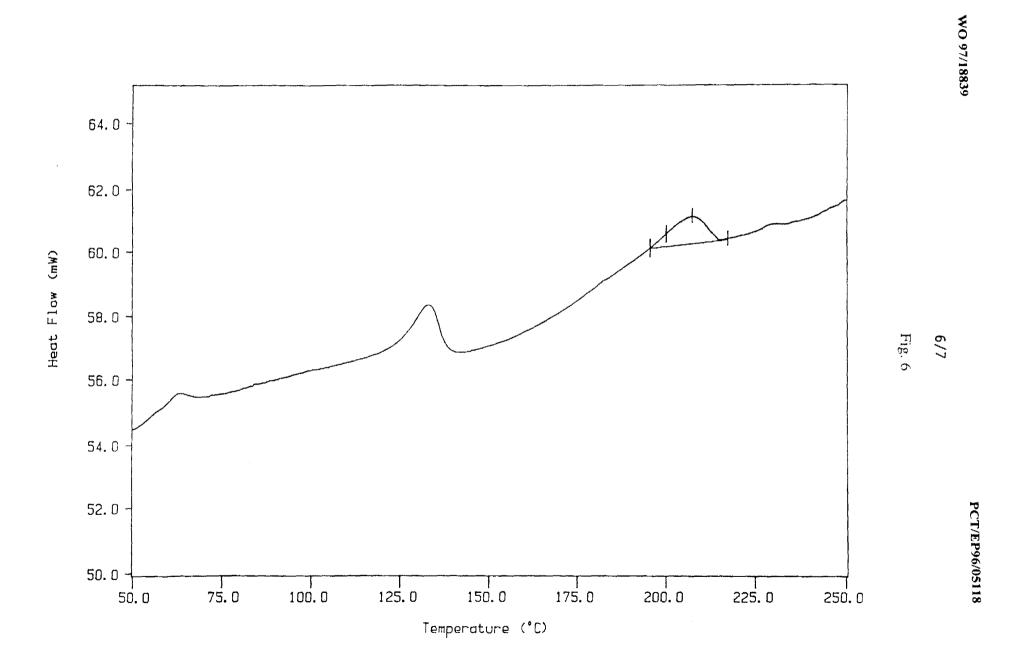
PROCESS DIRECTION

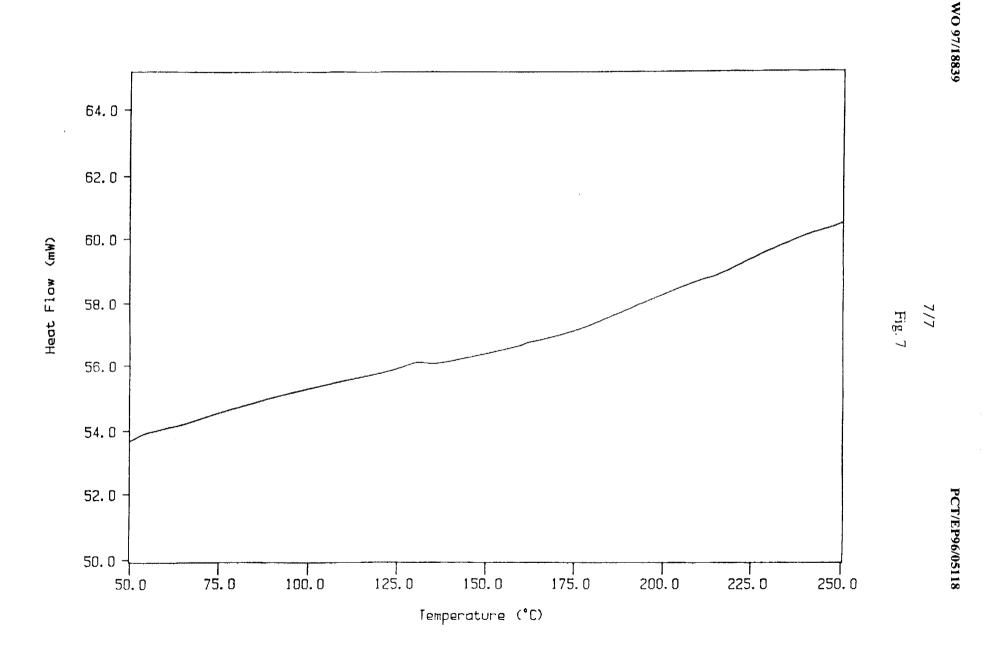












Int. onal Application No PCT/EP 96/05118

A. CLASS	IFICATION OF SUBJECT MATTER								
IPC 6	A61K47/48								
According	to International Patent Classification (IPC) or to both national class	sification and IPC							
	S SEARCHED	ation symbols)							
Minimum documentation searched (classification system followed by classification symbols)  I PC 6 A61K									
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Documenta	tion searched other than minimum documentation to the extent that	t such documents are included in the fields s	searched						
Electronic o	lata base consulted during the international search (name of data ba	ase and, where practical, search terms used)							
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT		· · · · · · · · · · · · · · · · · · ·						
Category °	Citation of document, with indication, where appropriate, of the	relevant passages	Relevant to claim No.						
,	LIO OA 11031 A (NIDDON CUINVAKIA C	0 170) 26	1 10						
X	WO 94 11031 A (NIPPON SHINYAKU C   May 1994	U LID) 26	1-12						
	cited in the application								
	see abstract								
l ,		3	1 10						
Y	FR 2 705 677 A (ROQUETTE FRERES) December 1994	2	1-12						
	cited in the application								
	see abstract								
	see examples								
	see claims								
х	EP 0 665 009 A (NIPPON SHINYAKU	COMPANY.	1-12						
	LIMITED.) 2 August 1995	•							
	cited in the application								
	see abstract see examples								
	see claims								
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X Furt	her documents are listed in the continuation of box C.	Patent family members are listed in	in annex.						
* Special ca	tegories of cited documents :	"T" later document published after the inte							
	ent defining the general state of the art which is not cred to be of particular relevance	or priority date and not in conflict wi cited to understand the principle or th							
"E" earlier	document but published on or after the international	'X' document of particular relevance; the							
filing of L* docume	ent which may throw doubts on priority claim(s) or	cannot be considered novel or cannot involve an inventive step when the do							
	is cited to establish the publication date of another  or other special reason (as specified)	"Y" document of particular relevance; the cannot be considered to involve an in							
"O" docume other n	ent referring to an oral disclosure, use, exhibition or neans	document is combined with one or mements, such combination being obvious	ore other such docu-						
	ent published prior to the international filing date but aan the priority date claimed	in the art.  *&* document member of the same patent	•						
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24	4 February 1997								
Name and m	nailing address of the ISA	Authorized officer							
	European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk								
	Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Dullaart, A							

Form PCT/ISA/210 (second sheet) (July 1992)

2

Int Ional Application No PCT/EP 96/05118

C10	DOCUMENTS COMMENTED TO BE DELEVANT	
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	J. PHARM. PHARMACOL., VOL. 44, NO. 2, PAGE(S) 73-8, 1992, XP002002172 UEKAMA, KANETO ET AL: "Inhibitory effect of 2- hydroxypropylbeta cyclodextrin on crystal growth of nifedipine during storage: superior dissolution and oral bioavailability compared with poly(vinylpyrrolidone) K-30" cited in the application * paragraph Materials and methods * see figures	1-12
Y	PHARM. WEEKBL. SCI. ED., 1988, VOL. 10, NO. 2, PAGE(S) 80-85, XP002002173  VAN DOORNE H. ET AL: "Formation and antimicrobial activity of complexes of beta- cyclodextrin and some antimycotic imidazole derivatives" cited in the application see abstract  * paragraph Results and discussion * see page 85, left-hand column	1-12
Y	J. ANTIMICROB. CHEMOTHER., 1993, VOL. 32, NO. 3, PAGE(S) 459-463, XP002002174 HOSTETLER J.S. ET AL: "Effect of hydroxypropyl-beta-cyclodextrin on efficacy of oral itraconazole in disseminated murine cryptococcosis" cited in the application see abstract * paragraph Results *	1-12
Y	JPN. J. MED. MYCOL., 1994, VOL. 35, NO. 3, PAGE(S) 263-267, XP002002175 MIKAMI Y. ET AL: "Effect of carrier solvents on the efficacy of oral itraconazole therapy in aspergillosis in mice" cited in the application see abstract see figures	1-12
Y	PHARM. RES., VOL. 11, NO. 12, PAGE(S) 1766-70, 1994, XP002002176 HIRAYAMA, FUMITOSHI ET AL: "Effect of 2-hydroxypropylbeta cyclodextrin on crystallization and polymorphic transition of nifedipine in solid state" cited in the application see abstract * paragraph Results and discussion *	1-12
	-/	

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In tional Application No
PCT/EP 96/05118

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	ation) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 5 009 900 A (LEVINE HARRY ET AL) 23 April 1991 cited in the application see examples see claims	1-12

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Form PCT/ISA/210 (continuation of second sheet) (July 1992)

ernational application No.

#### INTERNATIONAL SEARCH REPORT

PCT/EP 96/05118

Box I	Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)
This Int	ternational Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
1.	Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:
2. <b>X</b>	Claims Nos.: 1-7, 11-12 because they relate to parts of the International Application that do not comply with the prescribed requirements to such
	an extent that no meaningful International Search can be carried out, specifically:  In view of the large number of compounds, which are defined by the general definition of the active ingredient used in the claims, the search had to be restricted for economic reasons. The search was limited to the compounds for which pharmacological data was given and/or the compounds mentioned
3.	Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).
Box II	Observations where unity of invention is lacking (Continuation of item 2 of first sheet)
This Int	crnational Searching Authority found multiple inventions in this international application, as follows:
1.	As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2.	As all searchable claims could be searches without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3.	As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4.	No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark	The additional search fees were accompanied by the applicant's protest.
	No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/210	
in the claims, and to the general idea underlying the application (see Guidelines, chapter III, paragraph 2.3).	
Claims searched incompletely: 1-7,11-12	

Information on patent family members

In: Gonal Application No PCT/EP 96/05118

date	member(s)	date
26-05-94	AU 5376994 A	08-06-94
02-12-94	IT 1265964 B	16-12-96
02-08-95	AU 5160793 A WO 9408561 A	09-05-94 28-04-94
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Form PCT/ISA/210 (patent family annex) (July 1992)

# Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	Α	Appeal
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# Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

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	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	10551205	BODOR ET AL.
	Examiner	Art Unit
	Jonathan S Lau	1623

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U.S. Patent and Trademark Office Part of Paper No.: 20080325

## Search Notes



Application/Control No.	Applicant(s)/Patent Under Reexamination
10551205	BODOR ET AL.
Examiner	Art Unit

1623

	SEARCHED		
Class	Subclass	Date	Examiner

Jonathan S Lau

SEARCH NOTES		
Search Notes	Date	Examiner
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker)	3/26/2008	JSL
EAST - see attached notes	3/26/2008	JSL
Google Scholar - see attached notes	3/26/2008	JSL

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#### **BIB DATA SHEET**

#### **CONFIRMATION NO. 4092**

SERIAL NUM	1BER	FILING OF			CLASS	GR	OUP ART	UNIT	ATTC	DRNEY DOCKET
10/551,20	)5	11/14/2			514		1623			
		RUL	E							
Yogesh [	S. Bodo Dandike	or, Bal Harbo r, Toronto, C.	ANADA;							
** CONTINUING DATA ******************************  This application is a 371 of PCT/US04/09387 03/26/2004     which claims benefit of 60/458,922 03/28/2003     and claims benefit of 60/484,756 07/02/2003     and claims benefit of 60/541,247 02/04/2004  ** FOREIGN APPLICATIONS ************************************										
04/21/20		EIGN FILING	G LICENS	E GRA	WIED ""					
Foreign Priority claimed  35 USC 119(a-d) conditions met Yes No Verified and Acknowledged  Yes No  Met after Allowance Initials				STATE OR COUNTRY FL	_	SHEETS TOT CLAI			INDEPENDENT CLAIMS 6	
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Self-association of cyclodextrins and cyclodextrin complexes. - all 4 versions » **All Results** 

T Loftsson, M Masson, ME Brewster - J Pharm Sci, 2004 - ncbi.nlm.nih.gov A Berthod

... formation of noncovalent, dynamic inclusion complexes ... which regards drug-cyclodextrin

interactions as ... the important contribution of non-inclusion-based aspects ... D Armstrong

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W Li

M Másson

[CITATION] ... dynamics studies on inclusion and noninclusion phenomena between b-cyclodextrin and new anti-...

ME Amato, KB Lipkowitz, GM Lombardo, GC Pappalardo - J Chem Soc Perkin Trans, 1996

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Cyclodextrins in Polymer Synthesis: Supramolecular Cyclodextrin Complexes of Pyrrole and 3, 4- ...

J Storsberg, H Ritter, H Pielartzik, L Groenendaal - Advanced Materials, 2000 - doi.wiley.com

... a-Cyclodextrin/Pyrrole±Complex 1a (Inclusion Type) and a-Cyclodextrin/ EDT±Complex

2a (Non-Inclusion Type): Pyrrole (0.48 g, 7.2 mmol) or EDT (1.02 g, 7.2 ...

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Self-association and cyclodextrin solubilization of drugs - all 5 versions »

T Loftsson, A Magnusdottir, M Masson, JF ... - Journal of Pharmaceutical Sciences, 2002 - doi.wiley.com

... that only 1:1 drug/cyclodextrin complexes are formed and that the two events, inclusion complex formation and solubilization via non-inclusion complex for ...

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... enantioselective retention mechanisms on derivatized cyclodextrin gas chromatographic chiral ... - all 3 versions »

A Berthod, W Li, DW Armstrong - Analytical Chemistry, 1992 - pubs.acs.org

... Cyclodextrin inclusion complex- ation plays a major role in chiral recognition in

LC."J2 Early on, Smolkova-Keulemansova and co-workers published GC evidence ...

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The effects of organic salts on the **cyclodextrin** solubilization of drugs - all 6 versions »

T Loftsson, K Matthi asson, M Masson - International Journal of Pharmaceutics, 2003 - Elsevier

... have shown that  $\alpha$ -cyclodextrin ( $\alpha$ CD) forms both inclusion and non-inclusion complexes with dicarboxylic acids and that the two types of complexes coexist in ...

Cited by 9 - Related Articles - Web Search

### Self Association and Cyclodextrin Solubilization of NSAIDs - all 2 versions »

A Magnusdottir, M Másson, T Loftsson - Journal of **Inclusion** Phenomena and Macrocyclic Chemistry, 2002 - Springer ... that the solubility of drugs in a **cyclodextrin** solution is explained not only by **inclusion** complex formation but also by **non-inclusion** association of the ... Cited by 9 - Related Articles - Web Search

B Garcia, J Ramirez, S Wong, CB Lebrilla - International Journal of Mass Spectrometry, 2001 - Elsevier ... cyclodextrin hosts. The inclusion structure is the preferred state for the complexes.

Molecular dynamics calculations initiated with inclusion and noninclusion ...

Thermal dissociation of protonated **cyclodextrin**-amino acid complexes in the gas phase

Cited by 8 - Related Articles - Web Search

## Separation Behavior of Common Fullerenes in Cyclodextrin-HPLC Based on Computationally-Derived ... - all 2 versions

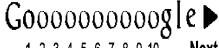
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CL Copper, KW Whitaker, MJ Sepaniak - Journal of Liquid Chromatography & Related Technologies, 1994 - informaworld.com ... the interaction between the fullerene solutes and the **cyclodextrin** ... (**inclusion** complex) positions. Other evidence of **non-inclusion** interactions is seen by the ...

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Result Page: 1 2 3 4 5 6 7 8 9 10 Ne

cyclodextrin non-inclusion inclusion

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### EAST Search History

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	97	((NICHOLAS) near2 (BODOR)).INV.	US-PGPUB; USPAT	ADJ	ON	2008/03/26 12:53
L2	1	((YOGESH) near2 (DANDIKER)).INV.	US-PGPUB; USPAT	ADJ	ON	2008/03/26 12:53
L3	2	I1 and cladribine and cyclodextrin	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 12:54
L4	6	Cadribine and cyclodextrin and (noninclusion or (non inclusion))	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 13:04
L5	409	cyclodextrin and (noninclusion or (non inclusion))	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 13:07
L6	137	15 and melt near9 extru\$7	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 13:07
L7	23	l6 and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 13:08
L8	6	l5 and cyclodextrin.ti,ab.	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/26 13:08
L9	3	"9718839"	US-PGPUB; USPAT; USOCR; EPO; DERWENT	ADJ	ON	2008/03/26 13:37
S1	3	Cadribine.ti,ab. and cyclodextrin.ti,ab,bsum.	US-PGPUB; USPAT; USOCR	ADJ	ON	2008/03/25 08:51

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Sheet

# INFORMATION DISC STATEMENT BY AP

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Complete if Known					
Application Number	10/551,205				
Filing Date	November 14, 2006				
First Named Inventor	Nicholas S. Bodor				
Examiner Name					
Attorney Docket No.	0056192-000024				

U.S. PATENT DOCUMENTS					
Examiner Initials	Document Number- Kind Code	Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines Where Relevant Passages or Figures Appear	
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	FOREIG	N PATENT DOCUME	NTS						
Foreign Patent Document			T			S.	TATUS		
Country Code <sup>1</sup> , Number, Kind Code	Publication Date (MM-DD-YYYY)	Name of Patentee or Applicant of Cited Document	Translation	Partial Translation	Eng. Lang. Summary	Search Report	IPER	Abstract	Cited in Spec. / Pg. No(s).
	Country Code <sup>1</sup> , Number,	Country Code <sup>1</sup> , Number, Publication Date	Foreign Patent Document  Country Code <sup>1</sup> , Number,  Publication Date  Applicant of Cited	Country Code <sup>1</sup> , Number, Publication Date Kind Code (MM-DD-YYYY) Name of Patentee or Applicant of Cited Document	Foreign Patent Document  Country Code <sup>1</sup> , Number, Kind Code  Code	Foreign Patent Document  Country Code¹, Number, Kind Code (MM-DD-YYYY)  Name of Patentee or Applicant of Cited Document  Name of Patentee or Applicant of Cited Document	Foreign Patent Document  Country Code¹, Number, Kind Code  Country Code¹, Number, Kind Code	Foreign Patent Document  Country Code¹, Number, Kind Code (MM-DD-YYYY)  Name of Patentee or Applicant of Cited Document  Name of Patentee or Applicant of Cited Document  Name of Patentee or Applicant of Cited Document	Foreign Patent Document  Country Code¹, Number, Kind Code  Country Code¹, Number, Kind Code

Enter Office that issued the document, by the two-letter code. **OTHER DOCUMENTS** Examiner Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, Initials serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published. PCT International Preliminary Report on Patentability and Written Opinion for International Application No. /J.L./ PCT/US2004/009387, International filing date March 26, 2004.

Signature	Examiner Signature	/Jonathan Lau/	Date Considered	03/26/2008	
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Attorney Docket No. 0056192-000024

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

re Patent Application of	)
Nicholas S. Bodor et al.	) Group Art Unit: 1614
Application No.: 10/551,205	) Examiner:
Filing Date: November 14, 2006	Confirmation No.: 4092
Title: ORAL FORMULATIONS OF CLADRIBINE	) ) )

#### **SECOND INFORMATION DISCLOSURE STATEMENT** TRANSMITTAL LETTER

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

form F	Enclosed is a Second Inform PTO-1449 for the above-identif		Disclosure Statement (IDS) and accompanying atent application.
$\boxtimes$	No additional fee for submiss	sion of	f an IDS is required.
	The fee of \$ 180 as set forth	in 37	C.F.R. § 1.17(p) is also enclosed.
	A statement under 37 C.F.R.	§ 1.9	7(e) is also enclosed.
	A statement under 37 C.F.R. 37 C.F.R. § 1.17(p) are also		7(e), and the fee of \$ 180 as set forth in sed.
	Charge to	Depo	sit Account No. 02-4800 for the fee due.
	A check in the amount of		is enclosed for the fee due.
	Chargeto	credit	card for the fee due. Form PTO-2038 is attached.
	37 C.F.R. §§ 1.16, 1.17 and	1.21 t	to charge any appropriate fees under hat may be required by this paper, and to credit ount No. 02-4800. This paper is submitted in
		Resp	pectfully submitted,
		Buch	HANAN INGERSOLL & ROONEY PC
Date	August 10, 2007	Ву:	Mary Katherine Baumeister Registration No. 26254
	3ox 1404 ndria, VA 22313-1404		

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#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re F	Patent Application of	)
Nichol	as S. Bodor et al.	) Group Art Unit: 1614
Applic	ation No.: 10/551,205	) Examiner:
Filed:	November 14, 2006	) Confirmation No.: 4092
For:	ORAL FORMULATIONS OF CLADRIBINE	) )

#### SECOND INFORMATION DISCLOSURE STATEMENT

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In accordance with the duty of disclosure as set forth in 37 C.F.R. § 1.56, the accompanying information is being submitted in accordance with 37 C.F.R. §§ 1.97 and 1.98.

Applicants request the Examiner's consideration of the enclosed publication cited during prosecution of the corresponding Chinese application in an Official Action dated May 11, 2007. The publication is listed on the accompanying Form PTO-1449. The reference is in Chinese; therefore, the cited section (page 105, lines 25-29) is accompanied by an English translation thereof. This reference was cited as relevant only to some of the process and product-by-process claims.

This statement, Form PTO-1449 and enclosed citation and translation are believed to be filed prior to an action on the merits. In addition, the undersigned hereby states under 37 C.F.R. § 1.97(e) that each item of information contained in this information disclosure statement was first cited in any communication from a patent office in a counterpart foreign application not more than three months prior to the filing of this information disclosure statement. Therefore, no fee is required to obtain consideration under 37 C.F.R. §1.97(b) or (c), whichever is applicable.

Second Information Disclosure Statement Application No. <u>10/551,205</u> Attorney Docket No. <u>0056192-000024</u> Page 2

It is respectfully requested that the Examiner returned an initialed copy of applicants' enclosed Form PTO-1449 with the next official communication or with the first Action on the merits.

Respectfully submitted,

**BUCHANAN INGERSOLL & ROONEY PC** 

Date: August 10, 2007

By:

Mary Katherine Baumeister Registration No. 26254

P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620

#### SECOND INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

	Complete if Known				
Application Number	10/551,205				
Filing Date	November 14, 2006				
First Named Inventor	Nicholas S. Bodor et al.				
Examiner Name					
Attorney Docket No.	0056192-000024				

Sheet 1 of 1

	U.S. PATENT DOCUMENTS						
Examiner Initials	Document Number	Kind Code (if known)	Name of Patentee or Applicant of Cited Document	Issue/Publication Date (MM-DD-YYYY)			

FOREIGN PATENT DOCUMENTS											
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Examiner Initials	Document Number	Kind Code (if known)	Country	Date of Publication (MM-DD-YYYY)	Translation	Partial Translation	Eng. Lang. Summary	Search Report	IPER	Abstract	Cited in Spec
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NON-PATENT LITERATURE DOCUMENTS						
Examiner Initials	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.					
/J.L./	Gao, Shen, New Dosage Form and New Technology of Modern Drugs, first edition, January 2002, Chapter 6, Section 3 (III) Procedures, page 105, lines 25-29 (published by People's Military Medical Publisher), and English translation thereof					

Examiner Signature	/Jonathan Lau/	Date Considered	03/26/2008

<sup>\*</sup>EXAMINER: Initial if reference considered, whether or not citation is in conformance with M.P.E.P. § 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to Applicant.

#### **THIRD INFORMATION DISCLOSURE** STATEMENT BY APPLICANT

(use as many sheets as necessary)

Application Number	10/551,20
Filing Date	November
First Named Inventor	Nicholas S
Examiner Name	

0056192-000024 Attorney Docket No.

Complete if Known

Sheet 1 of 2

U.S. PATENT DOCUMENTS						
Examiner Initials	Document Number- Kind Code	Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines Where Relevant Passages or Figures Appear		
	US-					
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FOREIGN PATENT DOCUMENTS										
	Foreign Patent Document						S <sup>-</sup>	ratus		
Examiner Initials	Country Code <sup>1</sup> , Number, Kind Code	Publication Date (MM-DD-YYYY)	Name of Patentee or Applicant of Cited Document	Translation	Partial Translation	Eng. Lang. Summary	Search Report	IPER	Abstract	Cited in Spec. / Pg. No(s).
Enter Office th	at issued the document, by the tw	o-letter code				<u> </u>	<u> </u>			

<sup>1</sup> Enter Office tha	at issued the document, by the two-letter code.					
	NON-PATENT LITERATURE DOCUMENTS					
Examiner Initials	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.					
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Examiner	/lanathan lau/	Date	03/26/2008
Signature	/Jonathan Lau/	Considered	00,20,200

# THIRD INFORMATION DISCLOSURE STATEMENT BY APPLICANT

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Sheet 2 of 2

	Complete if Known	4/3/
Application Number	10/551,205	NOV 0 8 2007 pg
Filing Date	November 14, 2006	MON O G TAO. B
First Named Inventor	Nicholas S. Bodor	\ .\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\
Examiner Name		To Describe
Attorney Docket No.	0056192-000024	TAUS

	NON-PATENT LITERATURE DOCUMENTS
Examiner Initials	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.
/J.L./	Nolan et al., "Preparation of Vesicles and Nanoparticles of Amphiphilic Cyclodextrins Containing Labile Disulfide Bonds", Langmuir, Vol. 19, pp. 4469-4472, 2003, American Chemical Society, US
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Examiner	/Jonathan Lau/	Date	03/26/2008	
Signature	/Jonaman Lau/	Considered	00/20/2000	

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re F	Patent Application of	) MAIL STOP PCT
Nichol	as S. Bodor et al.	) Group Art Unit:
Applic	ation No.: 10/551,205	) Examiner:
Filed:	PCT/US2004/009387 filed March 26, 2004	) Confirmation No.: 4092
For:	ORAL FORMULATIONS OF CLADRIBINE	) ) )

#### FIRST INFORMATION DISCLOSURE STATEMENT

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In accordance with the duty of disclosure as set forth in 37 C.F.R. § 1.56, the accompanying information is being submitted in accordance with 37 C.F.R. §§ 1.97 and 1.98. Applicants request the Examiner's consideration of the documents listed on the accompanying Form PTO-1449. All of these documents are cited in the instant specifiation, including the documents cited in the International Search Report (copy enclosed) which was issued in connection with PCT/US2004/009387, filed March 26, 2004, of which this application is the national phase.

Pursuant to 37 C.F.R. § 1.98, a copy of each of the documents cited is enclosed. However, copies of any listed U.S. patents and U.S. patent application publications are not enclosed since it is no longer required according to the July 11, 2003 waiver of the requirement for copies of cited U.S. patents and U.S. patent application publications in national patent applications filed after June 30, 2003 and international applications entering the national stage under 35 U.S.C. § 371 after June 30, 2003.

This Statement, Form PTO-1449 and documents are being submitted within three (3) months of the filing or entry of the national stage of this application or before the first Office Action on the merits, whichever is later. Since these documents are being filed within the time period set forth in 37 C.F.R. § 1.97(b), no fee or statement is required.

The following remarks are offered with respect to the listed documents which are not in English:

First Information Disclosure Statement Application No. <u>10/551,205</u> Attorney's Docket No. <u>0056192-000024</u> Page 2

DE 31 18 218 is in German. Applicants do not have an English translation. However, an English abstract, together with the citation of the document in the instant specification, are provided to serve as a brief statement of relevance. Applicants consider this a general state of the art reference.

<u>DE 33 17 064</u> is in German. Applicants do not have an English translation but provides herewith an English abstract. The abstract and citation of the document in the specification serve as a brief statement of relevance. Applicants consider this a general state of the art reference.

EP 0 149 197 B1 is in German, although the claims are also present in English. Applicants enclose a full English translation of this document, which is cited in the instant specification and which applicants consider to be a general state of the art reference.

It is respectfully requested that an Examiner-initialed copy of the accompanying Form PTO-1449 be returned to the undersigned.

Respectfully submitted,

BUCHANAN INGERSOLL AND ROONEY PC

Date: November 14, 2006

Mary Katherine Baumeister

Registration No. 26254

P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620

# FIRST INFORMATION DISCLOSURE STATEMENT BY APPLICANT

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	Complete if Known	
Application Number	10/551,205	
Filing Date		
First Named Inventor	Nicholas S. Bodor	
Examiner Name		
Attorney Docket No.	0056192-000024	

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Examiner Initials	Document Number	Kind Code (if known)	Name of Patentee or Applicant of Cited Document	Issue/Publication Date (MM-DD-YYYY)		
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							5	TATUS	<del></del>		
Examiner Initials	Document Number	Kind Code (if known)	Country	Date of Publication (MM-DD-YYYY)	Translation	Partial Translation	Eng. Lang. Summary	Search Report	IPER	Abstract	Cited in Spec
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800	31 18 218	A1	DE	04-22-1982						Х	Х
30000	33 17 064	A1	DE	11-15-1984						X	Х
1/	2 189 245	Α	GB	10-21-1987							Х
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	NON-PATENT LITERATURE DOCUMENTS						
Examiner Initials	molde name of the dame (in or a first terror, the of the article (which appropriate), the of the terr (book, magazine, journal,						
/J.L./	Tarasiuk et al., "Stability of 2-Chloro-2'-Deoxyadenosine at Various pH and Temperature", Archivum Immunologiae et Therapiae Experimentalis, Vol. 42, pp. 13-15, 1994, published by Birkhauser Publishers Ltd., Basel, Switzerland						
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Examiner Signature	/Jonathan Lau/	Date Considered	03/26/2008
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# FIRST INFORMATION DISCLOSURE STATEMENT BY APPLICANT

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Complete if Known				
Application Number	10/551,205			
Filing Date				
First Named Inventor	Nicholas S. Bodor			
Examiner Name				
Attorney Docket No.	0056192-000024			

NON-PATENT LITERATURE DOCUMENTS		
Examiner Initials	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.	
/J.L./	Tortorella et al., Current Opinion on Investigational Drugs, 2(12), pp. 1751-1756, 2001, published by PharmaPress Ltd., London, GB	
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#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	) MAIL STOP AMENDMENT
Nicholas Bodor et al.	) Group Art Unit: 1623 )
Application No.: 10/551,205	) Examiner: JONATHAN S LAU
Filed: November 14, 2006	) Confirmation No.: 4092
For: ORAL FORMULATIONS OF	)
CI ADRIBINE	)

#### **REPLY AND AMENDMENT**

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In response to the Office Action dated April 4, 2008, please first amend the above-identified patent application as follows:

#### AMENDMENTS TO THE SPECIFICATION:

Please replace the paragraph at page 22, lines 8-16 of the specification with the following amended paragraph:

The compositions of the invention are particularly suitable as modalities for the treatment of any cladribine-responsive disease. Several disease states responsive to cladribine are well-documented in the literature (see *infra*). For any target disease state, an effective amount of the complex cladribine-cyclodextrin complex complex, *i.e.* the amorphous mixture of the optimized amorphous saturated cladribine-amorphous cyclodextrin complex with amorphous free cladribine as described above is used (e.g., an amount affective effective for the treatment of multiple sclerosis, rheumatoid arthritis, or leukemia).

Please replace the paragraph at page 23, lines 7-28, of the specification with the following amended paragraph:

Moreover, the route of administration for which the therapeutically effective dosages are taught in the literature should be taken into consideration. While the instant compositions optimize the bioavailability of cladribine following oral administration, it will be appreciated that even optimal bioavailability from oral dosage forms is not expected to approach bioavailability obtain obtained after intravenous administration, particularly at early time points. Thus, it is often appropriate to increase a dosage suggested for intravenous administration to arrive at a suitable dosage for incorporation into a solid oral dosage form. At the present time, it is envisioned that, for the treatment of multiple sclerosis, 10 mg of cladribine in the instant complex cladribine-cyclodextrin complex in the instant solid dosage form would be administered once per day for a period of five to seven days in the first month, repeated for another period of five to seven days in the second month, followed by ten months of no treatment. Alternatively the patient would

Attorney's Docket No. 0056192-000024 Application No. 10/551,205 Page 3

be treated with 10 mg of cladribine in	the instant complex cladribine-
cyclodextrin complex in the instant do	sage form once per day for a period of
five to seven days per month for a total	al of six months, followed by eighteen
months of no treatment. For further d	osing information, see also U.S.
	Provisional Patent
Application No. [[	]] <del>[IVAX0021-P-</del>
USA/Attorney Docket No. 033935-011	l], and U.S. Provisional Patent
Application No. [[]] [IV/	X0022-P-USA/Attorney Docket No.
033935-012], both entitled "Cladribine	Regimen for Treating Multiple
Sclerosis", both filed on March 25, 20	04 and incorporated by reference
herein in their entireties.	

#### **AMENDMENTS TO THE CLAIMS:**

This listing of claims will replace all prior versions, and listings, of claims in the application:

#### **LISTING OF CLAIMS:**

- 1. (Currently Amended) A pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein.
- 2. (Currently Amended) The pharmaceutical composition according to Claim 1, wherein the <u>complex cladribine-cyclodextrin</u> complex is saturated with cladribine.
- 3. (Previously Presented) The composition according to Claim 1, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin, hydroxypropyl-γ-cyclodextrin, randomly methylated β-cyclodextrin, carboxymethyl-β-cyclodextrin or sulfobutyl-β-cyclodextrin.
- 4. (Previously Presented) The composition according to Claim 1, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 5. (Previously Presented) The composition according to Claim 1, wherein the amorphous cyclodextrin is hydroxypropyl-y-cyclodextrin.
- 6. (Previously Presented) The composition according to Claim 1, wherein the weight ratio of cladribine to amorphous cyclodextrin is from about 1:10 to about 1:16.

- 7. (Original) The composition according to Claim 6, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 8. (Original) The composition according to Claim 7, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 9. (Original) The composition according to Claim 7, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
- 10. (Original) The composition according to Claim 6, wherein the amorphous cyclodextrin is hydroxypropyl-y-cyclodextrin.
- 11. (Currently Amended) The composition according to Claim 1 Claim 2, wherein the approximate molar ratio of cladribine to amorphous cyclodextrin corresponds to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin.
- 12. (Previously Presented) The composition according to Claim 1, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 13. (Withdrawn and Currently Amended) A method for enhancing the oral bioavailability of cladribine comprising orally administering to a subject in need thereof a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein.

- 14. (Withdrawn and Currently Amended) The method according to Claim 13, wherein the <u>complex cladribine-cyclodextrin</u> complex is saturated with cladribine.
- 15. (Withdrawn) The method according to Claim 13, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin, hydroxypropyl-γ-cyclodextrin, randomly methylated β-cyclodextrin, carboxymethyl-β-cyclodextrin or sulfobutyl-β-cyclodextrin.
- 16. (Withdrawn) The method according to Claim 13, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 17. (Withdrawn) The method according to Claim 13, wherein the amorphous cyclodextrin is hydroxypropyl-y-cyclodextrin.
- 18. (Withdrawn) The method according to Claim 13, wherein the weight ratio of cladribine to amorphous cyclodextrin is from about 1:10 to about 1:16.
- 19. (Withdrawn) The method according to Claim 18, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 20. (Withdrawn) The method according to Claim 19, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 21. (Withdrawn) The method according to Claim 19, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
- 22. (Withdrawn) The method according to Claim 18, wherein the amorphous cyclodextrin is hydroxypropyl-γ-cyclodextrin.

- 23. (Withdrawn and Currently Amended) The method according to Claim 13 Claim 14, wherein the approximate molar ratio of cladribine to amorphous cyclodextrin corresponds to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin.
- 24. (Withdrawn) The method according to Claim 13, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 25. (Withdrawn and Currently Amended) A method for the treatment of symptoms of a cladribine-responsive condition in a subject suffering from said symptoms comprising orally administering to said subject a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein.
- 26. (Withdrawn and Currently Amended) The method according to Claim 25, wherein the <u>complex cladribine-cyclodextrin</u> complex is saturated with cladribine.
- 27. (Withdrawn) The method according to Claim 25, wherein the cladribine-responsive condition is selected from the group consisting of multiple sclerosis, rheumatoid arthritis and leukemia.
- 28. (Withdrawn) The method according to Claim 27, wherein the cladribine-responsive condition is multiple sclerosis.

- 29. (Withdrawn) The method according to Claim 25, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin, hydroxypropyl-γ-cyclodextrin, randomly methylated β-cyclodextrin, carboxymethyl-β-cyclodextrin or sulfobutyl-β-cyclodextrin.
- 30. (Withdrawn) The method according to Claim 25, wherein the weight ratio of cladribine to amorphous cyclodextrin is from about 1:10 to about 1:16.
- 31. (Withdrawn) The method according to Claim 25, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 32. (Withdrawn) The method according to Claim 31, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 33. (Withdrawn) The method according to Claim 31, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
- 34. (Withdrawn) The method according to Claim 25, wherein the amorphous cyclodextrin is hydropropyl-y-cyclodextrin.
- 35. (Withdrawn) The method according to Claim 25, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).

#### 36.-55. (Cancelled)

56. (Currently Amended) A complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex.

- 57. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, saturated with cladribine.
- 58. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin, hydroxypropyl-γ-cyclodextrin, randomly methylated β-cyclodextrin, carboxymethyl-β-cyclodextrin or sulfobutyl-β-cyclodextrin.
- 59. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 60. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, wherein the amorphous cyclodextrin is hydroxypropyl-γ-cyclodextrin.
- 61. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, wherein the weight ratio of cladribine to amorphous cyclodextrin is from about 1:10 to about 1:16.
- 62. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 61, wherein the amorphous cyclodextrin is hydroxypropyl-β-cyclodextrin.
- 63. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 62, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 64. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 62, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.

- 65. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 61, wherein the amorphous cyclodextrin is hydroxypropyl-γ-cyclodextrin.
- 66. (Currently Amended) The <u>complex cladribine-cyclodextrin</u> complex according to Claim 56, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 67. (Withdrawn and Currently Amended) A process for the preparation of a complex cladribine-cyclodextrin complex as claimed in Claim 56, which comprises the steps of:
- (i) combining cladribine and an amorphous cyclodextrin in water at a temperature of from about [[40]] <u>45</u> to about 80°C and maintaining said temperature for a period of from about 6 to about 24 hours;
  - (ii) cooling the resultant aqueous solution to room temperature; and
  - (iii) lyophilizing the cooled solution to afford an amorphous product.
- 68. (Withdrawn) A process according to Claim 67, further comprising a filtration step following step (ii).
- 69. (Withdrawn) A process according to Claim 67, wherein step (i) is performed at a temperature of from about 45 to about 60°C.
- 70. (Withdrawn) A process according to Claim 67, wherein step (i) is performed at a temperature of from about 45 to about 50°C.
- 71. (Withdrawn) A process according to Claim 69, wherein step (i) is performed with stirring.

- 72. (withdrawn) A process according to Claim 71, wherein step (i) is performed for a period of from about 6 to about 9 hours.
- 73. (Withdrawn) A process according to Claim 67, wherein step (ii) is performed for a period of from about 6 to about 9 hours.
- 74. (Withdrawn) A process according to Claim 67, wherein step (iii) comprises an initial freezing stage in which the solution is cooled to from about -40 to about -80° C, and held at said temperature for a period of from about 2 to about 4 hours.
- 75. (Withdrawn) A process according to Claim 74, wherein, in the initial freezing stage of step (iii), the solution is cooled to about -45°C.
- 76. (Withdrawn) A process according to Claim 67, wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-β-cyclodextrin are introduced in step (i).
- 77. (Withdrawn) A process according to Claim 67, wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-β-cyclodextrin are introduced in step (i).
- 78. (Withdrawn) A process according to Claim 76, wherein 825 parts by volume of water are introduced in step (i).
- 79. (Withdrawn) A process according to Claim 67, wherein the lyophilization step (iii) comprises:
- (a) an initial freezing stage in which the complexation solution is brought to from about -40°C to about -80°C for approximately 2 to 4 hours;
- (b) a primary drying stage at about -25°C for approximately 80 to 90 hours; and
  - (c) a secondary drying stage at about 30°C for approximately 15 to 20 hours.

- 80. (Withdrawn) A process according to Claim 79, wherein stage (a) of the lyophilization is conducted at about -45°C for approximately 3 to 4 hours.
- 81. (Withdrawn) A process according to Claim 79, wherein stage (b) of the lyophilization is conducted under a pressure of about 100 mTorr.
- 82. (Currently Amended) A pharmaceutical composition obtainable by a process comprising the steps of:
- (i) combining cladribine and an amorphous cyclodextrin in water at a temperature of from about [[40]] <u>45</u> to about 80°C and maintaining said temperature for a period of from about 6 to about 24 hours;
  - (ii) cooling the resultant aqueous solution to room temperature;
  - (iii) lyophilizing the cooled solution to afford an amorphous product; and
  - (iv) formulating the amorphous product into a solid oral dosage form.
- 83. (Original) A pharmaceutical composition according to Claim 82, wherein the process further comprises a filtration step following step (i) or (ii).
- 84. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (i) of the process is performed at a temperature of from about 45 to about 60°C.
- 85. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (i) of the process is performed at a temperature of from about 45 to about 50°C.
- 86. (Previously Presented) A pharmaceutical composition according to Claim 84, wherein step (i) of the process is performed with stirring.

- 87. (Original) A pharmaceutical composition according to Claim 86, wherein step (i) of the process is performed for a period of from about 6 to about 9 hours.
- 88. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (ii) of the process is performed for a period of from about 6 to about 9 hours.
- 89. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (iii) comprises an initial freezing stage in which the solution is cooled to from about -40 to about -80°C, and held at said temperature for a period of from about 2 to about 4 hours.
- 90. (Original) A pharmaceutical composition according to Claim 89, wherein, in the initial freezing stage of step (iii), the solution is cooled to about -45°C.
- 91. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process.
- 92. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process.
- 93. (Previously Presented) A pharmaceutical composition according to Claim 91, wherein 825 parts by volume of water are introduced in step (i) of the process.
- 94. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein the lyophilization step (iii) of the process comprises:
- (a) an initial freezing stage in which the complexation solution is brought to from about -40°C to about -80°C for approximately 2 to 4 hours;

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- (b) a primary drying stage at about -25°C for approximately 80 to 90 hours; and
  - (c) a secondary drying stage at about 30°C for approximately 15 to 20 hours.
- 95. (Original) A pharmaceutical composition according to Claim 94, wherein stage (a) of the lyophilization is conducted at about -45°C for approximately 3 to 4 hours.
- 96. (Previously Presented) A pharmaceutical composition according to Claim 94, wherein stage (b) of the lyophilization is conducted under a pressure of about 100 mTorr.
- 97. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein the formulation step (iv) of the process comprises blending the complex with magnesium stearate and compressing into tablets.
- 98. (Original) A pharmaceutical composition according to Claim 97, wherein magnesium stearate is pre-mixed with sorbitol powder before blending with the complex.

#### REMARKS

Entry of the foregoing and reexamination and reconsideration of the subject application, as amended, pursuant to and consistent with 37 C.F.R. § 1.112, are respectfully requested in light of the following remarks.

#### THE DRAWINGS

Applicants appreciate the Examiner's acceptance of the drawing as filed.

#### INFORMATION DISCLOSURE STATEMENTS

Applicants thank the Examiner for considering the four Information Disclosure Statements previously filed herein.

#### FILING DATES TO WHICH CLAIMS ARE ENTITLED

The Examiner has assessed the filing dates to which he believes the claims which he has examined are entitled. Thus, Claims 1-12, 56-66 and 82-98 have been assessed by the Examiner in regard to the earliest filing date to which he believes they are entitled.

Applicants have amended Claim 82 hereinabove so that step (i) is conducted at a temperature from about 45 to about 80°C rather than from about 40 to about 80°C as previously recited. This revised range is not only supported by the instant application (e.g., page 13, lines 21-25) but also by page 12, lines 20-23, of Provisional Appln. No. 60/541,247, filed February 4, 2004; moreover, step (ii) is disclosed at least on page 14, line 3 and in Example 2 of 60/541,247; step (iii) at least on page 14, line 6 and Example 2 of 60/541,247; and step (iv) at least on page 17, lines 25-27, page 18, lines 7-10 and Example 3 of 60/541,247. Claim 88 is supported at least by page 12, lines 20-22 of 60/541,247. Thus, applicants concur with the Examiner that the filing dates of Claims 1-11, 56-65, 84, 86 and 87 are the filing date of Application No. 60/541,247, filed February 4, 2004, but add that the filing dates of Claims 82 and 88 are also the February 4, 2004 filing date of Application No. 60/541,247.

Applicants concur with the Examiner's assessment that <u>Claims 12, 66, 83, 85</u> and 89 are entitled to the effective filing date of the present application; however, as

a national phase application, this application and thus Claims 12, 66, 83, 85 and 89 are entitled to the international filing date of PCT/US04/09387, that is, March 26, 2004. The Examiner's reference to November 14, 2006 as the filing date for these claims is incorrect, that date simply being the date on which the requirements of the last of the 371(c)(1), (c)(2) and (c)(4) requirements were received by the USPTO. In the official Notice of Acceptance of Application under 35 U.S.C. 371 and 37 C.F.R. 1.495, it is clearly stated: "The filing date of the above identified application is the international filing date of the international application (Article 11(3) and 35 U.S.C. 363)."

Despite the foregoing, it is not understood why the Examiner has found it necessary to assess the priority dates of the examined claims, as no art has been cited which would make it necessary to make such an assessment.

#### **ELECTIONS/RESTRICTIONS**

Applicants' election, with traverse, of the invention of Group I, Claims 1-12, 56-66 and 82-98 has been acknowledged and acted upon. Applicants continue to maintain that the amorphous nature of the various entities which make up the complex, that is the intimate amorphous admixture of (a) an amorphous inclusion complex of cladribine with amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex which is formulated into a solid dosage form is not disclosed or suggested by Schultz et al. even when read in conjunction with WO 97/18839, as Schultz et al.'s melt extrusion product would not be inherently the same as applicants' Claim 1 product. Applicants' reasons for so stating are set forth in the discussion of the references herein below. Based on the discussion below, applicants submit that because the elected claims are in fact patentable over the art of record, there is indeed the unifying feature to all of the claims which applicants pointed to earlier. Therefore, the withdrawn claims should be rejoined and examined.

#### **OBJECTIONS TO THE SPECIFICATION**

The disclosure has been objected to because of the blanks identifying provisional application numbers on page 23. By the foregoing amendment,

applicants have deleted the entire sentence containing the blanks because the applications in question have been abandoned.

The disclosure has also been objected to because of a typographical error on page 22, line 12. Applicants have corrected the error by the foregoing amendment.

It is believed that these amendments overcome the objections to the specification.

#### **CLAIM REJECTIONS - 35 U.S.C. § 112**

Claims 2, 11 and 57 are also rejected under 35 U.S.C. § 112, second paragraph, as being indefinite because of use of the term "saturated." Applicants submit that the claims are indeed clear and particularly point out and distinctly claim what applicants regard as their invention.

The Federal Circuit has made it very clear that definiteness of claim language must not be analyzed in a vacuum but rather (1) in light of applicants' specification, (2) in light of the prior art, and (3) in light of the manner in which the claims would be interpreted by one of ordinary skill in the relevant art. When analyzed in accord with Federal Circuit decisions, applicants' claims are definite. Applicants' claims are understandable and define what they regard as their invention; according to the C.C. P.A. decision In re Kamal et al. (CCPA 1968) 158 USPQ 120, such claims meet the requirements of the second paragraph of 35 U.S.C. § 112. In an effort to make these claims and others containing similar language even clearer, applicants have modified the language that refers to the saturated complexes to make it clear that it is the complex cladribine-cyclodextrin complexes which are saturated. Applicants have also modified the language of Claim 11; it would of course be apparent to the person of ordinary skill that applicants were referring to a point on the curve of the phase solubility diagram. The claims as amended have the same scope as prior to the amendment; these are not narrowing amendments but merely clarification of the subject matter to which the claims were previously directed.

The Examiner has noted that the term "saturated" is not defined in the claims, but applicants submit that it is the function of the specification, not the claims, to define terms. Applicants have certainly explained what they mean by saturated, not only by the disclosure at page 10, lines 1-13, but also by the disclosure at page 6,

line 20 to page 7, line 2; by the disclosure beginning at page 13, line 14 through page 14, line 16, which details the procedure used to develop the phase solubility curve; and by the disclosure at page 15, lines 5-29. Very specific information is given, not only as to time and temperature and subsequent filtration, on page 13, lines 21-26, but also in the discussion extending from page 16, line 1 to page 17, line 14. The phase solubility diagram and the discussion of the phase solubility diagram in Example 1 (and by reference, the complexation portion of Example 2) describe exactly how this phase solubility diagram/curve was generated. One of ordinary skill need only select a point on the phase solubility curve to identify the proportion of cladribine and cyclodextrin appropriate for the described saturated complexes for the conditions used in applicants' study. Alternatively, one of ordinary skill can repeat applicants' study to obtain the same curve, or can create his/her own phase solubility diagram for other conditions which he/she selects. The point is that applicants' work is reproducible, based on the teachings of their specification; selection of the same conditions as described will afford the same results; thus, the meaning of the claims which use the word "saturated" and which refer to the phase solubility diagram is clear to one of ordinary skill. As to the Examiner's complaint that no standard is given such as temperature, pressure or solvent, this is manifestly untrue for it is perfectly clear that the solvent disclosed in the specification is water and that the temperature and time are discussed with particularity in the specification, including the Examples, as already pointed out. Pressure is not mentioned because the work was carried out at atmospheric pressure, as would be understood by the skilled worker (who would know that pressure need be indicated only if it deviates from atmospheric pressure). Therefore, while there is no need to determine the amounts for each composition, at least when the cyclodextrin is hydroxypropyl-β-cyclodextrin or even hydroxypropyl-y-cyclodextrin (page 17, lines 9-14) and the phase solubility curve provided by applicants can be used, it would be a very routine matter for one of ordinary skill to create such a curve for other amorphous cyclodextrins or to merely combine cladribine with the chosen cyclodextrin using the conditions specified by applicants and then remove excess cladribine. This is a simple procedure given all of applicants' teachings; it is not rocket science but rather is well within the skill in the art.

For at least the reasons set forth above, applicants submit that the 35 U.S.C. § 112, second paragraph, rejection is untenable and should be withdrawn.

#### CLAIM REJECTIONS - 35 U.S.C. § 102

Claims 1-15, 11, 56-60, 82-90 and 84-98 have been rejected under 35 U.S.C. § 102(b) as being anticipated by Schultz et al. U.S. Patent No. 6,194,395, as evidenced by Baert et al. WO 97/18839.

Before discussing the cited references, applicants would like to discuss the amendments made to the claims hereinabove which make clearer what applicants regard as their invention. The Examiner is thanked for his very thorough review of the specification and the claim language, which has made it possible for applicants to see that some of their original language might have been open to misinterpretation while other language could be interpreted more broadly than they had intended. The amendments to Claims 2, 11 and 57 (as well as to withdrawn claims containing corresponding language) clarify that it is the entire complex cladribine-cyclodextrin complex which is saturated and that the point is located on the curve defining the saturated complexes as in the Figure. Applicants have also amended Claims 1 and 56 (and thus their dependent claims as well), as well as corresponding withdrawn claims, so that both Claims 1 and 56 now specify that the complex cladribinecyclodextrin complex is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a noninclusion complex. This of course excludes anything else from the complex. Claim 1, drawn to a pharmaceutical composition comprising the complex, has been further amended (as have the withdrawn claims containing corresponding language) to specify that the composition comprises no significant amount of free crystalline cladribine therein. This means that no significant amount of free cladribine can be detected considering the sensitivity of the analytical method; see Example 2, page 31, lines 3-13, where this language finds specific support. Applicants teach throughout this application that free crystalline cladribine which is not in the complex is excluded; see for example, page 13, lines 19-28; page 16, lines 1-12 and 13-28; page 20, line 28 to page 21, line 11; page 21, lines 24-29; Example 1, pages 26-28;

and Example 2, page 28, line 3 to page 29, line 26 and of course page 31, lines 3-13. As described, excess cladribine is typically removed from solutions of the complex by filtering it off after the complex complex has been formed in water; subsequent lyophilization of the filtered solution and minimal further processing affords the claimed solid oral dosage form. Therefore, the claim amendments clearly do not introduce new matter.

The Examiner states that Schultz et al. disclose a solid pharmaceutical oral dosage form comprising cladribine and cyclodextrin and applicants agree. However, the Examiner claims that Schultz et al.'s disclosure meets the limitations of instant Claims 1 and 56, which applicants regard as an unwarranted conclusion. Similarly, applicants find no evidence in Schultz et al. that the Schultz et al. solid product is substantially identical to a product-by-process meeting the limitations of instant Claims 82-90 and 94-96.

Schultz et al. disclose the use of either crystalline or amorphous cyclodextrins for their compositions, since some of those named by the patentees are known to be crystalline while others are known to be amorphous. Applicants' complexes and compositions utilize only amorphous cyclodextrins. Thus, many cyclodextrins disclosed by Schultz et al. would be inoperative in the present invention, as they would afford crystalline rather than amorphous products. In stating that the limitations of Claims 3-5 and 58-60 are met by Schultz et al., the Examiner is focusing only on the cyclodextrins in common; he does not address the basic differences between the Schultz et al. solid dosage form and applicants' products. Applicants will agree, however, that Schultz et al.'s preferred cyclodextrin is hydroxypropyl- β -cyclodextrin, which is a cyclodextrin also specified in many of applicants' claims. Again, applicants do not dispute that the excipients may be (but are not necessarily) the same, but this does not arrive at the products of instant Claims 3 and 58 or 97 or 98. As to the amounts of cladribine and cyclodextrin, Schultz et al. disclose weight ratios of from 1 to about 15 mg, of cladribine to about 100 to 500 mg. of a cyclodextrin; this can give a cladribine:cyclodextrin ratio of anywhere from 1:500 to 15:100, or from 1:500 to 1:6.67. If one took the lower limits of each in ratio to the upper limits of each, one would arrive at ratios from 1:100 to 15:500, or from 1:100 to 1:33.34. Most of the 1:500 to 1:6.67 ratio does not even

encompass applicants' ratio, while the 1:100 to 1:33.34 does not embrace it at all. Certainly no guidance in this respect is given by Schultz et al. Moreover, Claim 11 herein has been reworded to clarify that the point is on the curve, as described in the instant specification, and certainly this feature is not disclosed in any way by Schultz et al. Moreover, the instant claims no longer allow for the presence uncomplexed cladribine in either the composition or the complex.

The Examiner correctly states that the Schultz et al. patent incorporates by reference the method of making their solid oral dosage form by utilizing the meltextrusion process of Baert et al. The Baert et al. process is carried out by mixing the cyclodextrin and the active ingredient, heating until melting one of the components, forcing the mixture through one or more nozzles, and cooling until the mixture solidifies (page 5, line 24-29). Milling may follow. The term "melting" is used broadly by Baert et al. and includes transition to a glass; in particular cases, one component melts and the other dissolves therein forming solid solutions (page 5, lines 8-12). The extruded material may contain amorphous material or a solid solution (page 7, line 35 to page 8, line 7). While amorphous products are of interest, those which are mainly a solid solution are preferred (page 8, lines 11-23). In Table 1, on page 30, several different mixtures of hydroxypropyl- β -cyclodextrin and selected drugs were subjected to the Baert et al. process. As noted on page 13, lines 5-6, in every case, the mixture using this cyclodextrin gave a solid solution. The Examiner will note from Table 1 that the temperatures used, regardless of the identity of the drug, went as high as 292°C., with the temperatures for the itraconazole/HPβCD mixtures reaching 279°C-280°C. According to *The Merck Index* (copy of excerpt enclosed), itraconazole melts at 166.2°C while HPBCD melts at 278°C according to LookChem (copy of excerpt also attached).

In addition to the teachings of Baert et al. noted above, applicants draw the Examiner's attention to three of Baert et al.'s teachings which are of particular importance here:

1. On page 4, lines 5-7, Baert et al state:

The compounds that are suitable to be used in this technique are compounds that show no appreciable decomposition at the temperatures needed to melt and extrude the mixture of said

one or more active ingredients with the cyclodextrin or cyclodextrins.

2. On page 6, lines 14-19, Baert et al. state:

The possible formation of these solid solutions is one of the further advantages of the present invention. It will be appreciated by a person skilled in the art that mixing two or more solids, i.e., one or more cyclodextrins and the active ingredient or ingredients, and subsequently melting these solids together give rise to different products than when the said solids are first brought into contact with water or another solvent and then extruded.

3. While Baert et al. have general teachings regarding ratios of from 1:100 to 100:1, particularly 1:10 to 10:1, especially 1:5 to 5:1, 1:3 to 3:1, preferably 1:1, Table 1 therein uses ratios of active ingredient: HPβCD of 1:3 or 1:1. Table 2 utilizes 1:1 ratios.

The only solid dosage form envisioned by Schultz et al. is a melt-extrusion product of cladribine and cyclodextrin prepared according to Baert et al. There is no evidence that such a product was ever prepared. Indeed, cladribine melts at 220-235°C with decomposition; see the enclosed excerpt from Linscott's Directory (copy attached) as well as that from The Merck Index (also enclosed). Thus, cladribine decomposes well below the 278°C melting point of HPβCD and well below the temperature used by Baert et al. for their melt extrusion; cladribine is therefore not suitable for the Baert et al. process, according to Baert et al.'s teaching that suitable compounds show no appreciable decomposition at the temperature they use (point 1 above).

Furthermore, Baert et al.'s teaching on page 6 that their melt-extrusion process affords different products than when their solids are first brought into contact with water (point 2 above) militates against the Examiner's finding that a cladribine/cyclodextrin product prepared by Baert et al.'s process is the same as applicants' product, which is, in fact, prepared by first contacting cyclodextrin with water. Indeed, it is the use of water that enables the formation of cyclodextrin-drug complexes; it is by complex formation that the water solubility of many drugs has

been previously improved. There is no teaching by Baert et al. that would lead one of ordinary skill to conclude that Baert et al made solid <u>complexes</u>; indeed, Baert et al. specifically teach on page 6 that their products are different from products obtained by first dissolving cyclodextrin and drug in water. Thus, a melt-extrusion product of cladribine and cyclodextrin cannot anticipate applicants' product which comprises a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein. Applicants' product is prepared by preparing the complex in water; Baert et al. teach they obtain a different product than one that can be obtained from water; moreover, applicants' process uses temperatures up to only about 80°C, far below the decomposition temperature for cladribine and far below the temperatures used by Baert et al. See Claims 82-89 herein.

Further, for an anticipation to be inherent, the reference must <u>always</u> provide applicants' product. There is no reason to assume that Schultz et al's solid product <u>ever</u> is the same as applicants'; indeed, Baert et al. clearly teach that it is <u>different</u>.

For at least the reasons set forth above, the anticipation rejection of Claims 1-5, 11, 56-60, 82-90 and 94-98 based on Schultz et al. as evidenced by Baert et al. is untenable and should be withdrawn.

#### **CLAIM REJECTIONS - 35 U.S.C.§ 103**

Claims 1, 6-10, 12, 56, 61-66, 82 and 91-93 have been rejected under 35 U.S.C. § 103(a) as being unpatentable over Schultz et al. in view of Baert et al. We respectfully disagree.

Both Schultz et al. and Baert et al. are fully discussed above. In referring to the ratios disclosed by Schultz et al. for their melt extrusion solid dosage form, the Examiner has chosen to ignore the 1 mg. dosage at which the amount of cladribine begins; therefore, the range disclosed by Schultz et al. for cladribine:cyclodextrin varies from about 1:500 to 1:6.67. This is not even in accord with Baert et al.'s ratio of from about 1:1 to about 1:3. And if one uses Schultz et al's ratios proportionately,

low:low and high:high, one arrives at from 1:100 to 15:500 (1:33.34). In the former case, there is absolutely nothing in Schultz et al. that would lead to a specific ratio of from about 1:10 to about 1:16 or about 1:11 or about 1:14; it is applicants' own teachings which lead to these ratios. Likewise, the features of the other claims rejected under 35 U.S.C. § 103 are not disclosed by Schultz et al. Certainly Baert et al doesn't teach these ratios and if one looks at Schultz et al.'s suggested amounts proportionately, applicants' ratios are not even broadly encompassed by the reference. Still further, as noted earlier, Baert et al. clearly teach that the drug-cyclodextrin solid products of their melt extrusion process are distinctly different from products prepared in water; since applicants' products are prepared in water, they cannot possibly be the same as those obtained by the Baert et al. process incorporated by reference by Schultz et al.

Baert et al.'s ratios of active ingredient have been interpreted as mole ratios by the Examiner. There is no good reason for such an interpretation. The Examiner reasons that the fact that the active ingredients have different molecular weights leads to this interpretation, yet there are cyclodextrins of different molecular weights contemplated by Schultz et al. and by the present inventors and the ratios of Schultz et al. are clearly by weight (col. 6, lines 23-31), just as applicants' ratios are clearly weight ratios, e.g., Claim 8. Absent a teaching to the contrary, one of ordinary skill would assume that the ratios of Baert et al. are also weight ratios. At any rate, the Baert et al. melt-extrusion product is not one obtained by complexation in water; Baert et al. teach their melt-extrusion product is different from a product whose preparation begins by dissolving the drug and cyclodextrin in water. Therefore, any product that Schultz et al. might produce from cladribine and cyclodextrin subjected to Baert et al.'s melt extrusion product cannot be the same as applicants' complex cladribine-cyclodextrin complex which must be obtained from an aqueous solution which is treated in a specific manner. Baert et al. never suggests that they obtain a complex by their melt-extrusion process, much less one meeting the requirements of applicants' claims. Indeed, Baert et al. emphasize that their process, which is different, affords a different product than that obtained by first dissolving the drug and cyclodextrin. Likewise, applicants emphasize that applicants' process is strikingly different from Baert et al.'s process and thus logically would not afford the

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product which Schultz et al. would be expected to obtain by subjecting cladribine and cyclodextrin to Baert et al.'s process. Moreover, applicants have formed a very special complex which contains a large amount of cladribine as an amorphous inclusion complex and as amorphous free cladribine associated with the cyclodextrin as a non-inclusion complex. Note too that the <u>free</u> cladribine associated with the non-inclusion product is amorphous, in contrast to the cladribine starting material, which is crystalline. Note also that applicants produce their product by first complexing in water at temperatures of from about 45°C to about 80°C, far below the temperatures used by Baert et al. Cladribine actually decomposes at temperatures below that used by Baert et al.

It is clear from the foregoing that a molecular inclusion complexation process, let alone the particular inclusion process utilized by applicants to form their unique complex cladribine-cyclodextrin complex, is not inherent in Baert et al.'s melt extrusion process and that Baert et al.'s process gives a different product. To hold otherwise would be to ignore Baert et al.'s own teachings.

In view of the foregoing, it is submitted that the present application is free of all record rejections and objections. Further, favorable action in the form of a Notice of Allowance is believed to be next in order and is earnestly solicited.

Respectfully submitted,

BUCHANAN INGERSOLL & ROONEY PC

Date:October 3, 2008

Registration No. 26254

P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620

Attachments:

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"Cladribine", *The Merck Index*, (2001), pp. 407-408, Thirteenth Edition, Merck & Co., Inc., Whitehouse Station, NJ

"Itraconazole", *The Merck Index*, (2001), p. 938, Thirteen Edition, Merck & Co., Inc. Whitehouse Station, NJ

"Hydroxypropyl-beta-cyclodextrin, CAS No. 94035-02-6" *LookChem*, http://www.lookchem.com/cas-940/94035-02-6.html, September 23, 2008

"Non-antibody Products (Kits, Proteins, Microbial Antigens, Tissues, Services, etc.) Linscott's Directory of Immunological & Biological Reagents, http://www.linscottsdirectory.com/browse/products/page:36, Records 1,751-1,800 of 130,353, September 11, 2008

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angina and coronary artery disease: C. E. Handler, E. Sowton, ibid. 27, 415 (1984); in hypertension: E. B. Nelson et al., Clin. Pharmacol. Ther. 40, 694 (1986). Comparison of hemodynamic effects of enantiomers: R. P. Hof et al., J. Cardiovasc. Pharmacol. 8, 221 (1986). Series of articles on pharmacology and clinical use: Am. J. Med. 86, 1-146 (1989).

mp 168-170°.

B

mp 140°-170°. S(+)-Form. PN-205-033. Crystals from ether + hexane, mp 142°.  $[\alpha]_0^{20}$  +6.7° (c = 1.5 in ethanol). R(-)-Form. PN-205-034 Crystals from ether + hexane, mp 140°.  $[\alpha]_0^{20}$  -6.7° (c = 1.67 in ethanol). THERAP CAT: Antihypertensive; antianginal.

5263. Israpafant. [117279-73-9] 4-(2-Chlorophenyl)-6,-9-dimethyl-2-[2-[4-(2-methylpropyl)phenyl]ethyl]-6H-thieno-[3,2-f][1,2,4]triazolo[4,3-a][1,4]diazepine;  $(\pm)-4-(o\text{-chlorophen-}$ yl)-2-(p-isobutylphenethyl)-6,9-dimethyl-6H-thieno[3,2-f]-striazolo[4,3-a][1,4]diazepine; Y-24180; Pafnol. C<sub>28</sub>H<sub>29</sub>ClN<sub>4</sub>S; mol wt 489.09. C 68.76%, H 5.98%, Cl 7.25%, N 11.46%, S 6.56%. Platelet activating factor (PAF) antagonist. Prepn: T. Tahara et al., EP 268242; eidem, US 4820703 (1988, 1989 both to Yoshitomi). Pharmacology: M. Terasawa et al., Prostaglandins 40, 553 (1990). Receptor binding study: S. Takehara et al., ibid. 571. Clinical evaluation in asthma: S. Hozawa et al., Am. J. Respir. Crit. Care Med. 152, 1198 (1995).

Colorless crystals from isopropyl ether, mp 129.5-131.5°. Sol in propylene glycol.

THERAP CAT: Antiasthmatic.

5264. Itaconic Acid. [97-65-4] Methylenesuccinic acid; propylenedicarboxylic acid.  $C_5H_6O_4$ ; mol wt 130.10. C 46.16%, H 4.65%, O 49.19%. Obtained by dry distillation of citric acid and subsequent treatment of the anhydride with water. Produced on a large scale by submerged aerobic fermentation using Aspergillus terreus and low cost carbohydrates from beet or cane: Kane et al., US 2385283 (1945 to Pfizer). Synthesis from propargyl chloride, carbon monoxide, nickel carbonyl and water: Chiusoli, US 3025320 (1962 to Montecatini).

Hygroscopic crystals; characteristic odor, d 1.63, mp 162with decompn. Also reported as mp 172° [Kinoshita, Acra Phytochem. (Japan) 5, 273 (1931)]. One gram dissolves in 12 ml water, 5 ml alcohol; very slightly sol in benzene, chloroform, ether, carbon disulfide, petr ether. Keep well closed.

[123258-84-4] 2,3-Dihydro-N-[(3endo)-8-methyl-8-azabicyclo[3.2.1]oct-3-yl]-2-oxo-1Hbenzimidazole-1-carboxamide; 2-oxo-N-1 $\alpha H$ ,5 $\alpha H$ -tropan-3 $\alpha$ yl-1-benzimidazoline-1-carboxamide. C<sub>16</sub>H<sub>20</sub>N<sub>4</sub>O<sub>2</sub>; mol wt yl-1-benzimidazonne-1-candokanide. — 1611-2014-2-2. moi wi 300.35. C 63.98%, H 6.71%, N 18.65%, O 10.65%. Serotonin (5-HT<sub>3</sub>) receptor antagonist. Prepri: M. Turconi *et al.*, **EP** 309423 (1989 to Istituto De Angeli); eidem, US 5223511 (1993 to Boehringer, Ing.); M. Turconi et al., J. Med. Chem. 33, 2101 (1990). Pharmacology: idem et al., Eur. J. Pharmacol. 203, 203 (1991). Mode of action: M. B. Passani et al., Brit. J. Pharmacol. 112, 695 (1994). Clinical efficacy and tolerability: H. Goldschmidt et al., Anti-Cancer Drugs 8, 436 (1997). Review of therapeutic potential: M. B. Passani, R. Corradetti, CNS Drug Reviews 2, 195-213 (1996).

Crystals from acetonitrile, mp 205-207°. LD<sub>50</sub> in mice, rats

(mg/kg): 56, 62 i.v. (Passani). **Hydrochloride.** [127618-28-4] DAU 6215. C<sub>16</sub>H<sub>20</sub>N<sub>4</sub>-O2.HCl; mol wt 336.82. Colorless crystals, mp 270°.

5266. Itraconazole. [84625-61-6] 4-[4-[4-[4-[4-[2-(2,4-Dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1methylpropyl)-3H-1,2,4-triazol-3-one; ( $\pm$ )-1-sec-butyl-4-[p-[4-[p-[(2R\*,4S\*)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1)]1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]- $\Delta^2$ -1,2,4-triazolin-5-one; oriconazole; R-51211; Itrizole; Sporanox; Triasporin. C<sub>35</sub>H<sub>38</sub>Cl<sub>2</sub>N<sub>8</sub>O<sub>4</sub>; mol wt 705.65. C 59.57%, H 5.43%, Cl 10.05%, N 15.88%, O 9.07%. Orally active antimycotic structurally related to ketoconazole, Prepn: J. Heeres, L. J. J. Backx, EP 6711; eidem, US 4267179 (1980, 1981 both to Janssen); J. Heeres et al., J. Med. Chem. 27, 894 (1984). In vitro activity: A. Espinel-Ingroff et al., Antimicrob. Ag. Chemother. 26, 5 (1984). HPLC determn in biological samples: R. Woestenborghs et al., J. Chromatos. 413, 332 (1987). Symposium on pharmacology and clinical efficacy: Rev. Infect. Dis. 9, Suppl 1, S1-S152 (1987). Toxicity data: H. Van Cauteren et al., ibid. S43. Review of clinical pharmacokinetics: J. Heykants et al., Mycoses 32, Suppl 1, 67-87 (1989); of clinical efficacy in dermatophytosis: P. De Doncker, G. Cauwenbergh, *Brit. J. Clin. Pract.* Suppl. 71, 118-122 (1990). *Review:* A. M. Sugar, *Curr. Clin. Topics Inf. Dis.* 13, 74-98 (1993).

Crystals from toluene, mp 166.2°. pKa 3.7. Lipophilic; par tition coefficient (n-octanol/aq buffer of pH 8.1): 5.66. Practically insol in water and dil acidic solns. LD<sub>50</sub> (14 day) in mice, rats, dogs (mg/kg): >320, >320, >200 orally (Van Cauteren) THERAP CAT: Antifungal.

5267. Itramin Tosylate. [13445-63-1] 2-Aminoethanol nitrate mono(4-methylbenzenesulfonate); 2-aminoethanol mitrate mono-p-toluenesulfonate; 2-nitratoethylaminotoluene-p-sulfonate; Cardisan; Tostram; Nilatil. C<sub>0</sub>H<sub>14</sub>N<sub>2</sub>O<sub>6</sub>S; mol wt 278.28. C 38.85%, H 5.07%, N 10.07%, O 34.50%, S 11.52%. Prepn: SE 168308 (1959 to Aktiebolaget Pharmacia), C.A. 54, 244054 (1969) 24405d (1960).

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ely sol in chlobenzene, dec uv max: 260, ne, chloroform;

260, 334 nm

ydroxy-2-meth--10-carboxylic L C 57.94%, H ced by Penicil-Bainier and by Trans. Roy. Soc Nature 167, 995 oc. 1951, 2013, Money, Nature i et al., J. Chem.

at 155°, dec 290zely sol in ethasparingly sol in table to acid and

nethyl-6-octenal 6%, O 10.37% I in many other lissa: Tieman, 54, 236 (1958) 151, 505; Eschi

d 0.848-0.856

α-citronellal. [141-26-4] 3,7-Dimethyl-7-octenal; rhodinal. Liquid. bp<sub>1.4</sub> 51°,  $n_{\rm p}^{20}$  1.4410. [α]<sub>0</sub><sup>20</sup> +9.75°. USE: In soap perfumes; insect repellent

2354. β-Citronellol. [106-22-9] 3,7-Dimethyl-6-octen-1-ol; 2,6-dimethyl-2-octen-8-ol; citronellol; cephrol. C<sub>10</sub>H<sub>20</sub>O; mol wt 156.26. C 76.86%, H 12.90%, O 10.24%. *1*-Form is a constituent of rose and geranium oils. d-Form occurs in Ceylon and Java citronella oils. History: J. L. Simonsen, L. N. Owen, and Java the dependence of the temperature of the Terpenes vol. I (University Press, Cambridge, 2nd ed. 1947). Prepn of (±)-form: Adams, Garvey, J. Am. Chem. Soc. 48, 477 (1926); Ofner et al., Helv. Chim. Acta 42, 2577 (1959). Prepn of (+)-form: Rienäcker, Ohloff, Angew. Chem. 73, 240 (1961); Naves, Tullen, Helv. Chim. Acta 44, 1867 (1961); Eschinazi, J. Org. Chem. 26, 3072 (1961); Rienäcker, Chimia 27, 97 (1973); C. G. Overberger, J. L. Weise, J. Am. Chem. Soc. 90, 3525 (1968); T. Sato et al., Tetrahedron Letters 1980, 3377. Prepn of (-)-form: Ohloff, loc. cit.; Rienäcker, loc. cit.; Shono et al., Tetrahedron Letters 1974, 1295; K. Mori, T. Sugai, Synthesis 1982, 752. Synthesis of (+) or (-)-form from isoprene: Hidai et al., Chem. Commun. 1975, 170. Stereospecific prepn via microbiological (Saccharomyces cerevisiae) reduction: P. Gramatica et al., Experientia 38, 775 (1982). Manuf: Woroch et al.; Bain; Webb, US 2990422; US 3005845; US 3028431 (1961, 1961, 1962, all to Glidden); Eschinasi, US 3052730 (1962 to Givaudan). Abs config of the (+)-form: Freudenberg, Hohmann, Ann. 584, 54 (1953); Freudenberg, Lwowski, ibid. 587, 213 (1954). NMR, HPLC determin of R/S enantiomer ratios: D. Valentine et al., J. Org. Chem. 41, 62 (1976). See also Rhodinol.

R-(+)-β-Citronellol

(+)-Form. Oily liquid, bp 224.5°, bp<sub>10</sub> 108.4°,  $d_4^{20}$  0.8550.  $n_0^{20}$  1.4559.  $[\alpha]_0^{20}$  +5.22°. Very slightly sol in water, miscible Very slightly sol in water, miscible with with alcohol, ether.

(-)-Form.  $\beta$ -Rhodinol; Levocitrol. bp<sub>10</sub> 108-109°. d<sub>4</sub><sup>18</sup> 1.4576.  $[\alpha]_0^{20}$  -4.76°.

(±)-Form. Dihydrogeraniol.  $d_4^{23.5}$  0.851.  $n_D^{23.5}$  1.454. USE: In perfumery.

2355. Citrulline. [372-75-8] N<sup>5</sup>-(Aminocarbonyl)-L-ornithine;  $\delta$ -ureidonorvaline;  $\alpha$ -amino- $\delta$ -ureidovaleric acid;  $N^{\delta}$ carbamylornithine. C<sub>6</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>; mol wt 175.19. C 41.13%, H 7.48%, N 23.99%, O 27.40%.  $H_2$ NCONH(C $H_2$ )<sub>3</sub>CH(N $H_2$ )-COOH. An amino acid, first isolated from the juice of watermelon, Citrullus vulgaris Schrad., Cucurbitaceae: Biochem. Z. 224, 420 (1930); isoln from casein: Wada, ibid. 257, 1 (1933). Synthesis from ornithine through copper complexes: Kurtz, J. Biol. Chem. 122, 477 (1938); by alkaline hydrolysis of arginine: Fox, ibid. 123, 687 (1938); from cyclopentanone oxime: Fox et al., J. Org. Chem. 6, 410 (1941). Crystallization: Matsuda et al., JP 71 174 (1971 to Ajinomoto), C.A. 74, 126056u (1971). Crystal and molecular structure: Naganathan, Venkatesan, Acta Crystallogr. 27B, 1079 (1971); Ashida et al., ibid. 28B, 1367 (1972). Use in asthenia and he patic insufficiency: FR 2198739 (1974 to Hublot & Vallet), C.A. 82, 144952e (1975). Clinical trial in treatment of lysinuric protein intolerance: J. Rajantie et al., J. Pediarr. 97, 927 (1980); T. O. Carpenter et al., N. Engl. J. Med. 312, 290 (1985).

Prisms from methanol + water, mp 222°  $[\alpha]_D^{20} + 3.7$ ° (c = 2). pK<sub>1</sub> 2.43; pK<sub>2</sub> 9.41. Sol in water. Insol in methanol, ethanol

Hydrochloride. [34312-10-2] C<sub>6</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub> HCl. Crystals, dec 185°.  $[\alpha]_D^{22} + 17.9$ ° (c = 2).

Malate (salt). [54940-97-5] Stimol.  $C_6H_{13}N_3O_3.C_4H_6O_5$ ; mol wt 309.27

IHERAP CAT: Treatment of asthenia.

2356. Citrullol. [1390-93-8] C<sub>22</sub>H<sub>38</sub>O<sub>4</sub>; mol wt 366.53. C 72.09%, H 10.45%, O 17.46%. From fruit pulp of Citrullus colocynthis Schrad., Cucurbitaceae: Power, Moore, J. Chem. Soc. 97, 99 (1910); Power, Salway, ibid. 103, 399, 1022 (1913);

Khadem, Rahman, Tetrahedron Letters 1962, 1137. Crystals, mp 282-283°. uv max: 242, 272, 282 nm (log & 2.85, 2.68, 2.68). Sol in pyridine; practically insol in usual or-

ganic solvents

Diacetate. C<sub>26</sub>H<sub>42</sub>O<sub>6</sub>. Crystals, mp 162°.

2357. Citrus Red 2. [6358-53-8] 1-[(2,5-Dimethoxyphenyl)azo]-2-naphthalenol; C.I. Solvent Red 80; C.I. 12156. C<sub>18</sub>-H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>; mol wt 308.33. C 70.12%, H 5.23%, N 9.09%, O 15.57%. Prepn: H. W. Elley, H. W. Daudt, US 2224904 (1940 to Du Pont). Metabolism: J. L. Radomski, J. Pharmacol. Exp. Ther. 134, 100 (1961); 136, 378 (1962). Toxicology: M. Shartatt et al., Food Cosmet. Toxicol. 4, 493 (1966). Review of carcinogenicity studies: IARC Monographs 8, 101-106. See also Colour Index vol. 4 (3rd ed., 1971) p 4033.

Crystals, mp 155-157°. Slightly sol in water; partially sol in ethanol and vegetable oils

USE: To color orange skins.

2358. Civet. Zibeth. Unctuous secretion from receptacles between the anus and genitalia of both male and female civet cat. Constit. Civetone and similar compds.

Semi-solid, yellowish to brown unctuous substance; unpleasant, subacrid, bitter taste; fusible and burns without leaving much residue. Insol in water; partly sol in hot alcohol or in ether.

USE: As a fixative in perfumery.

2359. Civetone. [542-46-1] (Z)-9-Cycloheptadecen-1one. C<sub>17</sub>H<sub>30</sub>O; mol wt 250.42. C 81.54%, H 12.07%, O 6.39%. 17-Membered macrocyclic musk, constituent of civet: Ruzicka, Helv. Chim. Acta 9, 230 (1926); Ruzicka et al., ibid. 10, 695 (1927). Occurs in nature as cis-form. Synthesis of cis-civetone: Stoll et al., ibid. 31, 543 (1948); J. Tsuji, T. Mondai, Tetrahedron Letters 1977, 3285; E. Seoane et al., Chem. & Ind. (London) 1978, 165. Synthesis of trans-form: H. Hunsdiecker, Ber. 77, 185 (1944); H. H. Mathur, S. C. Bhattacharyya, J. Chem. Soc. 1968, 114. Crystal and molecular structure of cis-civetone: G. Bernardinelli, R. Gerdil, Helv. Chim. Acta 65, 558 (1982).

Crystals, mp 31-32°. Musky odor becoming pleasant in extreme dilns.  $d_4^{33}$  0.917.  $bp_{742}$  342°;  $bp_2$  59°.  $n_D^{33}$  1.4830. USE: In perfumery.

2360. Cladribine. [4291-63-8] 2-Chloro-2'-deoxyadenosine; 2-chloro-6-amino-9-(2-deoxy-β-D-erythro-pentofuran-

Consult the Name Index before using this section.

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osyl)purine; 2-chlorodeoxyadenosine; 2-CdA; CldAdo; NSC-105014-F; Leustatin.  $C_{10}H_{1,2}\mathrm{ClN}_3O_3$ ; mol wt 285.69. C 42.04%, H 4.23%, Cl 12.41%, N 24.51%, O 16.80%. Substituted purine nucleoside with antileukemic activity. Prepn as intermediate in synthesis of 2-deoxynucleosides: H. Venner, Ber. 93, 140 (1960); M. Ikehara, H. Tada, J. Am. Chem. Soc. 85, 2344 (1963); eidem, ibid. 87, 606 (1965). Synthesis and biological activity: L. F. Christensen et al., J. Med. Chem. 15, 735 (1972). Stereospecific synthesis: Z. Kazimierczuk et al., J. Am. Chem. Soc. 106, 6379 (1984); R. K. Robins, G. R. Revankar, EP 173059; eidem, US 4760137 (1986, 1988 both to Brigham Young Univ). Specific toxicity to lymphocytes: D. A. Carson et al., Proc. Nat. Acad. Sci. USA 77, 6865 (1980); eidem, Blood 62, 737 (1983). Mechanism of action: S. Seto et al., J. Clin. Invest. 75, 377 (1985). Clinical evaluation in chronic lymphocytic leukemia: L. D. Piro et al., Blood 72, 1069 (1988); in hairy cell leukemia: eidem, N. Engl. J. Med. 322, 1117 (1990).

Crystals from water, softens at 210-215°, solidifies and turns brown (Christensen). Also reported as crystals from ethanol, mp 220° (softens), resolidifies, turns brown and does not melt below 300° (Kazimierczuk).  $[\alpha]_D^{25} - 18.8^\circ$  (c = 1 in DMF). uv max in 0.1N NaOH: 265 nm; in 0.1N HCl: 265 nm.

THERAP CAT: Antineoplastic

2361. Clanobutin. [30544-61-7] 4-[(4-Chlorobenzoyl)-(4-methoxyphenyl)amino]butanoic acid; 4-[p-chloro-N-(p-methoxyphenyl)benzamido]butyric acid; N-(p-chlorobenzoyl)-y-(p-anisidino)butyric acid; Bykahepar. C<sub>18</sub>H<sub>18</sub>ClNO<sub>4</sub>; mol wt 347.80. C 62.16%, H 5.22%, Cl 10.19%, N 4.03%, O 18.40%. Prepn: K. Klemm et al., DE 1917036 corresp to US 3780095 (1971, 1973 both to Byk-Gulden). Series of articles on synthesis, physical and pharmacological properties: Arzneimittel-Forsch. 29, 1-15 (1979). In vitro biochemical study: H. Wolf et al., Biochem. Pharmacol. 29, 1649 (1980). Effect on bile excretion in rats, dogs: P. Berchtold et al., Arzneimittel-Forsch. 30, 1878 (1980).

Cryst from ethyl acetate, mp 115-116°. pKa 5.04. Soly in water at 37°:  $4.02\times10^{-2}$  mol/l at pH 7. LD<sub>50</sub> in rats (mg/kg): >2000 orally; 570 i.v. (Klemm).

THERAP CAT: Choleretic.

THERAP CAT (VET): Choleretic; in treatment of piroplasmosis and anaplasmosis.

2362. Clarithromycin. [81103-1I-9] 6-O-Methylerythromycin; A-56268; TE-031; Biaxin; Clathromycin; Cyllind; Klacid; Klaricid; Macladin; Naxy; Veclam; Zeclar. C<sub>38</sub>H<sub>69</sub>-NO<sub>13</sub>; mol wt 747.95. C 61.02%, H 9.30%, N 1.87%, O 27.81%. Semisynthetic macrolide antibiotic; derivative of erythromycin, q.v. Prepn: Y. Watanabe et al., EP 41355; eidem, US 4331803 (1981, 1982 both to Taisho); and in vitro antibacterial activity: S. Morimoto et al., J. Antibiot. 37, 187 (1984). In vitro and in vivo antibacterial activity: P. B. Fernan-

des et al., Antimicrob. Ag. Chemother. 30, 865 (1986). Comparative antibacterial spectrum in vitro: C. Benson et al., Eur. J. Clin. Microbiol. 6, 173 (1987); H. M. Wexler, S. M. Finegold ibid. 492. HPLC determn in biological fluids: D. Croteau et al., J. Chromatog. 419, 205 (1987). Acute toxicity study: 8, Abe et al., Chemotherapy (Tokyo) 36, Suppl. 3, 274 (1988). Symposium on pharmacology and comparative clinical studies. J. Antimicrob. Chemother. 27, Suppl. A, 1-124 (1991). Comprehensive description: I. I. Salem, Anal. Profiles Drug Subs. Excip., 24, 45-85, (1996).

Colorless needles from chloroform + diisopropyl ether (1:2), mp 217-220° (dec). Also reported as crystals from ethanol, mp 222-225° (Morimoto). uv max (CHCl<sub>3</sub>): 288 nm ( $\epsilon$  27.9). uv max (CHCl<sub>3</sub>): 288 nm, (methanol): 211, 288 nm. [ $\epsilon$ l<sub>1</sub>]  $\epsilon$ l<sub>2</sub> + 90.4° (c = 1 in CHCl<sub>3</sub>). Stable at cidic pH. LD<sub>50</sub> in male, female mice, male, female rats (mg/kg): 2740, 2700, 3470, 2700 orally, 1030, 850, 669, 753 i.p., >5000 all s.c. (Abe).

THERAP CAT: Antibacterial.

2363. Clathrates. Compounds that are capable of trapping other substances within their own crystal lattices. The cavities of the host molecules are classified as cages, tunnels, or layered types, depending on the way they include guest molecules. The geometry of the cavities limits the guest molecules by size and shape, rather than by chemical similarity with the host molecules. Among common clathrates are molecules sieves, cyclotriphosphazenes, and Dianin's compound, as well as hydroquinone, cyclodextrins, o-thymotide, and deoxycholic acid, q.q.v. Cavitands are organic hosts with enforced (rigid) cavities: D. J. Cram, Science 219, 1177 (1983); R. C. Helgeson et al., Chem. Commun. 1983, 101. Comprehensive book: Clathrate Compounds, V. M. Bhatnagar, Ed. (Chemical Pub. Co., New York, 1970) 244 pp. Reviews: D. D. MacNicol et al., Chem. Soc. Rev. 7, 65-87 (1978); E. C. Makin, "Clathration" in Kirk-Othmer Encyclopedia of Chemical Technology Vol. 6 (Wiley-Interscience, New York, 3rd ed., 1979) pp 178-189.

USE: As complexing agent; stabilizing agent. In analytical separations.

2364. Clavulanic Acid. [58001-44-8] [2R-(2α,3Z,5α]] 3-(2-Hydroxyethylidene)-7-oxo-4-oxa-1-azabicyclo-[3.2.0]heptane-2-carboxylic acid; MM 14151. C<sub>8</sub>H<sub>8</sub>NO<sub>3</sub>; mol wt 199.16. C 48.25%, H 4.55%, N 7.03%, O 40.17%. β-Lactamase inhibitor. Antibiotic produced by Streptomyces claviligerus; first reported naturally occurring fused β-lactam containing oxygen. Isoln: M. Cole et al., DE 2517316 (1975 to Beecham), C.A. 84, 72635t (1976); A. G. Brown et al., J. Antibiot. 29, 668 (1976). Structure, x-ray crystallography: T.T. Howarth et al., Chem. Commun. 1976, 266. Total synthesis of (±)-form: P. H. Bentley et al., ibid. 1977, 748, 905; eidem. Tetrahedron Letters 1979, 1889. β-Lactamase inhibition and antibacterial spectrum: C. Reading, M. Cole, Antimicrob. Ag. Chemother. 11, 852 (1977). Mechanism of action: B. G. Spratt et al., ibid. 12, 406 (1977). Antibacterial activity, pharmacology and clinical efficacy of combination with amoxicillin: A. P. Ball et al., Lancet 1, 620 (1980); R. N. Brogden et al., Drugs 23, 337-362 (1981). In vitro and in vivo synergism with ticarcilling. R. Sutherland et al., Am. J. Med. 79, Suppl. 5B, 13 (1985).

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nyl)-4acetoni 2(3H)-1 H<sub>10</sub>Cl<sub>2</sub> N 15.0 eidem. in pige

> THE 23 oxy-N (1-ber peridy mol v 8.56% mide, A. V. J. Pri macc (1980 biotie Phari 214 ( tinal (198 Yaku 2080 radic 1491

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 $Current\ position: \underline{Home} > \underline{Products} > Hydroxypropyl-beta-cyclodextrin$ 

### Hydroxypropyl-beta-cyclodextrin CAS No:94035-02-6

Name: Hydroxypropyl-beta-cyclodextrin

Synonyms: beta-Hydroxypropylcyclodextrin

beta-Cyclodextrin, 2-hydroxypropyl ether

**HPB** 

2-Hydroxypropyl-beta-cyclodextrin

128446-35-5

CAS Number: 94035-02-6

Molecular Formula:  $C_{42}(H)_{70-n}O_{35}(C_3H_7)_n$ 

Melting Point: 278 °C

Safety Description: S24/25 Details

Inquire now List of Suppliers for Hydroxypropyl-beta-cyclodextrin Country

Onbio Inc.
Introduction:HYDROXYPROPYL-BETA-CYCLODEXTRIN

United Sta

Yiming Fine Chemicals Co., Ltd.

Introduction:mp : 267 °C (dec.)

China (Mair

storage temp. : 2-8°C

solubility: H2O: 45 % (w/v)

form solution (clear, colorless)

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Name	Description	Size	Catalog #	Supplier	
СКВВ	Recombinant Human Creatine Kinase BB Isoenzyme	10μg, 50μg, 1mg	CKI- 268	PROSPEC-TANY TECHNOGENE LTD.	More In
<u>Ckdk6</u>	The RP-39008 Ckdk6 protein is a partial length (aa 1-327) bacterially expressed recombinant protein.RP-39008 is suitable for use as a control in ELISA and Western blot applications.The RP-39008 protein is GST-tagged.	10 ug	RP-39008	ABR - AFFINITY BIOREAGENTS INC.	More In
СКММ	Human Creatine Kinase MM	200µg, 1mg, 10mg	CKI- 273	PROSPEC-TANY TECHNOGENE LTD.	More In
CKS-17	Sequence: Leu-Gln-Asn-Arg-Arg-Gly-L eu-Asp-Leu-Phe-Leu-Ly s-Glu-Gly-Gly-LeuStorage and Stability: Lyophilized powder may be stored at 4?C for short-term only. Reconstitute to nominal volume by adding sterile 40-50% glycerol and store at -20?C. R	1mg	C5818-05	UNITED STATES BIOLOGICAL	More In
CKS-17 (7-12)	Sequence: Leu-Asp-Leu-Leu-Phe-LeuSt orage and Stability: Lyophilized powder may be stored at 4?C for short-term only. Reconstitute to nominal volume by adding sterile 40-50% glycerol and store at -20?C. Reconstituted product is stable for 12 months	25mg	C5818-05A	UNITED STATES BIOLOGICAL	More In

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9/11/2008

CKS-17	This Peptide CKS-17 is considered as the major immunosuppressive site of retroviral TM protein. It can suppress T effector cell function in vitro. It has been shown that CKS-17 causes an imbalance of human types 1 and 2 cytokines and inhibition of	0.5 mg	06-271- 83162	GENWAY BIOTECH INC.	More In
CKS-17	This Peptide CKS-17 is considered as the major immunosuppressive site of retroviral TM protein. It can suppress T effector cell function in vitro. It has been shown that CKS-17 causes an imbalance of human types 1 and 2 cytokines and inhibition of	1 mg	06-271- 83162	GENWAY BIOTECH INC.	More In
CARCINOEMBRYONIC ANTIGEN (CL)	testing/assay service	n/a	n/a	RDL REFERENCE LABORATORY INC.	More In
CASPASE-3/7 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE-8 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE-9 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CL 218872	Benzodiazepine agonist displaying selectivity for a1 subunit-containing GABAA receptors (Ki values are 130, 1820, 1530, > 10000, 490 and > 10000 nM for a1, a2, a3, a4, a5 and a6-subunit containing re	10mg, 50mg	1709	TOCRIS BIOSCIENCE	More In
CL-387,785	Irreversibly inhibits EGF- receptor (EGFR) kinase activity in vivo (IC50 = 250-490 pM) as well as EGF-stimulated autophosphorylation of tyrosine residues in the EGFR in vivo (IC50 = 5 nM). Blocks EGF- mediated growth in A431 cells. Inhibits prolifer	n/a	233100	CALBIOCHEM/EMD BIOSCIENCES	More In
	A selective inhibitor of MMP-13 (IC50 = 10 $\mu$ M).	A Company of the Comp	The state of the s		

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CL-82198	Binds to the S1' pocket of MMP-13 with its morpholine ring adjacent to the catalytic zinc atom. Does not inhibit MMP-1, MMP-9, and TACE.	n/a	233105	CALBIOCHEM/EMD BIOSCIENCES	More In
Calphostin C, Cladosporium cladosporioides	A cell permeable, highly specific inhibitor of protein kinase C (IC50 = 50 nM) that interacts with the protein's regulatory domain by competing at the binding site of diacylglycerol and phorbol esters. Does not compete with Ca2+ or phospholi	n/a	208725	CALBIOCHEM/EMD BIOSCIENCES	More In
Cladribine	It is a substituted purine nucleoside with antileukemic activity. Melting Point: 220-235?C dec. Solubility: Methanol, Water	50mg	C5819-75	UNITED STATES BIOLOGICAL	More In
Clarithromycin	A semi-synthetic macrolide antibiotic. A derivative of erythromycin.Melting Point: 217-220?C dec.Solubility: Chloroform, Ethanol	50mg	C5829	UNITED STATES BIOLOGICAL	More In
Clavulanic Acid	A B-Lactamase inhibitor.	10mg	C5836	UNITED STATES BIOLOGICAL	More In
CLIC3	The RP-39009 CLIC3 protein is a full length bacterially expressed recombinant protein.RP-39009 is suitable for use as a control in ELISA and Western blot applications.The RP-39009 amino acid sequence corresponds to the NCBI accession number NP_004	10 ug	RP-39009	ABR - AFFINITY BIOREAGENTS INC.	More In
Clidinium Bromide	An anticholinergic. Used as an antispasmodic.Melting Point: 240-241?C	5g	C5840-75	UNITED STATES BIOLOGICAL	More In
	A metal ion chelator that crosses the blood brain barrier and acts as a neurotoxic antibiotic. Reported to dissolve				

<u>Clioquinol</u>	senile plaques and reduce amyloid's ability to clump together, apparently by trapping the Cu2+ and Zn2+ that stud these depos	n/a	233165	CALBIOCHEM/EMD BIOSCIENCES	More In
CLK3, active	n/a	10 ug	14-724	MILLIPORE	More In
CLK2, active	n/a	10 ug	14-774	MILLIPORE	More In
Clofarabine	ISecond generation purine nucleoside analog; antimetabolite that inhibits DNA synthesis and resists deamination by adenosine deaminase. Antineoplastic.Melting Point: 225-227?C	10mg	C5843-55	UNITED STATES BIOLOGICAL	More In
Clofarabine	Deoxycytidine kinase (dCK) substrate. Phosphorylated to form clofarabine triphosphate, which competes with dATP for DNA polymerase- a and - ε and potently inhibits ribonucleotide reductase (IC50 = 65 nM). Induces apoptosis by directl	10mg, 50mg	2600	TOCRIS BIOSCIENCE	More In
CLOFIBRATE	n/a	n/a	n/a	CAYMAN CHEMICAL CO.	More In
Clofibrate	PPAR agonist (EC50 values are 50, 500 and > 100 μM at PPAR α, PPAR γ and PPAR δ respectively). Antihyperlipoproteinemic.	1g	0824	TOCRIS BIOSCIENCE	More In
Clofibric acid	PPAR agonist. Antihyperlipoproteinemic.	1g	0825	TOCRIS BIOSCIENCE	More In
Clofibrate	An anti- hyperlipoproteinemic agent believed to act by inhibiting cholesterol biosynthesis. Activates PPARa and induces cytochrome P450 4A1 and 4A3. Imparts protection against acetaminophen toxicity and increases hepatic glutathione levels.	n/a	231405	CALBIOCHEM/EMD BIOSCIENCES	More In
Clofulbicyne	n/a	1 mg.	TXL9001-1	ACCURATE CHEMICAL & SCIENTIFIC CO.	More In

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Clofulbicyne	n/a	5x1 mg.	TXL9001-5	ACCURATE CHEMICAL & SCIENTIFIC CO.	More In
Clomifene citrate	International Chemical Reference Substances are established upon the advice of the WHO Expert Committee on Specifications for Pharmaceutical Preparations. They are supplied primarily for use in physical and chemical tests and assays described in t	100 mg	9930259	W.H.O. COLLABORATING CENTRE	More In
Clomiphene, Citrate	An unducer of ovulation. A gonad-stimulating principle.Melting Point: 116.5-118?CSolubility: Methanol	10g	C5843-65	UNITED STATES BIOLOGICAL	More In
Cloning	>1500 bp into 3 different expression vectors	n/a	PE05-0003	HYPEROMICS FARMA INC.	More In
Cloning	<1500 bp into 3 different expression vectors	n/a	PE05-0002	HYPEROMICS FARMA INC.	More In
Clopidogrel Carboxylic Acid	A metabolite of the drug Clopidogrel.Solubility: Methanol, Water	5mg	C5849-01	UNITED STATES BIOLOGICAL	More In
CLOSTRIPAIN Clostridium	n/a	n/a	n/a	PROMEGA CORPORATION	More In
<u>Clotrimazole</u>	An antifungal agent that acts as a potent and specific inhibitor of the Ca2+-activated K+ channel (Gardos channel; IC50 = 650 nM).  Prevents K+ loss and dehydration of sickled erythrocytes.	n/a	233230	CALBIOCHEM/EMD BIOSCIENCES	More In
Clozapine	An antipsychotic.Melting Point: 183-184? CSolubility: Acetone, Ether	250mg	C5866	UNITED STATES BIOLOGICAL	More In
Clozapine	Atypical antipsychotic drug, with a much lower tendency to cause extrapyramidal side effects than conventional neuroleptics. Displays a broad range of pharmacological actions; the antipsychotic effects are thought to be mediated principally by 5-H	50mg, 500mg	0444	TOCRIS BIOSCIENCE	More In

<u>CLTB</u>	The RP-39010 CLTB protein is a full length bacterially expressed recombinant protein.RP-39010 is suitable for use as a control in ELISA and Western blot applications.The RP-39010 amino acid sequence corresponds to the NCBI accession number NP_0018	10 ug	RP-39010	ABR - AFFINITY BIOREAGENTS INC.	More In
Aldosterone-3 CMO (BSA)	The major mineralcorticoid, which is secreted almost independently of ACTH from the pitutitary, is aldosterone. Aldosterone secretion is controlled mostly by the levels of potassium and sodium in serum and a blood pressure control system called th	<b>5</b> mg	A1350-04	UNITED STATES BIOLOGICAL	More In
Androstenedione-3 (CMO)	Androstenedione was discovered in 1935. It is naturally produced in men and women. It is a direct precursor to the hormone testosterone. The liver converts androstenedione to testosterone.Precursor:4-Androsten-3,17-dione-3Sto rage and Stability:Lyo	10mg	A2292-02	UNITED STATES BIOLOGICAL	More In
CMPD-1	Non-ATP-competitive, selective inhibitor of p38 a-mediated MK2a (mitogen-activated protein kinase-activated protein kinase 2a) phosphorylation (apparent Ki = 330 nM). Does not inhibit p38 a-mediated phosphorylation of the two other kno	10mg, 50mg	2186	TOCRIS BIOSCIENCE	More In
CMV	Glycine Extract	mL	0810003GE	ZEPTOMETRIX CORP.	More In
CMV	Cytomegalovirus (AD169) Infected Cell Extract. Used for IgG assays - Control is NHDF AV043	n/a	CV001	EASTCOAST BIO INC.	More In

сму	Gradient Purified. Used for IgM assays.	n/a	CV046	EASTCOAST BIO INC.	More In
сму	Cytomegalovirus Ag slides for FA. Made to Order	n/a	CG015	EASTCOAST BIO INC.	More In
CMV	Part Pure	n/a	J43010	BIOSPACIFIC INC.	More In

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Name	Description	Size	Catalog #	Supplier	
СКВВ	Recombinant Human Creatine Kinase BB Isoenzyme	10μg, 50μg, 1mg	CKI- 268	PROSPEC-TANY TECHNOGENE LTD.	More In
<u>Ckdk6</u>	The RP-39008 Ckdk6 protein is a partial length (aa 1-327) bacterially expressed recombinant protein.RP-39008 is suitable for use as a control in ELISA and Western blot applications.The RP-39008 protein is GST-tagged.	10 ug	RP-39008	ABR - AFFINITY BIOREAGENTS INC.	More In
СКММ	Human Creatine Kinase MM	200µg, 1mg, 10mg	CKI- 273	PROSPEC-TANY TECHNOGENE LTD.	More In
CKS-17	Sequence: Leu-Gln-Asn-Arg-Arg-Gly-L eu-Asp-Leu-Leu-Phe-Leu-Ly s-Glu-Gly-Gly-LeuStorage and Stability: Lyophilized powder may be stored at 4?C for short-term only. Reconstitute to nominal volume by adding sterile 40-50% glycerol and store at -20?C. R	1mg	C5818-05	UNITED STATES BIOLOGICAL	More In
CKS-17 (7-12)	Sequence: Leu-Asp-Leu-Leu-Phe-LeuSt orage and Stability: Lyophilized powder may be stored at 4?C for short-term only. Reconstitute to nominal volume by adding sterile 40-50% glycerol and store at -20?C. Reconstituted product is stable for 12 months	25mg	C5818-05A	UNITED STATES BIOLOGICAL	More In

CKS-17	This Peptide CKS-17 is considered as the major immunosuppressive site of retroviral TM protein. It can suppress T effector cell function in vitro. It has been shown that CKS-17 causes an imbalance of human types 1 and 2 cytokines and inhibition of	0.5 mg	06-271- 83162	GENWAY BIOTECH INC.	More In
CKS-17	This Peptide CKS-17 is considered as the major immunosuppressive site of retroviral TM protein. It can suppress T effector cell function in vitro. It has been shown that CKS-17 causes an imbalance of human types 1 and 2 cytokines and inhibition of	1 mg	06-271- 83162	GENWAY BIOTECH INC.	More In
CARCINOEMBRYONIC ANTIGEN (CL)	testing/assay service	n/a	n/a	RDL REFERENCE LABORATORY INC.	More In
CASPASE-3/7 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE-8 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE-9 (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CASPASE (CL)	n/a	n/a	n/a	PROMEGA CORPORATION	More In
CL 218872	Benzodiazepine agonist displaying selectivity for a1 subunit-containing GABAA receptors (Ki values are 130, 1820, 1530, > 10000, 490 and > 10000 nM for a1, a2, a3, a4, a5 and a6-subunit containing re	10mg, 50mg	1709	TOCRIS BIOSCIENCE	More In
CL-387,785	Irreversibly inhibits EGF-receptor (EGFR) kinase activity in vivo (IC50 = 250-490 pM) as well as EGF-stimulated autophosphorylation of tyrosine residues in the EGFR in vivo (IC50 = 5 nM). Blocks EGF-mediated growth in A431 cells. Inhibits prolifer	n/a	233100	CALBIOCHEM/EMD BIOSCIENCES	More In
The state of the s	A selective inhibitor of MMP-13 (IC50 = 10 $\mu$ M).		The state of the s		

<u>CL-82198</u>	Binds to the S1' pocket of MMP-13 with its morpholine ring adjacent to the catalytic zinc atom. Does not inhibit MMP-1, MMP-9, and TACE.	n/a	233105	CALBIOCHEM/EMD BIOSCIENCES	More In
Calphostin C, Cladosporium cladosporioides	A cell permeable, highly specific inhibitor of protein kinase C (IC50 = 50 nM) that interacts with the protein's regulatory domain by competing at the binding site of diacylglycerol and phorbol esters. Does not compete with Ca2+ or phospholi	n/a	208725	CALBIOCHEM/EMD BIOSCIENCES	More In
Cladribine	It is a substituted purine nucleoside with antileukemic activity. Melting Point: 220-235?C dec. Solubility: Methanol, Water	50mg	C5819-75	UNITED STATES BIOLOGICAL	More In
Clarithromycin	A semi-synthetic macrolide antibiotic. A derivative of erythromycin.Melting Point: 217-220?C dec.Solubility: Chloroform, Ethanol	50mg	C5829	UNITED STATES BIOLOGICAL	More In
Clavulanic Acid	A B-Lactamase inhibitor.	10mg	C5836	UNITED STATES BIOLOGICAL	More In
CLIC3	The RP-39009 CLIC3 protein is a full length bacterially expressed recombinant protein.RP-39009 is suitable for use as a control in ELISA and Western blot applications.The RP-39009 amino acid sequence corresponds to the NCBI accession number NP_004	10 ug	RP-39009	ABR - AFFINITY BIOREAGENTS INC.	More In
Clidinium Bromide	An anticholinergic. Used as an antispasmodic.Melting Point: 240-241?C	5g	C5840-75	UNITED STATES BIOLOGICAL	More In
	A metal ion chelator that crosses the blood brain barrier and acts as a neurotoxic antibiotic. Reported to dissolve				

<u>Clioquinol</u>	senile plaques and reduce amyloid's ability to clump together, apparently by trapping the Cu2+ and Zn2+ that stud these depos	n/a	233165	CALBIOCHEM/EMD BIOSCIENCES	More In
CLK3, active	n/a	10 ug	14-724	MILLIPORE	More In
CLK2, active	n/a	10 ug	14-774	MILLIPORE	More In
Clofarabine	ISecond generation purine nucleoside analog; antimetabolite that inhibits DNA synthesis and resists deamination by adenosine deaminase. Antineoplastic.Melting Point: 225-227?C	10mg	C5843-55	UNITED STATES BIOLOGICAL	More In
Clofarabine	Deoxycytidine kinase (dCK) substrate. Phosphorylated to form clofarabine triphosphate, which competes with dATP for DNA polymerase- α and - ε and potently inhibits ribonucleotide reductase (IC50 = 65 nM). Induces apoptosis by directl	10mg, 50mg	2600	TOCRIS BIOSCIENCE	More In
CLOFIBRATE	n/a	n/a	n/a	CAYMAN CHEMICAL CO.	More In
Clofibrate	PPAR agonist (EC50 values are 50, 500 and > 100 μM at PPAR α, PPAR γ and PPAR δ respectively). Antihyperlipoproteinemic.		0824	TOCRIS BIOSCIENCE	More In
Clofibric acid	PPAR agonist. Antihyperlipoproteinemic.	<b>1</b> g	0825	TOCRIS BIOSCIENCE	More In
Clofibrate	An anti- hyperlipoproteinemic agent believed to act by inhibiting cholesterol biosynthesis. Activates PPARa and induces cytochrome P450 4A1 and 4A3. Imparts protection against acetaminophen toxicity and increases hepatic glutathione levels.	n/a	231405	CALBIOCHEM/EMD BIOSCIENCES	More In
Clofulbicyne	n/a	1 mg.	TXL9001-1	ACCURATE CHEMICAL & SCIENTIFIC CO.	More In

Clofulbicyne	n/a	5x1 mg.	TXL9001-5	ACCURATE CHEMICAL & SCIENTIFIC CO.	More In
Clomifene citrate	International Chemical Reference Substances are established upon the advice of the WHO Expert Committee on Specifications for Pharmaceutical Preparations. They are supplied primarily for use in physical and chemical tests and assays described in t	100 mg	9930259	W.H.O. COLLABORATING CENTRE	More In
Clomiphene, Citrate	An unducer of ovulation. A gonad-stimulating principle.Melting Point: 116.5-118?CSolubility: Methanol	10g	C5843-65	UNITED STATES BIOLOGICAL	More In
Cloning	>1500 bp into 3 different expression vectors	n/a	PE05-0003	HYPEROMICS FARMA INC.	More In
Cloning	<1500 bp into 3 different expression vectors	n/a	PE05-0002	HYPEROMICS FARMA INC.	More In
Clopidogrel Carboxylic Acid	A metabolite of the drug Clopidogrel.Solubility: Methanol, Water	5mg	C5849-01	UNITED STATES BIOLOGICAL	More In
CLOSTRIPAIN Clostridium	n/a	n/a	n/a	PROMEGA CORPORATION	More In
Clotrimazole	An antifungal agent that acts as a potent and specific inhibitor of the Ca2+-activated K+ channel (Gardos channel; IC50 = 650 nM).  Prevents K+ loss and dehydration of sickled erythrocytes.	n/a	233230	CALBIOCHEM/EMD BIOSCIENCES	More In
Clozapine	An antipsychotic.Melting Point: 183-184? CSolubility: Acetone, Ether	250mg	C5866	UNITED STATES BIOLOGICAL	More In
Clozapine	Atypical antipsychotic drug, with a much lower tendency to cause extrapyramidal side effects than conventional neuroleptics. Displays a broad range of pharmacological actions; the antipsychotic effects are thought to be mediated principally by 5-H	50mg, 500mg	0444	TOCRIS BIOSCIENCE	More In

<u>CLTB</u>	The RP-39010 CLTB protein is a full length bacterially expressed recombinant protein.RP-39010 is suitable for use as a control in ELISA and Western blot applications.The RP-39010 amino acid sequence corresponds to the NCBI accession number NP_0018	10 ug	RP-39010	ABR - AFFINITY BIOREAGENTS INC.	More In
Aldosterone-3 CMO (BSA)	The major mineralcorticoid, which is secreted almost independently of ACTH from the pitutitary, is aldosterone. Aldosterone secretion is controlled mostly by the levels of potassium and sodium in serum and a blood pressure control system called th	5mg	A1350-04	UNITED STATES BIOLOGICAL	More In
Androstenedione-3 (CMO)	Androstenedione was discovered in 1935. It is naturally produced in men and women. It is a direct precursor to the hormone testosterone. The liver converts androstenedione to testosterone.Precursor:4-Androsten-3,17-dione-3Sto rage and Stability:Lyo	10mg	A2292-02	UNITED STATES BIOLOGICAL	More In
CMPD-1	Non-ATP-competitive, selective inhibitor of p38 a-mediated MK2a (mitogen-activated protein kinase-activated protein kinase 2a) phosphorylation (apparent Ki = 330 nM). Does not inhibit p38 a-mediated phosphorylation of the two other kno	10mg, 50mg	2186	TOCRIS BIOSCIENCE	More In
СМУ	Glycine Extract	mL	0810003GE	ZEPTOMETRIX CORP.	More In
CMV	Cytomegalovirus (AD169) Infected Cell Extract. Used for IgG assays - Control is NHDF AV043	n/a	CV001	EASTCOAST BIO INC.	More In
	Cytomegalovirus				

CMV	Gradient Purified. Used for IgM assays.	n/a	CV046	EASTCOAST BIO INC.	More In
CMV	Cytomegalovirus Ag slides for FA. Made to Order	n/a	CG015	EASTCOAST BIO INC.	More In
CMV	Part Pure	n/a	J43010	BIOSPACIFIC INC.	More In

Page 36 of 2,608

Records 1,751 - 1,800 of 130,353

[1] <u>« 29 | 30 | 31 | 32 | 33 | 34 | 35 | 36 | 37 | 38 | 39 | 40 |</u>

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Electronic Patent Application Fee Transmittal										
Application Number:	10	551205								
Filing Date:	14-Nov-2006									
Title of Invention:	Oral formulations of cladribine									
First Named Inventor/Applicant Name:	Nicholas S. Bodor									
Filer:	Mary Katherine Baumeister/Diana Francis									
Attorney Docket Number:	0056192-000024									
Filed as Large Entity										
U.S. National Stage under 35 USC 371 Filing	Fee	s								
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)					
Basic Filing:										
Pages:										
Claims:										
Miscellaneous-Filing:										
Petition:										
Patent-Appeals-and-Interference:										
Post-Allowance-and-Post-Issuance:										
Extension-of-Time:										
Extension - 3 months with \$0 paid		1253	1	1110	1110					

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Miscellaneous:				
	Tot	al in USD	(\$)	1110

Electronic Acknowledgement Receipt					
EFS ID:	4054303				
Application Number:	10551205				
International Application Number:					
Confirmation Number:	4092				
Title of Invention:	Oral formulations of cladribine				
First Named Inventor/Applicant Name:	Nicholas S. Bodor				
Customer Number:	21839				
Filer:	Mary Katherine Baumeister/Diana Francis				
Filer Authorized By:	Mary Katherine Baumeister				
Attorney Docket Number:	0056192-000024				
Receipt Date:	03-OCT-2008				
Filing Date:	14-NOV-2006				
Time Stamp:	10:37:12				
Application Type:	U.S. National Stage under 35 USC 371				

### **Payment information:**

Submitted with Payment	yes
Payment Type	Deposit Account
Payment was successfully received in RAM	\$1110
RAM confirmation Number	5296
Deposit Account	024800
Authorized User	

The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

Charge any Additional Fees required under 37 C.F.R. 1.492 (National application filing, search, and examination fees)

Charge any Additional Fees required under 37 C.F.R. Section ' (Patent application and reexamination processing fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.19 (Document supply fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.20 (Post Issuance fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.21 (Miscellaneous fees and charges)

### File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)				
1	Maio and discount of the control of	005610224TL 45	106535						
l	Miscellaneous Incoming Letter	005619224TL.pdf	802ec79ff5c0d1ec5dba3514562c642abd4a 9181	no	2				
Warnings:									
Information:									
,	Factor of Time	00FC10224F0T w.lf	46723		1				
2	Extension of Time	005619224EOT.pdf	dc7eb25d6272e05224cf13c539e286ac0a2 e3259	- no					
Warnings:									
Information:									
3	Amendment/Req. Reconsideration-After	005619224AMEND.pdf	4488479	no	48				
3	Non-Final Reject	005019224AMEND.pdf	b527124e797383c9425f8fece79d8cd57c4e e563						
Warnings:	·								
Information:									
4	Fee Worksheet (PTO-06)	fee-info.pdf	30075	no	2				
7	ree worksneet (r 10-00)	ree-imo.pui	a665357a4ed17d1ebac948c63e3acee571a e5f56	110					
Warnings:				'					
Information:									
		Total Files Size (in bytes)	: 46	71812					

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

) MAIL STOP AMENDMENT

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of

	•			
Nichol	as Bodor et al.	) Group Art Unit: 1623		
Applic	ation No.: 10/551,205	Examiner: JONATHAN S LAU		
Filing	Date: November 14, 2006	Confirmation No.: 4092		
Title:	ORAL FORMULATIONS OF CLADRIBINE			
	AMENDMENT/REPLY TRA	NSMITTAL LETTER		
P.O. E	nissioner for Patents Box 1450 ndria, VA 22313-1450			
Sir:				
Enclos	sed is a reply for the above-identified patent	application.		
$\boxtimes$	A Petition for Extension of Time is enclose	ed.		
	Terminal Disclaimer(s) and the [under 37 C.F.R. § 1.20(d) are enclosed.	\$ 65 S \$ 130 fee per Disclaimer due		
$\boxtimes$	Also enclosed is/are: copies of the attach Amendment.	ments listed on page 26 of Reply and		
	Small entity status is hereby claimed.			
	Applicant(s) requests continued examination the \$\infty\$ \$405 \$\infty\$ \$810 fee due under 37			
	Applicant(s) requests that any previously usentered. Continued examination is requestidentified above.			
	Applicant(s) previously submittedcontinued examination is requested.	on for which		
	Applicant(s) requests suspension of action which does not exceed in accordance with 37 C.F.R. § 1.103(c). is enclosed.	three months from the filing of this RCE,		
	A Request for Entry and Consideration of (1809/2809) is also enclosed.	Submission under 37 C.F.R. § 1.129(a)		

No additional claim fee is required.							
	An additional cl	aim fee is	required, and is	calculated	as shown below:		
			AMENDE	D CLAIMS			
		No. of Claims	Highest No. of Claims Previously Paid For	Extra Claims	Rate	Additional Fee	
Total (	Claims	78	78	0	x \$ 50 (1202)	\$	
Indepe	endent Claims	5	5	0	x \$ 210 (1201)		
☐ If A	mendment adds m	ultiple depe	endent claims, ad	d \$ 370 (120	03)	\$	
Total	Claim Amendmen	t Fee	.,,_,,			\$	
☐ Sm	nall Entity Status cla	aimed - sub	tract 50% of Tota	l Claim Ame	ndment Fee		
TOTA	L ADDITIONAL CL	AIM FEE	OUE FOR THIS A	MENDMEN	T	\$	
			•		2-4800 for the fee o		
	Charge		to credit card fo	or the fee d	ue. Form PTO-20	38 is attached.	
$\boxtimes$	The Director is hereby authorized to charge any appropriate fees under 37 C.F.R. §§ 1.16, 1.17 and 1.20(d) and 1.21 that may be required by this paper, and to credit any overpayment, to Deposit Account No. 02-4800.						
			Respectfully	y submitted	,		
			BUCHANAN I	INGERSOLL	& ROONEY PC		
Date	October 3, 2008	i.	Mary	Katherine E	in Jauneveter Baumeister 26254	<u>,                                     </u>	

P.O. Box 1404

703 836 6620

Alexandria, VA 22313-1404

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	) MAIL STOP AMENDMENT				
Nicholas Bodor et al.	) Group Art Unit: 1623				
Application No.: 10/551,205	) Examiner: JONATHAN S LAU				
Filing Date: November 14, 2006	) Confirmation No.: 4092				
Title: ORAL FORMULATIONS OF CLADRIBINE	) ) )				
PETITION FOR E	XTENSION OF TIME				
Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450					
Sir:					
The following extension of time is reques Action dated April 4, 2008 for	sted to: extend the period for response to the Office				
Three Months to October 6, 2008					
The shortened statutory period ha	as been reset by an Advisory Action dated				
An Extension fee in the amount o	f is enclosed.				
	nt No. 02-4800.				
Chargeto cred	it card. Form PTO-2038 is attached.				
The Director is hereby authorized to char 1.17 and 1.21 that may be required by this pape Account No. 02-4800.	rge any appropriate fees under 37 C.F.R. §§1.16, r, and to credit any overpayment, to Deposit				
Resp	ectfully submitted,				
Висн	ANAN INGERSOLL & ROONEY PC				
	Mary Katherine Baumeister  Registration No. 26254				
P.O. Box 1404 Alexandria, VA 22313-1404					

Buchanan Ingersoll & Rooney PC
Attorneys & Government Relations Professionals

703 836 6620

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

P	PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875								Docket Number 51,205		ing Date 14/2006	To be Mailed
	APPLICATION AS FILED - PART I (Column 1) (Column 2)								ENTITY	OR		HER THAN ALL ENTITY
	FOR	N	UMBER FII	.ED	NUN	MBER EXTRA		RATE (\$)	FEE (\$)		RATE (\$)	FEE (\$)
	BASIC FEE (37 CFR 1.16(a), (b),	or (c))	N/A			N/A		N/A			N/A	
	SEARCH FEE (37 CFR 1.16(k), (i), (	or (m))	N/A			N/A		N/A			N/A	
	EXAMINATION FE (37 CFR 1.16(o), (p),		N/A			N/A		N/A			N/A	
	TAL CLAIMS CFR 1.16(i))		mir	nus 20 = *				x \$ =		OR	x \$ =	
	EPENDENT CLAIM CFR 1.16(h))	S	m	inus 3 = *				x \$ =			x \$ =	
	APPLICATION SIZE (37 CFR 1.16(s))	shee is \$2 addi	ts of pap 50 (\$125 tional 50 s	er, the appl for small e sheets or fr	lication ntity) action	gs exceed 100 in size fee due for each i thereof. See CFR 1.16(s).						
	MULTIPLE DEPEN	IDENT CLAIM PF	ESENT (3	7 CFR 1.16(j)	)							
* If	the difference in colu	ımn 1 is less than	zero, ente	r "0" in colun	nn 2.			TOTAL			TOTAL	
	APP	(Column 1)	AMEND	(Column		(Column 3)		SMAL	L ENTITY	OR		ER THAN ALL ENTITY
AMENDMENT	10/03/2008	CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOU: PAID FOR		PRESENT EXTRA		RATE (\$)	additional Fee (\$)		RATE (\$)	ADDITIONAL FEE (\$)
Ĭ	Total (37 CFR 1.16(i))	* 78	Minus	** 78		= 0		x \$ =		OR	X \$52=	0
١X	Independent (37 CFR 1.16(h))	* 4	Minus	***6		= 0		x \$ =		OR	X \$220=	0
Ĭ	Application Si	ze Fee (37 CFR 1	.16(s))									
	FIRST PRESEN	ITATION OF MULTI	PLE DEPEN	DENT CLAIM (	(37 CFF	R 1.16(j))				OR		
								TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	0
		(Column 1)		(Column		(Column 3)						
L		CLAIMS REMAINING AFTER AMENDMENT		HIGHES NUMBE PREVIOU PAID FO	ER JSLY	PRESENT EXTRA		RATE (\$)	ADDITIONAL FEE (\$)		RATE (\$)	ADDITIONAL FEE (\$)
Ä	Total (37 CFR 1.16(i))	*	Minus	**		=		x \$ =		OR	x \$ =	
AMENDMENT	Independent (37 CFR 1.16(h))	*	Minus	***		=		x \$ =		OR	x \$ =	
Ä	Application Si	ze Fee (37 CFR 1	.16(s))									
FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))								OR				
							• !	TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	
** If	* If the entry in column 1 is less than the entry in column 2, write "0" in column 3.  ** If the "Highest Number Previously Paid For" IN THIS SPACE is less than 20, enter "20".  *** If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, enter "3".  The "Highest Number Previously Paid For" (Total or Independent) is the highest number found in the appropriate box in column 1.											

This collection of information is required by 37 CFR 1.16. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.		
10/551,205	11/14/2006	Nicholas S. Bodor	0056192-000024	4092		
	7590 01/07/200 INGERSOLL & ROOI	EXAM	EXAMINER			
POST OFFICE		LAU, JONATHAN S				
ALEAANDKIA	x, v A 22313-1404	ART UNIT PAPER NUMBI				
		1623				
			NOTIFICATION DATE	DELIVERY MODE		
			01/07/2009	ELECTRONIC		

### Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

ADIPFDD@bipc.com

		Application No.	Applicant(s)				
	Office Action Comments	10/551,205	BODOR ET AL.				
	Office Action Summary	Examiner	Art Unit				
		Jonathan S. Lau	1623				
Period fo	The MAILING DATE of this communication app or Reply	ears on the cover sheet with the c	orrespondence address				
WHIC - Exter after - If NO - Failui Any r	A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.  - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).						
Status							
1) 又	Responsive to communication(s) filed on <u>03 Oc</u>	ctober 2008.					
·	• • • • • • • • • • • • • • • • • • • •	action is non-final.					
7—	Since this application is in condition for allowar		secution as to the merits is				
٠/١	closed in accordance with the practice under <i>E</i>						
	·		0 0.0.2.2.0.				
Dispositi	on of Claims						
4)🛛	Claim(s) 1-35 and 56-98 is/are pending in the a	application.					
	4a) Of the above claim(s) <u>13-35 and 67-81</u> is/aı	re withdrawn from consideration.					
5)	Claim(s) is/are allowed.						
6)🖂	Claim(s) 1-12, 56-66, and 82-98 is/are rejected	l.					
7)	Claim(s) is/are objected to.						
8)	Claim(s) are subject to restriction and/or	election requirement.					
Applicati	on Papers						
		_					
-	The specification is objected to by the Examine		·				
· ·	The drawing(s) filed on is/are: a) ☐ acce						
	Applicant may not request that any objection to the						
	Replacement drawing sheet(s) including the correcti		, ,				
11)	The oath or declaration is objected to by the Ex	aminer. Note the attached Office	Action or form PTO-152.				
Priority u	ınder 35 U.S.C. § 119						
<ul> <li>12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).</li> <li>a) All b) Some * c) None of:</li> <li>1. Certified copies of the priority documents have been received.</li> <li>2. Certified copies of the priority documents have been received in Application No</li> <li>3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).</li> <li>* See the attached detailed Office action for a list of the certified copies not received.</li> </ul>							
Attachment(s)							

### **DETAILED ACTION**

Page 2

This Office Action is responsive to Applicant's Amendment and Remarks, filed 03 Oct 2008, in which claims 1, 2, 11, 13, 14, 23, 25, 26, 56-67 and 82 are amended to change the scope and breadth of the claim.

This application is the national stage entry of PCT/US04/09387, filed 26 Mar 2004; and claims benefit of provisional application 60/458,922, filed 28 Mar 2003; and claims benefit of provisional application 60/484,756, filed 02 July 2003; and claims benefit of provisional application 60/541,247, filed 04 Feb 2004.

The filing date of the instant claims 12, 66, 83, 85 and 89 are deemed to be the filing date of the instant application which is the filing date of PCT/US04/09387, 26 Mar 2004. The filing date of instant claims 1-11, 56-65, 82, 84 and 86-88 are deemed to be the filing date of provisional application 60/541,247, filed 04 Feb 2004.

Claims 1-35 and 56-98 are pending in the current application. Claims 13-35 and 67-81, drawn to non-elected inventions, are withdrawn.

### Election/Restrictions

Applicant's remarks regarding the requirement for restriction are moot as the requirement was made FINAL in the Office Action mailed 04 Apr 2008.

As recited in the Office Action mailed 06 Dec 2006, where applicant elects claims directed to the product, and the product claims are subsequently found allowable,

withdrawn process claims that depend from or otherwise require all the limitations of the allowable product claim will be considered for rejoinder. All claims directed to a nonelected process invention must require all the limitations of an allowable product claim for that process invention to be rejoined.

### Objections Withdrawn

Applicant's Amendment, filed 03 Oct 2008, with respect to objections to the specification has been fully considered and is persuasive, as the identified informalities are corrected.

This rejection has been withdrawn.

### Rejections Withdrawn

Applicant's Amendment, filed 03 Oct 2008, with respect to claims 2, 11 and 57 rejected under 35 U.S.C. 112, second paragraph, as being indefinite has been fully considered and is persuasive, as amended claims 2 and 11 recite the complex cladribin-cyclodextrin complex and amended claim 11 recites the definite term of a point on a specifically defined curve on a phase solubility diagram for saturated complexes of cladribine in varying concentrations of the cyclodextrin and Applicant's remarks regarding the definition of saturated is persusaive.

This rejection has been withdrawn.

Applicant's Amendment, filed 03 Oct 2008, with respect to claims 1-5, 11, 56-60, 82-90 and 94-98 rejected under 35 U.S.C. 102(b) as being anticipated by Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) has been fully considered and is persuasive, as Schultz et al. is not seen to disclose the composition comprising no significant amount of free crystalline cladribine therein (amended claim 1) or the complex consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex (amended claim 56) explicitly, implicitly or inherently.

This rejection has been withdrawn.

The following are new or modified grounds of rejection necessitated by Applicant's Amendment, filed 03 Oct 2008, in which claims 1, 2, 11, 13, 14, 23, 25, 26, 56-67 and 82 are amended to change the scope and breadth of the claim. Claims 2-12, 57-66 and 82-83 depend from claims 1 and 56 directly or indirectly, and incorporate all limitations therein, including changes to the scope and breadth of the claim.

### Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Amended claims 1-12, 56-66 and 82-83 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Baert et al. (WIPO Publication WO97/18839, published 29 May 1997, of record).

Schultz et al. discloses a solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin (column 2, lines 31-39). The disclosed product is substantially identical to the product-by-process. Schultz et al. discloses the use of  $\beta$ - and  $\gamma$ -cyclodextrins (column 2, lines 56-58) and derivatives wherein one or more cyclodextrin hydroxy groups are replaced with groups such as methyl, hydroxypropyl, carboxymethyl (column 3, lines 26-27) or sulfobutylcyclodextrins (column 4, lines 22-24). The phrase "one or more cyclodextrin hydroxy groups" combined with the absence of specific structural details of which hydroxyl group is substituted with a methyl group meets the limitation of "randomly methylated  $\beta$ -cyclodextrins". Schultz et al. discloses the solid oral dosage form in the form of a tablet (column 5, lines 37-38) including the

excipients sorbitol and magnesium stearate (column 6, lines 2-7), disclosing a product that is substantially identical to a product-by-process meeting the limitations of the instant claims invention. Schultz et al. discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, which would lead one of skill in the art to instantly envision a cladribine to cyclodextrin ratio ranging from 15 mg:100 mg to 15mg:500 mg, or 1:6.67 to 1:33.3 by weight (column 6, lines 23-31). The instant specification suggests that maximum amount of cladribine which can be complexed gives a weight ratio of 1:10 for the cladribine:cyclodextrin complex. Therefore a composition comprising the cladribine:cyclodextrin complex that contains a cladribine to cyclodextrin ratio of 1:6.67 describes a composition that comprises a "saturated" complex and meets the limitations of instant claims 2 and 57. Schultz et al. incorporates-by-reference the method of making said solid oral dosage form (Schultz et al.column 5, lines 50-52) disclosed in WIPO Publication WO97/18839, Baert et al., which provides evidence in the embodiment wherein the melt-extruded forms consist essentially of amorphous material (Baert et al. page 8, lines 14-15). Therefore Baert et al. provides evidence that it was recognized in the prior art that the product disclosed by Schultz et al. inherently includes amorphous cladribine-cyclodextrin complex in a solid oral dosage form. Schultz et al. implicitly discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, or a cladribine to cyclodextrin ratio ranging from 1:6.67 to 1:33.3 by weight (column 6, lines 23-31).

To address the scientific issue regarding the equilibrium presence of both the inclusion and non-inclusion complex, while the equation for the equilibrium of the

cladribine-cyclodextrin inclusion complex and the cladribine-cyclodextrin non-inclusion complex would be different for cladribine and cyclodextrin in a solvent versus cladribine and cyclodextrin in a molten state due to the lack of a solvent, the equilibrium and thus equilibrium products, the cladribine-cyclodextrin inclusion complex and the cladribine-cyclodextrin non-inclusion complex, would still be inherent in the product disclosed by Schultz et al.

Schultz et al. does not specifically disclose the composition comprising no significant amount of free crystalline cladribine therein (instant claims 1). Schultz et al. does not specifically disclose the composition corresponding to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin (instant claim 11). Schultz et al. does not specifically disclose the complex consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex (instant claim 56). Schultz et al. does not specifically disclose the composition comprising a cladribine to cyclodextrin ratio from about 1:10 to about 1:16 (instant claims 6, 7, 10, 61, 62 and 65), or a ratio of about 1:14 (instant claims 8 and 63) or about 1:11 (instant claims 9 and 64). Schultz et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b) (instant claims 12 and 66). Schultz et al. does not specifically disclose the product-by-process wherein

12.00 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl- $\beta$ -cyclodextrin are introduced in step (i) of the process (instant claim 91 and 93), to give a cladribine to cyclodextrin ratio of 1:14.38. Schultz et al. does not specifically disclose the product-by-process wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl- $\beta$ -cyclodextrin are introduced in step (i) of the process (instant claim 92), to give a cladribine to cyclodextrin ratio of 1:10.55.

Baert et al. discloses a solid mixture comprising one or more cyclodextrins and an insoluble active ingredient embedded into the cyclodextrin carrier (abstract), and teaches ratios of active ingredient to cyclodextrin of from about 1:100 to 100:1, from about 1:5 and 5:1 and from about 1:3 to 3:1 (page 11, lines 1-5). These ratios are interpreted as mole ratios because Baert et al. teaches the use of different active ingredients with different molecular weights. A mole ratio of active ingredient to cyclodextrin of about 1:3 for cladribine (MW 285.7 g/mol) and β-cyclodextrin (MW 1135 g/mol) gives a ratio by weight of approximately 1:11.9. The ratio of 1:11.9 meets the limitation of both a ratio of about 1:11 and a ratio of about 1:14 according to the nonlimiting definition of "about" as a variance of 20% provided in the instant specification page 9, lines 6-11. Such a saturated complex would consist of only (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex, and being a saturated complex corresponds to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve

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defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin.

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It would have been obvious to one of ordinary skill in the art at the time of the invention to practice the solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin disclosed by Schultz et al. in the ratios of cladribine and cyclodextrin taught by Baert et al. One of ordinary skill in the art would be motivated to combine the Schultz et al. and Baert et al. because Schultz et al. incorporates-byreference Baert et al. and because Baert et al. suggests that improving a similar product according to the teachings of Baert et al. has beneficial properties such as high bioavailability and dissolution rate (Baert et al. page 7, lines 25-27). One of ordinary skill in the art would have an expectation of success because the ratios taught by Baert et al. fall within the range of ratios that is implicitly disclosed by Schultz et al. Schultz et al. in view of Baert et al. does not teach the specific cladribine to cyclodextrin ratios of 1:14.38 or 1:10.55, however these ratios are encompassed by the prior art and Baert et al. suggests optimization of the ratio (Baert et al. page 11, lines 1-5). See also MPEP 2144.05 II.A, "Generally, differences in concentration or temperature will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical." One of ordinary skill in the art would be motivated to optimize the cladribine to cyclodextrin ratio to give the composition comprising no significant amount of free crystalline cladribine therein because Schultz et al. teaches undesireable recrystallization of cladribine in tissue may occur and damage the surround tissue and that complexes with cyclodextrin are known

to solubilize the compound (Schultz et al. column 2, lines 1-15). Schultz et al. in view of Baert et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b). However, it is well known in the art that the formation of an inclusion complex from a non-inclusion complex is an equilibrium process, and the position of this equilibrium is dependent on the concentrations of the cladribine and cyclodextrin. This molecular inclusion equilibrium is a process inherent in the formation of the inclusion complex in both aqueous solutions and hot melt liquid mixtures, and Baert et al. teaches variation of the ratio of cladribine to cyclodextrin and hence their relative concentration.

It is noted that In re Best (195 USPQ 430) and In re Fitzgerald (205 USPQ 594) discuss the support of rejections wherein the prior art discloses subject matter which there is reason to believe inherently includes functions that are newly cited or is identical to a product instantly claimed, namely the position of the equilibrium process governing formation of an inclusion complex and a non-inclusion complex. In such a situation the burden is shifted to the applicants to "prove that subject matter shown to be in the prior art does not possess characteristic relied on" (205 USPQ 594, second column, first full paragraph).

Claims 82-90 and 94-98 are drawn to a product-by-process. The disclosed product is substantially identical to the instantly claimed product-by-process, an amorphous solid pharmaceutical oral dosage form comprising cladribine and cyclodextrin. "[E]ven though product-by-process claims are limited by and defined by

the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted) (Claim was directed to a novolac color developer. The process of making the developer was allowed. The difference between the inventive process and the prior art was the addition of metal oxide and carboxylic acid as separate ingredients instead of adding the more expensive pre-reacted metal carboxylate. The product-by-process claim was rejected because the end product, in both the prior art and the allowed process, ends up containing metal carboxylate. The fact that the metal carboxylate is not directly added, but is instead produced in-situ does not change the end product.). See MPEP 2113.

### **Response to Applicant's Remarks:**

Applicant's Remarks, filed 03 Oct 2008 have been fully considered and not found to be persuasive.

Applicant notes that the method of Baert et al. is drawn to compounds that show no appreciable decomposition at the temperatures needed to melt and extrude the mixture. Applicant provides evidence that cladribine melts with decomposition at 220-235 °C, below the temperatures of the working examples of Baert et al. and below the melting point of HPβCD at 278 °C. This evidence regarding the applicability of the method disclosed by Baert et al. incorporated into Schultz et al. applied to the

disclosure of Schultz et al. has been carefully considered in view of the absence of disclosed working examples within Schultz et al. However it is well known in that art that a mixture of compounds exhibits some magnitude of freezing point depression due to the colligative properties of the mixture (entry for liquid, Britannica Online Encyclopedia, cited in PTO-892), and conversely there is a depression in the melting point. Suzuki et al. 1988 (Chem. Pharm. Bull., 1988, 36(2), p720-725, cited in PTO-892) provides evidence that further freezing point depression is observed for mixtures of butanol and sucrose with cyclodextrins (page 720, abstract and paragraph 2-3) due to the formation of the complex. Suzuki et al. 1993 (Chem. Pharm. Bull., 1993, 41(8), p1444-1447, cited in PTO-892) discloses this freezing point depression is observed in complexes such as barbiturate/CD (page 1444, left column, paragraphs 2-3). Therefore as the evidence provided concerns the melting point of the compounds as pure compounds rather than as the mixture, and in view of the presumption of validity afforded to the issued patent Schultz et al., this remark is not persuasive.

Applicant remarks that Baert et al. teaches the method of Baert et al. gives rise to different products than when said solids are first brought into contact with water or another solvent and then extruded. However, the invention of Baert et al. suggests that the different products given rise to are the solid solutions of the immediately prior statement (page 6, lines 14-15), or a mixture consisting of amorphous material and no crystalline material implied at page 8, lines 10-20.

Applicant's remarks that there is no reason for the interpreting the ratios taught by Baert et al. as mole ratios in view of the teachings of Schultz et al. and the instant

application regarding the disclosure of weight ratios. However, Suzuki et al. 1988 discloses cyclodextrin complexes in terms of the mole ratio (page 722, paragraph 3) and Suzuki et al. 1993 discloses the cyclodextrin complexes in terms of the stoichimetic ratios (page 1444, left column, paragraphs 1-3), or mole ratios. Therefore the prior art teaches both the interpretation as a mole ratio and a weight ratio. Absent a teaching within Baert et al. specifying the type of ratio intended and in view of the use of both mole ratios and a weight ratios in the prior art in the area of cyclodextrin complexes, one of skill in the art would find either interpretation equally reasonable.

Applicant notes that the product made obvious by Schultz et al. in view of Baert et al. is produced by a different process than the instant product-by-process. However, absent factual evidence of how this process necessarily makes a different product, it is found that the amorphous solid pharmaceutical oral dosage form comprising cladribine and cyclodextrin made obvious by Schultz et al. in view of Baert et al. is substantially identical to the same product may be the process of the instant invention.

### Conclusion

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within

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TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan S. Lau whose telephone number is 571-270-3531. The examiner can normally be reached on Monday - Thursday, 9 am - 4 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Shaojia Anna Jiang can be reached on 571-272-0627. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jonathan Lau Patent Examiner Art Unit 1623 /Shaojia Anna Jiang/ Supervisory Patent Examiner Art Unit 1623

## Notice of References Cited Application/Control No. 10/551,205 Examiner Jonathan S. Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Page 1 of 1

### U.S. PATENT DOCUMENTS

*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Name	Classification
	Α	US-			
	В	US-			
	С	US-			
	D	US-			
	Е	US-			
	F	US-			
	G	US-			
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### FOREIGN PATENT DOCUMENTS

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	N					
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### **NON-PATENT DOCUMENTS**

*		Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)										
	J	entry for liquid, Britannica Online Encyclopedia, http://www.search.eb.com/, accessed online on 31 Dec 2008.										
	٧	Suzuki et al. Chem. Pharm. Bull., 1988, 36(2), p720-725.										
	8	Suzuki et al. Chem. Pharm. Bull., 1993, 41(8), p1444-1447.										
	х											

\*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

# Index of Claims Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

<b>✓</b>	Rejected	-	Cancelled	N	N Non-Elected		Α	Appeal
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U.S. Patent and Trademark Office

Part of Paper No.: 20081231

# Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	Α	Appeal
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	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	10551205	BODOR ET AL.
	Examiner	Art Unit
	Jonathan S Lau	1623

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# Search Notes

Application/Control No.	Applicant(s)/Patent Under Reexamination
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Examiner	Art Unit
Jonathan S Lau	1623

SEARCHED							
Subclass	Date	Examiner					

SEARCH NOTES						
Search Notes	Date	Examiner				
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker)	3/26/2008	JSL				
EAST - see attached notes	3/26/2008	JSL				
Google Scholar - see attached notes	3/26/2008	JSL				

INTERFERENCE SEARCH						
Class	Subclass	Date	Examiner			

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.			
10/551,205	10/551,205 11/14/2006 Nicholas S. Bodor		11/14/2006 Nicholas S. Bodor		0056192-000024	4092	
	7590 06/18/200 INGERSOLL & ROOI	EXAMINER					
POST OFFICE	BOX 1404	LAU, JONATHAN S					
ALEAANDKIA	A, VA 22313-1404		ART UNIT	PAPER NUMBER			
			1623				
			NOTIFICATION DATE	DELIVERY MODE			
			06/18/2009	ELECTRONIC			

#### Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

ADIPFDD@bipc.com

	Application No.	Applicant(s)					
Interview Summary	10/551,205	BODOR ET AL.					
interview Summary	Examiner	Art Unit					
	Jonathan S. Lau	1623					
All participants (applicant, applicant's representative, PTO	personnel):						
(1) <u>Jonathan S. Lau</u> .	(3) <u>Mary Katherine Baumei</u>	ster.					
(2) <u>Shaojia Anna Jiang</u> .	(4) <u>Nicholas Bodor</u> .						
Date of Interview: 10 Jun 2009.							
Type: a)☐ Telephonic b)☐ Video Conference c)☒ Personal [copy given to: 1)☒ applicant 2	)∏ applicant's representative	·]					
Exhibit shown or demonstration conducted: d) ☐ Yes If Yes, brief description: <u>n/a</u> .	e)⊠ No.						
Claim(s) discussed: <u>1</u> .							
Identification of prior art discussed: Van Axel Castelli et al.	( <u>J. Pharm. Sci. 2008)</u> .						
Agreement with respect to the claims f) was reached. g	)⊠ was not reached. h)⊡ N	I/A.					
Substance of Interview including description of the general reached, or any other comments: <u>Applicant will consider filliclarify claim language</u> . <u>Applicant explained how the physic complex</u> . <u>Applicant discussed how the data of Van Axel Caller description</u> , if necessary, and a copy of the amend allowable if available must be attached. Also, where no content is a validable must be attached.	ng an RCE. Applicant will con al mixture of cladribine-CD is astelli et al. applies to the prior ments which the examiner agi	nsider filing an ar distinguished from r art of record. reed would rende	mendment to m the er the claims				
allowable, if available, must be attached. Also, where no coallowable is available, a summary thereof must be attached		ould render the t	ciaims				
THE FORMAL WRITTEN REPLY TO THE LAST OFFICE ACTION MUST INCLUDE THE SUBSTANCE OF THE INTERVIEW. (See MPEP Section 713.04). If a reply to the last Office action has already been filed, APPLICANT IS GIVEN A NON-EXTENDABLE PERIOD OF THE LONGER OF ONE MONTH OR THIRTY DAYS FROM THIS INTERVIEW DATE, OR THE MAILING DATE OF THIS INTERVIEW SUMMARY FORM, WHICHEVER IS LATER, TO FILE A STATEMENT OF THE SUBSTANCE OF THE INTERVIEW. See Summary of Record of Interview requirements on reverse side or on attached sheet.							
	/Shaojia Anna Jiang/						

U.S. Patent and Trademark Office PTOL-413 (Rev. 04-03)

Interview Summary

Supervisory Patent Examiner, Art Unit 1623

In re l	⊃ate		I THE UNITED STATES PATENT plication of	AND TRADEMARK OFFICE MAIL STOP RCE
Nicho	las S	S. Boo	dor et al.	Group Art Unit: 1623
Applic	atio	n No.	: 10/551,205	Examiner: JONATHAN S LAU
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Sir:				
identii § 1.17	fied		nt(s) requests continued examination and encloses the 🏻 \$405 [	on under 37 C.F.R. § 1.114 of the above- ⊠ \$810  fee due under 37 C.F.R.
1.		A.		eviously unentered after final amendments ination is requested based on the enclosed low.
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			Consider the amendment(s)/reply on	under 37 C.F.R. § 1.116 previously filed
			Consider the arguments in the Apon	opeal Brief or Reply Brief previously filed
			Other:	
2.	The	follow	ving documents are enclosed with	this submission:
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#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

MAIL STOP RCE
Group Art Unit: 1623
Examiner: JONATHAN S LAU
Confirmation No.: 4092

## REPLY AND AMENDMENT ACCOMPANYING FILING OF REQUEST FOR CONTINUED EXAMINATION (RCE) PURSUANT TO 37 C.F.R. § 1.114

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In response to the Office Action dated January 7, 2009, and in connection with Applicants' Request for Continued Examination, please first amend the above-identified patent application as follows:

#### **AMENDMENTS TO THE CLAIMS:**

This listing of claims will replace all prior versions, and listings, of claims in the application:

#### **LISTING OF CLAIMS:**

- 1. (Currently Amended) A pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16.
- 2. (Previously Presented) The pharmaceutical composition according to Claim 1, wherein the complex cladribine-cyclodextrin complex is saturated with cladribine.
  - 3-7. (Cancelled))
- 8. (Currently Amended) The composition according to Claim 7 Claim 1, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 9. (Currently Amended) The composition according to Claim 7 Claim 1, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
  - 10. (Cancelled)

- 11. (Currently Amended) The composition according to Claim 2, wherein the approximate molar ratio of cladribine to <u>said</u> amorphous cyclodextrin corresponds to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin.
- 12. (Previously Presented) The composition according to Claim 1, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 13. (Withdrawn and Currently Amended) A method for enhancing the oral bioavailability of cladribine comprising orally administering to a subject in need thereof a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16.
- 14. (Withdrawn) The method according to Claim 13, wherein the complex cladribine-cyclodextrin complex is saturated with cladribine.

15-19. (Cancelled)

20. (Withdrawn and Currently Amended) The method according to Claim 19 Claim 13, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.

- 21. (Withdrawn and Currently Amended) The method according to Claim 19 Claim 13, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
  - 22. (Cancelled)
- 23. (Withdrawn and Currently Amended) The method according to Claim 14, wherein the approximate molar ratio of cladribine to <u>said</u> amorphous cyclodextrin corresponds to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin.
- 24. (Withdrawn) The method according to Claim 13, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 25. (Withdrawn and Currently Amended) A method for the treatment of symptoms of a cladribine-responsive condition in a subject suffering from said symptoms comprising orally administering to said subject a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16.
- 26. (Withdrawn) The method according to Claim 25, wherein the complex cladribine-cyclodextrin complex is saturated with cladribine.

- 27. (Withdrawn) The method according to Claim 25, wherein the cladribine-responsive condition is selected from the group consisting of multiple sclerosis, rheumatoid arthritis and leukemia.
- 28. (Withdrawn) The method according to Claim 27, wherein the cladribine-responsive condition is multiple sclerosis.

29-31. (Cancelled)

- 32. (Withdrawn and Currently Amended) The method according to Claim 31 Claim 25, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 33. (Withdrawn and Currently Amended) The method according to Claim 31 Claim 25, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
  - 34. (Cancelled)
- 35. (Withdrawn) The method according to Claim 25, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).

#### 36.-55. (Cancelled)

56. (Currently Amended) A complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with an the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, said complex cladribine-

cyclodextrin complex having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16.

57. (Previously Presented) The complex cladribine-cyclodextrin complex according to Claim 56, saturated with cladribine.

58-62. (Cancelled)

- 63. (Currently Amended) The complex cladribine-cyclodextrin complex according to Claim 62 Claim 56, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:14.
- 64. (Currently Amended) The complex cladribine-cyclodextrin complex according to Claim 62 Claim 56, wherein the weight ratio of cladribine to hydroxypropyl-β-cyclodextrin is about 1:11.
  - 65. (Cancelled)
- 66. (Previously Presented) The complex cladribine-cyclodextrin complex according to Claim 56, wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b).
- 67. (Withdrawn and Currently Amended) A process for the preparation of a complex cladribine-cyclodextrin complex as claimed in Claim 56, which comprises the steps of:
- (i) combining cladribine and an the amorphous cyclodextrin in water at a temperature of from about 45 to about 80°C and maintaining said temperature for a period of from about 6 to about 24 hours;
  - (ii) cooling the resultant aqueous solution to room temperature; and
  - (iii) lyophilizing the cooled solution to afford an amorphous product.

- 68. (Withdrawn) A process according to Claim 67, further comprising a filtration step following step (ii).
- 69. (Withdrawn) A process according to Claim 67, wherein step (i) is performed at a temperature of from about 45 to about 60°C.
- 70. (Withdrawn) A process according to Claim 67, wherein step (i) is performed at a temperature of from about 45 to about 50°C.
- 71. (Withdrawn) A process according to Claim 69, wherein step (i) is performed with stirring.
- 72. (Withdrawn) A process according to Claim 71, wherein step (i) is performed for a period of from about 6 to about 9 hours.
- 73. (Withdrawn) A process according to Claim 67, wherein step (ii) is performed for a period of from about 6 to about 9 hours.
- 74. (Withdrawn) A process according to Claim 67, wherein step (iii) comprises an initial freezing stage in which the solution is cooled to from about -40 to about -80° C, and held at said temperature for a period of from about 2 to about 4 hours.
- 75. (Withdrawn) A process according to Claim 74, wherein, in the initial freezing stage of step (iii), the solution is cooled to about -45°C.
- 76. (Withdrawn) A process according to Claim 67, wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-β-cyclodextrin are introduced in step (i).

- 77. (Withdrawn) A process according to Claim 67, wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-β-cyclodextrin are introduced in step (i).
- 78. (Withdrawn) A process according to Claim 76, wherein 825 parts by volume of water are introduced in step (i).
- 79. (Withdrawn) A process according to Claim 67, wherein the lyophilization step (iii) comprises:
- (a) an initial freezing stage in which the complexation solution is brought to from about -40°C to about -80°C for approximately 2 to 4 hours;
- (b) a primary drying stage at about -25°C for approximately 80 to 90 hours; and
  - (c) a secondary drying stage at about 30°C for approximately 15 to 20 hours.
- 80. (Withdrawn) A process according to Claim 79, wherein stage (a) of the lyophilization is conducted at about -45°C for approximately 3 to 4 hours.
- 81. (Withdrawn) A process according to Claim 79, wherein stage (b) of the lyophilization is conducted under a pressure of about 100 mTorr.
- 82. (Currently Amended) A pharmaceutical composition <u>according to</u>

  <u>Claim 1</u> obtainable by a process comprising the steps of:
- (i) combining cladribine and an the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin in water at a temperature of from about 45 to about 80°C and maintaining said temperature for a period of from about 6 to about 24 hours;
  - (ii) cooling the resultant aqueous solution to room temperature;
  - (iii) lyophilizing the cooled solution to afford an amorphous product; and
  - (iv) formulating the amorphous product into a solid oral dosage form.

- 83. (Original) A pharmaceutical composition according to Claim 82, wherein the process further comprises a filtration step following step (i) or (ii).
- 84. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (i) of the process is performed at a temperature of from about 45 to about 60°C.
- 85. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (i) of the process is performed at a temperature of from about 45 to about 50°C.
- 86. (Previously Presented) A pharmaceutical composition according to Claim 84, wherein step (i) of the process is performed with stirring.
- 87. (Original) A pharmaceutical composition according to Claim 86, wherein step (i) of the process is performed for a period of from about 6 to about 9 hours.
- 88. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (ii) of the process is performed for a period of from about 6 to about 9 hours.
- 89. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein step (iii) comprises an initial freezing stage in which the solution is cooled to from about -40 to about -80°C, and held at said temperature for a period of from about 2 to about 4 hours.
- 90. (Original) A pharmaceutical composition according to Claim 89, wherein, in the initial freezing stage of step (iii), the solution is cooled to about -45°C.

- 91. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process.
- 92. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process.
- 93. (Previously Presented) A pharmaceutical composition according to Claim 91, wherein 825 parts by volume of water are introduced in step (i) of the process.
- 94. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein the lyophilization step (iii) of the process comprises:
- (a) an initial freezing stage in which the complexation solution is brought to from about -40°C to about -80°C for approximately 2 to 4 hours;
- (b) a primary drying stage at about -25°C for approximately 80 to 90 hours; and
  - (c) a secondary drying stage at about 30°C for approximately 15 to 20 hours.
- 95. (Original) A pharmaceutical composition according to Claim 94, wherein stage (a) of the lyophilization is conducted at about -45°C for approximately 3 to 4 hours.
- 96. (Previously Presented) A pharmaceutical composition according to Claim 94, wherein stage (b) of the lyophilization is conducted under a pressure of about 100 mTorr.
- 97. (Previously Presented) A pharmaceutical composition according to Claim 82, wherein the formulation step (iv) of the process comprises blending the complex with magnesium stearate and compressing into tablets.

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98. (Original) A pharmaceutical composition according to Claim 97, wherein magnesium stearate is pre-mixed with sorbitol powder before blending with the complex.

#### **REMARKS**

In connection with applicants' Request for Continued Examination (RCE), applicants respectfully request entry of the foregoing and reexamination and reconsideration of the subject application, as amended, pursuant to and consistent with 37 C.F.R. § 1.114, and in light of the remarks which follow.

#### STATUS OF CLAIMS

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 66-98 remain in this application. Claims 3-7, 10, 15-19, 22, 29-31, 34, 58-62 and 65 have been cancelled by the foregoing amendment, without prejudice or disclaimer, which Claims 36-55 were previously cancelled. Claims 1, 8, 9, 11, 13, 20, 21, 23, 25, 32, 33, 56, 63, 64, 67 and 82 have been amended hereinabove. Claims 1, 2, 8, 9, 11, 12, 56, 57, 63, 64, 66 and 82-98 are under examination. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81 have been withdrawn from consideration as drawn to non-elected subject matter. However, the withdrawn claims have been amended to be commensurate in scope with the examined claims as amended so that they may ultimately be rejoined.

#### STATEMENT OF SUBSTANCE OF INTERVIEW

Applicants acknowledge and thank Examiners Lau and Jiang for the courtesy of the personal interview granted to the inventor, Nicholas S. Bodor, and to applicants' undersigned representative, on June 10, 2009.

At the interview, applicants' representative indicated that an RCE would be filed. The claim language was discussed and it was agreed that support for the word "free" would be investigated and the claim clarified in this respect, if necessary. It was also proposed that in order to expedite prosecution, the independent claims be amended to recite only a particular amorphous cyclodextrin, namely hydropropyl-β-cyclodextrin, for which data was discussed, as well as a weight ratio range of cladribine to the cyclodextrin of from about 1:10 to about 1:16 (as set forth in, for

example, Claim 6); these suggestions were looked upon favorably by the Examiners following a detailed discussion, first of the references relied upon by the Examiner and then of the Van Axel Castelli et al., *J. Pharm.Sci.* 2008 submitted at the interview. The Van Axel Castelli et al. publication, a further copy of which is submitted herewith and listed on the accompanying Form PTO-1449, was shown to fully support the data in the specification for the claimed subject matter, to distinguish the instant complex from physical mixtures and to show the correctness of applicants' previous arguments with respect to the obviousness rejection based on Schultz et al. in view of Baert et al. Rather than repeating these discussions in detail, in this interview summary, applicants will repeat these discussions in detail in the remarks below. Finally, applicants provided several recent internet news reports concerning clinical trials of the product, further copies of which are appended.

#### **DISCUSSION OF CLAIM AMENDMENTS**

Claim 1 has been amended to specify "the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin," rather than "an amorphous cyclodextrin"; this feature was previously recited, for example, in original Claim 4. Claim 1 has been further amended to specify that the weight ratio of cladribine to said amorphous cyclodextrin in said composition is from about 1:10 to about 1:16; this feature was previously recited, for example, in original Claim 6. As for the use of the word 'free" in association with "cladribine," the expression "free cladribine" simply means cladribine not in the inclusion complex; see page 7, lines 24-25, where "free cladribine" is clearly defined. Therefore, it is appropriate to retain this word in the claims; amorphous free cladribine is associated with the amorphous cyclodextrin as the non-inclusion complex (b) while there is no significant amount of free crystalline cladribine in the composition. As for the subject matter cancelled from Claim 1 or from any of the other claims, applicants of course reserve the right to file a continuing application thereon.

Claims 3-7 have been cancelled as either outside the scope of the claims or redundant in light of the amendment of Claim 1.

Claims 8 and 9 have been amended so that they depend from Claim 1, which contains the features of original Claim 7, and Claim 10 has been cancelled as outside the scope of amended Claim 1.

A minor linguistic amendment has been made to Claim 11 to make it more consistent with amended Claim 1.

Claim 13 has been amended to be commensurate in scope with Claim 1.

Claims 15-19 have been cancelled as either outside the scope of Claim 13 as amended or redundant in light of the amendment of Claim 13; Claims 20 and 21 have been amended so that they depend from Claim 13; and Claim 22 has been cancelled as outside the scope of amended Claim 13. All of these amendments are consistent with the amendment of Claim 1.

A minor linguistic amendment has been made to Claim 23 to make it more consistent with amended Claim 13.

Claim 25 has been amended to be commensurate in scope with Claim 1.

Claims 29-31 and 34 have been cancelled as either outside the scope of Claim 25 as amended or redundant in light of the amendment of Claim 25.

Claims 32-33 have been amended to depend from Claim 25, which contains the features of the claim upon which they previously depended.

Claim 56 has been amended to be consistent with Claim 1. Claims 58-62 and 65 have accordingly been cancelled as outside the scope of amended Claim 56 or redundant in light of the amendment of Claim 56. The dependencies of Claims 63 and 64 have been amended accordingly.

A minor linguistic amendment has been made to Claim 67 to make it more consistent with amended Claim 56.

Finally, Claim 82 has been amended so that it depends from Claim 1 and the language has been amended to specify the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin consistent with Claim 1.

It is apparent from the foregoing that no new matter has been introduced by the amendments made.

#### FILING DATES ACCORDED TO CLAIMS

Applicants thank the Examiner for considering their remarks, particularly with respect to the international filing date. The PCT filing date of March 26, 2004 has been accorded to Claims 12, 66, 83, 85 and 89, while all other claims are deemed to be entitled to the February 4, 2004 filing date of U.S. Provisional Appln. No. 60/541,247.

#### **ELECTIONS/RESTRICTIONS**

As noted above, applicants have amended the withdrawn claims to be commensurate in scope with the product claims, so that the withdrawn claims can be ultimately rejoined.

#### **OBJECTIONS WITHDRAWN**

The Examiner's withdrawal of the previous objections to the specification is noted, with appreciation.

#### **REJECTIONS WITHDRAWN**

The withdrawal of the previous rejections under 35 U.S.C. §§ 112, second paragraph, and 102(b) are likewise noted, with appreciation.

#### CLAIM REJECTIONS - 35 U.S.C. § 103

Claims 1-12, 56-66 and 82-83 have been rejected under 35 U.S.C. § 103(a) as being unpatentable over Schultz et al. US Patent No. 6,194,395 in view of Baert et al. WO 97/18839, both previously made of record. It is submitted that this rejection cannot be maintained against any of the claims now in this application.

As explained at the interview, Schultz et al. describe two kinds of cyclodextrin formulations of cladribine, i.e., "soluble aqueous formulations of cladribine with

cyclodextrin solubilizers which are injectable in humans, as well as oral solid dosage forms containing a mixture of cladribine and cyclodextrins." (Emphasis added). See col., 1, lines 6-10, of the Schultz et al. patent. Applicants do not dispute the fact that Schultz et al. disclose hydroxypropyl-β-cyclodextrin (HPβCD), which is amorphous, but Schultz et al. disclose crystalline cyclodextrins as well. Indeed, Schultz et al. teach aqueous formulations containing a cladribine/HPBCD complex in solution for injectable use. On the other hand, Schultz et al. clearly teach that their solid oral dosage forms are mixtures of cladribine and cyclodextrin as set forth not only in col. 1, lines 8-10 ("oral solid dosage forms containing a mixture of cladribine and cyclodextrin") but also in col. 5, lines 50-64. There it is indicated that the solid oral dosage forms may be prepared as disclosed in Baert et al. WO 97/18837; this is in fact the only method disclosed by Schultz et al. for preparing the solid oral dosage forms. In col. 5, beginning at line 52, it is stated that "solid mixtures of the cyclodextrins with the active ingredient are prepared via melt-extrusion....the cladribine active ingredient and the cyclodextrins are mixed with other optional ingredients and then heated until melting occurs. The mixture is then extruded through an extruder having one or more nozzles." As set forth in col. 6 of the Schultz et al. patent, a typical oral dosage form has a formulation containing, as a milled extrudate, 1 mg to 15 mg of cladribine and 100 to 500 mg of cyclodextrin, and as excipients, 100 to 300 mg of microcrystalline cellulose, 10 to 200 mg of crospovidone, 1 to 5 mg of colloidal silicon dioxide and 2 to 10 mg of sterotex. Schultz et al. do not disclose or suggest to the ordinary skilled pharmaceutical scientist solid oral formulation of cladribine/cyclodextrin complexes as claimed herein. As noted at the interview, and as will be explained in detail below with reference to the Van Axel Castelli et al. article, the characteristics of a complex and a physical mixture are distinctly different; Schultz et al.'s solid mixtures of cladribine and cyclodextrin cannot be assumed to be the same as applicants' product, which is a complex cladribine-cyclodextrin complex. Even if the broad ratios of Schultz et al.'s mixtures encompass the ratios in applicants' complexes, having the same ratio does not give a physical mixture the same properties as a complex, a fact which is clearly shown by the Van Axel Castelli et al. document discussed in detail below.

As further noted at the interview, the Baert et al. WO document, incorporated by reference in Schultz et al., for its melt-extrusion method of making solid oral dosage forms, describes amorphous materials and solid solutions but does not teach or suggest that its solid dosage form products include amorphous cyclodextrin-drug complexes. On the contrary Baert et al. teach that:

- (a) Prior art problems are solved by Baert et al. by the use of a melt-extrusion process to form solid <u>mixtures</u> comprising one or more cyclodextrins and insoluble active ingredients (Emphasis added; see page 3, lines 7-9).
- (b) The compounds suitable for use in Baert et al.'s process "are compounds that show no appreciable decomposition at the temperatures needed to melt and extrude the mixture of said one or more active ingredients with the cyclodextrin or cyclodextrins" (page 4, lines 5-7).
- (c) Baert et al. teach that the characteristics of their products are different from those of a product obtained in water, since it is stated on page 6, lines 15-19:

It will be appreciated by a person skilled in the art that mixing two or more solids, i.e., one or more cyclodextrins and the active ingredient or ingredients, and subsequently melting these solids together will give rise to different products than when the said solids are first brought into contact with water or another solvent and then extruded.

- (d) Baert et al.'s process uses very high temperatures; note Table 1 on page 12, where several different drugs are mixed with HPβCD and melt-extruded, with temperatures of around 280°C being utilized and the products in every case as noted on page 13, lines 5-6, being solid solutions.
- (e) Baert et al. do not mention cladribine; indeed, applicants have previously shown that cladribine is known to decompose at the high temperatures used by Baert et al. and thus cladribine falls in the group of compounds Baert et al. teaches are unsuitable for use in their process [point (b) above].
- (f) Baert et al. in no way teach or suggest that their products contain complexes.

The unsuitability of Baert et al.'s temperatures and hence for the Baert et al. process as applied to cladribine with cyclodextrin, specifically with HPβCD, is furthermore proved by data in the Van Axel Castelli et al. article, as discussed in more detail below.

Also at the interview, to provide background with respect to drug/cyclodextrin complexation, applicants brought to the Examiners' attention, the Loftsson and Brewster cyclodextrin review article previously made of record in applicants' Third Information Disclosure Statement; see Loftsson et al., "Pharmaceutical Application of Cyclodextrins. 1. Drug Solubilization and Stabilization," *Journal of Pharmaceutical Sciences*, Vol. 85, No. 10, pp. 1017-1025, 1996, American Pharmaceutical Association and the American Chemical Society, US. The Loftsson and Brewster review article shows that it was known in the art that:

- (a) Cyclodextrin/drug complexation typically is carried out in aqueous media, not by mixing in the absence of water (page 1020, left column, first full paragraph).
- (b) This complexation involves many different forces (van der Waals, hydrogen bonding, etc.) and the use of water is essential to the formation of complexes (page 1018, right column, second full paragraph to page 1020, left column, noting in particular the mention of release of water molecules from the cyclodextrin cavity as a driving force for drug-cyclodextrin complex formation).
- (c) The complexes have different properties from mere physical mixtures of drugs and cyclodextrins, for example in terms of drug solubilization and drug stabilization (pages 1020-1024), a fact shown for HPβD and cladribine in the Van Axel Castelli et al. article discussed below.
- (d) The amorphous cyclodextrins such as HPβCD have almost countless isomeric and variably substituted forms and upon complexation result in amorphous products which are mixtures of countless complexes (page 1018, left column, line 12 from the bottom, to page 1018, right column line 2).

To address applicants' position that use of cladribine in Baert et al.'s melt extrusion product is inappropriate because of the fact that the decomposition temperature of cladribine is lower than the Baert et al. process's temperature, the Examiner has cited several additional documents (the Britannica Online excerpt and

the two Suzuki et al. articles). Applicants believe that these documents are irrelevant for at least the following reasons:

- (a) These references relate to freezing point depression and teach nothing about decomposition of cladribine at the high temperatures used in the Baert et al. process.
- (b) These references relate only to crystalline cyclodextrins ( $\alpha$  and  $\beta$ -cyclodextrin), which give crystalline complexes from water and not to amorphous cyclodextrins such as HP $\beta$ CD, which give amorphous complexes from water, these different kinds of complexes exhibiting different properties; moreover, neither Schultz et al. nor Baert et al. even remotely suggest that their solid oral dosage forms contain cyclodextrin/drug complexes, both characterizing their solid products as mixtures.
- (c) The experimental data provided in the Van Axel Castelli et al. article and discussed below prove that cladribine and an amorphous cyclodextrin such as HP $\beta$ CD do not form an eutectic mixture; rather, Van Axel Castelli et al. as discussed below, shows that cladribine, whether in a complex or in a mixture with HP $\beta$ CD, decomposes at temperatures far below those used for this cyclodextrin in the Baert et al. process.

Turning to the Van Axel Castelli et al. article, which was discussed in detail by Dr. Bodor at the interview, a further copy of which is provided herewith and which is listed on the accompanying form PTO-1449, the following remarks are offered:

The Van Axel Castelli et al. article shows that an inclusion complex of cladribine and 2-hydroxypropyl-β-cyclodextrin has properties which are different from those of the cyclodextrin, those of cladribine and those of physical mixtures of the cyclodextrin with cladribine. These differences were discussed at length by Dr. Bodor at the interview with respect to various analyses conducted by Van Axel Castelli et al. and can be summarized as follows:

(a) Thermo gravimetric analysis (TGA) was conducted over the temperature range from 25°C to 360°C for (a) cladribine, (b) HPβCD, (c) their inclusion complex and (d) their physical mixture and the results are shown in Fig. 2. Fig. 2a, the TGA curve for cladribine itself, shows decomposition of cladribine starting at about 200°C. Fig. 2b, the TGA curve for HPβCD, shows a mass loss of

6% at 30°C to about 140°C, due to dehydration, and decomposition at about 300°C. Fig. 2c, the TGA curve for the cladribine/HPβCD complex, shows water loss between 20°C and 100°C and a decomposition process starting at about 250°C. Fig. 2d, the TGA curve for the cladribine plus HPβCD physical mixture, shows a multi-stage decomposition pathway; the first decomposition stage between room temperature and 100°C is due to loss of water from the cyclodextrin, whereas the second decomposition stage, observed at temperatures above 200°C, is due to the decomposition of cladribine. Fig. 2d also shows that heating the mixture to high temperatures does <u>not</u> lead to complexation but rather to <u>decomposition of cladribine</u>. Comparing the TGA for the complex with that of the physical mixture shows a slower degradation for the complex than for the mixture. Nevertheless, cladribine, whether in a complex or in a mixture with HPβCD, decomposes at temperatures far below those used for HPβCD-containing products in the Baert et al. process.

(b) Differential scanning calorimetry (DSC) analysis was conducted and the results shown in Fig. 3 for (a) cladribine, (b) HPβCD, (c) cladribine/HPβCD complex, (d) cladribine plus HPBCD physical mixtures, and (e) cladribine plus HPβCD kneading product. The authors note that the DSC trace of cladribine shows two endothermic events (Fig. 3a), the first at 206.3°C being close to the cladribine decomposition onset temperature and corresponding to the melting transition, and the second at 211.9°C, which is during the decomposition process, and probably is due to a decomposition product of cladribine. The DSC profile for the cyclodextrin (Fig. 3b) confirms an endothermic event corresponding to water loss from about 40°C to about 100°C. The authors further note that the DSC curves of the inclusion complex (Fig. 3c), physical mixture (Fig. 3d) and kneading product (Fig. 3e) all show a broad thermal event from about 40 to 140°C, due to water loss in the cyclodextrin. In addition, the physical mixture and the kneading product are observed to feature two endothermic events around 200°C; these can be attributed to free cladribine in the mixture and kneading product. In contrast, Fig. 3c, the DSC trace for the complex, shows only one endothermic event, which occurs in the high temperature region, at 234.5°C; this is considerably higher than the degradation onset of pure cladribine around 200°C (Fig.2a), while the latter also characterizes the mixtures. As noted by the authors, the absence of thermal events typical of pure cladribine shows a loss of cladribine crystalline character for the complex. This also confirms DSC data for the products of instant Examples 1 and 2 reported on page 31 of the instant specification and correlates well with applicants' x-ray diffraction traces for the products of Examples 1 and 2 reported on page 31 of the specification, where no peaks for crystalline cladribine were found in the complexes;

(c) FT-IR and FT-Raman spectroscopy were also recorded by Van Axel Castelli et al. In Figures 5 and 6, the FT-IR and FT-Raman spectra of cladribine alone (a), cladribine/cyclodextrin physical mixture (b), cladribine/cyclodextrin complex (c) and cyclodextrin alone (d) were compared. The authors indicate that the IR spectrum of the physical mixture (Fig. 5b) can be interpreted as the sum of the spectra for pure crystalline cladribine (Fig. 5a) and pure HPβCD (Fig. 5d), also supported by Fig. 6. The authors continue:

In contrast, both IR and Raman spectra of the inclusion complex show clear differences with respect to those of the physical mixture. In particular, markers of the crystalline phase of cladribine (arrows on Figs. 4 and 5) cannot be found in the spectra of the inclusion complex.

The authors further find that their data suggest that, when part of an inclusion complex, cladribine is present in a different (non-crystalline) phase relative to that of pure crystalline cladribine and that direct interaction between cladribine molecules is prevented. Further, they note that the amorphous phase has to be attributed to the formation of molecular complexes where the interaction between cladribine and HPβCD shields cladribine molecules, thus preventing crystallization.

(d) Van Axel Castelli et al. also used nuclear magnetic resonance spectroscopy to better understand the molecular interactions in the cladribine/cyclodextrin complex. To obtain direct proof of complex formation, the authors conducted a 2D ROESY experiment, the results of which showed a typical inner portion of HPβCD, confirming that a host-guest inclusion complex had formed between cladribine and HPβCD. Further, the authors conducted <sup>13</sup>C CP-MAS NMR experiments and reported the spectra in Fig. 10 for cladribine (a), cladribine +

HPβCD physical mixture (b) and cladribine/ HPβCD complex (c). In the cladribine spectrum (Fig. 10a), the authors observed sharp peaks due to cladribine's high degree of crystallinity. The spectrum for the physical mixture (Fig. 10b) corresponds to the sum of the spectra of the two components, with no resonance peaks or line broadening, showing no intermolecular interaction in the mixture, the solid being composed of distinct ordered domains of each component. In the spectrum for the complex, no shift in the HPβCD signals are detected, whereas the cladribine resonances are broadened and only slightly detectable. The authors note that this indicates that no crystalline domains of cladribine are present.

- (e) Van Axel Castelli et al. also conducted DSC and TGA thermal profiles for tablets of the cladribine/ HP $\beta$ CD complex and found them comparable to those for the cladribine/ HP $\beta$ CD complex itself. Moreover, even stressed tablets showed no notable differences in the DSC thermal profile, which demonstrated that they were storage stable even under less than ideal conditions.
- (f) Van Axel Castelli et al. conclude that thermal analyses, vibrational analyses, and solid-state NMR all indicate that cladribine behaves differently when in the complex compared with the physical mixture or kneading product, while ROESY provides evidence for the existence of an internal complex between cladribine and HPβCD. They further conclude that tablets of the complex are not affected by their manufacturing from the complex itself and are storage stable. These tablets have been used in a successful clinical trial for oral treatment of patients with MS (CLARITY trial).

Information about the clinical trials of this product was presented at the interview and copies of several articles appearing on the internet are appended and listed on the accompanying Form PTO-1449. The product is expected to be the first marketed oral product for the treatment of multiple sclerosis.

In summary, applicants submit that the obviousness rejection based on Schultz et al. in view of Baert et al. is untenable and should be withdrawn. The data provided by Van Axel Castelli et al. conclusively show that cladribine, whether alone, in a physical mixture with HP $\beta$ CD, or even in a cladribine/HP $\beta$ CD complex, decomposes at temperatures far lower than those used by Baert et al. for melt extruding drugs with the same cyclodextrin. The data further show that heating a

Attorney's Docket No. 0056192-000024 Application No. 10/551,205

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physical mixture of cladribine with HPβCD to high temperatures does not result in a complex of cladribine and in fact the cladribine in the mixture decomposes long before the melting point for HPβCD is reached. Thus, the Baert et al. process is not suitable for making a melt extrudate of cladribine with hydroxypropyl-β-cyclodextrin and moreover such a product prepared according to Baert et al. would not contain a complex of cladribine with the cyclodextrin as claimed in this application. The Schultz et al. oral dosage form prepared by the Baert et al. process simply cannot contain a cladribine/cyclodextrin complex.

Claims 82-98 are drawn to a product-by-process. These claims have been amended above and now depend directly or indirectly from Claim 1 and thus contain all of the Claim 1 limitations. Applicants have shown that the Claim 1 composition is free of the art, therefore, the product-by-process claims are also patentable over the art.

#### CONCLUSION

In view of the foregoing, it is believed that all record rejections have been overcome. Further, favorable action in the form of a Notice of Allowance is believed to be in order and is earnestly solicited.

In the event that there are any remaining issues which could be resolved in a telephone discussion, the Examiner is urged to telephone the undersigned at the number given below so that such issues can be promptly resolved.

Respectfully submitted,

**BUCHANAN INGERSOLL & ROONEY PC** 

Date: <u>July 6, 2009</u>

By: Mary Keller

Mary Katherine Baumeister

Registration No. 26254

P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620

Attorney's Docket No. 0056192-000024

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	) MAIL STOP RCE						
Nicholas S. Bodor et al.	) Group Art Unit: 1623						
Application No.: 10/551,205	) ) Examiner: JONATHAN S LAU						
Filing Date: November 14, 2006	) Confirmation No.: 4092						
Title: ORAL FORMULATIONS OF CLADRIBINE	) ) )						
PETITION FOR	EXTENSION OF TIME						
Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450							
Sir:							
The following extension of time is requ- Action dated January 7, 2009 for	ested to: extend the period for response to the Office						
Three Months to July 7, 2009							
The shortened statutory period	has been reset by an Advisory Action dated						
An Extension fee in the amount	t of is enclosed.						
Chargeto De	eposit Account No. 02-4800.						
☐ Charge <u>\$ 1110</u> to credit card.							
The Director is hereby authorized to ch 1.17 and 1.21 that may be required by this pap Account No. 02-4800.	narge any appropriate fees under 37 C.F.R. §§1.16, per, and to credit any overpayment, to Deposit						
Res	spectfully submitted,						
Bud	CHANAN INGERSOLL & ROONEY PC						
Date: <u>July 6, 2009</u> By:	Mary Katherine Baumeister  Registration No. 26254						
P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620							

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	) MAIL STOP RCE
Nicholas S. Bodor et al.	) Group Art Unit: 1623
Application No.: 10/551,205	) )    Examiner: JONATHAN S LAU
Filed: November 14, 2006	) Confirmation No.: 4092
For: ORAL FORMULATIONS OF CLADRIBINE	) ) ) )

#### FIFTH INFORMATION DISCLOSURE STATEMENT

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In accordance with the duty of disclosure as set forth in 37 C.F.R. § 1.56, the accompanying information is being submitted in accordance with 37 C.F.R. §§ 1.97 and 1.98. Applicants request the Examiner's consideration of the documents listed on the accompanying Form PTO-1449.

Pursuant to 37 C.F.R. § 1.98, a copy of each of the documents cited is enclosed.

This Statement, Form PTO-1449 and the listed documents are submitted before the mailing of a first Office Action after the filing of a Request for Continued Examination under 37 C.F.R. § 1.114. Continued examination is requested and the fee required under 37 C.F.R. § 1.17(e) accompanies the present submission.

These documents are not prior art. They are submitted to support positions taken and statements made at the interview of June 10, 2009, and in the accompanying Reply and Amendment.

It is respectfully requested that an Examiner-initialed copy of the accompanying Form PTO-1449 be returned to the undersigned with the next official communication.

Fifth Information Disclosure Statement Application No. 10/551,205
Attorney Docket No. 0056192-000024
Page 2

The Director is hereby authorized to charge any appropriate fees under 37 C.F.R. §§ 1.16, 1.17 and 1.21 that may be required by this paper, and to credit any overpayment, to Deposit Account No. 02-4800.

Respectfully submitted,

**BUCHANAN INGERSOLL & ROONEY PC** 

Date: July 6, 2009

Mary Katherine Baumeister

Registration No. 26254

P.O. Box 1404 Alexandria, VA 22313-1404 703 836 6620 Substitute for form 1449/PTO & 1449B/PTO

## FIFTH INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

Application Number	10/551,205						
Filing Date	November 14, 2006						
First Named Inventor	Nicholas Bodor et al.						
Examiner Name	JONATHAN S LAU						
Attorney Docket No.	0056192-000024						

Complete if Known

Sheet 1 of 1

	U.S. PATENT DOCUMENTS								
Examiner Initials	Document Number- Kind Code	Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines Where Relevant Passages or Figures Appear					
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	FOREIGN PATENT DOCUMENTS												
	Foreign Patent Document					STATUS							
Examiner Initials	Country Code <sup>1</sup> , Number, Kind Code	Publication Date (MM-DD-YYYY)	Name of Patentee or Applicant of Cited Document	Translation	Partial Translation	Eng. Lang. Summary	Search Report	IPER	Abstract	Cited Po	l in Spec. / j. No(s).		
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Enter Office that issued the document, by the two-letter code. NON-PATENT LITERATURE DOCUMENTS Examiner Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, Initials serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published. VAN AXEL CASTELLI et al. "Characterisation of an Inclusion Complex Between Cladribine and 2-Hydroxypropyl-β-Cyclodextrin," J. Pharm. Sci., Vol. 97, No. 9, September 2008, pp. 3897-3906, Wiley InterScience and the American Pharmacists Association, US Drugs.com, "Oral Investigational Treatment Cladribine Tablets for Multiple Sclerosis Significantly Reduced Relapse Rate in Phase III Pivotal Trial," accessed online February 3, 2009, at http://www.drugs.com/clinical\_trials/oral-investigational-cladribine-multiple-sclerosis "Serono's Oral Cladribine for the Treatment of Multiple Sclerosis Awarded Fast Track Status by FDA", accessed online February 3, 2009 at http://prnewswire.com Merck Serono News Release, "Two-year Phase III Data Presented at AAN 61st Annual Meeting Show Positive Outcome of Cladribine Tablets in Patients with Multiple Sclerosis", April 29/30, 2009, available online.

Examiner	Date	
Signature	Considered	

<sup>\*</sup>EXAMINER: Initial if reference considered, whether or not citation is in conformance with M.P.E.P. § 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to Applicant.

Electronic Patent	App	olication Fee	Transm	ittal					
Application Number:	10:	551205							
Filing Date:	14	-Nov-2006							
Title of Invention:  Oral formulations of cladribine									
First Named Inventor/Applicant Name:	Nicholas S. Bodor								
Filer:	Mary Katherine Baumeister/Diana Francis								
Attorney Docket Number:	0056192-000024								
Filed as Large Entity									
U.S. National Stage under 35 USC 371 Filing	Fee	s							
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)				
Basic Filing:									
Pages:									
Claims:									
Miscellaneous-Filing:									
Petition:									
Patent-Appeals-and-Interference:									
Post-Allowance-and-Post-Issuance:									
Extension-of-Time:									
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Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Miscellaneous:				
Request for continued examination	1801	1	810	810
	Tot	al in USD	(\$)	1920

Electronic Ac	Electronic Acknowledgement Receipt				
EFS ID:	5646212				
Application Number:	10551205				
International Application Number:					
Confirmation Number:	4092				
Title of Invention:	Oral formulations of cladribine				
First Named Inventor/Applicant Name:	Nicholas S. Bodor				
Customer Number:	21839				
Filer:	Mary Katherine Baumeister/Diana Francis				
Filer Authorized By:	Mary Katherine Baumeister				
Attorney Docket Number:	0056192-000024				
Receipt Date:	06-JUL-2009				
Filing Date:	14-NOV-2006				
Time Stamp:	14:56:32				
Application Type:	U.S. National Stage under 35 USC 371				

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#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875						Δ	Application or Docket Number 10/551,205		Filing Date 11/14/2006		To be Mailed	
	APPLICATION AS FILED – PART I  (Column 1)  (Column 2)  SMALL ENTITY OR  SMALL ENTITY											
	FOR	N	JMBER FIL	.ED	NUM	IBER EXTRA		RATE (\$)	FEE (\$)		RATE (\$)	FEE (\$)
	BASIC FEE (37 CFR 1.16(a), (b),	or (c))	N/A			N/A		N/A			N/A	
	SEARCH FEE (37 CFR 1.16(k), (i), (ii)	or (m))	N/A			N/A		N/A			N/A	
	EXAMINATION FE (37 CFR 1.16(o), (p),		N/A			N/A		N/A			N/A	
	TAL CLAIMS CFR 1.16(i))		mir	us 20 = *				x \$ =		OR	x \$ =	
IND	EPENDENT CLAIM CFR 1.16(h))	S	m	inus 3 = *				X \$ =			x \$ =	
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	MULTIPLE DEPEN	IDENT CLAIM PR	ESENT (3	7 CFR 1.16(j))	ı							
* If t	he difference in colu	umn 1 is less than	zero, ente	r "0" in colum	ın 2.			TOTAL			TOTAL	
	APP	(Column 1)	AMEND	(Column		(Column 3)		SMAL	L ENTITY	OR		ER THAN ALL ENTITY
AMENDMENT	07/06/2009	CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUS PAID FOR		PRESENT EXTRA		RATE (\$)	ADDITIONAL FEE (\$)		RATE (\$)	ADDITIONAL FEE (\$)
ME	Total (37 CFR 1.16(i))	* 56	Minus	** 78		= 0		x \$ =		OR	X \$52=	0
볿	Independent (37 CFR 1.16(h))	* 5	Minus	***6		= 0		x \$ =		OR	X \$220=	0
√ME	Application S	ize Fee (37 CFR 1	.16(s))									
	FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))								OR			
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AM	FIRST PRESEN	NTATION OF MULTIF	PLE DEPEN	DENT CLAIM (	37 CFF	R 1.16(j))				OR		
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This collection of information is required by 37 CFR 1.16. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.		
10/551,205	11/14/2006	Nicholas S. Bodor	0056192-000024	4092		
	7590 09/17/200 INGERSOLL & ROO		EXAM	IINER		
POST OFFICE	BOX 1404		LAU, JONATHAN S			
ALEXANDRIA	A, VA 22313-1404		ART UNIT	PAPER NUMBER		
		1623				
			NOTIFICATION DATE	DELIVERY MODE		
			09/17/2009	ELECTRONIC		

#### Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

ADIPFDD@bipc.com

	Application No.	Applicant(s)				
Office Action Comments	10/551,205	BODOR ET AL.				
Office Action Summary	Examiner	Art Unit				
	Jonathan S. Lau	1623				
The MAILING DATE of this communicate Period for Reply	tion appears on the cover sheet v	vith the correspondence ad	dress			
A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.  - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).						
Status						
1)⊠ Responsive to communication(s) filed c	on <i>06 July 200</i> 9.					
· <u> </u>	☐ This action is non-final.					
3) Since this application is in condition for		tters, prosecution as to the	merits is			
closed in accordance with the practice	·	•	· ····o····o			
·						
Disposition of Claims						
4a) Of the above claim(s) <u>13,14,20,21,2</u> 5) ☐ Claim(s) is/are allowed. 6) ☒ Claim(s) <u>1,2,8,9,11,56,57,63,64 and 82</u> 7) ☐ Claim(s) is/are objected to.	6) Claim(s) 1,2,8,9,11,56,57,63,64 and 82-98 is/are rejected.  7) Claim(s) is/are objected to.					
Application Papers						
9)  The specification is objected to by the E 10)  The drawing(s) filed on is/are: a) Applicant may not request that any objection	☐ accepted or b)☐ objected to	-				
Replacement drawing sheet(s) including the 11) The oath or declaration is objected to by	e correction is required if the drawing	g(s) is objected to. See 37 CF	, ,			
Priority under 35 U.S.C. § 119						
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  a) All b) Some * c) None of:  1. Certified copies of the priority documents have been received.  2. Certified copies of the priority documents have been received in Application No  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.						
Attachment(s)  1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 1 page / 06 July 2009.	.948) Paper No	Summary (PTO-413) (s)/Mail Date Informal Patent Application 				

#### **DETAILED ACTION**

Page 2

#### Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 06 July 2009 has been entered.

This Office Action is responsive to Applicant's Amendment and Remarks, filed 06 July 2009, in which claims 1 and 56 are amended to change the scope and breadth of the claim; claims 8, 9, 11, 63, 64 and 82 are amended to change dependency; claims 3-7, 10, 15-19, 22, 29-31, 34, 58-62 and 65 are canceled; and withdrawn claims 13, 20, 21, 23, 25, 32, 33 and 67 are amended.

This application is the national stage entry of PCT/US04/09387, filed 26 Mar 2004; and claims benefit of provisional application 60/458,922, filed 28 Mar 2003; and claims benefit of provisional application 60/484,756, filed 02 July 2003; and claims benefit of provisional application 60/541,247, filed 04 Feb 2004.

The filing date of the instant claims 12, 83, 85 and 89 are deemed to be the filing date of the instant application which is the filing date of PCT/US04/09387, 26 Mar 2004.

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The filing date of instant claims 1, 2, 8, 9, 11, 56, 57, 63, 64, 82, 84 and 86-98 are deemed to be the filing date of provisional application 60/541,247, filed 04 Feb 2004.

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 67-98 are pending in the current application. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81, drawn to non-elected inventions, are withdrawn. Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 are examined on the merits herein.

#### Rejections Withdrawn

Applicant's Amendment and Remarks, filed 06 July 2009, with respect to claims 1-12, 56-66 and 82-83 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Baert et al. (WIPO Publication WO97/18839, published 29 May 1997, of record) has been fully considered and is persuasive, as claims 3-7, 10, 58-62 and 65 are canceled and Applicant's remarks supported by the evidence of Van Axel Castelli et al. provided by Applicant in IDS mailed 06 July 2009 is persuasive that the product taught by Schultz et al. in view of Baert et al. is structurally different from the instant invention as claimed and therefore the product taught by Schultz et al. in view of Baert et al. does not teach all limitations of the of instant invention as claimed.

This rejection has been withdrawn.

The following are new grounds of rejection.

#### Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Amended Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Wrenn Jr. (US Patent 6,174,873, issued 16 Jan 2001, cited in PTO-892) and in view of Loftsson et al. (US Patent 6,699,849, filed 16 Feb 1999, cited in PTO-892).

Schultz et al. discloses a solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin (column 2, lines 31-39). Schultz et al. teaches β-cyclodextrins are known to possess the ability to form inclusion complexes and to have concomitant solubilizing properties (column 2, lines 10-15). Schultz et al. discloses the use of β-cyclodextrins (column 2, lines 56-58) and derivatives wherein one or more cyclodextrin hydroxy groups are replaced with groups such as hydroxypropyl (column 3, lines 26-27). Schultz et al. discloses the solid oral dosage form in the form of a tablet (column 5, lines 37-38) including the excipients sorbitol and magnesium stearate (column 6, lines 2-7). Schultz et al. discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, which would lead one of skill in the art to instantly envision a cladribine to cyclodextrin ratio ranging from 15 mg:100 mg to 15mg:500 mg, or 1:6.67 to 1:33.3 by weight (column 6, lines 23-31). Schultz et al.

implicitly discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, or a cladribine to cyclodextrin ratio ranging from 1:6.67 to 1:33.3 by weight (column 6, lines 23-31).

Schultz et al. does not specifically disclose the composition comprising no significant amount of free crystalline cladribine therein (instant claims 1). Schultz et al. does not specifically disclose the composition corresponding to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin (instant claim 11). Schultz et al. does not specifically disclose the complex consisting of (a) an amorphous inclusion complex of cladribine with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex (instant claim 56). Schultz et al. does not specifically disclose the composition comprising a cladribine to cyclodextrin ratio from about 1:10 to about 1:16 (instant claims 6, 7, 10, 61, 62 and 65), or a ratio of about 1:14 (instant claims 8 and 63) or about 1:11 (instant claims 9 and 64). Schultz et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b) (instant claims 12 and 66). Schultz et al. does not specifically disclose the product-by-process wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-βcyclodextrin are introduced in step (i) of the process (instant claim 91 and 93), to give a cladribine to cyclodextrin ratio of 1:14.38. Schultz et al. does not specifically disclose

the product-by-process wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process (instant claim 92), to give a cladribine to cyclodextrin ratio of 1:10.55.

Wrenn Jr. teaches solid formulations for oral administration of adenosine analogs (abstract). Wrenn Jr. teaches the adenosine analogs include cladribine (column 6, lines 35-40). Wrenn Jr. teaches it is desirable to improve the solubility and absorption characteristics of poorly water soluble drugs by formulating the adenosine analog in an amorphous form together with solubilizing excipients (column 12, lines 25-30). Wrenn Jr. teaches the stabilization by absorption using a polymer that prevents recrystallization and the combination of the amorphous form and the solubilizing characteristics of the excipients enhances the solubility of the adenosine analog, and the amorphous drug complex may be formulated into a tablet system (column 12, lines 30-40).

Loftsson et al. teaches it is known in the art that substituted cyclodextrins show an increased aqueous solubility and that such chemical modification transforms crystalline cyclodextrins into amorphous mixtures increasing their aqueous solubility (column 1, lines 35-45). Loftsson et al. teaches it is known in the art that in aqueous solution cyclodextrins form complexes with many drugs (column 2, lines 1-10). Loftsson et al. teaches various methods of preparation of drug-cyclodextrin complexes are known in the art, including preparation of a solid complex by evaporation or freeze-drying following formation of the complex by equilibration (column 2, lines 20-40). Loftsson et al. teaches purine derivatives are compatible with said cyclodextrin complexes (column 9, lines 50-55).

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It would have been obvious to one of ordinary skill in the art at the time of the invention to combine Schultz et al. in view of Wrenn Jr. and in view of Loftsson et al. Schultz et al. teaches solid formulations for oral administration of cladribine and cyclodextrin. Wrenn Jr. is drawn the field of solid formulations for oral administration of adenosine analogs such as cladribine. Loftsson et al. teaches the level of skill in the art with regard to cyclodextrin complexes including complexes with purine derivatives. One of ordinary skill in the art would have been motivated to combine Schultz et al. in view of Wrenn Jr. and in view of Loftsson et al. because Schultz et al. teaches Schultz et al. teaches undesirable recrystallization of cladribine in tissue may occur and damage the surrounding tissue and that complexes with cyclodextrin are known to solubilize the compound, Wrenn Jr. teaches it is desirable to improve the solubility and absorption characteristics of poorly water soluble drugs by formulating the adenosine analog in an amorphous form together with solubilizing excipients, and Loftsson et al. teaches it is known in the art that complexes with substituted cyclodextrin give amorphous mixtures increasing their aqueous solubility. One of ordinary skill in the art would have a reasonable expectation of success in combining Schultz et al. in view of Wrenn Jr. and in view of Loftsson et al. to render obvious a product that meets all limitation of the instant invention because Wrenn Jr. teaches the stabilization of the adenosine analog by absorption using a polymer that prevents recrystallization and Loftsson et al. teaches modified cyclodextrins that form amorphous mixtures and preparation of a solid complex by evaporation or freeze-drying, which is expected to give a non-crystalline product. Schultz et al. in view of Wrenn Jr. and in view of Loftsson et al. does not teach the

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specific cladribine to cyclodextrin ratios of 1:14.38 or 1:10.55, however these ratios are encompassed by the prior art and Schultz et al. teaches it is within the level of skill in the art to optimize the ratio of cyclodextrin relative too cladribine (column 4, lines 35-45). See also MPEP 2144.05 II.A, "Generally, differences in concentration or temperature will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical." One of ordinary skill in the art would be motivated to optimize the cladribine to cyclodextrin ratio to give the composition comprising no significant amount of free crystalline cladribine therein because Schultz et al. teaches undesirable recrystallization of cladribine in tissue may occur and damage the surrounding tissue and that complexes with cyclodextrin are known to solubilize the compound (Schultz et al. column 2, lines 1-15). Schultz et al. in view of Wrenn Jr. and in view of Loftsson et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b). However, Loftsson et al. teaches formation of an inclusion complex from a non-inclusion complex in an aqueous solution is an equilibrium process, and the position of this equilibrium is dependent on the concentrations of the cladribine and cyclodextrin.

Claims 82-90 and 94-98 are drawn to a product-by-process. The disclosed product is substantially identical to the instantly claimed product-by-process, a pharmaceutical solid oral dosage form comprising an amorphous inclusion complex of cladribine and cyclodextrin and a non-inclusion complex of an amorphous cladribine

and an amorphous cyclodextrin. "[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted) (Claim was directed to a novolac color developer. The process of making the developer was allowed. The difference between the inventive process and the prior art was the addition of metal oxide and carboxylic acid as separate ingredients instead of adding the more expensive pre-reacted metal carboxylate. The product-by-process claim was rejected because the end product, in both the prior art and the allowed process, ends up containing metal carboxylate. The fact that the metal carboxylate is not directly added, but is instead produced in-situ does not change the end product.). See MPEP 2113.

#### Conclusion

No claim is found to be allowable.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan S. Lau whose telephone number is 571-270-3531. The examiner can normally be reached on Monday - Thursday, 9 am - 4 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Shaojia Anna Jiang can be reached on 571-272-0627. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jonathan Lau Patent Examiner Art Unit 1623 /Shaojia Anna Jiang/ Supervisory Patent Examiner Art Unit 1623

## Notice of References Cited Application/Control No. 10/551,205 Examiner Jonathan S. Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit Page 1 of 1

#### U.S. PATENT DOCUMENTS

*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Name	Classification
*	Α	US-6,174,873	01-2001	Wrenn, Jr., Simeon M.	514/45
*	В	US-6,699,849	03-2004	Loftsson et al.	514/58
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#### FOREIGN PATENT DOCUMENTS

*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Country	Name	Classification
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#### **NON-PATENT DOCUMENTS**

*		Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)					
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\*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

**Notice of References Cited** 

Part of Paper No. 20090910

# Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

✓	Rejected	-	Cancelled
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N	Non-Elected	Α	Appeal
ı	Interference	0	Objected

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Final	Original	11/26/2007	03/26/2008	01/02/2009	09/10/2009					
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Part of Paper No.: 20090910

# Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

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U.S. Patent and Trademark Office

	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	10551205	BODOR ET AL.
	Examiner	Art Unit
	Jonathan S Lau	1623

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	Α	Appeal
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☐ Claims	renumbered	in the same	order as pr	esented by	applicant		□ СРА	□ т.	D. 🗆	R.1.47
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Final	Original	11/26/2007	03/26/2008	01/02/2009	09/10/2009					
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U.S. Patent and Trademark Office Part of Paper No.: 20090910

### Search Notes

Application/Control No.	Applicant(s)/Patent Under Reexamination
10551205	BODOR ET AL.
Examiner	Art Unit
Jonathan S Lau	1623

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SEARCH NOTES						
Search Notes	Date	Examiner				
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker)	3/26/2008	JSL				
EAST - see attached notes	3/26/2008	JSL				
Google Scholar - see attached notes	3/26/2008	JSL				
EAST - see attached notes	9/10/2009	JSL				
Google Scholar - see attached notes	9/10/2009	JSL				
STN - CAPlus file - see attached notes	9/10/2009	JSL				

	INTERFERENCE SEARCH		
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Absorbable stent comprising coating for controlling degradation and maintaining pH ...

JZ Zhao - US Patent App. 11/297,944, 2005 - Google Patents

... everoli- mus, pimecrolimus, tacrolimus, paclitaxel, cladribine as well as ... zirconia, yittrium tetragonal polycrystalline zirconia, amorphous silicon, amorphous ...

All 4 versions

#### USE OF TRI-SUBSTITUTED GLYCEROL COMPOUNDS FOR THE TREATMENT OF ...

A ZANDER, F AYUKETANG, W Richter, L Weber ... - 2008 - freepatentsonline.com ... The term "amorphous", as used herein, refers to a solid in which ... non-Hodgkin lymphomas, AML), clofarabine (ALL), pentostatine (CLL), and cladribine (CLL), and ...

All 3 versions

Polymeric stent having modified molecular structures in both the hoops and selected ...

R Burgermeister, JH Contiliano, V Dave, Y ... - US Patent App. 11/440,807, 2006 - Google Patents ... internal structure modifications may be utilized to create devices having specific gross characteristics such as crys- talline and Amorphous morphology and ... All 6 versions

#### Polymeric stent having modified molecular structures in the flexible connections

R Burgermeister, JH Contiliano, V Dave, Y ... - US Patent App. 11/441,370, 2006 - Google Patents ... internal structure modifications may be utilized to create devices having specific gross characteristics such as crys- talline and amorphous morphology and ... All 5 versions

#### Drug-eluting articles with improved drug release profiles

R Falotico, JZ Zhao - US Patent App. 11/300.821, 2005 - Google Patents ... prepared from D-, L-PLA and PGA are typically amorphous in nature ... molecular weight drugs, such as mycophenolate acid (MPA), estradiol, cladribine, probucol, etc ... All 4 versions

#### Oral administration of adenosine analogs- \*pharmcast.com

SM Wrenn Jr - US Patent 6.174,873, 2001 - Google Patents

... deaminase and 2-chloro-2'-deoxyadenosine (also known as cladribine or 2CDA ... the solubility of other- wise insoluble adenosine analogs in an amorphous state in ...

Cited by 3 - Related articles - All 7 versions

#### Polymeric stent having modified molecular structures in the flexible connectors and in the ...

R Burgermeister, JH Contiliano, V Dave, Y ... - US Patent App. 11/440,774, 2006 - Google Patents ... internal structure modifications may be utilized to create devices having specific gross characteristics such as crys-talline and amorphous morphology and ...

All 6 versions

#### Polymeric stent having modified molecular structures in the flexible connectors and the radial ...

R Burgermeister, JH Contiliano, V Dave, Y ... - US Patent App. 11/440,764, 2006 - Google Patents ... internal structure modifications may be utilized to create devices having specific gross characteristics such as crys- talline and **amorphous** morphology and ... All 6 versions

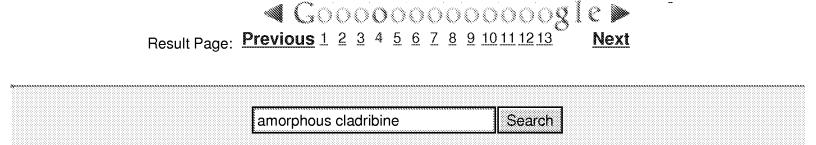
#### Hoogsteen vs. Watson-Crick Base Pairing: Incorporation of 2-Substituted Adenine-and 7- ...

N Ramzaeva, E Michalek, Z Kazimierczuk, F ... - Chemistry & Biodiversity, 2007 - interscience.wiley.com ... Synthetic 2-chloro-2'-deoxyadenosine (**cladribine**, Cl 2 A d , Leustatin ; 1a) [18–22] has gained considerable interest due to its toxicity toward T- and B ...

Related articles - BL Direct - All 2 versions

#### USE OF ROLL COMPACTED PYROGENICALLY PRODUCED SILICON DIOXIDE IN ...

R Hofmann, A Gray, M Drechsier, P It - 2007 - freepatentsonline.com ... content, specific gravity, refractive index, color or **amorphous** form ... cholera vaccine; chorionic gonadotropin; cidofovir; cisplatin; **cladribine**; clidinium bromide ... All 3 versions



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Results 1 - 10 of about 132. (0.09 sec)

#### Cyclodextrin cladribine formulations

TW Schultz, R Naeff... - US Patent 6,194,395, 2001 - Google Patents ... For example, suitable tablets may be prepared in the 65 conventional way having one or more of the following excipients: Cladribine Cyclodextrin 1 mg to 15 mg ... All 2 versions

#### Oral formulations of cladribine

NS Bodor, Y Dandiker - US Patent App. 10/551,205, 2004 - Google Patents ... CI. A61K 31/724 (2006.01) C07H 3/06 (2006.01) (52) US CI (57) ABSTRACT 514/58; 536/123.1 Provided are compositions of cladribine and cyclodextrin which are ... All 6 versions

#### **[CITATION] ORAL FORMULATIONS OF CLADRIBINE**

Y DANDIKER, NS BODOR

#### Pharmaceutical compositions comprising cyclodextrins

RPG Vandecruys - US Patent App. 09/445,297, 1999 - Google Patents ... to the total weight of drug compound, acid, cyclodextrin and organic ... antineoplastic agents and antimetabo- lites (adriamycine, cladribine, dactinomycin, dauno ... Cited by 1 - Related articles - All 7 versions

#### Cyclodextrin-based materials, compositions and uses related thereto

SH Pun, NC Bellocg, ME Davis... - US Patent App. 10/681,745, 2003 - Google Patents ... 10,2004 (54) CYCLODEXTRIN-BASED MATERIALS, COMPOSITIONS AND USES RELATED THERETO (75) Inventors: Suzie Hwang Pun, Torrance, CA (US); Nathalie C. Bellocq ... All 2 versions

#### Pregelatinized starch in a controlled release formulation

RPG Vandecruys, EMJ Jans - US Patent App. 10/674,701, 2003 - Google Patents ... capecitabine, gemcitab- ine, mercaptopurine, thioguanine, cladribine, methotrexate; [0042 ... this purpose, the recommended amount of cyclodextrin or derivatives ... All 9 versions

#### Pharmaceutical formulation of decitabine

C Tang, R Joshi-Hangai - US Patent App. 11/009,540, 2004 - Google Patents ... formulations. In particular, decitabine is formulated with a cyclodextrin compound to stabilize and/or enhance solubility of the drug. ... All 6 versions

#### Inhibitor of tumor metastasis or recurrence

K Sudo, T Houkan, P It - 1998 - freepatentsonline.com ... an emulsifier, a buffer, a preservative, cyclodextrin, sodium hydroxide etc. ... butosin, calcium folinate, calcium levofolinate, cladribine, emitefur, fludarabine ... All 2 versions

cladribine cyclodextrin - Google Scholar

#### Pharmaceutical formulation of cytidine analogs and derivatives

C Tang, R Joshi-Hangal - US Patent App. 11/010,189, 2004 - Google Patents ... In particular, the cytidine analog or derivative is formulated with a **cyclodextrin** compound to stabilize and/or enhance solubility of the drug. ... All 6 yersions

#### Methods for treating carbonic anhydrase mediated disorders

JL Masferrer, JM O'neal - US Patent App. 10/367,384, 2003 - Google Patents Page 1. US 20030220376A1 (19) United States (12) Patent Application Publication (io> Pub. NO.: US 2003/0220376 Al Masferrer et al. (43) Pub. Date: Nov. ... Ali 6 versions

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	cladribine		rin	Search	

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#### **EAST Search History**

#### **EAST Search History (Prior Art)**

Ref#	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
S1	1036	purine and cyclodextrin and complex and amorphous	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:47
S2	348	S1 and @ad<="20040204"	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:48
S3	1	"6,699,849".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:48
S4	1	purine and cyclodextrin and complex and amorphous	EPO; DERWENT	ADJ	ON	2009/09/10 13:48
S5	1	"20040186075".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:49
S8	1	S3 and purine and cyclodextrin and complex and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:53
S9	1	S5 and purine and cyclodextrin and complex and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:53
S10	1	"5,024,998".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:55
S11	1	S10 and purine and cyclodextrin and complex and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:56
S12	1	amorphous near3 cladribine	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 13:59
S13	1	"6174873".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:03
S15	1	S13 and (purine or cyclodextrin or complex or amorphous or solid)	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:03
S16	1	(amorphous near9 cladribine) or (amorphous same cladribine)	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:08
S17	4260	cladribine	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:23
S18	299	S17 and cyclodextrin and complex and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:23
S19	66	S18 and @ad<="20040204"	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:23
S20	1	cladribine and cyclodextrin and amorphous	EPO; DERWENT	ADJ	ON	2009/09/10 14:25

S21	420	(adenosine or purine or cladribine).ti,ab,bsum. and cyclodextrin and complex and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:28
S22	429	(adenosine or purine or cladribine).ti,ab,bsum. and cyclodextrin and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:28
S23	236	(adenosine or purine or cladribine).ti,ab,bsum. and cyclodextrin.ti,ab,bsum,clm. and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:29
S24	124	\$23 and @ad<="20040204"	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:29
S25	1	"20040127404".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:45
S26	1	"5,773,423".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:55
S27	1	\$26 and (purine or cyclodextrin or complex or amorphous or solid)	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:55
S28	235	(adenosine or purine or cladribine).ti,ab,bsum. and cyclodextrin.ti,ab,bsum. and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:57
S29	123	\$28 and @ad<="20040204"	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:57
S30	6	(adenosine or purine or cladribine).ti,ab. and cyclodextrin.ti,ab,bsum. and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:57
S31	1	(adenosine or purine or cladribine).ti,ab. and cyclodextrin.ti,ab. and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:57
S32	5	(adenosine or purine or cladribine).ti,ab,bsum. and cyclodextrin.ti,ab. and (complex or inclusion) and amorphous and solid	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 14:58
L1	1	"6,194,395".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2009/09/10 17:12

#### **EAST Search History (Interference)**

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Sheet

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### FIFTH INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

1

Complete if Known						
Application Number	10/551,205					
Filing Date	November 14, 2006					
First Named Inventor	Nicholas Bodor et al.					
Examiner Name	JONATHAN S LAU					
Attorney Docket No.	0056192-000024					

	U.S. PATENT DOCUMENTS							
Examiner Initials	Document Number- Kind Code	Publication Date MM-DD-YYYY	Name of Patentee or Applicant of Cited Document	Pages, Columns, Lines Where Relevant Passages or Figures Appear				
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		FOREIG	N PATENT DOCUME	ENTS							
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**NON-PATENT LITERATURE DOCUMENTS** Examiner Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, Initials serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published. VAN AXEL CASTELLI et al. "Characterisation of an Inclusion Complex Between Cladribine and 2-/J.L./ Hydroxypropyl-β-Cyclodextrin," J. Pharm. Sci., Vol. 97, No. 9, September 2008, pp. 3897-3906, Wiley InterScience and the American Pharmacists Association, US Drugs.com. "Oral Investigational Treatment Cladribine Tablets for Multiple Sclerosis Significantly Reduced /J.L./ Relapse Rate in Phase III Pivotal Trial," accessed online February 3, 2009, at http://www.drugs.com/clinical trials/oral-investigational-cladribine-multiple-sclerosis "Serono's Oral Cladribine for the Treatment of Multiple Sclerosis Awarded Fast Track Status by FDA", /J.L./ accessed online February 3, 2009 at http://prnewswire.com Merck Serono News Release. "Two-year Phase III Data Presented at AAN 61st Annual Meeting Show /J.L./Positive Outcome of Cladribine Tablets in Patients with Multiple Sclerosis". April 29/30, 2009, available online.

Examiner Signature	/Jonathan Lau/	Date Considered	09/10/2009
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<sup>\*</sup>EXAMINER: Initial if reference considered, whether or not citation is in conformance with M.P.E.P. § 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to Applicant.

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Did you mean: pyrene cyclodextrin complex amorphous

#### Cyclodextrins in the pharmaceutical field

O Bekers, EV Uijtendaal, JH Beijnen, A Bult ... - Drug Development and Industrial Pharmacy, 1991 - informaworld.com ... action of the cyclodextrin-trans-glycosidase enzyme on a medium containing starch. ... entirely or at least partially into the cavity, an inclusion complex may be ...

Cited by 106 - Related articles - All 3 versions

#### High-energy cyclodextrin complexes

T Loffsson, M Masson, E Stefansson - US Patent App. 10/750,940, 2004 - Google Patents ... administering the cyclodextrin-drug complex thus obtained ... trin, hydroxypropyl-ycyclodextrin and y-cyclodextrin. ... for use herein are purine derivatives, which ... Related articles - All 6 versions

#### Cyclodextrin complexes of benzodiazepines

T Loftsson, M Masson, E Stefansson - US Patent 6,699,849, 2004 - Google Patents ... can transform the crystal- line cyclodextrins into amorphous mixtures increasing ... are formed or 15 broken during the drug-cyclodextrin complex formation, the ... Related articles - All 2 versions

#### ... and characterisation of sulfated amphiphilic α-, β-and γ-cyclodextrins: application to the ...

A Dubes, G Degobert, H Fessi, H Parrot- ... - Carbohydrate research, 2003 - Elsevier ... 1) is an acyclic synthetic analogue of **purine** nucleosides with ... and coworkers[28.] who demonstrated that β-cyclodextrin forms a 1:1 complex with acyclovir ... Cited by 6 - Related articles - All 4 versions

#### Production of podophyllotoxin from Podophyllum hexandrum: a potential natural product for ...

A Giri, M Lakshmi Narasu - Cytotechnology, 2000 - Springer ... purine synthesis and inhibition of purine incorporation into ... Podophyllum resin is an amorphous powder, light ... β-cyclodextrin complex resulted in enhanced podo ... Cited by 27 - Related articles - BL Direct - All 5 versions

#### 2, 6, 9-Substituted purine derivatives and their use in the treatment of proliferative disorders

PM Fischer, M Jarman, T McDonald, B Nutley ... - US Patent App. 10/742,237, 2003 - Google Patents ... seeks to provide new 2,6,9- substituted **purine** derivatives, particularly ... Formation of a drug- cyclodextrin complex may modify the solubility, dissolution rate ... All 6 versions

#### Pharmaceutical formulations for parenteral use

NS Bodor - US Patent 5,024,998, 1991 - Google Patents ... from sweetener hydrolysis 'Tuttle also describes use of 2,6-di-0-methyl-/3-cyclodextrin and 2,3,6-tri-0-methyl-/3-cyclodextrin to form the inclusion complex. ... Cited by 13 - Related articles - All 6 versions

#### COMBINATION OF CRYSTALLINE FORM OF A RIBOFURANOSYLURONAMIDE ...

T SILK, J SMITH - 2003 - freepatentsonline.com

... is capable of dissolving both **amorphous** 6- [ (2, 2 ... 4-piperidyl] ureido}

ethyl)-9H-purine- 2-carboxamide ... Formation of a drug-cyclodextrin complex may modify ...

All 4 versions

#### CRYSTALLINE FORM OF A RIBOFURANOSYLURONAMIDE DERIVATIVE; A HUMAN ...

T SILK, J SMITH - 2003 - freepatentsonline.com

... is capable of dissolving both amorphous 6- [ (2, 2 ... 4-piperidyl] ureido} ethyl)-9Hpurine-2-carboxamide ... Formation of a drug-cyclodextrin complex may modify the ... All 4 versions

#### Redox systems for brain-targeted drug delivery

NS Bodor - US Patent 5.017,566, 1991 - Google Patents

... from sweetener hydrolysis 'Turtle also describes use of 2,6-di-O-methyl-£-cyclodextrin and 2,3,6-tri-O-methyl £-cyclodextrin to form the inclu- sion complex. ...

Cited by 1 - Related articles - All 5 versions

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L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN RN 4291-63-8 REGISTRY ED Entered STN: 16 Nov 1984 CN Adenosine, 2-chloro-2'-deoxy- (CA INDEX NAME) OTHER NAMES:						

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CN
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CN
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     2-Chloro-6-amino-9-(2-deoxy-\beta-D-erythro-pentofuranosyl) purine
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47 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
1491 REFERENCES IN FILE CAPLUS (1907 TO DATE)

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L3 ANSWER 1 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2009:1016541 CAPLUS

TITLE: Implantable biodegradable medical good impregnated with magnetic particles and optionally drugs for

treatment following tumor surgery

INVENTOR(S):
Jordan, Andreas

PATENT ASSIGNEE(S): Magforce Nanotechnologies AG, Germany

SOURCE: PCT Int. Appl., 45pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: German FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

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PATENT NO.
                      KIND DATE
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    WO 2009100716
                       A2 20090820
                                         WO 2009-DE196
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            ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH,
            PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TJ,
            TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW
        RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU,
            IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, SE, SI,
            SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
            TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,
            ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
    DE 102008008522
                    A1 20090813
                                         DE 2008-102008008522 20080211
PRIORITY APPLN. INFO.:
                                          DE 2008-102008008522A 20080211
                                          US 2008-71084P P 20080411
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AB The present invention relates to implantable and preferably biol.

metabolizable medical products comprising nanoparticles, and the use
thereof for thermotherapeutic treatment following surgical removal of
tumors and cancers. ABSThe medical good is implanted after tumor surgery;
magnetic field causes the beads to heat the wound area; in combination
with a drug the antitumor and antimicrobial activity can be effected.
Thus iron oxide magnetic particles were prepared from iron dichloride and
iron trichloride solution by precipitation in sodium hydroxide; the suspension
was

diluted to 5 weight% iron oxide. A wound pad composed of calcium alginate and sodium CM-cellulose was impregnated with the nanoparticle-containing suspension.

L3 ANSWER 2 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2009:971041 CAPLUS

TITLE: Implantable biodegradable medical good impregnated

with magnetic particles and optionally drugs for

treatment following tumor surgery

INVENTOR(S):
Jordan, Andreas

PATENT ASSIGNEE(S): Magforce Nanotechnologies AG, Germany

SOURCE: Ger. Offen., 19pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 102008008522 WO 2009100716	A1 A2	20090813 20090820	DE 2008-102008008522 WO 2009-DE196	20080211 20090211
W: AE, AG,	AL, AM, AO	, AT, AU, A	AZ, BA, BB, BG, BH, BR,	BW, BY, BZ,
CA, CH,	CN, CO, CR	, CU, CZ, E	DE, DK, DM, DO, DZ, EC,	EE, EG, ES,
FI, GB,	GD, GE, GH	, GM, GT, H	HN, HR, HU, ID, IL, IN,	IS, JP, KE,
KG, KM,	KN, KP, KR	, KZ, LA, I	LC, LK, LR, LS, LT, LU,	LY, MA, MD,
ME, MG,	MK, MN, MW	, MX, MY, M	MZ, NA, NG, NI, NO, NZ,	OM, PG, PH,
PL, PT,	RO, RS, RU	, SC, SD, S	SE, SG, SK, SL, SM, ST,	SV, SY, TJ,
TM, TN,	TR, TT, TZ	, UA, UG, U	US, UZ, VC, VN, ZA, ZM,	ZW
RW: AT, BE,	BG, CH, CY	, CZ, DE, D	DK, EE, ES, FI, FR, GB,	GR, HR, HU,
IE, IS,	IT, LT, LU	, LV, MC, M	MK, MT, NL, NO, PL, PT,	RO, SE, SI,
SK, TR,	BF, BJ, CF	, CG, CI, C	CM, GA, GN, GQ, GW, ML,	MR, NE, SN,

TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.:

DE 2008-102008008522A 20080211 US 2008-71084P P 20080411

AB The invention concerns biodegradable medical goods that contain magnetic micro— or nanoparticles and optionally drugs. The medical good is implanted after tumor surgery; magnetic field causes the beads to heat the wound area; in combination with a drug the antitumor and antimicrobial activity can be effected. Thus iron oxide magnetic particles were prepared from iron dichloride and iron trichloride solution by precipitation in sodium hydroxide; the suspension was diluted to 5 weight% iron oxide. A wound pad composed of calcium alginate and sodium CM-cellulose was impregnated with the nanoparticle-containing suspension.

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 3 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:674934 CAPLUS

DOCUMENT NUMBER: 149:17767

TITLE: Compositions of Chk1 kinase inhibitor for cancer

treatment

INVENTOR(S): Colvin, Anita A.; Koppenol, Sandy; Wisdom, Wendy A.

PATENT ASSIGNEE(S): Icos Corporation, USA SOURCE: PCT Int. Appl., 107 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.					KIND DATE			APPLICATION NO.						D.	ATE		
	_	2008				A2		2008								2	0071	002
	WO	2008	0670.	27		А3		2009	0416									
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			GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	AZ,
								TJ,						·	·	·	·	·
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	CA	2673	483			A1		2008	0605	CA 2007-2673483						2	0071	002
	ΕP	2063	879			A2		2009	0603		EP 2	007-	8711	06		2	0071	002
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			IS,	IT,	LI,	LT,	LU,	LV,	MC,	MT,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,
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	IN 2009MN00498				·	A		2009	0522		IN 2	009-1	MN49	8		2	0090.	309
	KR 2009065537							2009	0622							2	0090	417
PRIO	RIORITY APPLN. INFO.:										US 2						0061	020
											WO 2	007-1	JS80:	150	Ţ	w 2	0071	002

OTHER SOURCE(S): MARPAT 149:17767

AB Compns. containing at least one Chk1 kinase inhibitor and at lease one cyclodextrin are disclosed. Also disclosed are methods of treating a

proliferative disorders, especially cancer or potentiating a cancer treatment with a composition comprising at least one Chk1 inhibitor and at least one cyclodextrin. Thus, an injection solution was formulated containing a disubstituted urea Chk1 inhibitor 50 mg, Captisol 16.66 mg, HCl and NaOH to pH 4.5, and water to 1 mL. Captisol improved chemical stability of the Chk1 inhibitor compared to a solution containing a Chk1 inhibitor mesylate salt and dextrose. Degradation of Chk1 inhibitor was found to be accelerated by moisture and heat. After storage at  $40^{\circ}/75^{\circ}$  RH, the Captisol-containing formulation contained 3.06 and 4.96% of related impurities after 1 and 2 mo, resp., while the non-Captisol containing formulation contained 4.41 and 7.10% of impurities at the resp. time points.

L3 ANSWER 4 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2007:993749 CAPLUS

DOCUMENT NUMBER: 147:330433

TITLE: Composition and method for topical treatment of

tar-responsive dermatological disorders

INVENTOR(S): Yu, Ruey J.; Van Scott, Eugene J.; Lee, Yaling

PATENT ASSIGNEE(S): Tristrata, Inc., USA

SOURCE: U.S. Pat. Appl. Publ., 15pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.					KIN	D	DATE 			APPL	ICAT	ION I	NO.		D.	ATE	
	US	2007	0207	 222		A1	_	2007	0906		US 2	007-	6802	27		2	0070	228
	ΑU	2007	2235	60		A1		2007	0913		AU 2	007-	2235	60		2	0070.	228
	AU	2007	2235	60		A2		2008	1016									
	CA	2644	311			A1		2007	0913		CA 2	007-	2644.	311		2	0070	228
	WO	2007	1036	87		A2		2007	0913		WO 2	007-1	US62	975		2	0070	228
	WO	2007	1036	87		А3		2008	1211									
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			MN,	MW,	MX,	MY,	MΖ,	NA,	NG,	NI,	NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,
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			TZ,	UA,	UG,	US,	UΖ,	VC,	VN,	ZA,	ZM,	ZW						
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			CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	NE,	SN,	TD,	TG,	BW,	GH,
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			KG,	KΖ,	MD,	RU,	ТJ,	TM,	AP,	EA,	EP,	OA						
	EP	1998	788			A2		2008	1210		EP 2	007-	7576.	36		2	0070	228
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			BA,	HR,	MK,	RS												
	JP	2009	5283	82		Τ		2009	0806		JP 2	008-	5574	87		2	0070	228
	CN 101460060					Α		20090617 CN 2007-80015758						20081031				
PRIO	RIORITY APPLN. INFO.:										US 2	006-	7781.	28P		P 2	0060	301
											WO 2	007-	US62	975	1	W 2	0070	228

AB The present invention relates to a composition including a wax and a therapeutically effective amount of tar for topical treatment of a tar-responsive dermatol. disorder, the composition being in liquid or light gel form when at a temperature selected from room temperature and a temperature of skin of a

mammal upon application of the composition to the skin of the mammal. The invention also relates to a method of treating a tar-responsive dermatol. disorder by topically applying the composition to skin of a mammal, preferably a human, that is affected by the disorder. Thus, a fast-drying liquid tar composition was formulated containing coal tar solution 15 g, ethanol 42 g, propylene

glycol 5 g, cyclomethicone (DC 345) 15 g, tri-Et citrate 5 g, Brij 93 10 g, liquid wax DIADD (dioctyldodecyl dodecanedioate) 5 g, and an optional fragrance 3 g. Topical application of the composition for 4 mo to a human subject having plaque psoriasis resulted in 90% improvement of clin. signs of disorder.

L3 ANSWER 5 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2006:1202261 CAPLUS

DOCUMENT NUMBER: 145:495768

TITLE: Soft tissue implants, anti-scarring agents, and

therapeutic compositions

INVENTOR(S): Hunter, William L.; Toleikis, Philip M.; Gravett,

David M.; Maiti, Arpita; Liggins, Richard T.;

Takacs-Cox, Aniko; Avelar, Rui; Signore, Pierre E.; Loss, Troy A. E.; Hutchinson, Anne; McDonald-Jones,

Gaye; Lakhani, Fara

PATENT ASSIGNEE(S): Angiotech International A.-G., Switz.

SOURCE: PCT Int. Appl., 2979 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 3

PATENT INFORMATION:

PAT	PATENT NO.					KIND DATE			APPLICATION NO.						D.	ATE	
WO WO	2006 2006				A2 A3		2006 2007	_	1	wo 2	006-	us11	690		2	0060	331
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		IS,	ΙΤ,	LT,	LU,	LV,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	ВJ,
		CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	ΤG,	BW,	GH,
		GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	ΑZ,	BY,
		KG,	KΖ,	MD,	RU,	ТJ,	TM										
WO	2006	1215.	22		A2		2006	1116	1	WO 2	006-	US11	726		2	0060	331
WO	2006	1215.	22		А3		2008	0502									
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GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG

US 2005-679293P P 20050510 US 2005-679962P P 20050510 US 2005-679291P P 20050510 PRIORITY APPLN. INFO.:

Soft tissue implants (e.g., breast, pectoral, chin, facial, lip, and nasal implants) are used in combination with an anti-scarring agent in order to inhibit scarring that may otherwise occur when the implant is placed within an animal.

ANSWER 6 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:493530 CAPLUS

143:32415 DOCUMENT NUMBER:

TITLE: Soft tissue implants and anti-scarring agents INVENTOR(S): Hunter, William L.; Gravett, David M.; Toleikis,

Philip M.; Maiti, Arpita

PATENT ASSIGNEE(S): Angiotech International A.-G., Switz.

PCT Int. Appl., 2592 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 19

PATENT INFORMATION:

PA:	PATENT NO.					KIND DATE									D.	ATE	
WO	2005				A2		2005			 WO 2	 004-	 US39	 465		2	0041	122
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		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KP,	KR,	KΖ,	LC,
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MΖ,	NA,	NI,
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US	2005	0148	512		A1		2005	0707		US 2	004-	9862	30		2	0041	110
US	2005	977		A1		2005	0818		US 2	004-	9862	31		2	0041	110	
CN	1010		Α		2007	1226		CN 2	004-	8003	1664		2	0041	110		
ΑU	2004		A1		2005	0609		AU 2	004-	2930	75		2	0041	122		
CA	A 2536192						2005	0609		CA 2	004-	2536	192		2	0041	122
WO	2005	0512	32		A2		2005	0609		WO 2	004-	US39	346		2	0041	122
WO	2005	0512	32		А3		2005	1208									
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                              20060809
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                               20061213
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    CN 1878514
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                       A1
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    CN 101420970
                        A
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PRIORITY APPLN. INFO.:
                                                              P 20031120
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                                                              A 20041110
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                                           US 2003-518785P
                                                                 20031110
                                           US 2004-582833P
                                                             P 20040624
                                           US 2004-986450
                                                              A1 20041110
                                           WO 2004-US37930
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                                           WO 2004-US39183
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                                                                 20041122
                                           WO 2004-US39346
                                                              W 20041122
                                           WO 2004-US39353
                                                              W 20041122
                                           WO 2004-US39465
                                                              W 20041122
    The invention relates to soft tissue implants for use in cosmetic or
    growth by inflammatory scar tissue. Thus, a silicone gel containing
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AΒ reconstructive surgery and to compns. to make the implants resistant to paclitaxel was used as a filling in breast implant.

OS.CITING REF COUNT: THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD 1 (1 CITINGS)

ANSWER 7 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN L3 ACCESSION NUMBER: 2004:1036851 CAPLUS

DOCUMENT NUMBER: 142:696 TITLE: Synergistic treatment of cancer using immunomers in

conjunction with chemotherapeutic agents

INVENTOR(S): Kandimalla, Ekambar R.; Agrawal, Sudhir; Wang, Daqin

PATENT ASSIGNEE(S): Hybridon, Inc., USA SOURCE: PCT Int. Appl., 106 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIND DATE				APPI	ICAT	ION	NO.		D.	ATE		
	2004 2004								,	WO 2	2004-	 US15	313		2	0040	 514
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		NO,	NΖ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
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		EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	ΙΤ,	LU,	MC,	NL,	PL,	PT,	RO,	SE,
		SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	ΝE,
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ΑU	2004	2410	93		A1		2004	1202		AU 2	004-	2410	93		2	0040	514
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US	2008	0206	265		A1		2008	0828			008-						-
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											004-				A1 2		
									•	WO 2	004-	US15	313	•	W 2	0040	514
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OTHER SOURCE(S): MARPAT 142:696

AB The invention discloses the therapeutic use of immunostimulatory oligonucleotides and/or immunomers in combination with chemotherapeutic agents to provide a synergistic therapeutic effect.

OS.CITING REF COUNT: 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD

(4 CITINGS)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 8 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2004:857358 CAPLUS

DOCUMENT NUMBER: 141:337747

TITLE: Oral formulations of cladribine INVENTOR(S): Bodor, Nicholas S.; Dandiker, Yogesh

PATENT ASSIGNEE(S): Ivax Corporation, USA SOURCE: PCT Int. Appl., 56 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

#### PATENT INFORMATION:

PATENT NO.

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PATENT NO. KIND DATE APPLICATION NO.
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WO 2004087101
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                NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
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BR 2004008848 A 20060404 BR 2004-8848 20040326
CN 1787809 A 20060614 CN 2004-80012713 20040326
CN 100408028 C 20080806
JP 2006521403 T 20060921 JP 2006-509371 20040326
ZA 2005007935 A 20070328 ZA 2005-7935 20040326
ZA 2005007939 A 20070328 ZA 2005-7939 20040326
US 20070197468 A1 20070823 US 2004-551205 20040326
MX 2005010329 A 20060531 MX 2005-10329 20050927
NO 2005004945 A 20051124 NO 2005-4945 20051025
RITY APPLN. INFO.:
US 2003-458922P P 20030328
US 2004-541247P P 20040204
WO 2004-US9387 W 20040326
Provided are compns. of cladribine and cyclodextrin which are especially
                IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK
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ΔR
      Provided are compns. of cladribine and cyclodextrin which are especially suited
      for the oral administration of cladribine. The formulations may be used
      to treat patients with multiple sclerosis.
                                      THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD
OS.CITING REF COUNT:
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REFERENCE COUNT:
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L3 ANSWER 9 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 2004:857357 CAPLUS
                              141:337746
DOCUMENT NUMBER:
TITLE:
                               Cladribine formulations for improved oral and
                              transmucosal delivery
INVENTOR(S):
                         Ivax Corporation, USA
                             Bodor, Nicholas S.
PATENT ASSIGNEE(S):
SOURCE:
                               PCT Int. Appl., 63 pp.
                               CODEN: PIXXD2
DOCUMENT TYPE:
                              Patent
                              English
FAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION:
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KIND DATE APPLICATION NO.

DATE

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WO 2004087100 A2
WO 2004087100 A3
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              NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
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    CN 1787810

JP 2006526009

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ZA 2005007935

A 20070328

ZA 2005007939

MX 2005010330

A 20060531

MX 2005-10330

US 20070065492

A1 20070322

US 2005-551094

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IN 2005DN04555

A 20070817

IN 2005-DN4555

20051006

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A 20051124

NO 2005-4944

20051025

WS 2003-458922P

P 20030328

US 2003-484756P

P 20030702

US 2004-541246P

WO 2004-US9384

W 20040204

WO 2004-US9384

W 20040326
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PRIORITY APPLN. INFO.:
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REFERENCE COUNT:
                                 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
     ANSWER 10 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 2004:780831 CAPLUS
DOCUMENT NUMBER:
                          141:282824
TITLE:
                          Controlled release implant formulations for
                          cell-schedule dependent anticancer agents
INVENTOR(S):
                          Warren, Stephen L.; Dadey, Eric J.; Zhou, Mingxing;
                          Dunn, Richard L.
PATENT ASSIGNEE(S):
                          Atrix Laboratories, Inc., USA
                          PCT Int. Appl., 127 pp.
SOURCE:
                          CODEN: PIXXD2
DOCUMENT TYPE:
                          Patent
LANGUAGE:
                          English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                         KIND DATE APPLICATION NO. DATE
     PATENT NO.
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                         ____
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     WO 2004081196
                                              WO 2004-US7650
                          A2 20040923
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     WO 2004081196
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GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,

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            SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
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     EP 1622540
                        A2
                              20060208 EP 2004-719856
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                    T 20070906
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     JP 2007525429
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                                          US 2005-222668
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                                          US 2003-454100P P 20030311
PRIORITY APPLN. INFO.:
                                          US 2003-505124P
                                                            P 20030922
                                          WO 2004-US7650 W 20040311
AΒ
     The present invention provides a flowable composition suitable for use as a
     controlled release implant. The composition includes: (a) a biodegradable,
     biocompatible thermoplastic polymer that is at least substantially insol.
     in aqueous medium, water or body fluid; (b) a cell-cycle dependent biol.
     agent, a schedule-dependent biol. agent, a metabolite thereof, a
     pharmaceutically acceptable salt thereof, or a prodrug thereof; and (c) a
     biocompatible organic liquid, at standard temperature and pressure, in which
the
     thermoplastic polymer is soluble  The present invention also provides a
     method of treating cancer in a mammal. The present invention also
     provides a method of blocking, impeding, or otherwise interfering with
     cell cycle progression at the G1-phase, G1/S interphase, S-phase, G2/M
     interface or M-phase of the cell cycle in a mammal. The methods includes
     administering to a mammal an effective amount of a flowable composition of the
     present invention. Examples demonstrate the feasibility and efficacy
     potential for intratumoral delivery of Floxuridine in the Atrigel
     (glycolide-lactide copolymer) delivery system to an animal tumor model.
                              THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD
OS.CITING REF COUNT:
                        1
                              (1 CITINGS)
REFERENCE COUNT:
                        1
                              THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS
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    ANSWER 11 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                        2002:521462 CAPLUS
DOCUMENT NUMBER:
                        137:88442
TITLE:
                        Incensole and furanogermacrens and compounds in
                        treatment for inhibiting neoplastic lesions and
                        microorganisms
INVENTOR(S):
                        Shanahan-Pendergast, Elisabeth
PATENT ASSIGNEE(S):
                       Ire.
SOURCE:
                       PCT Int. Appl., 68 pp.
                        CODEN: PIXXD2
DOCUMENT TYPE:
                       Patent
LANGUAGE:
                        English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
    PATENT NO.
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                                         APPLICATION NO.
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                               _____
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                       A2
     WO 2002053138
                               20020711
                                          WO 2002-IE1
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     WO 2002053138
                        A3 20020919
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W: AE, AG, AT, AU, BB, BG, CA, CH, CN, CO, CU, CZ, LU, LV, MA, MD,

UA, UG, US, VN, YU, RU, TJ, TM RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, AT, BE, CH, CY, DE, ES, FI, ML, MR, NE, SN, TD, TG AU 2002219472 A1 20020716 AU 2002-219472 20020102 EP 1351678 A2 20031015 EP 2002-727007 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR A1 20040513 US 20040092583 US 2004-250535 20040102 PRIORITY APPLN. INFO.: IE 2001-2 A 20010102 WO 2002-IE1 W 20020102 OTHER SOURCE(S): MARPAT 137:88442 The invention discloses the use of incensole and/or furanogermacrens, derivs. metabolites and precursors thereof in the treatment of neoplasia, particularly resistant neoplasia and immunodysregulatory disorders. These compds. can be administered alone or in combination with conventional chemotherapeutic, antiviral, antiparasite agents, radiation and/or surgery. Incensole and furanogermacren and their mixture showed antitumor activity against various human carcinomas and melanomas and antimicrobial activity against Staphylococcus aureus and Enterococcus faecalis. OS.CITING REF COUNT: 19 THERE ARE 19 CAPLUS RECORDS THAT CITE THIS RECORD (19 CITINGS) REFERENCE COUNT: THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT ANSWER 12 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2001:300514 CAPLUS DOCUMENT NUMBER: 134:331617 Oil-in-water emulsion compositions for polyfunctional TITLE: active ingredients INVENTOR(S): Chen, Feng-jing; Patel, Mahesh V. PATENT ASSIGNEE(S): Lipocine, Inc., USA PCT Int. Appl., 82 pp. SOURCE: CODEN: PIXXD2 DOCUMENT TYPE: Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: KIND DATE APPLICATION NO. DATE PATENT NO. \_\_\_\_\_ \_\_\_\_ \_\_\_\_\_ \_\_\_\_\_ WO 2000-US28835 A1 20010426 WO 2001028555 20001018 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU,

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WO 2001028555

A1 20010426 WO 2000-US28835 20001018

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

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US 20020107265 A1 20020808 US 1999-420159 19991018

US 6720001 B2 20040413

PRIORITY APPLN. INFO:

US 1999-420159 A 19991018

AB Pharmaceutical oil-in-water emulsions for delivery of polyfunctional active ingredients with improved loading capacity, enhanced stability, and reduced irritation and local toxicity are described. Emulsions include an
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aqueous phase, an oil phase comprising a structured triglyceride, and an emulsifier. The structured triglyceride of the oil phase is substantially

free of triglycerides having three medium chain (C6-C12) fatty acid

moieties, or a combination of a long chain triglyceride and a polarity-enhancing polarity modifier. The present invention also provides methods of treating an animal with a polyfunctional active ingredient, using dosage forms of the pharmaceutical emulsions. For example, an emulsion was prepared, with cyclosporin A as the polyfunctional active ingredient dissolved in an oil phase including a structured triglyceride (Captex 810D) and a long chain triglyceride (safflower oil). The composition contained (by weight) cyclosporin A 1.0, Captex 810D 5.0, safflower oil 5.0, BHT 0.02, egg phospholipid 2.4, dimyristoylphosphatidyl glycerol 0.2, glycerol 2.25, EDTA 0.01, and water up to 100%, resp.

OS.CITING REF COUNT: 17 THERE ARE 17 CAPLUS RECORDS THAT CITE THIS RECORD (17 CITINGS)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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(FILE 'HOME' ENTERED AT 13:38:46 ON 10 SEP 2009)

FILE 'REGISTRY' ENTERED AT 13:38:55 ON 10 SEP 2009

E CLADRIBINE/CN

L1 1 S E3

E CYCLODEXTRIN/CN

L2 1 S E3

FILE 'CAPLUS' ENTERED AT 13:39:31 ON 10 SEP 2009

L3 12 S L1 AND L2

L4 6 S L3 AND PY<=2004

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COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 40.24 55.74

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE TOTAL ENTRY SESSION

CA SUBSCRIBER PRICE -9.84 -9.84

SESSION WILL BE HELD FOR 120 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 13:42:04 ON 10 SEP 2009

Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID:ssptajs11623

PASSWORD:

\* \* \* \* \* RECONNECTED TO STN INTERNATIONAL \* \* \* \* \* \* SESSION RESUMED IN FILE 'CAPLUS' AT 14:31:02 ON 10 SEP 2009 FILE 'CAPLUS' ENTERED AT 14:31:02 ON 10 SEP 2009 COPYRIGHT (C) 2009 AMERICAN CHEMICAL SOCIETY (ACS)

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 40.24 55.74 DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -9.84 -9.84 => d his (FILE 'HOME' ENTERED AT 13:38:46 ON 10 SEP 2009) FILE 'REGISTRY' ENTERED AT 13:38:55 ON 10 SEP 2009 E CLADRIBINE/CN 1 S E3 L1E CYCLODEXTRIN/CN L2 1 S E3 FILE 'CAPLUS' ENTERED AT 13:39:31 ON 10 SEP 2009 L3 12 S L1 AND L2 L46 S L3 AND PY<=2004 => s 12 and (purine or adenosine) and (inclusion or complex) and amorphous 7414 L2 42133 PURINE 12214 PURINES 46736 PURINE (PURINE OR PURINES) 98553 ADENOSINE 819 ADENOSINES 98749 ADENOSINE (ADENOSINE OR ADENOSINES) 135543 INCLUSION 73659 INCLUSIONS 181483 INCLUSION (INCLUSION OR INCLUSIONS) 1507827 COMPLEX 816567 COMPLEXES 1831504 COMPLEX (COMPLEX OR COMPLEXES) 301262 AMORPHOUS 5 AMORPHOUSES 301266 AMORPHOUS (AMORPHOUS OR AMORPHOUSES) L5 0 L2 AND (PURINE OR ADENOSINE) AND (INCLUSION OR COMPLEX) AND AMOR PHOUS => s 12 and (purine or adenosine) and (inclusion or complex) 7414 L2 42133 PURINE 12214 PURINES 46736 PURINE (PURINE OR PURINES) 98553 ADENOSINE 819 ADENOSINES 98749 ADENOSINE (ADENOSINE OR ADENOSINES) 135543 INCLUSION 73659 INCLUSIONS 181483 INCLUSION (INCLUSION OR INCLUSIONS)

1507827 COMPLEX 816567 COMPLEXES

1831504 COMPLEX

(COMPLEX OR COMPLEXES)

L6 13 L2 AND (PURINE OR ADENOSINE) AND (INCLUSION OR COMPLEX)

=> s 16 and py<=2004 25141550 PY<=2004

L7 8 L6 AND PY<=2004

=> d 17 1-8 ibib abs

L7 ANSWER 1 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2002:521462 CAPLUS

DOCUMENT NUMBER: 137:88442

TITLE: Incensole and furanogermacrens and compounds in treatment for inhibiting neoplastic lesions and

microcraniana

microorganisms

INVENTOR(S): Shanahan-Pendergast, Elisabeth

PATENT ASSIGNEE(S): Ire.

SOURCE: PCT Int. Appl., 68 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	PATENT NO.					KIND DATE			APPLICATION NO.						D	ATE		
						_									_			
WO	2002	0531	38		A2		2002	0711	1	WO 2	002-	IE1			2	0020	102 ·	<
WO	2002	0531	38		А3		2002	0919										
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		UA,	UG,	US,	VN,	YU,	RU,	ТJ,	TM									
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		${ m ML}$ ,	MR,	ΝE,	SN,	TD,	ΤG											
AU	2002	2194	72		A1		2002	0716		AU 2	002-	2194	72		2	0020	102 ·	<
EP	1351	678			A2		2003	1015		EP 2	002-	7270	07		2	0020	102 -	<
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		ΙE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL,	TR							
US	2004	0092	583		A1		2004	0513	1	JS 2	004-	2505.	35		2	0040	102 -	<
PRIORITY APPLN. INFO.:				.:				IE 2001-2				Ž	A 2	010	102			
									1	WO 2	002-	IE1		I	W 2	0020	102	

OTHER SOURCE(S): MARPAT 137:88442

AB The invention discloses the use of incensole and/or furanogermacrens, derivs. metabolites and precursors thereof in the treatment of neoplasia, particularly resistant neoplasia and immunodysregulatory disorders. These compds. can be administered alone or in combination with conventional chemotherapeutic, antiviral, antiparasite agents, radiation and/or surgery. Incensole and furanogermacren and their mixture showed antitumor activity against various human carcinomas and melanomas and antimicrobial activity against Staphylococcus aureus and Enterococcus faecalis.

OS.CITING REF COUNT: 19 THERE ARE 19 CAPLUS RECORDS THAT CITE THIS RECORD (19 CITINGS)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 2 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:95053 CAPLUS

DOCUMENT NUMBER: 132:242544

TITLE: Advanced statistical evaluation of complex

formation constant from electrophoretic data

AUTHOR(S): Bartak, P.; Bednar, P.; Kubacek, L.; Stransky, Z. CORPORATE SOURCE: Trida Svobody 8, Centre of Bioanalytical Research,

Palacky University, Olomouc, 771 46, Czech Rep.

SOURCE: Analytica Chimica Acta (2000), 407(1-2),

327-336

CODEN: ACACAM; ISSN: 0003-2670

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB A new method for the estimation of <u>complex</u> formation consts. is presented. The method is based on electrophoretically measured effective mobilities and applied to the estimation of the <u>complex</u> formation constant in respect to interactions between nitrogen heterocyclic bases and cyclodextrines. The calcn. of consts. is based on the linearization of the dependence between effective mobility and the cyclodextrine concentration

and

the application of an advanced statistical evaluation procedure. <u>Complex</u> formation consts. 14.8 and 63.2 l/mol were obtained for the interaction of pyridinium and benzylaminopurinium with dimethyl- $\beta$ -cyclodextrin (DM- $\beta$ -CD), resp. Consts. in the order of magnitude 101-102 l/mol were obtained for some other <u>purine</u> derivs. The proposed procedure, in connection with the math. software for matrix operations, is rather simple and gives much more valuable outputs than commonly used concepts.

OS.CITING REF COUNT: 8 THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD

(8 CITINGS)

REFERENCE COUNT: 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 3 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:65552 CAPLUS

DOCUMENT NUMBER: 132:127462

TITLE: Particles, in particular micro- or nanoparticles, of

crosslinked mono- and oligosaccharides, their production, and cosmetic, pharmaceutical, or food

compositions containing them

INVENTOR(S): Perrier, Eric; Rey-Goutenoire, Sylvie; Buffevant,

Chantal; Levy, Marie-Christine; Pariot, Nadine;

Edwards, Florence; Andry, Marie-Christine

PATENT ASSIGNEE(S): Coletica, Fr.

SOURCE: Ger. Offen., 34 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 19932216	A1	20000127	DE 1999-19932216	19990709 <
DE 19932216	B4	20051208		
FR 2780901	A1	20000114	FR 1998-8809	19980709 <
FR 2780901	B1	20000929		
NL 1012517	C2	20000111	NL 1999-1012517	19990705 <
KR 2000011579	A	20000225	KR 1999-27476	19990708 <
KR 799407	В1	20080130		
JP 2000038402	A	20000208	JP 1999-196705	19990709 <

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JP 3437797
                       В2
                              20030818
                      B1 20010306 US 1999-350131
A1 20010516 ES 1999-1547
    US 6197757
                      B1
                                                               19990709 <--
    ES 2155793
                                                               19990709 <--
    ES 2155793
                       B1
                            20011201
    IT 1311514
                        B1 20020313 IT 1999-T0599
                                                               19990709 <--
                                        FR 1998-8809 A 19980709
PRIORITY APPLN. INFO.:
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AB Particles consisting of ≥1 mono- or oligosaccharide, which are surface-crosslinked in emulsion by esterification of primary OH groups on the saccharides with a polyfunctional acylating agent, are useful as carriers or encapsulating agents for various hydrophilic or lipophilic active substances in preparation of cosmetic, pharmaceutical, or food compns. The particles are biocompatible, biodegradable, and suitable for stabilization and protection of sensitive active substances or for their sustained release. The crosslinking reaction preferably occurs in a water-in-oil emulsion at room temperature and results in formation of a membrane

of crosslinked saccharide surrounding an aqueous phase. The saccharide may be a cyclodextrin; by forming an <code>inclusion</code> compound with an active substance, it can be used to remove or harvest the latter from a liquid medium, or alternatively can slowly release an active substance from an <code>inclusion</code> compound Thus, 6 mL of a 10% solution of dihydroxyacetone (a ketose) in 1M carbonate buffer (pH 11) was emulsified in 30 mL cyclohexane containing 5% Span 85, and with continued stirring, 40 mL of a 5% solution of terephthaloyl chloride in CHCl3-cyclohexane (1:4 by volume); after 30 min, the microcapsules were collected and washed. These microcapsules dissolved slowly in 1% Na2CO3 solution or in PEG owing to alcoholysis of the ester bonds; the released dihydroxyacetone reacted with glycine to form a brown color. The microcapsules can therefore be used in cosmetic tanning prepns.

OS.CITING REF COUNT: 9 THERE ARE 9 CAPLUS RECORDS THAT CITE THIS RECORD (10 CITINGS)

 ${\tt L7}$   $\,$  ANSWER 4 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1999:549161 CAPLUS

DOCUMENT NUMBER: 131:175082

TITLE: High-energy cyclodextrin-drug complexes with

increased bioavailability

INVENTOR(S): Loftsson, Thorsteinn; Masson, Mar; Stefansson, Einar

PATENT ASSIGNEE(S): Cyclops, Iceland

SOURCE: PCT Int. Appl., 51 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATE	PATENT NO.				KIN	D	DATE			APPL	ICAT	ION 1	NO.		DZ	ATE	
 WO 9	942	 111			 A1	_	 1999	0826	,	 WO 1	 999-	 IS3			19	 9990:	216 <
	W:	AL,	AM,	AT,	AU,	AZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CU,	CZ,	DE,
		DK,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,
		ΚE,	KG,	KP,	KR,	KΖ,	LC,	LK,	LR,	LS,	LT,	LU,	LV,	MD,	MG,	MK,	MN,
		MW,	MX,	NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,
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		CM,	GΑ,	GN,	GW,	ML,	MR,	NE,	SN,	TD,	TG						
CA 2	CA 2320772			A1 19990826			1	CA 1999-2320772					19	9990:	216 <		
AU 9	AU 9926385				А					AU 1999-26385					19	9990:	216 <

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B2 20030410
A1 20010117 EP 1999-906440
       AU 759280
       EP 1067942
                                                                                                  19990216 <--
            R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
                  IE, FI
      NZ 505951 A 20030228 NZ 1999-505951 19990216

JP 2003522207 T 20030722 JP 2000-532126 19990216

US 6699849 B1 20040302 US 1999-250185 19990216

US 20040186075 A1 20040923 US 2004-750940 20040105

RITY APPLN. INFO.: US 1998-75544P P 19980223
                                                                                                19990216 <--
                                                                                                19990216 <--
                                                                                                  19990216 <--
                                                                                                  20040105 <--
PRIORITY APPLN. INFO.:
                                                                US 1999-250185
                                                                                           A1 19990216
                                                                WO 1999-IS3
                                                                                           W 19990216
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AB Methods for enhancing the complexation efficiency of a drug with cyclodextrin and for enhancing the availability of a drug following administration of a cyclodextrin-drug *complex*.

Phenytoin-2-hydroxypropyl  $\beta$ -cyclodextrin complexes were

prepared, lyophilized to a powder which can be formulated into tablets. The bioavailability of phenytoin was enhanced.

OS.CITING REF COUNT: 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD

(3 CITINGS)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 5 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1997:727151 CAPLUS

DOCUMENT NUMBER: 128:23072

ORIGINAL REFERENCE NO.: 128:4531a,4534a

TITLE: Oligosaccharide analogs of polysaccharides. Part 14.

Carbocyclic cyclodextrin analogs. Synthesis of all

trimeric and tetrameric isomers by homo- and

heterocoupling of 1,4-cis-diethynylated

1,5-anhydroglucitols

AUTHOR(S): Burli, Roland; Vasella, Andrea

CORPORATE SOURCE: Lab. Organische Chemie, ETH-Zentrum, Zurich, CH-8092,

Switz.

SOURCE: Helvetica Chimica Acta (1997), 80(7),

2215-2237

CODEN: HCACAV; ISSN: 0018-019X Verlag Helvetica Chimica Acta

DOCUMENT TYPE: Journal LANGUAGE: English

GΙ

PUBLISHER:

AB Hetero- or homocoupling of protected 1,4-cis-diethynylated 1,5-anhydroglucitols leads to 2 isomeric cyclotrimers and to 4 isomeric cyclotetramers. The C1-sym. cyclotrimer I and the C1- and the C2-sym. cyclotetramers II and III, resp., were prepared The cyclotrimer I was prepared by intramol., oxidative homocoupling and, alternatively, by a 1-pot trimerization/cyclization of the monomer. Oxidative homocoupling was used for the cyclization of appropriate tetramers to II and III. The acyclic tetramers were made by sequential Cadiot-Chodkiewicz coupling or by a combination of a Cadiot-Chodkiewicz reaction and an intermol., oxidative homocoupling. The solid-state conformation of a C4-sym. cyclotetramer corresponds well to the one predicted by force-field calcns. The water-solubilities of cyclotrimers and -tetramers, their calculated

<sup>\*</sup> STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT \*

conformations, and the D-<u>adenosine</u> binding properties of the cyclotetramers were compared.

OS.CITING REF COUNT: 20 THERE ARE 20 CAPLUS RECORDS THAT CITE THIS RECORD (20 CITINGS)

L7 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1994:711646 CAPLUS

DOCUMENT NUMBER: 121:311646

ORIGINAL REFERENCE NO.: 121:56853a,56856a

TITLE: Proton Transfer and  $n \to \pi^*$  Transition in the

Photophysics of 1,N6-Ethenoadenosine

AUTHOR(S): Agbaria, Rezik A.; Parola, Abraham H.; Gill, David

CORPORATE SOURCE: Department of Physics, Ben-Gurion University,

Beer-Sheva, 84105, Israel

SOURCE: Journal of Physical Chemistry (1994),

98(50), 13280-5

CODEN: JPCHAX; ISSN: 0022-3654

DOCUMENT TYPE: Journal LANGUAGE: English

The photophys. characteristics of 1,N6-enthenoadenosine ( $\epsilon$ Ado) AΒ show irregularities in terms of the expected photophysics from a pH equilibrium between two forms that absorb light at different wavelengths. Furthermore, a comparison between the absorption spectra of purine , adenine, and  $\epsilon$ Ado leads to the conclusion that  $\epsilon$ Ado does not follow the adenine, but rather has more in common with the purine. The adenine itself does not follow its parent compound, purine. We, therefore, reinterpret the absorption of  $\varepsilon\overline{A}$ do, such as the unprotonated form has two absorption bands, the second of which is an n  $\rightarrow$   $\pi^*$  transition, whereas the protonated form has only one  $\pi \to \pi^*$  absorption band, which overlaps with the first absorption band of the unprotonated form. The n  $\rightarrow$   $\pi^{\star}$ absorption "disappeared" upon protonation, apparently due to stabilization of the lone-pair electrons. Under these presumptions, the photophysics of  $\epsilon$ Ado is no longer peculiar. Transitions to and from both excited singlet states,  $\text{S}\pi\pi^{\star}$  and  $\text{S}\text{n}\pi^{\star},$  along with the relative order of their resp. triplets, are shown to play an active role in the photophysics of  $\epsilon Ado$ . Therefore, the reported multiple emissions from εAdo, at low temperature, are to be expected. The reported observations in the literature provide evidence for the multiple excited states of εAdo. In the present work, cyclodextrins provide a powerful tool in the photophys. study of &Ado. In particular, cyclodextrin host isolation matrix (CHIM) provides a unique environment that can be applied to mimic the photophysics of the isolated mol. in the gas phase or at low temps.

OS.CITING REF COUNT: 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS)

L7 ANSWER 7 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1994:69602 CAPLUS

DOCUMENT NUMBER: 120:69602

ORIGINAL REFERENCE NO.: 120:12359a,12362a

TITLE: Preparation and use of polyanionic polymer-based

conjugates targeted to vascular endothelial cells

INVENTOR(S): Thorpe, Philip E.

PATENT ASSIGNEE(S): University of Texas System, USA; Imperial Cancer

Research Technology Ltd.

SOURCE: PCT Int. Appl., 117 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

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PATENT NO.
                  KIND DATE APPLICATION NO. DATE
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                             _____
                       A1 19930930 WO 1993-US2619 19930322 <--
    WO 9318793
        W: AT, AU, BB, BG, BR, CA, CH, CZ, DE, DK, ES, FI, GB, HU, KP, KR,
           LU, MG, MN, MW, NL, NO, PL, PT, US
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            BJ, CF, CG, CI, CM, GA, GN, ML, MR
                       A 19951212 US 1992-856018
A 19931021 AU 1993-38166
A1 19950111 EP 1993-907633
    US 5474765
                                                              19920323 <--
    AU 9338166
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    EP 632728
       R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, LU, MC, NL, PT
    US 5762918 A 19980609
                                        US 1994-307745 19941205 <--
PRIORITY APPLN. INFO.:
                                         US 1992-856018
                                                          A2 19920323
                                         WO 1993-US2619 A 19930322
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AB An anionic polymer (e.g. a heparin derivative) is linked to an active agent (especially a steroid), preferably by a selectively hydrolyzable bond, for delivery of the active agent to vascular endothelial cells. The conjugates are useful as angiogenesis inhibitors for treatment of e.g. cancer, arthritis, and diabetic blindness. Thus, heparin was condensed with adipic dihydrazide and then with cortisol; the cortisol:heparin mol ratio in the product was 8-9. This conjugate was markedly acid labile, suppressed DNA synthesis and cell migration in human umbilical vein endothelial cells, retarded or abolished the vascularization of sponges in vivo, and retarded lung tumor growth in mice by 65%. No adverse effects of the conjugate were detected, and equivalent treatments with a mixture of heparin and cortisol were significantly less effective in all cases.

OS.CITING REF COUNT: 11 THERE ARE 11 CAPLUS RECORDS THAT CITE THIS RECORD (11 CITINGS)

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 8 OF 8 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:637813 CAPLUS

DOCUMENT NUMBER: 119:237813

ORIGINAL REFERENCE NO.: 119:42169a, 42172a

TITLE: Dye transfer thermal printing process. VI. Prevention

of image decoloration in dye transfer recording

AUTHOR(S): Kusakawa, Hideaki; Enmanji, Koe

CORPORATE SOURCE: Kanazawa Inst. Technol., Nonoichi, 721, Japan SOURCE: Denshi Shashin Gakkaishi (1993), 32(1), 3-6

CODEN: DSHGDD; ISSN: 0387-916X

DOCUMENT TYPE: Journal LANGUAGE: Japanese

AB The thermal dye transfer color ink, which is developed to have same sensitivity as the com. used thermal printing paper for G-II type facsimile, is composed of dyes such as SOT-Blue 2, -Red 2G, and -Yellow 5 with suitable binder polymers. The light fastness of these dyes is low. Thus, it is necessary to improve it, especially, for -Blue 2. Decoloration of the dye is prevented either by charge-transfer <code>complex</code> formation or the <code>inclusion</code> of the dyes. For binder polymers such as PMMA, in which the dye is dissolved rather than dispersed, it is not possible to form charge-transfer <code>complexes</code> and improvement of light fastness is not observed For polar binder polymers such as poly(vinyl alc.), in which the dye and electron-acceptor particles are dispersed rather than dissolved, it was necessary to add electron-acceptor to form

complexes. The dye mol. is too large for cyclodextrin to enclose
it, and, accordingly, the improvement in light fastness was not so
remarkable.

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CA SUBSCRIBER PRICE	-16.40	-16.40

SESSION WILL BE HELD FOR 120 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 14:32:22 ON 10 SEP 2009

# IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	MAIL STOP AMENDMENT
Nicholas Bodor et al.	Group Art Unit: 1623
Application No.: 10/551,205	Examiner: JONATHAN S LAU
Filed: November 14, 2006	Confirmation No.: 4092
For: ORAL FORMULATIONS OF CLADRIBINE	) ) )
<b>(</b>	

# **REPLY TO OFFICIAL ACTION**

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In response to the Office Action dated September 17, 2009, the following remarks are offered:

#### **REMARKS**

Applicants request reexamination and reconsideration of the subject application pursuant to and consistent with 37 C.F.R. § 1.112 in light of the following:

# STATUS OF CLAIMS

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 66-98 remain in this application. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81 have been withdrawn from consideration. Claims 1, 2, 8, 9, 11, 12, 56, 57, 63, 64, 66 and 82-98 are under examination.

It is respectfully pointed out that, while the other claims under examination have been rejected, Claims 12 and 66 are not indicated to be rejected in the Office Action Summary or in the rejection set forth in page 4 of the Office Action.

Clarification is requested.

#### **INFORMATION DISCLOSURE STATEMENTS**

Applicants thank the Examiner for considering the documents cited in their Fifth Information Disclosure Statement filed July 6, 2009. A Sixth Information Disclosure Statement is filed herewith. The document listed on the accompanying form PTO-1449 is discussed in the remarks which follow.

#### **REJECTION WITHDRAWN**

The Examiner's previous rejection under 35 U.S.C. § 103(a) based on Schultz et al. in view of Baert et al. has been withdrawn.

#### **CLAIM REJECTIONS - 35 U.S.C. § 103**

Amended Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 have been rejected under 35 U.S.C. § 103(a) as being unpatentable over Schultz et al. US Patent No. 6,194,395, in view of Wrenn Jr. US Patent No. 6,174,873 and in view of Loftsson et al. US Patent 6,699,849. Applicants believe that this rejection is untenable against any of the claims in this application.

#### **APPLICANTS' INVENTION**

While the Examiner has withdrawn his previous obviousness rejection based on Schultz et al. in view of Baert et al., the Examiner nevertheless clings to his interpretation of particular features of Schultz et al. to the exclusion of what the Schultz et al. patent as a whole teaches to one of ordinary skill in the art.

First, applicants would like to once again draw the Examiner's attention to the essential features of applicants' invention as set forth in independent Claims 56 and 1, as well as in Claim 82.

Claim 56 is drawn to a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of: (a) an amorphous inclusion complex of cladribine with the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin (HPβCD) and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, said complex cladribine-cyclodextrin complex having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16. This is a complex complex of cladribine and hydroxypropyl-β-cyclodextrin (HPβCD) in which there is an intimate mixture consisting of two different complexes, first an amorphous inclusion complex of cladribine and HPβCD (itself an amorphous cyclodextrin) and secondly a non-inclusion complex in which amorphous free cladribine is associated with the amorphous cyclodextrin HPβCD, and moreover this complex complex has a very particular weight ratio of cladribine to HPβCD of from about 1:10 to about 1:16.

The cladribine/ HPβCD complex of the invention has many properties that distinguish it from a mere mixture of hydroxypropyl-β-cyclodextrin and cladribine as was fully and convincingly shown by the data provided in the Van Axel Castelli et al. document provided with applicants' previous response and discussed in great detail therein.

The unique cladribine/hydroxypropyl-β-cyclodextrin complex defined in Claim 56 is an essential feature of applicants' unique pharmaceutical composition as claimed in Claim 1. Claim 1 is drawn to a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with the amorphous

cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16. Thus, in addition to the special features of the complex complex itself as already discussed, it is essential that the pharmaceutical composition comprise no significant amount of free crystalline cladribine therein and that the complex complex be formulated into a solid oral dosage form. Neither the complex complex nor the pharmaceutical composition comprises any significant amount of free crystalline cladribine; this is excluded from the complex complex by use of the closed "consisting of" language in defining the components of the admixture therein.

Claim 82 is a product-by-process claim which specifies the steps applicants have found to provide the unique pharmaceutical composition of Claim 1, which in turn contains as an essential feature the precisely defined unique complex complex of Claim 56. Thus, Claim 82 is drawn to a pharmaceutical composition according to Claim 1 obtainable by a process comprising the steps of:

- (i) combining cladribine and the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin in water at a temperature of from about 45 to about 80°C and maintaining said temperature for a period of from about 6 to about 24 hours;
  - (ii) cooling the resultant aqueous solution to room temperature;
  - (iii) lyophilizing the cooled solution to afford an amorphous product; and
- (iv) formulating the amorphous product into a solid oral dosage form.

  The time and temperature conditions set forth in part (i) are especially critical to formation of the Claim 56 complex complex and ultimately to formation of the Claim 1 pharmaceutical composition.

The Examiner relies upon Schultz et al. for disclosing a solid pharmaceutical dosage form comprising cladribine and cyclodextrin at column 2, lines 31-39. These lines do not indicate the nature of the relationship between the cladribine and the cyclodextrin. However, as applicants pointed out to the Examiner previously, Schultz et al. elsewhere disclose that their oral <u>solid</u> dosage forms contain a <u>mixture</u> of cladribine and cyclodextrin; see column 1, lines 8-10. Reference to <u>solid mixtures</u> is

also made by Schultz et al. in column 5, lines 50-64. Schultz et al. teach inclusion complex formation in solution but <u>only</u> to form <u>injectable solutions</u>. As to ratios, Schultz et al.'s weight ratios for their solid oral dosage form are <u>1</u> mg to 15 mg of cladribine and 100 mg to 500 mg of cyclodextrin (col. 6, lines 23-31). This does not lead one of ordinary skill to "instantly envision" a cladribine:cyclodextrin ratio ranging from 15 mg:100mg to 15 mg:500 mg, but rather from 1 mg:500 mg to 15 mg:100 mg. This is <u>not</u> 1:6.67 to 1:33.3, it is 1:500::1:6.67, a much broader ratio range than that stated by the Examiner. Further, as applicants previously pointed out, Schultz et al.'s ratio is for a <u>mixture</u>, <u>not</u> for a complex. Virtually any ratio could be present in a mixture. Such does not suggest what ratios would be not only possible but also advantageous in a complex.

While applicants agree that Schultz et al. do not disclose any of the many features noted by the Examiner in the paragraph spanning pages 5-6 of the Official Action, it is pointed out that Schultz et al. also do not suggest applicants' complex complex (as defined in Claim 56) or a pharmaceutical composition comprising applicants' complex complex formulated into a solid oral dosage form comprising no significant amount of free crystalline cladribine (as defined in Claim 1) or a process for preparing such a pharmaceutical formulation (as defined in Claim 82). However, what is missing from Schultz et al. is not supplied by the secondary references, Wrenn, Jr. and Loftsson et al.

It is agreed that the Wrenn, Jr. patent is directed to solid formulations for oral administration and that the adenosine analogs therein include cladribine. Wrenn, Jr.'s teaching in column 12, lines 25-30 is part of his discussion of an INDAS system. As noted at line 25, INDAS takes the form of a high energy matrix tablet. Production of that matrix tablet involves including adenosine analogs in an amorphous form together with a combination of energy, excipients and unique processing procedures. Wrenn, Jr. goes on to state (col. 12, lines 30-40) that once included in the desirable physical form, the resultant high energy complex may be stabilized by an absorption process that utilizes a novel polymer cross-linked technology to prevent recrystallization. The combination of the change in the physical state of the adenosine analogs coupled with the solubilizing characteristics of the excipients employed enhances the solubility of the adenosine analogs. The resulting absorbed

amorphous drug complex granulate may then, according to Wrenn, Jr., be formulated with a gel-forming erodable tablet system to promote substantially smooth or continuous absorption. As set forth in Wrenn, Jr.'s Claim 1, his oral dosage form comprises an acid-labile 2-deoxyadenosine analog which chemically decomposes in the acid environment and one or more components which inhibit that decomposition selected from the group consisting of erodible matrix, enteric coating, solid dispersion and ion exchange resin. In Claim 8, Wrenn, Jr. specifies that the composition is in a controlled-release mechanism. In Claim 10, the controlledrelease mechanism may be an INDAS system, among others. Thus, while Wrenn, Jr. suggests amorphous forms of the drug, this is in the context of its being only one part of the INDAS system he is describing. Wrenn, et al. in no way suggests cyclodextrin complexation, much less how to make or how to use the specific complex complex of applicants' claims, which has nothing to do with novel polymer cross-linking technology. Indeed, cyclodextrins are not polymers and do not provide cross-linking! Wrenn, Jr. neither discloses nor suggests what applicants have done, which is to provide an intimate admixture consisting of (a) an amorphous inclusion complex of cladribine with HPβCD, and (b) amorphous free cladribine associated with said HPBCD as a non-inclusion complex, which is formulated into a solid oral dosage form comprising no significant amount of free crystalline cladribine therein, the cladribine: HP&CD ratio being from about 1:10 to about 1:16. There is no cladribine/cyclodextrin inclusion complex in Wrenn, Jr. and there is no non-inclusion complex there either, much less the remotest suggestion of applicants' invention as claimed herein. Indeed, Wrenn, Jr. is not remotely relevant to the present invention.

We turn now to Loftsson et al., which does indeed relate to cyclodextrins. Indeed, the very Loftsson and Brewster cyclodextrin review article referenced on page 18 of applicants' July 6, 2009 response and previously made of record in applicants' Third Information Disclosure Statement is referenced in column 2, lines 20-24 of the Loftsson et al. patent relied upon by the Examiner. Applicants already acknowledged many basic teachings in the cyclodextrin art, including the amorphous nature of HPβCD. Applicants agree that the statements made by the Examiner on page 6 of the Official Action about cyclodextrins are correct, with one notable exception. Loftsson et al do not teach that purine derivatives are, without

<u>qualification</u>, compatible with cyclodextrin for forming complexes. The Examiner has taken page 9, lines 50-55 completely out of context, as explained below.

The Loftsson et al. patent is aimed at enhancing the cyclodextrin complexation efficiency of certain structural classes of drugs by relying on reversible ring opening. In the OBJECTS AND SUMMARY OF THE INVENTION in columns 4-6, the Examiner's attention is drawn in particular to column 5, line 43 to column 6. line 49, where various aspects of the Loftsson et al. invention are summarized. In each of these aspects, the drug is defined as "having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen. oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal..." (Emphasis added). See also the wording of the claims of Loftsson et al.'s PCT counterpart, WO 99/42111, submitted with the accompanying IDS, which repeat this language. (The US claims recite only benzodiazepines, all of which have a cycle imine structure.) Thus, when Loftsson et al. disclose in columns 8-9 groups of preferred drugs, the patentees are speaking in the context of the quoted language; in other words, their purines are not any purines but are only ones which are susceptible to reversible ring opening. In the case of purines, Loftsson et al. disclose that the drugs are preferably caffeine, theophylline, etophylline, proxyphylline or theobromine. It is immediately apparent that cladribine, whose structure is depicted on page 1 of the instant application and in column 1 of Schultz et al., does not have the requisite imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal structure and thus is not susceptible to reversible ring opening and it not appropriate for use in Loftsson et al.'s invention. Moreover, as would be readily apparent to one of ordinary skill, cladribine is a nucleoside analog; it is a purine base in glycosidic linkage with a sugar, which is a ribofuranose. Loftsson et al. do not remotely suggest nucleosides or nucleoside analogs.

Secondly, the process described by Loftsson et al. for complexing their drugs with cyclodextrin is conducted at a pH level below about 5, preferably between about 3 and about 5, in the case of basic drugs such as benzodiazepine. See column 10 and the Examples, especially beginning with Example 3. The work described is

carried out in solution. Generally, the suspension of drug in aqueous cyclodextrin solution is heated in a sealed container in an autoclave, presumably to encourage ring openings at acidic pH. There is no specific disclosure of a solid oral dosage form that applicants can locate Furthermore, it is well-known that cladribine is acidlabile; as noted by Schultz et al., use of the compound orally has been limited by this fact. See col. 1, lines 36-51 of Schultz et al. Thus, one of ordinary skill would not be motivated to subject cladribine to the methods of Loftsson et al.'s patent, first because it does not even meet the structural requirements for use therein and secondly because it would decompose at the low pH levels favored by Loftsson et al. There is indeed no suggestion of cladribine in the Loftsson et al. patent.

While various methods of preparing drug-cyclodextrin complexes have been known in the art, applicants have found that a very carefully controlled series of steps are required to produce the instantly claimed complex complex and pharmaceutical composition containing it. The time and temperature specified in Claim 82 in step (i) are essential, for example, for production of applicants' unique products. The sum total of the conditions used by applicants together with the fact that cladribine is a nucleoside analog and thus has not only a particular purine ring but also a ribofuranose sugar ring surprisingly gives a very complex product when combined with HPBCD, because the hydroxyl groups on the ribofuranose ring in cladribine are able to hydrogen-bond to the hydroxyls on the exterior of the cyclodextrin ring, while the purine ring of cladribine is able to be at least partially included in the cavity of the cyclodextrin ring. This provides a very unique result which could not have been predicted. The present invention takes advantage of both parts (purine base + sugar) of the structure of cladribine so as to provide an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with HPβCD and (b) amorphous free cladribine associated with HPβCD as a non-inclusion complex (which is multiple hydrogen bond mediated as explained above), which then is further formulated into a solid oral dosage form, the composition comprising no significant amount of free crystalline cladribine therein, the composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16. There is no suggestion in the art of this complex complex or of how to obtain it; the cited references in combination fail to lead one of

Attorney Docket No. 0056192-000024 Application No. 10/551,205 Page 9

ordinary skill to this result. Thus, the subject matter of Claims 1, 5 and 82 is free of the outstanding rejection. The same is true of independent Claims 13 and 25 which contain all of the limitations of Claim 1. The more specific claims herein are even more remote from the prior art.

With respect to the product-by-process claims, applicants agree that the product of Claim 82 is the same as the product of Claim 1. Claims 82-90 and 94-98 are, however, patentable for all of the reasons set forth above with respect to Claim 1.

#### CONCLUSION

In view of the foregoing, it is submitted that all claims in this application are free of the record rejections. Further, favorable action in the form of a Notice of Allowance is believed to be next in order and is earnestly solicited.

Respectfully submitted,

BUCHANAN INGERSOLL & ROONEY PC

Date: Sumber 16, 2009

Bv:

Mary Katherine Baumeister

Registration No. 26254

Customer No. 21839

703 836 6620

Attorney Docket No. <u>0056192-000024</u>

# IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	MAIL STOP AMENDMENT
Nicholas Bodor et al.	) Group Art Unit: 1623
Application No.: 10/551,205	Examiner: JONATHAN S LAU
Filing Date: November 14, 2006	Confirmation No.: 4092
Title: ORAL FORMULATIONS OF CLADRIBINE	) ) )

# SIXTH INFORMATION DISCLOSURE STATEMENT TRANSMITTAL LETTER

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

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		Resp	ectfully submitted,
		BUC	HANAN INGERSOLL AND ROONEY PC
Date	<u>December 16, 2009</u>	Ву:	Mary Katherine Baumeister Registration No. 26254
Custo	omer No. 21839		

Attorney Docket No. <u>0056192-000024</u>

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

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Nicholas Bodor et al.	) Group Art Unit: 1623
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For: ORAL FORMULATIONS OF CLADRIBINE	) ) ) )

### SIXTH INFORMATION DISCLOSURE STATEMENT

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In accordance with the duty of disclosure as set forth in 37 C.F.R. § 1.56, the accompanying information is being submitted in accordance with 37 C.F.R. §§ 1.97 and 1.98. Applicants request the Examiner's consideration of the document listed on the accompanying Form PTO-1449.

Pursuant to 37 C.F.R. § 1.98, a copy of the document cited is enclosed.

This Statement, Form PTO-1449 and document are being submitted after the issuance of an Official Action on the merits but prior to the final action, therefore under 37 C.F.R. § 1.97(c), the fee set forth in 37 C.F.R. § 1.17(p) is enclosed.

A fee of \$ 180 as set forth in 37 C.F.R. § 1.17(p) is enclosed.

It is respectfully requested that an Examiner-initialed copy of Form PTO-1449 be returned to the undersigned.

SIXTH Information Disclosure Statement Application No. <u>10/551,205</u> Attorney Docket No. <u>0056192-000024</u> Page 2

The Director is hereby authorized to charge any appropriate fees under 37 C.F.R. §§ 1.16, 1.17 and 1.21 that may be required by this paper, and to credit any overpayment, to Deposit Account No. 02-4800.

Respectfully submitted,

**BUCHANAN INGERSOLL & ROONEY PC** 

Date: December 16, 2009

Mary Katherine Baumeister

Registration No. 26254

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Substitute for form 1449/PTO & 1449B/PTO

# SIXTH INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

Complete if Known					
Application Number	10/551,205				
Filing Date	November 14, 2006				
First Named Inventor	Nicholas Bodor et al.				
Examiner Name	JONATHAN S LAU				
Attorney Docket No.	0056192-000024				

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<ul> <li>72) Inventors: LOFTSSON, Thorsteinn;         Reykjavik (IS). MASSON, Mar;         Reykjavik (IS). STBFANSSON         IS-110 Reykjavik (IS).</li> <li>74) Agent: LOFTSSON, Thorsteinn; Uni         of Pharmacy, Hagi at Hofsvallagate</li> </ul>	Fjolnisvegur I I, Einar; Fjard versity of Icelar	, IS-10 laras 1 nd, De <sub>l</sub>	Published  With international search report.  Before the expiration of the time lin  claims and to be republished in the e	
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#### HIGH-ENERGY CYCLODEXTRIN COMPLEXES

#### **BACKGROUND OF THE INVENTION**

#### Field of the Invention:

The invention relates to methods for enhancing the complexation of a heterocyclic drug with cyclodextrin and to methods for enhancing the availability of a heterocyclic drug following administration of a cyclodextrin-drug complex.

#### **Background Art:**

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Cyclodextrins are a group of structurally related saccharides which are formed by enzymatic cyclization of starch by a group of amylases termed glycosyltransferases. Cyclodextrins are cyclic oligosaccharides, consisting of ( $\alpha$ -1,4)-linked  $\alpha$ -D-glucopyranose units, with a somewhat lipophilic central cavity and a hydrophilic outer surface. The most common naturally occurring cyclodextrins are  $\alpha$ -cyclodextrin,  $\beta$ -cyclodextrin and  $\gamma$ -cyclodextrin consisting of 6, 7 and 8 glucopyranose units, respectively. Of these three derivatives,  $\beta$ -cyclodextrin appears to be the most useful pharmaceutical complexing agent due to its cavity size, availability, low cost and other properties.

The natural cyclodextrins, in particular  $\beta$ -cyclodextrin, have limited aqueous solubility and their complex formation with lipophilic drugs often results in precipitation of solid drug-cyclodextrin complexes. Thus, the solubility of  $\beta$ -cyclodextrin in water is only about 18.5 mg/ml at room temperature. This low aqueous solubility is, at least partly, associated with strong intramolecular hydrogen bonding in the cyclodextrin crystal lattice. Substitution of any of the hydrogen bond-forming hydroxyl groups, even by hydrophobic moieties such as methoxy groups, will increase the aqueous solubility of  $\beta$ -cyclodextrin. In addition, since these manipulations frequently produce large numbers of isomeric

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products, chemical modification can transform the crystalline cyclodextrins into amorphous mixtures increasing their aqueous solubility.

Cyclodextrin derivatives of current pharmaceutical interest include the hydroxypropyl derivatives of  $\alpha$ -,  $\beta$ - and  $\gamma$ -cyclodextrin, sulfoalkylether cyclodextrins such as sulfobutylether β-cyclodextrin, alkylated cyclodextrins such as the randomly methylated β-cyclodextrin, and various branched cyclodextrins such as glucosyl- and maltosyl-β-cyclodextrin (T. Loftsson and M.E. Brewster, "Cyclodextrins as pharmaceutical excipients", Pharm. Technol. Eur., 9(5), 26-34 (1997); T. Loftsson and M.E. Brewster, "Pharmaceutical applications of cyclodextrins. I. Drug solubilization and stabilization", J. Pharm. Sci. 85(10), 1017-1025 (1996); R.A. Rajewski and V.J. Stella, "Pharmaceutical applications of cyclodextrins. 2. In vivo drug delivery", J. Pharm. Sci. 85(11), 1142-1169 (1996); T. Irie and K. Uekama, "Pharmaceutical applications of cyclodextrins. 3. Toxicological issues and safety evaluation", J. Pharm. Sci., 86(2), 147-162 (1997); V.J. Stella and R.A. Rajewski, "Cyclodextrins: their future in drug formulation and delivery", Pharm. Res., 14(5), 556-567 (1997); T. Loftsson, "Increasing the cyclodextrin complexation of drugs and drug bioavailability through addition of water-soluble polymers", *Pharmazie*, 53, 733-740 (1998)).

#### Preparation of cyclodextrin inclusion complexes

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In aqueous solutions, cyclodextrins form complexes with many drugs through a process in which the water molecules located in the central cavity are replaced by either the whole drug molecule, or more frequently, by some lipophilic portion of the drug structure. Once included in the cyclodextrin cavity, the drug molecules may be dissociated through complex dilution, by replacement of the included drug by some other suitable molecule (such as dietary lipids or bile salts in the GI tract) or, if the complex is located in close approximation to a lipophilic biological membrane (such as the mucosal membrane of the GI tract),

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the drug may be transferred to the matrix for which it has the highest affinity. Importantly, since no covalent bonds are formed or broken during the drug-cyclodextrin complex formation, the complexes are in dynamic equilibrium with free drug and cyclodextrin molecules (R.A. Rajewski and V.J. Stella, "Pharmaceutical applications of cyclodextrins. 2. In vivo drug delivery", *J. Pharm. Sci.* 85(11), 1142-1169 (1996)).

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Various methods have been applied to the preparation of drug-cyclodextrin complexes (T. Loftsson and M.E. Brewster, "Pharmaceutical applications of cyclodextrins. I. Drug solubilization and stabilization", J. Pharm. Sci. 85(10), 1017-1025 (1996); T. Loftsson and M.E. Brewster, "Cyclodextrins as pharmaceutical excipients", Pharm. Technol. Eur., 9(5), 26-34 (1997)). In solution, the complexes are usually prepared by addition of an excess amount of the drug to an aqueous cyclodextrin solution. The suspension formed is equilibrated (for periods of up to one week at the desired temperature) and then filtered or centrifuged to form a clear drug-cyclodextrin complex solution. Since the rate determining step in complex formation is often the phase to phase transition of the drug molecule, it is sometimes possible to shorten this process by formation of supersaturated solutions through sonication followed by precipitation. For preparation of the solid complexes, the water is removed from the aqueous drug-cyclodextrin solutions by evaporation or sublimation, e.g. spray-drying or freeze-drying. Other methods can also be applied to prepare solid drugcyclodextrin complexes including kneading methods, co-precipitation, neutralization and grinding techniques. In the kneading method, the drug is added to an aqueous slurry of a poorly water-soluble cyclodextrin such as β-cyclodextrin. The mixture is thoroughly mixed, often at elevated temperatures, to yield a paste which is then dried. This technique can frequently be modified so that it can be accomplished in a single step with the aid of commercially available mixers which can be operated at temperatures over 100 °C and under vacuum. The kneading

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method is a cost-effective means for preparing solid cyclodextrin complexes of poorly water-soluble drugs. Co-precipitation of a cyclodextrin complex through addition of organic solvent is also possible. Unfortunately, the organic solvents used as precipitants can interfere with complexation which makes this approach less attractive than the kneading method. However, we have discovered that some organic solvents under some specific conditions, e.g. 10% (v/v) aqueous acetic acid solution, can enhance the complexation. Solid complexes of ionizable drugs can sometimes be prepared by the neutralization method wherein the drug is dissolved in an acidic (for basic drugs) or basic (for acidic drugs) aqueous cyclodextrin solution. The solubility of the drug is then lowered through appropriate pH adjustments (i.e. formation of the unionized drug) to force the complex out of solution. Finally, solid drug-cyclodextrin complexes can be formed by the grinding of a physical mixture of the drug and cyclodextrin and then heating the mixture in a sealed container to 60 to 90 °C.

# 15 Methods for enhancing cyclodextrin complexation

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For a variety of reasons including cost, production capabilities and toxicology, the amounts of cyclodextrin which can be used in most drug formulations is limited (T. Loftsson and M.E. Brewster, "Cyclodextrins as pharmaceutical excipients", *Pharm. Technol. Eur.*, 9(5), 26-34 (1997); T. Loftsson, "Increasing the cyclodextrin complexation of drugs and drug bioavailability through addition of water-soluble polymers", *Pharmazie*, 53, 733-740 (1998)).

If one drug molecule (D) forms a complex with one cyclodextrin molecule (CD), then the complexation efficiency ([D-CD]/[CD]) will be equal to the intrinsic solubility of the drug ( $S_0$ ) times the stability constant of the drug-cyclodextrin complex ( $K_C$ ). In aqueous cyclodextrin solutions saturated with drug, the concentration of free drug ([D]) is approximately equal to  $S_0$ . Thus,

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increased complexation efficiency can be obtained by either increasing  $S_0$  or by increasing K<sub>C</sub> or by increasing both simultaneously. Addition of organic solvents, such as ethanol, to the aqueous complexation media can result in enhanced complexation efficiency through increase in  $S_0$ . Drug ionization can increase the complexation efficiency through increase in S<sub>0</sub>. Addition of certain low molecular weight acids, such as acetic, citric, malic, or tartaric acid, to aqueous complexation media can enhance cyclodextrin solubilization of basic drugs through increase in S<sub>0</sub> (i.e. salt formation, pH changes and lowering melting point) and/or increase in the apparent K<sub>C</sub>. Water-soluble polymers can increase the complexation efficiency through increase in the apparent K<sub>C</sub>. Furthermore, it is often possible to enhance cyclodextrin complexation even further by using several different methods simultaneously to enhance the cyclodextrin complexation. Pharmaceutical applications of these and other methods have been reviewed (See T. Loftsson, "Increasing the cyclodextrin complexation of drugs and drug bioavailability through addition of water-soluble polymers", *Pharmazie*, 53, 733-740 (1988); T. Loftsson and M.E. Brewster, "Cyclodextrins as pharmaceutical excipients", Pharm. Technol. Eur., 9(5), 26-34 (1997); T. Loftsson and M.E. Brewster, "Pharmaceutical applications of cyclodextrins. I. Drug solubilization and stabilization", J. Pharm. Sci. 85(10), 1017-1025 (1996)).

#### 20 Permeability of drugs through biological membranes

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The cyclodextrin molecules are relatively large (molecular weight ranging from almost 1000 to over 1500), with a hydrated outer surface, and under normal conditions, cyclodextrin molecules will only permeate biological membranes with considerable difficulty (R.A. Rajewski and V.J. Stella, "Pharmaceutical applications of cyclodextrins. 2. In vivo drug delivery", *J. Pharm. Sci.* 85(11), 1142-1168 (1996); T. Irie and K. Uekama, "Pharmaceutical applications of cyclodextrins. 3. Toxicological issues and safety evaluation", *J. Pharm. Sci.* 

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86(2), 147-162 (1997); K.-H. Frömming and J. Szeitli. Cyclodextrins in pharmacy, Kluwer Academic Publishers, Dordrecht, The Netherlands, 1994; T. Loftsson and J.H. Ólafsson, "Cyclodextrins: new drug delivery systems in dermatology", Int. J. Dermatol., 37, 241-246 (1998); T. Loftsson and E. 5 Stefánsson, "Effect of cyclodextrins on topical drug delivery to the eye", Drug Dev. Ind. Pharm. 23(5), 473-481 (1997)). It is generally recognized that cyclodextrins act as true carriers by keeping the hydrophobic drug molecules in solution and deliver them to the surface of the biological membrane, e.g. skin, mucosa or the eye cornea, where they partition into the membrane. The relatively lipophilic membrane has low affinity for the hydrophilic cyclodextrin molecules and therefore they remain in the aqueous membrane exterior, e.g. the aqueous vehicle system, salvia or the tear fluid. Conventional penetration enhancers, such as alcohols and fatty acids, disrupt the lipid layers of the biological barrier. Cyclodextrins, on the other hand, act as penetration enhancers by increasing drug availability at the surface of the biological barrier. Furthermore, addition of water-soluble polymer, such as polyvinylpyrrolidone, apparently increase even further the availability of the drug molecules at the surface of the biological membrane resulting in enhanced drug bioavailability (T. Loftsson, "Increasing the cyclodextrin complexation of drugs and drug bioavailability through addition of water-soluble polymers", *Pharmazie*, 53, 733-740 (1998); T. Loftsson, M. Másson and E. Stefánsson, "Cyclodextrins as Permeation enhancers", Proceedings of the 17<sup>th</sup> Pharmaceutical Technology Conference and Exhibition, Volume 2, Dublin, 24-26 March, 1998, pp. 313-324).

#### **OBJECTS AND SUMMARY OF THE INVENTION**

#### 25 Enhancing complexation efficiency

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It is possible to enhance the cyclodextrin (CD) complexation efficacy, or efficiency, of drugs (D), and other "guest" molecules, by either increasing the

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apparent stability constant (K<sub>C</sub>) of the drug-cyclodextrin complex (D-CD) or increasing the apparent intrinsic solubility (S<sub>0</sub>) of the drug. For example, K<sub>C</sub> can be increased by addition of water-soluble polymers to the aqueous complexation media and S<sub>0</sub> can be increased by ionization of the drug molecule, as described previously. However, increased complexation efficiency by itself does not necessarily result in increased drug availability in the aqueous complexation media or increased drug availability from solid drug-cyclodextrin complexes. On the other hand, if the drug-cyclodextrin complexes are prepared under conditions which ensure enhanced complexation and if the complexation efficiency decreases upon administration, then enhanced drug availability will be observed. Thus, the present invention involves: i) enhancement of the complexation efficiency and ii) reduction of the complexation efficiency after administration. For example, it is possible to enhance the complexation efficiency of many ionizable drugs by preparing the complexes at a pH where the drug is ionized but obtain decreased efficiency upon administration due to pH changes and consequent decreased ionization. One example of such a drug is phenytoin (pKa 8.1). Its solubility in water at room temperature (25 °C) is only 18  $\mu$ g/ml at pH 5 and 32  $\mu$ g/ml at pH 8 (P.A. Schwartz, C.T. Rhodes and J.W. Cooper, "Solubility and ionisation characters of phenytoin", J. Pharm. Sci., 66, 994-997 (1977)). Addition of 25% (w/v) 2-hydroxypropyl- $\beta$ -cyclodextrin to the aqueous solutions increases the solubility of phenytoin to 5.0 mg/ml at pH 5 and 6.4 mg/ml at pH 8, which is 280and 200-fold solubility enhancement, respectively. Although the apparent stability constant (K<sub>C</sub>) of the phenytoin-cyclodextrin complex is much larger for the drug in the unionized form than for the anionic form, it is possible to obtain much higher total solubility by increasing the apparent intrinsic solubility  $(S_0)$  of the drug  $(T, S_0)$ Loftsson and N. Bodor, "Effects of 2-hydroxypropyl-β-cyclodextrin on the aqueous solubility of drugs and transdermal delivery of 17β-estradiol", Acta Pharm. Nord., 1, 185-194 (1989)). However, if the pH 8.0 solution was placed in

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an environment which would decrease the pH from 8 to 5 (e.g. topical application to the skin), then a supersaturated solution would be formed which would result in enhanced drug availability (e.g. it would result in enhanced transdermal drug delivery). Other means to enhance  $S_0$  include reversible derivation (e.g. prodrug formation) of the guest molecule and addition of certain low molecular weight acids. The value of K<sub>C</sub> can, for example, be increased by addition of certain low molecular weight acids, by addition of water-soluble polymers to the aqueous complexation media or by using mixed solvent systems such as aqueous 10% (v/v) acetic acid. For example, addition of the polymers and heating in an autoclave (to 120-140°C for 20-40 minutes) does not only increase the complexation but it has also been shown to enhance transdermal and transcorneal drug delivery (T. Loftsson and A.M. Sigurdardottir, "Cyclodextrins as skin penetration enhancers", in J. Szejtli and L. Szente (Eds.) Proceedings of the Eighth International Symposium on Cyclodextrins, Kluwer Academic Publishers, 1996, pp. 403-406; T. Loftsson and E. Stefansson, "Effect of cyclodextrins on topical drug delivery to the eye", Drug Devel. Ind. Pharm., 23(5), 473-481 (1997)). As shown in Table 1 below, it is not enough to add the polymers to the complexation medium. Addition of polymers to the unheated vehicles did not enhance the transdermal delivery of enalaprilat. However, heating the vehicles after addition of the polymers resulted in significant enhancement. The effect of the polymers on the transdermal delivery of enalaprilat can, at least partly, be explained by decreased complexation efficiency (i.e. decrease in  $K_C$ ) at the skin surface.

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Table 1. The effect of heating on transdermal delivery of enalaprilat from 10% (w/v) HP $\beta$ CD solutions at pH 5.0 containing 2.5% enalaprilat in a suspension. The concentration of dissolved enalaprilat was between 2.0 and 2.3% (w/v).

Donor phase	Flux (mg h <sup>-1</sup> cm	Ratio	
(w/v per cent)	Un-heated	Heated	
нрвср	18±2	. <del>प</del>	-
HPβCD, 0.25% PVP	16±6	$23\pm7$	1.4
HPβCD, 0.10% HPMC	14±3	$37\pm12$	2.6

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In one aspect of the present invention there is provided a method for enhancing the complexation efficacy, i.e. efficiency, of a drug with cyclodextrin, said drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said drug to chemically reversible ring-opening so that at least a portion (at least 0.1% by weight) thereof is in ring-opened form, and complexing said drug with cyclodextrin.

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In a related aspect of the invention, there is provided a method for enhancing the complexation efficiency of a drug with cyclodextrin, said drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal

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or hemiketal, said method comprising complexing said drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least a portion (at least 0.1% by weight) of said drug.

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In another aspect of the invention, there is provided a method for enhancing the availability of a drug following administration of a cyclodextrindrug complex to a warm-blooded animal in need of same, said drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imine, hemiacetal or hemiketal, said method comprising complexing said drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least a portion (at least 0.1% by weight) of said drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to said animal under conditions which reduce the complexation efficiency.

In still another aspect, the present invention provides a method for enhancing the availability of a basic drug (i.e. a proton acceptor) following administration of a cyclodextrin-drug complex to a warm-blooded animal in need of same, said basic drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said basic drug to complexation in an aqueous medium at a pH level below the pKa+2 value of said basic drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to said animal under conditions which reduce the complexation efficiency.

In yet another aspect, the present invention provides a method for enhancing the availability of an acidic drug following administration of a cyclodextrin-drug complex to a warm-blooded animal in need of same, said acidic drug having a structure comprising at least one heterocyclic ring having a total of 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said acidic drug to complexation in an aqueous medium at a pH level above the pKa-2 value of said acidic drug to enhance the complexation efficiency, followed by administering the cyclodextrindrug complex thus obtained to said animal under conditions which reduce the complexation efficiency.

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#### BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a graph illustrating the effect of pH on the phase-solubility of phenytoin (pKa 8.1) in aqueous hydroxypropyl- $\beta$ -cyclodextrin (HP $\beta$ CD) solutions at 25°C at pH 10.19 ( $\spadesuit$ ); pH 7.55 ( $\blacksquare$ ) and pH 2.74 ( $\spadesuit$ );

Fig. 2 is a graph illustrating the effect of pH on the solubility of alprazolam (pKa 2.4) in aqueous 10% (w/v) HP $\beta$ CD solutions at room temperature;

Fig. 3 is a graph illustrating the effect of pH (i.e. the diazepine ring-opening) on the solubility of midazolam (pKa 6.2) in pure aqueous buffer solutions ( $\bullet$ ), aqueous buffer solutions containing 10% (w/v) HP $\beta$ CD ( $\blacksquare$ ) and aqueous buffer solutions containing both 10% (w/v) HP $\beta$ CD and 0.10% (w/v) hydroxypropyl methylcellulose (HPMC) ( $\bullet$ ) at room temperature;

Fig. 4 is a graph illustrating the effects of cyclodextrins, pH and 10% (v/v) acetic acid on the solubility of midazolam in aqueous solutions: pure aqueous buffer solution (♠); aqueous 10% (v/v) acetic acid solution (♠); 10% w/v HPβCD solution containing 0.10% (w/v) HPMC in aqueous 10% (v/v) acetic acid solution

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( $\blacksquare$ ); 10% (w/v) aqueous sulfobutyl ether- $\beta$ -cyclodextrin (SBE $\beta$ CD) solution in aqueous 10% (v/v) acetic acid solution ( $\spadesuit$ ); and

Fig. 5 is a graph plotting the concentration in ng/ml of midazolam in serum after intravenous administration of 2 mg of a commercial intravenous formulation of midazolam ( $\bigcirc$ ) and nasal administration of 4.8 mg of a nasal formulation of midazolam prepared in accord with the present invention ( $\triangle$ ), against time in minutes, where each point represents the mean value and error bars represent standard deviation.

# **DETAILED DESCRIPTION OF THE INVENTION**

The following table (**Table 2**) lists some of the currently available cyclodextrins contemplated for use in the present invention.

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Table 2. Some of the currently available cyclodextrins obtained by substitution of the OH-groups located on the edge of the cyclodextrin ring. Since both the number of substituents and their location will affect the physicochemical properties of the cyclodextrin molecules, such as their aqueous solubility and complexing abilities, each derivative listed should be regarded as a group of closely related cyclodextrin derivatives.

Туре	α-Cyclodextrin derivatives	β-Cyclodextrin derivatives	γ-Cyclodextrin derivatives
Alkylated:			
	Methyl	Methyl Ethyl	Methyl
	Butyl	Butyl	Butyl Pentyl
Hydroxylalkylated:			•
	2-Hydroxypropyl	Hydroxyethyl 2-Hydroxypropyl 2-Hydroxybutyl	Hydroxyethyl 2-Hydroxypropyl
Esterified:		• • •	
	Acetyl	Acetyl Propionyl Butyryl	Acetyl
	Succinyl	Succinyl Benzoyl Palmityl	Succinyl
		Toluenesulfonyl	
Esterified and alkyla	ited:		
		Acetyl methyl Acetyl butyl	
Branched:			
	Glucosyl	Glucosyl	Glucosyl
	Maltosyl	Maltosyl	Maltosyl
Ionic:	O-1	O- 1	. (C-ul
	Carboxymethyl ether	Carboxymethyl ether Carboxymethyl ethyl	
	Phosphate ester	Phosphate ester 3-Trimethylammoniu 2-hydroxypropyl ethe	
		Sulfobutyl ether	
Polymerized:	Ciman la materia	Cimula malrina	Cimple nel-
	Simple polymers Carboxymethyl	Simple polymers Carboxymethyl	Simple polymers Carboxymethyl

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Particularly preferred cyclodextrins for use herein are hydroxypropyl- $\beta$ -cyclodextrin,  $\beta$ -cyclodextrin sulfobutyl ether, the branched  $\beta$ -cyclodextrins (especially glucosyl- $\beta$ -cyclodextrin and maltosyl- $\beta$ -cyclodextrin, hydroxypropyl- $\gamma$ -cyclodextrin and  $\gamma$ -cyclodextrin.

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In preferred aspects of the present invention, the drug for use herein is one having a structure comprising at least one heterocyclic ring. The heterocyclic ring generally has a total of 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms. While each hetero ring atom can be nitrogen, oxygen or sulfur, heterocycles having at least one nitrogen or oxygen ring atom are preferred. Preferably, the drug has at least one heterocyclic ring which is a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal.

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Especially desirable drugs for use in accord with the present invention are benzodiazepines. Benzodiazepines contain a benzene ring fused with a diazepine ring which is a 7-membered ring with nitrogen atoms in positions 1 and 4. By way of example, the chemical name of alprazolam is 8-chloro-1-methyl-6-phenyl-4H-s-triazolo[4,3-a][1,4]benzodiazepine, the chemical name of midazolam is 8chloro-6-(2-fluorophenyl)-1-methyl-4H-imidazo[1,5-a][1,4]benzodiazepine and that of triazolam is 8-chloro-6-(o-chlorophenyl)-1-methyl-4H-s-triazolo[4,3al[1,4]benzodiazepine. Thus, all of these compounds have the 1,4-benzodiazepine structure with a double bond between nitrogen atom number 4 and carbon atom number 5 (which gives the molecule a cyclic imine structure). The benzodiazepines are cyclic imines. They are all basic, i.e. they are proton acceptors. Preferred benzodiazepines for use herein are alprazolam, brotizolam, chlordiazepoxide, clobazam, clonazepam, clorazepam, demoxazepam, flumazenil, flurazepam, halazepam, midazolam, nordazepam, medazepam, diazepam, nitrazepam, oxazepam, midazepam, lorazepam, prazepam, quazepam, triazolam, temazepam and lorazolam. Especially preferred are midazolam, alprazolam, clonazepam, lorazepam and triazolam.

Another group of preferred drugs for use herein consists of the barbituric acid derivatives. The barbituric acids contain a 2,4,6-trihydroxypyrimidine (also called 2,4,6-trioxohexahydropyrimidine) ring in their structure, a 6-member ring with nitrogen in positions 1 and 3. Thus, the chemical name of barbital is 5,5-diethyl-2,4,6(1H,3H,5H)-pyrimidinetrione and that of phenobarbital is 5-ethyl-5-phenyl-2,4,6(1H,3H,5H)-pyrimidinetrione. The barbituric acids can be characterized as cyclic amides or lactams (cyclic amides are called lactams) or imides (which are nitrogen analogues of cyclic anhydrides). Barbituric acids are weak acids. Preferred barbituric acid derivatives are barbital, butobarbital, amobarbital, phenobarbital, aprobarbital, secobarbital, crotylbarbital, cyclobarbital, phenobarbital, hexobarbital, methylphenobarbital, thiopental, isopropylbromallylbarbituric acid, cyclohexenylallylthiobarbituric acid and their salts. Thiopental is 5-ethyldihydro-5-(1-methylbutyl)-2-thioxo-4,6(1H,5H)-pyrimidinedione, i.e. one = O moiety in the barbituric acid structure has been replaced by = S.

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Yet another group of preferred drugs for use in the present invention consists of the hydantoins. Hydantoins are, like barbituric acids, cyclic urea derivatives. The ring-opened acyl derivatives of hydantoins and barbituric acids are sometimes called ureides. Both hydantoins and barbituric acids can form urea upon hydrolysis. Hydantoins contain a 2,4-imidazolidinedione ring in their structure, a 5-membered ring with nitrogen in positions 1 and 3. The chemical name of, for example, phenytoin, is 5,5-diphenyl-2,4-imidazolidinedione. Hydantoins are closely related to barbituric acids and are acids like them.

Still another group of preferred drugs for use in the present invention consists of pyrazole derivatives. The expression "pyrazole derivatives" as used herein includes drugs containing a pyrazole ring, 3-pyrazoline ring or pyrazolidine ring in their structure, all of which are 5-membered rings with nitrogens in positions 1 and 2. These compounds are either basic or acidic. Preferred pyrazole

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derivatives for use herein include phenazone, phenylphenazone, metamidazole, phenylbutazone, oxyphenbutazone and sulfinpyrazone.

Yet another group of drugs preferred for use herein consists of imidazole derivatives. The expression "imidazole derivatives" as used herein includes drugs containing an imidazole, imidazoline or imidazolidine ring in their structure. These are 5-membered rings with nitrogen atoms in positions 1 and 3. These compounds are either basic or acidic. Preferred compounds of this type include histamine, miconazole, pilocarpine, naphazoline and clonidine.

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Another group of preferred drugs for use in this invention are pyrimidine derivatives. These drugs contain a 6-membered ring with nitrogen atoms in positions 1 and 3. These derivatives are usually basic. Preferred pyrimidine derivatives include thiamine, trimethoprim, orotic acid, methylthiouracyl and prothiouracyl.

Still another group of preferred drugs for use herein are purine derivatives, which contain purine, that is, imidazo(4,5-d)pyrimidine, in their structures. These drugs are frequently basic but some of them are acidic. Preferred purine derivatives include caffeine, theophylline, etophylline, proxyphylline and theobromine.

Cyclic drugs having heterocyclic rings characterized as enamines, lactones, lactams, thiolactams, anhydrides, imides, imines, hemiacetals and hemiketals are thus appropriate for use in preferred embodiments of the invention, in which ring opening of the heterocyclic ring takes place.

In various aspects of the present invention, the drug is subjected to chemically reversible ring-opening so that at least a portion thereof is in ring-opened form. The portion in ring-opened form is at least 0.1% by weight, preferably at least 1 or 2% by weight, more preferably at least 5% by weight of said drug. In aqueous formulations, the amount of drug in ring-opened form is frequently from about 5 to about 10% by weight and usually no more than about

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50%. In solid formulations, the portion of drug in ring-opened form can generally be much higher, frequently about 50% or more, and sometimes even about 90 to 95%.

When the method of the invention comprises complexing the drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least a portion (at least 0.1% by weight) of the drug, the complexation is advantageously conducted at a pH level which affords ring-opening of at least 5% by weight of said drug. Preferably the complexation is conducted at a pH level of below about 5.

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In one preferred embodiment, the drug is a basic drug, especially a benzodiazepine, and the complexation is conducted at a pH level of below about 5. It is also preferred that the cyclodextrin is hydroxypropyl-β-cyclodextrin, βcyclodextrin sulfobutyl ether, a branched  $\beta$ -cyclodextrin (especially glucosyl  $\beta$ cyclodextrin or maltosyl-β-cyclodextrin), β-cyclodextrin, hydroxypropyl-γcyclodextrin or  $\gamma$ -cyclodextrin. It is also preferred that the benzodiazepine is alprazolam, brotizolam, chlordiazepoxide, clobazam, clonazepam, clorazepam, demoxazepam, flumazenil, flurazepam, halazepam, midazolam, nordazepam, medazepam, diazepam, nitrazepam, oxazepam, midazepam, lorazepam, prazepam, quazepam, triazolam, temazepam or loprazolam; and that the cyclodextrin-drug complex thus obtained be formulated as a nasal spray, sublingual tablet or parenteral solution, especially when formulated suitable for use in producing a sedative, anti-anxiety, anticonvulsant or muscle relaxant effect, most especially as a pre-anaesthetic medication, or to supplement anaesthesia, to induce and/or maintain anaesthesia or to induce a hypnotic effect. In especially preferred embodiments, the benzodiazepine is midazolam, alprazolam, clonazepam, lorazepam or triazolam; the cyclodextrin is hydroxypropyl-β-cyclodextrin, βcyclodextrin sulfobutyl ether, a branched β-cyclodextrin (especially glucosyl βcyclodextrin or maltosyl β-cyclodextrin), β-cyclodextrin, hydroxypropyl-γ-

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cyclodextrin or  $\gamma$ -cyclodextrin; and the complexation is conducted at a pH level below about 5, preferably between about 3 and about 5.

In another embodiment of the present method utilizing chemically reversible ring-opening described above, the drug is an acidic drug.

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In yet another embodiment of the present method utilizing chemically reversible ring-opening described above, the drug is a barbituric acid derivative, a hydantoin, a pyrazole derivative, an imidazole derivative, a pyrimidine derivative or a purine derivative. When the drug is a barbituric acid derivative, it is preferably barbital, butobarbital, amobarbital, phenobarbital, aprobarbital, secobarbital, crotylbarbital, cyclobarbital, phenobarbital, hexobarbital, methylphenobarbital, thiopental, isopropylbromallylbarbituric acid, or cyclohexenylallylthiobarituric acid, or a salt thereof. When the drug is a hydantoin, it is preferably phenytoin. When the drug is a pyrazole derivative, it is preferably phenazone, propylphenazone, metamidazole, phenylbutazone, oxyphenbutazone or sulfinpyrazone. When the drug is an imidazole derivative, it is preferably histamine, miconazole, pilocarpine, naphazoline or clonidine. When the drug is a pyrimidine derivative, it is preferably thiamine, trimethoprim, orotic acid, methylthiouracyl or prothiouracyl. When the drug is a purine derivative, it

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When the present invention comprises complexing the drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least a portion (at least 0.1% by weight) of the drug to enhance the complexation efficacy, followed by administering the cyclodextrin-drug complex thus obtained to said animal under conditions which reduce the complexation efficacy, the complexation is generally conducted at a pH level which affords ring-opening of at least 5% by weight of the drug. Preferably, the complexation is conducted at a pH level of below about 5, especially between about 3 and about 5. The cyclodextrin is preferably hydroxypropyl-β-

is preferably caffeine, theophylline, etophylline, proxyphylline or theobromine.

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cyclodextrin, β-cyclodextrin sulfobutyl ether, a branched β-cyclodextrin (especially glucosyl-β-cyclodextrin or maltosyl-β-cyclodextrin), β-cyclodextrin, hydroxypropyl-γ-cyclodextrin or γ-cyclodextrin. The drug is preferably a benzodiazepine, especially midazolam, alprazolam, clonazepam, lorazepam or triazolam. The cyclodextrin-drug complex is preferably administered in the form of an aqueous solution or a hydrogel, particularly as a nasal spray or nasal drops, or as a parenteral solution. As a nasal spray of a benzodiazepine, the aqueous solution is advantageously brought to a pH level of below about 6, preferably below about 4.7, most especially to a pH between about 3 and about 4.7. When administered as a solid, the cyclodextrin-drug complex is preferably formulated as a tablet for oral, buccal or sublingual administration. The water may be removed from the aqueous complexation medium after formation of the cyclodextrin-drug complex.

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When the present invention comprises subjecting a basic drug to complexation in an aqueous medium at a pH level below the pKa+2 value of said basic drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to an animal under conditions which reduce the complexation efficiency, the basic drug is preferably a benzodiazepine. Benzodiazepines of particular interest are alprazolam, brotizolam, chlordiazepoxide, clobazam, clonazepam, clorazepam, demoxazepam, flumazenil, flurazepam, halazepam, midazolam, nordazepam, medazepam, diazepam, nitrazepam, oxazepam, midazolam, lorazepam, prazepam, quazepam, triazolam, temazepam and loprazolam. Particularly preferred benzodiazepines are alprazolam, midazolam, clonazepam, lorazepam and triazolam. The cyclodextrin-benzodiazepine complex obtained in the complexation step is preferably formulated as a nasal spray, sublingual tablet or parenteral solution, which is preferably administered in an effective sedative, anti-anxiety, anticonvulsant or muscle relaxant amount, particularly as a pre-anaesthetic

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medication, or to supplement anaesthesia, to induce and/or maintain anaesthesia or to induce a hypnotic effect. In this general aspect of the invention, the pH level of the aqueous complexation medium is advantageously selected so that it also affords ring-opening of at least 5% by weight of the drug. For the benzodiazepines, the complexation is preferably conducted at a pH level of below about 5, most preferably between about 3 and about 5. Also in this general aspect of the invention, in one preferred embodiment, the complexation is carried out in the presence of from about 0.001 to about 5% (weight/volume) of a pharmacologically inactive, pharmaceutically acceptable water-soluble polymer at a temperature of from about 30°C to about 150°C. Preferably, the polymer is a cellulose derivative or a polyvinyl polymer; more preferably, the polymer is methylcellulose, hydroxyethyl cellulose, hydroxypropyl cellulose, hydroxypropyl methylcellulose, hydroxyethyl methylcellulose, hydroxypropyl ethylcellulose, hydroxyethyl ethylcellulose, sodium carboxymethylcellulose or polyvinylpyrrolidone. An especially preferred cellulose derivative is hydroxypropyl methylcellulose. A method for enhancing drug-cyclodextrin complexation utilizing a pharmacologically inactive water-soluble polymer is described in Loftsson United States Patents No. 5,324,718 and No. 5,472,954. In another preferred embodiment of this general aspect of the invention, the complexation is also carried out in the presence of acetic acid and/or one or more pharmaceutically acceptable salts of acetic acid, the acetate-water ratio of the aqueous complexation medium being from about 1:1000 to about 2:1, preferably from about 1:100 to about 1:1, more preferably from about 1:20 to about 1:4. Preferably, the drug is midazolam and the cyclodextrin is hydroxypropyl-βcyclodextrin,  $\beta$ -cyclodextrin sulfobutyl ether, a branched  $\beta$ -cyclodextrin (especially glucosyl- $\beta$ -cyclodextrin or maltosyl- $\beta$ -cyclodextrin),  $\beta$ -cyclodextrin, hydroxypropyl- $\gamma$ -cyclodextrin or  $\gamma$ -cyclodextrin.

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When the present invention comprises subjecting an acidic drug to complexation in an aqueous medium at a pH level above the pKa-2 value of said acidic drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to an animal under conditions which reduce the complexation efficiency, preferably the pH level of the aqueous complexation medium is selected such that it also affords ring-opening of at least 5% by weight of said drug.

In order to further illustrate the present invention and the advantages thereof, the following specific examples are given, it being understood that same are intended only as illustrative and in no way limitative of the invention.

# Example 1

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Phenytoin (5,5-diphenylhydantoin) is a water-insoluble weak acid (pKa 8.1) which forms a somewhat water-soluble anion in alkaline solution. Solubility (S) of phenytoin at three different pH levels was determined in aqueous solutions containing various amounts of 2-hydroxypropyl-β-cyclodextrin (HPβCD) of molar substitution (MS) = 0.9, i.e. (a) pH  $2.74\pm0.18$  (SD), (b) pH  $7.55\pm0.12$ , and (c) pH 10.19±0.14. Excess amount of the drug was added to the aqueous HPβCD solution and the suspension formed sonicated for one hour at room temperature (23°C). After equilibration at 25°C in a water-bath for three days, the suspension was filtered through a 0.45  $\mu$ m membrane filter, diluted with aqueous methanolic solution and the amount of dissolved phenytoin determined by a high pressure liquid chromatographic method (HPLC). FIG. 1 illustrates the effect of pH on the phase-solubility of phenytoin (pKa 8.1) in aqueous HPβCD solutions at 25°C. The results set forth in FIG. 1 show significant enhancement in the HP $\beta$ CD solubilization (i.e. the efficiency of the complexation) of the drug at pH 10.19 (•) where the drug is mainly in the ionized form. Formation of phenytoin-HPβCD complexes at pH 10.19 can result in enhanced bioavailability of phenytoin. For

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example, topical application of such a solution to the skin will result in lowering of pH, which will decrease the efficiency of the complexation, which again will result in enhanced permeability of phenytoin into and through the skin. Also, formation of phenytoin-HP $\beta$ CD complexes at pH of about 10 (e.g. in aqueous ammonia solutions) and lyophilization of the complex will result in phenytoin-HP $\beta$ CD complex powder which can, for example, be formulated into tablets. The bioavailability of phenytoin from such tablets will be enhanced compared to the phenytoin availability from tablets containing phenytoin-HP $\beta$ CD complex prepared at lower pH, e.g. at pH 2.7 ( $\bullet$ ) or 7.6 ( $\blacksquare$ ).

### 10 Example 2

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Alprazolam is a water-insoluble weak base (pKa 2.41) which forms a somewhat water-soluble cation in acidic solution. Solubility (S) of alprazolam at several different pH levels was determined in aqueous solutions containing 10% (w/v) 2-hydroxypropyl- $\beta$ -cyclodextrin (HP $\beta$ CD) of molar substitution (MS) = 0.3. Excess amount of the drug was added to the aqueous HPBCD solution and the suspension formed heated in a sealed container in an autoclave (120-140°C for 20-40 minutes). After equilibration at room temperature (22-23°C) for seven days, the suspension was filtered through a 0.45 µm membrane filter, diluted with aqueous methanolic solution and the amount of dissolved alprazolam determined by a high pressure liquid chromatographic method (HPLC). FIG. 2 illustrates the effect of pH on the solubility of alprazolam (pKa 2.4) in aqueous 10% (w/v) HP $\beta$ CD solutions at room temperature. The results set forth in FIG. 2 show significant enhancement in the HPBCD solubilization (i.e. the efficiency of the complexation) of the drug at a pH at which the drug is mainly in the ionized form. The sharp increase in the solubility can, however, only partly be explained by the ionization of the alprazolam molecule.

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# Example 3

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Several drugs which have a nitrogen-containing heterocycle in their structure are known to undergo reversible ring-opening which frequently is pH dependent. For example, barbituric acids undergo reversible ring cleavage (H.J. Roth, K. Eger and R. Troschütz, *Pharmaceutical Chemistry*. *Volume 2. Drug Analysis*. Ellis Horwood, 1991, pp. 308-309):

HO NH<sub>2</sub> +H<sub>2</sub>O 
$$R_1$$
 NH  $H_2$ O  $H_2$ N OH  $H_$ 

Another example of such reversible ring-opening is the opening of cyclic imines through formation of an aldehyde or ketone and a primary amine:

Another example of such reversible ring-opening is the opening of cyclic imines through formation of an aldehyde or ketone and a primary amine:

An example of such structure is the 1H-1,4-diazepine ring which, for example, is an essential structure of the benzodiazepine derivatives. These structural changes are pH-dependent and reversible, and it is known that the open form frequently coexists with the closed one in several commercial products. One example is the iv solution of midazolam (Dormicum<sup>TM</sup> from F. Hoffmann-LaRoche & Ltd,

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Switzerland) where the drug is partly in the open form (M. Gerecke, "Chemical structure and properties of midazolam compared with other benzodiazepines", *Br. J. Clin. Pharmac.*, 11S-16S (1983)). However, the open form of midazolam is rapidly converted to the closed one upon *iv* administration.

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We have determined the effect of pH and cyclodextrins, i.e. HP $\beta$ CD MS 0.3, sulfobutylether  $\beta$ -cyclodextrin (SBE $\beta$ CD) with degree of substitution (DS) = 6.4,  $\alpha$ -cyclodextrin ( $\alpha$ CD) and  $\gamma$ -cyclodextrin ( $\gamma$ CD) on the ring-opening of several benzodiazepines. The cyclodextrin concentration was 10% (w/v) and the benzodiazepine concentration was 1x10<sup>-4</sup> M. The concentration of the closed form was determined immediately after dissolving the benzodiazepine in the aqueous cyclodextrin solution and again 24 hours later (i.e. after equilibration at 23°C). Preliminary experiments had shown that equilibrium between the closed and the open form was attained within 3 hours at 23°C.

It is clear from the results displayed in **Table 3** below that a large fraction of the benzodiazepines (over 50% at pH below 2) are in the open form at low pH and that the fraction of open form frequently increases upon addition of cyclodextrin to the aqueous solution. For example, at pH 3 about 60% of alprazolam in aqueous HP $\beta$ CD solution is in the open form. This will increase the apparent intrinsic solubility (S $_0$ ). This increase in S $_0$  will result in enhanced complexation efficiency. The observed increase in the complexation efficiency will result in enhanced cyclodextrin solubilization of the benzodiazepines in aqueous solutions.

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**Table 3**. The effect of pH and cyclodextrins on the fraction of the open form of several benzodiazepines at room temperature (approx. 23 °C).

Benzodiazepine	Cyclodextrin	pН	Fraction open
Alprazolam (pKa 2.4)	None	2	0.82
		3	0.56
		4	0.33
	HPβCD	2	0.89
		3	0.60
		4	0.23
	SBEβCD	2	0.96
		3	0.84
		. 4	0.33
	$\alpha CD$	2	0.94
		3	0.79
		4	0.25
	γCD	2	0.81
		2 3	0.41
		4	0.42
Diazepam (pKa 3.3)	None	2	0.30
, ,		3	0.23
		. 4	0.15
	HPβCD	2	0.65
	•	3	0.29
		4	0.15
	SBEβCD	2	0.63
	•	3	0.56
		4	0.22
	αCD	2	0.67
		3	0.51
		4	0.13
	γCD	2	0.41
	•	3	0.17
		4	0.13

Table cont. on next page.

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Benzodiazepine	Cyclodextrin	pН	Fraction open
Midazolam (pKa 6.2)	None	2 3	0.74 0.28
	нрвср	4 2 3 4	0.18 0.56 0.18 0.23
	SBEβCD	2 3 4	0.81 0.39 0.11
	αCD	. 2 3 4	0.79 0.32 0.10
	γCD	2 3 4	0.61 0.21 0.17
Triazolam (pKa between 2 and 3)	None	2 3 4	0.53 0.08 0.00
	нрвсо	2 3 4	0.51 0.09 0.00
	SBEβCD	2 3 4	0.71 0.25 0.00
	αCD	2 3 4	0.75 0.23 0.00
	γCD	2 3 4	0.33 0.01 0.00

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# Example 4

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Midazolam is a water-insoluble weak base (pKa 6.2) which forms a somewhat water-soluble cation in acidic solution. Solubility (S) of midazolam at several different pH levels was determined in: a) pure aqueous buffer solutions (i.e. without HPβCD and HPMC); b) aqueous buffer solutions containing 10% (w/v) 2-hydroxypropyl- $\beta$ -cyclodextrin (HP $\beta$ CD) of molar substitution (MS) = 0.3; and c) aqueous solutions containing 10% (w/v) 2-hydroxypropyl- $\beta$ cyclodextrin (HP $\beta$ CD) of molar substitution (MS) = 0.3 and 0.10% (w/v) hydroxypropyl methylcellulose (HPMC) 4000. Excess amount of the drug was added to the aqueous solution and the suspension formed was heated in a sealed container in an autoclave (120-140°C for 20-40 minutes). After equilibration at room temperature (22-23°C) for seven days, the suspension was filtered through a  $0.45 \mu m$  membrane filter, diluted with aqueous methanolic solution and the amount of dissolved midazolam determined by a high pressure liquid chromatographic method (HPLC). FIG. 3 illustrates the effect of pH (i.e. the ring-opening) on the solubility of midazolam (pKa 6.2) in pure aqueous buffer solutions (●), aqueous buffer solutions containing 10% (w/v) HPBCD (■), and aqueous buffer solutions containing both 10% (w/v) HPβCD and 0.10% (w/v) HPMC (♦) at room temperature. The results set forth in FIG. 3 show significant enhancement in the HPβCD solubilization (i.e. the efficiency of the complexation) of the drug at pH levels where the drug exists partly in the open form. Addition of HPMC significantly improves the efficiency.

### Example 5

Solubility (S) of midazolam at several different pH levels was determined in: a) pure aqueous buffer solutions (i.e. without cyclodextrin, polymer or acetic acid); b) aqueous buffer solutions containing 10% (v/v) acetic acid as a co-solvent; c) aqueous buffer solutions containing 10% (w/v) sulfobutylether β-cyclodextrin

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(SBEβCD) and 10% (v/v) acetic acid as a co-solvent; and d) aqueous buffer solutions containing 10% (w/v) 2-hydroxypropyl-β-cyclodextrin (HPβCD), 0.10% (w/v) hydroxypropyl methylcellulose (HPMC) and 10% (v/v) acetic acid as a cosolvent. Excess amount of the drug was added to the aqueous HPβCD solution and the suspension formed was heated in a sealed container in an autoclave (120-140°C for 20-40 minutes). After equilibration at room temperature (22-23°C) for seven days, the suspension was filtered through a 0.45 µm membrane filter, diluted with aqueous methanolic solution and the amount of dissolved midazolam determined by a high pressure liquid chromatographic method (HPLC). FIG. 4 illustrates the effects of cyclodextrins, pH and 10% (v/v) acetic acid on the solubility of midazolam in aqueous solutions: pure aqueous buffer solution ( $\triangle$ ); aqueous 10% (v/v) acetate solution (•); 10% (w/v) HPβCD solution containing 0.10% (w/v) HPMC in aqueous 10% (v/v) acetic acid solution (■); 10% (w/v) aqueous SBE $\beta$ CD solution in aqueous 10% (v/v) acetate ( $\spadesuit$ ). The results set forth in FIG. 4 show that addition of 10% (v/v) acetic acid significantly improves the complexation. Addition of the acetic acid increases the value of S<sub>0</sub> without having any significant effect on the value of K<sub>C</sub>, which significantly improves the complexation efficiency and, consequently, enhances the cyclodextrin solubilization of the drug. Midazolam carries a positive charge at acidic pH and, thus, the negatively charged SBEβCD forms a more stable complex than the uncharged HPBCD with midazolam at these conditions. Addition of 10% (v/v) acetic acid as a co-solvent resulted in a small decrease in the fraction of the open ring form of the drug.

#### Example 6

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Female hairless mice were sacrificed by cervical dislocation and their full-thickness skins removed. The outer surface of the skin was rinsed with 35% (v/v) methanol in water and subsequently with distilled water to remove any

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contamination. The skin was placed in Franz diffusion cells. The receptor phase consisted of phosphate buffer saline pH 7.4. The skin diffusion cells were stirred with a magnetic bar and kept at 37°C by circulating water through an external jacket. The donor phase (2.0 ml) consisted of a solution of the drug in aqueous 7% (w/v) SBE $\beta$ CD solution pH 3.3, or aqueous cyclodextrin solution where the pH had been raised from 3.3 to 4.1 (by addition of NaOH) before it was applied to the skin. The alprazolam concentration in the donor phase was 1.85 mg/ml at pH 3.3. Samples (200  $\mu$ l) of receptor phase were removed from the cells at various time intervals up to 48 hours and replaced with a fresh buffer solution. The samples were kept frozen until analyzed by HPLC. The flux was calculated from the linear part of each permeability profile and the permeability coefficient obtained by dividing the flux with the concentration of dissolved drug in the donor phase. The results set forth in Table 4 show clearly that raising the pH from 3.3 to 4.1 increases the flux though biological membranes such as hairless mouse skin.

Table 4. The flux of alprazolam through hairless mouse skin. The donor phase consisted of aqueous pH 3.3 buffer solution containing 7% (w/v) SBEβCD saturated with the drug. In one case the pH of the donor phase was kept constant at pH 3.3, but in the other case the pH was raised to 4.1 (by addition of NaOH) before it was applied to the skin. The alprazolam concentration in the donor phase was 1.85 mg/ml at pH 3.3.

Donor phase	Flux (mg/cm <sup>2</sup> /h)	Ratio	
Without increasing the pH	3.91 x 10 <sup>-4</sup>	1.0	
Increasing the pH from 3.3 to 4.1	4.56 x 10 <sup>-4</sup>	1.2	

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#### Example 7

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The effect of cyclodextrins and organic solvents on the rate of diazepine ring-closure of several selected benzodiazepines was investigated. Stock solutions containing 1.0x10<sup>-3</sup> M of the benzodiazepine in 0.10 M aqueous hydrochloric acid solution (pH approx. 1.1) were prepared and stored at 37.0°C. The benzodiazepines were in the ring-open form in these stock solutions. Aqueous 0.50 M tris buffer (pH 7.50) solution was prepared. The observed first-order rate constant for the closing (i.e. formation) of the benzodiazepine ring was determined in the following reaction media: a) pure aqueous tris buffer solution; b) aqueous tris buffer solution containing 10% (w/v) cyclodextrin; c) tris buffer solution containing 10% (w/v) cyclodextrin and 10% (v/v) ethanol (EtOH); d) tris buffer solution containing 10% (w/v) cyclodextrin and 50% (v/v) EtOH; e) tris buffer solution containing 10% (w/v) cyclodextrin and 10% (v/v) dimethylsulfoxide (DMSO); and f) tris buffer solution containing 10% (w/v) cyclodextrin and 50% (w/v) DMSO. The stock solution (30  $\mu$ l) was added to 1.50 ml of the reaction media which had previously been equilibrated to 37.0°C and the first-order rate constant for the appearance of the closed form determined from the appearance of the closed form as observed on HPLC. Tables 5, 6 and 7 show the effects of cyclodextrins, EtOH and DMSO on the observed first-order rate constant for the regeneration of alprazolam, triazolam and midazolam, respectively. In pure aqueous buffer solutions, addition of EtOH and DMSO decreases somewhat the rate of ring closure, at least in the case of alprazolam and midazolam. Addition of cyclodextrin or the organic solvents have insignificant effect on the pH under these conditions. The dielectric constant of the reaction medium will, however, decrease upon addition of the organic solvents. It is possible that this decrease in the dielectric constant will reduce the ability of the reaction media to stabilize the transition state which could explain the decrease in the observed rate constant. Addition of cyclodextrin decreased significantly, in all cases, the rate of ring

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closure. The cyclodextrins formed stable complexes with the ring-open form of the drug and, thus, the rate decreased upon addition of cyclodextrin. Addition of EtOH or DMSO to the cyclodextrin-containing reaction media resulted in increase in the rate, compared to reaction media containing only cyclodextrin, which could be due to decreased complexation of the diazepine ring-open form. EtOH and DMSO will compete with the diazepine ring-open form for a space in the cyclodextrin cavity resulting in decreased complexation.

Table 5. The effect of cyclodextrins, ethanol (EtOH) and dimethylsulfoxide (DMSO) on the first-order rate constant for the formation of the diazepine ring, i.e. regeneration of alprazolam, at pH 7.5 and 37°C.

	The observed first-order rate constant x10 <sup>2</sup> (min <sup>-1</sup> )				
Cyclodextrin	Pure water	10% EtOH	50% EtOH	10% DMSO	50% DMSO
No CD	14.2	11.5	7.24	9.68	10.7
10% RMβCD	2.97	4.90	6.70	3.97	7.92
10% HPβCD	3.30	5.23	7.07	4.44	8.57
10% SBEβCD	3.11	5.18	5.82	4.77	9.36

**Table 6.** The effect of cyclodextrins, ethanol (EtOH) and dimethylsulfoxide (DMSO) on the first-order rate constant for the formation of the diazepine ring, i.e. regeneration of triazolam, at pH 7.5 and 37°C.

	The observed first-order rate constant x10 <sup>-2</sup> (min <sup>-1</sup> )				
Cyclodextrin	Pure water	10% EtOH	50% EtOH	10% DMSO	50% DMSO
No CD	1.32	1.31	1.84	1.28	1.37
10% RMβCD	0.64	0.92	1.00	0.78	1.12
10% HPβCD	0.66	0.92	1.02	0.79	1.14
10% SBEβCD	0.58	0.82	0.97	0.73	1.13

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**Table 7.** The effect of cyclodextrins, ethanol (EtOH) and dimethylsulfoxide (DMSO) on the first-order rate constant for the formation of the diazepine ring, i.e. regeneration of midazolam, at pH 7.5 and 37°C.

	The observed first-order rate constant x10 <sup>-2</sup> (min <sup>-1</sup> )				nin <sup>-1</sup> )
Cyclodextrin	Pure water	10% EtOH	50% EtOH	10% DMSO	50% DMSO
No CD	17.9	12.6	8.41	13.8	10.9
10% RMβCD	3.05	4.24	6.99	4.94	8.48
10% HPβCD	2.77	3.86	6.53	3.36	8.40
10% SBEβCD	1.30	3.30	6.50	2.24	8.55

# Example 8

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The bioavailability of midazolam in a nasal spray according to the invention was evaluated. The composition of the midazolam nasal spray was as follows: midazolam 1.70% (w/v), sulfobutylether β-cyclodextrin sodium salt (Captisol®) 14.00% (w/v), benzalkonium chloride 0.02% (w/v), sodium edetate (EDTA tetrasodium) 0.10% (w/v), hydroxypropyl methylcellulose 0.10% (w/v), phosphoric acid 0.50% (v/v), sodium hydroxide quantum satis ad pH 4.35 in purified water. The intravenous (iv) dose was fixed at 2 mg (Dormicum™ 5 mg/ml iv solution from F. Hoffmann-La Roche & Ltd., Switzerland) but the intranasal (in) dose was 0.06 mg/kg or 4.84 mg (285  $\mu$ l nasal spray) on the average. This was a cross-over study where each individual received both the iv and in formulation (via nasal spray) with a one week resting period between administrations. Serum samples were collected at various time points after administration of the drug and the midazolam concentration determined with an HPLC method. Fig. 5 illustrates the concentration profile of midazolam in serum after administration of 2 mg of midazolam intravenously (O) or 4.8 mg of midazolam intranasally ( $\Delta$ ). Each point represents the mean value; error bars represent standard deviation. The bioavailability of midazolam after intranasal

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administration was determined to be 61% and the mean  $C_p^{\ max}$  was determined to be 52 ng/ml at 12 min after intranasal administration of the drug. Sedation was not observed after the iv administration but sedation was observed in all three individuals within 10 min after intranasal administration of the drug. This sedation lasted for about one and one-half hours. Insignificant irritation was observed in the three individuals tested after intranasal administration of the drug.

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While the invention has been described in terms of various preferred embodiments, the person skilled in the art will appreciate that various modifications, substitutions, omissions and changes can be made without departing from the spirit thereof. Accordingly, it is intended that the scope of the present invention be limited solely by the scope of the following claims, including equivalents thereof.

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#### WHAT IS CLAIMED IS:

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- 1. A method for enhancing the complexation efficiency of a drug with cyclodextrin, said drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said drug to chemically reversible ring-opening so that at least 0.1% by weight thereof is in ring-opened form, and complexing said drug with cyclodextrin.
- 2. A method according to Claim 1, comprising complexing said drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least 0.1% by weight of said drug.
  - 3. A method for enhancing the availability of a drug following administration of a cyclodextrin-drug complex to a warm-blooded animal in need of same, said drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imine, hemiacetal or hemiketal, said method comprising complexing said drug with cyclodextrin in an aqueous medium under conditions which effect chemically reversible ring-opening of at least 0.1% by weight of said drug to enhance the complexation efficiency, followed by administering the cyclodextrindrug complex thus obtained to said animal under conditions which reduce the complexation efficiency.

4. A method for enhancing the availability of a basic drug (i.e., a proton acceptor) following administration of a cyclodextrin-drug complex to a warm-blooded animal in need of same, said basic drug having a structure comprising at least one heterocyclic ring having a total of from 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said basic drug to complexation in an aqueous medium at a pH level below the pKa+2 value of said basic drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to said animal under conditions which reduce the complexation efficiency.

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- 5. A method for enhancing the availability of an acidic drug following administration of a cyclodextrin-drug complex to a warm-blooded animal in need of same, said acidic drug having a structure comprising at least one heterocyclic ring having a total of 4 to 7 ring atoms, of which from 1 to 3 are hetero ring atoms, each of said hetero ring atoms being selected from nitrogen, oxygen and sulfur, said ring being a cyclic imine, enamine, lactone, lactam, thiolactam, anhydride, imide, hemiacetal or hemiketal, said method comprising subjecting said acidic drug to complexation in an aqueous medium at a pH level above the pKa-2 value of said acidic drug to enhance the complexation efficiency, followed by administering the cyclodextrin-drug complex thus obtained to said animal under conditions which reduce the complexation efficiency.
- 6. A method according to any one of Claims 2 to 4, wherein the complexation is conducted under conditions which effect chemically reversible ring-opening of at least 1% by weight of said drug, preferably wherein the

complexation is conducted at a pH level which affords ring-opening of at least 5% by weight of said drug.

7. A method according to Claim 5, wherein the complexation is conducted under conditions which effect chemically reversible ring-opening of at least 1% by weight of said drug, preferably wherein the complexation is conducted at a pH level which affords ring-opening of at least 5% by weight of said drug.

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- 8. A method according to any one of Claims 1-4 and 6, wherein the complexation is conducted at a pH level of below about 5.
- 9. A method according to any one of Claims 1-3, 6 and 8, wherein the drug is a basic drug.
  - 10. A method according to any one of Claims 1-4, 6, 8, and 9, wherein the drug is a benzodiazepine.
  - 11. A method according to Claim 10, wherein the benzodiazepine is alprazolam, brotizolam, chlordiazepoxide, clobazam, clonazepam, clorazepam, demoxazepam, flumazenil, flurazepam, halazepam, midazolam, nordazepam, medazepam, diazepam, nitrazepam, oxazepam, midazepam, lorazepam, prazepam, quazepam, triazolam, temazepam or loprazolam.
  - 12. A method according to any one of Claims 1-4, 6 and 8-11, wherein the cyclodextrin is hydroxypropyl- $\beta$ -cyclodextrin,  $\beta$ -cyclodextrin sulfobutyl ether,  $\beta$ -cyclodextrin,  $\gamma$ -cyclodextrin or hydroxypropyl- $\gamma$ -cyclodextrin.

- 13. A method according to Claim 5 or 7, wherein the cyclodextrin is hydroxypropyl- $\beta$ -cyclodextrin,  $\beta$ -cyclodextrin sulfobutyl ether,  $\beta$ -cyclodextrin,  $\gamma$ -cyclodextrin or hydroxypropyl- $\gamma$ -cyclodextrin.
- 14. A method according to any one of Claims 10-12, followed by
  5 formulating the cyclodextrin-drug complex thus obtained as a nasal spray,
  sublingual tablet or parenteral solution.
  - 15. A method according to Claim 14, wherein the nasal spray, sublingual tablet or parenteral solution is formulated to be suitable for use in producing a sedative, anti-anxiety, anticonvulsant or muscle relaxant effect, preferably for use as a pre-anaesthetic medication, or to supplement anaesthesia, to induce and maintain anaesthesia or to induce a hypnotic effect.

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- 16. A method according to Claim 15, wherein the benzodiazepine is alprazolam, clonazepam, lorazepam, midazolam or triazolam.
- 17. A method according to any one of Claims 1-4, 6 and 8-12, wherein the complexation is conducted at a pH level between about 3 and about 5.
  - 18. A method according to Claim 1 or 2, wherein the drug is an acidic drug and the complexation is conducted under conditions which effect chemically reversible ring-opening of at least 1% by weight of said drug, preferably wherein the complexation is conducted at a pH level which affords ring-opening of at least 5% by weight of said drug.
  - 19. A method according to Claim 1 or 2, wherein the drug is a barbituric acid derivative, a hydantoin, a pyrazole derivative, an imidazole

-38-

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derivative, a pyrimidine derivative or a purine derivative, and the complexation is conducted under conditions which effect chemically reversible ring-opening of at least 1% by weight of said drug, preferably wherein the complexation is conducted at a pH level which affords ring-opening of at least 5% by weight of said drug.

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- 20. A method according to Claim 19, wherein the barbituric acid derivative is barbital, butobarbital, amobarbital, phenobarbital, aprobarbital, secobarbital, crotylbarbital, cyclobarbital, phenobarbital, hexobarbital, methylphenobarbital, thiopental, isopropylbromallylbarbituric acid, or cyclohexenylallylthiobarituric acid, or a salt thereof; or wherein the hydantoin is phenytoin; or wherein the pyrazole derivative is phenazone, propylphenazone, metamidazole, phenylbutazone, oxyphenbutazone or sulfinpyrazone; or wherein the imidazole derivative is histamine, miconazole, pilocarpine, naphazoline or clonidine; or wherein the pyrimidine derivative is thiamine, trimethoprim, orotic acid, methylthiouracyl or prothiouracyl; or wherein the purine derivative is caffeine, theophylline, etophylline, proxyphylline or theobromine.
- 21. A method according to Claim 3, wherein the cyclodextrin-drug complex is administered in the form of an aqueous solution or a hydrogel.
- 22. A method according to Claim 21, wherein the cyclodextrin-drug complex is administered as a nasal spray or nasal drops.
- 20
- 23. A method according to Claim 21, wherein the cyclodextrin-drug complex is administered as a parenteral solution.
- 24. A method according to Claim 21, wherein the aqueous solution is at a pH level of below about 6 and is administered as a nasal spray.

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- 25. A method according to Claim 24, wherein the pH level of the nasal spray is below about 4.7, preferably wherein the pH level of the nasal spray is between about 3 and about 4.7.
- 26. A method according to Claim 3, wherein the drug is a benzodiazepine, the complexation is conducted at a pH level which affords ring-opening of at least 5% by weight of said drug, said pH level being below about 5, and the cyclodextrin-drug complex is administered as a solid.
- 27. A method according to Claim 26, wherein the solid cyclodextrindrug complex is administered as a tablet formulated for oral, buccal or sublingual administration.
  - 28. A method according to Claim 3, wherein the water is removed from the aqueous complexation medium after formation of the cyclodextrin-drug complex.
- 29. A method according to Claim 4, wherein the drug is a benzodiazepine and the complexation is carried out at a pH level which affords ring-opening of at least 5% by weight of said drug and in the presence of from about 0.001 to about 5% (weight/volume) of a pharmacologically inactive, pharmaceutically acceptable water-soluble polymer at a temperature of from about 30°C to about 150°C.
- 20 30. A method according to Claim 29, wherein the polymer is a cellulose derivative or a polyvinyl polymer.

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- 31. A method according to Claim 30, wherein the polymer is methylcellulose, hydroxyethyl cellulose, hydroxypropyl cellulose, hydroxyethyl methylcellulose, hydroxypropyl ethylcellulose, hydroxyethyl ethylcellulose, sodium carboxymethylcellulose or polyvinylpyrrolidone.
- 32. A method according to Claims 4 or 29, wherein the drug is a benzodiazepine and the complexation is carried out at a pH level which affords ring-opening of at least 5% by weight of said drug and in the presence of at least one member of the group consisting of acetic acid and its pharmaceutically acceptable salts, the acetate-water ratio of the aqueous complexation medium being from about 1:1000 to about 2:1.
  - 33. A method according to Claim 32, wherein the drug is midazolam and the cyclodextrin is hydroxypropyl- $\beta$ -cyclodextrin,  $\beta$ -cyclodextrin sulfobutyl ether,  $\beta$ -cyclodextrin,  $\gamma$ -cyclodextrin or hydroxypropyl- $\gamma$ -cyclodextrin.

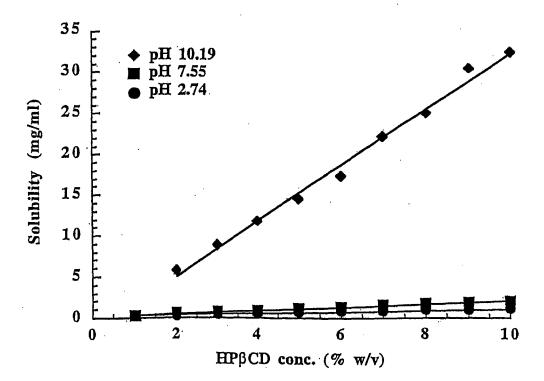


Figure 1

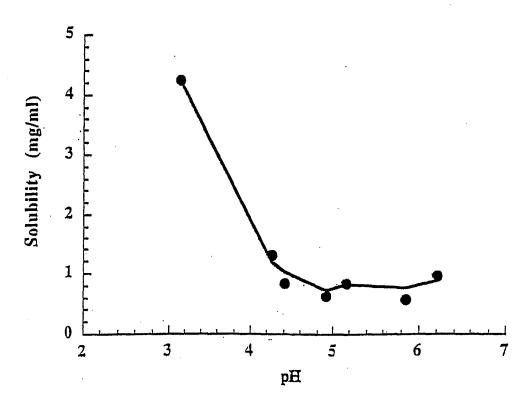


Figure 2

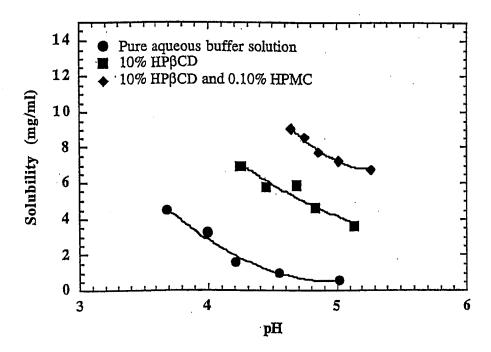


Figure 3

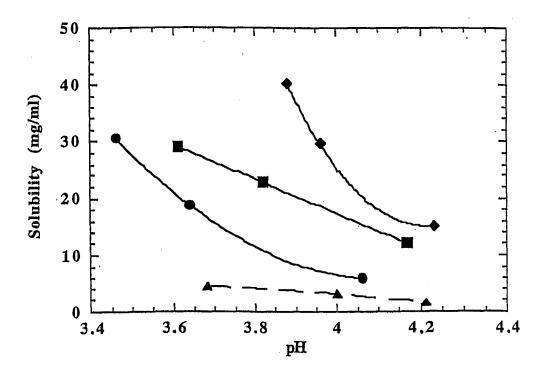


Figure 4

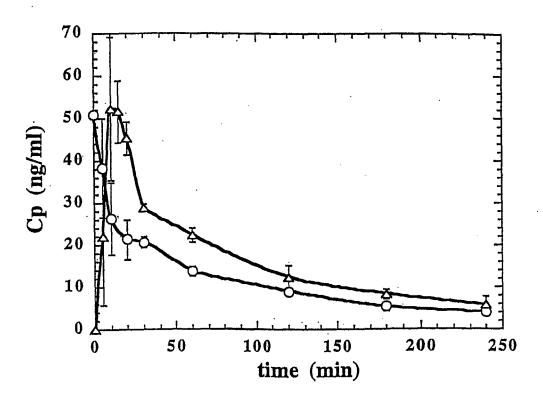


Figure 5

International application No.

PCT/IS 99/00003

		PC1/13 99/0	30003				
A, CLAS	SIFICATION OF SUBJECT MATTER						
IPC6: According t	A61K 31/715, C08B 37/16 o International Patent Classification (IPC) or to both n	ational classification and IPC					
B. FIELDS SEARCHED							
Minimum d	ocumentation searched (classification system followed b	y classification symbols)					
IPC6:	A61K, C08B						
Documenta	tion searched other than minimum documentation to th	e extent that such documents are included	in the fields searched				
SE,DK,	FI,NO classes as above						
Electronic d	ata base consulted during the international search (name	e of data base and, where practicable, sear	ch terms used)				
C. DOCL	IMENTS CONSIDERED TO BE RELEVANT						
Category*	Citation of document, with indication, where app	propriate, of the relevant passages	Relevant to claim No.				
X	International Journal of Pharmac 1991, Injoon Oh et al, "Stat solubilization of oxathiin o anti-HIV agent" page 23 - pa	1-9,12-13, 17-18,21,23, 28,30,31,33					
A		10,11,14-16, 19,20,22, 24-27,32					
	<del></del>						
A	US 5324718 A (THORSTEINN LOFTSSO (28.06.94)	1-33					
A	US 5472954 A (THORSTEINN LOFTSSO 5 December 1995 (05.12.95)	),	1-33				
X Furth	er documents are listed in the continuation of Box	C. X See patent family anne	X.				
"A" docume	categories of cited documents: int defining the general state of the art which is not considered f particular relevance	"T" later document published after the in date and not in conflict with the applithe principle or theory underlying the	ication but cited to understand				
"E" erlier d	ocument but published on or after the international filing date the which may throw doubts on priority claim(s) or which is establish the publication date of another citation or other	"X" document of particular relevance: the considered novel or carnot be considered when the document is taken alon	lered to involve an inventive				
"O" document referring to an oral disclosure, use, exhibition or other means "Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination							
"P" docume the price	int published prior to the international filing date but later than only date claimed	"&" document member of the same pater					
	e actual completion of the international search	Date of mailing of the international 1 9 -06- 1999	search report				
7 June	1999 mailing address of the ISA	Authorized officer					
	Patent Office	Avanotived outce.					
	S-102 42 STOCKHOLM	Eva Johansson/Els					
racsimile.	Facsimile No. + 46 8 666 02 86 Telephone No. + 46 8 782 25 00						

Form PCT/ISA/210 (second sheet) (July 1992)

International application No.
PCT/IS 99/00003

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
A	WO 9402518 A1 (THE UNIVERSITY OF KANSAS), 3 February 1994 (03.02.94)	1-33
	<del></del>	

International application No. PCT/IS99/00003

Box I	Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)
This inter	national search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
1. 🔀	Claims Nos.: 3-5,7,13,21-33 because they relate to subject matter not required to be searched by this Authority, namely:  See PCT Rule 39.1(iv): Methods for treatment of the human or animal body by surgery or therapy, as well as diagnostic methods.
2.	Claims Nos.: because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
Box II	Observations where unity of invention is lacking (Continuation of item 2 of first sheet)
This Inte	mational Searching Authority found multiple inventions in this international application, as follows:
1.	As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2.	As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3.	As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4.	No required additional search fees were timely paid by the applicant. Consequently, this interzetional search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark	on Protest  The additional search fees were accompanied by the applicant s protest.  No protest accompanied the payment of additional search fees.

Form PCT/ISA/210 (continuation of first sheet (1)) (July1992)

Information on patent family members

03/05/99

International application No.
PCT/IS 99/00003

Patent document cited in search report	Publication date		Patent family member(s)	Publication date	
US 5324718 A	28/06/94	AT DE EP SG US	177647 T 69323937 D 0579435 A,B 49182 A 5472954 A	15/04/99 00/00/00 19/01/94 18/05/98 05/12/95	
US 5472954 A	05/12/95	AT DE EP SG US	177647 T 69323937 D 0579435 A,B 49182 A 5324718 A	15/04/99 00/00/00 19/01/94 18/05/98 28/06/94	
WO 9402518 A1	03/02/94	AU AU CA EP JP US	672814 B 4779993 A 2119154 A,C 0620828 A 6511513 T 5376645 A	17/10/96 14/02/94 03/02/94 26/10/94 22/12/94 27/12/94	

Form PCT/ISA/210 (patent family annex) (July 1992)

Electronic Patent Application Fee Transmittal							
Application Number:	10:	551205					
Filing Date:	14	Nov-2006					
Title of Invention:	Oral formulations of cladribine						
First Named Inventor/Applicant Name:	Nicholas S. Bodor						
Filer:	Ma	ry Katherine Baume	eister/Diana Fran	ıcis			
Attorney Docket Number:	00:	56192-000024					
Filed as Large Entity							
U.S. National Stage under 35 USC 371 Filing	Fee	s					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Basic Filing:							
Pages:							
Claims:							
Miscellaneous-Filing:							
Petition:							
Patent-Appeals-and-Interference:							
Post-Allowance-and-Post-Issuance:							
Extension-of-Time:							

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Miscellaneous:						
Submission- Information Disclosure Stmt	1806	1	180	180		
Total in USD (\$)						

Electronic Acl	Electronic Acknowledgement Receipt						
EFS ID:	6649992						
Application Number:	10551205						
International Application Number:							
Confirmation Number:	4092						
Title of Invention:	Oral formulations of cladribine						
First Named Inventor/Applicant Name:	Nicholas S. Bodor						
Customer Number:	21839						
Filer:	Mary Katherine Baumeister/Diana Francis						
Filer Authorized By:	Mary Katherine Baumeister						
Attorney Docket Number:	0056192-000024						
Receipt Date:	16-DEC-2009						
Filing Date:	14-NOV-2006						
Time Stamp:	15:40:55						
Application Type:	U.S. National Stage under 35 USC 371						

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1	Transmittal Letter	005619224TL.pdf	50718	no	1
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Warnings:					
Information:					
2	Amendment/Req. Reconsideration-After	005619224AMEND.pdf	481410	no	9
-	Non-Final Reject	ooso 1922 Millet B.pai	1ce21e4cbf70b501cc1e5fb52f53eba7dd3a 6b6a	110	
Warnings:					
Information:					
3	Transmittal Letter	005619224TLIDS.pdf	40409	no	1
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Warnings:			•		-
Information:					
4	Transmittal Letter	005619224lDS.pdf	47767	no	2
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Warnings:					
Information:					
5	Information Disclosure Statement (IDS)	0056192241449.pdf	64666	no	1
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Information:	:				
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6	Foreign Reference	005619224ref.pdf	3848a30c436978462410f72650d3b601e39 8a9a7	no	51
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		Total Files Size (in bytes	): 25	78931	
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#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re F	Patent Applicatio	n of		) MAIL STOP AMENDMENT					
Nicho	las Bodor et al.			Group Art Unit: 1623					
Applic	ation No.: 10/55	51,205		) Examiner:	JONATHAN S LAU				
Filing	Date: Novembe	r 14, 2006		) Confirmati	on No.: 4092				
Title:	ORAL FORM	IULATIONS C	F CLADRIBINE	) ) )					
		AM	ENDMENT/REPLY TR	RANSMITTAL LE	TTER				
P.O. E	nissioner for Pate Box 1450 ndria, VA 22313-	1							
Sir:						$F^* = \{A_i\}$			
Enclo	sed is a reply for	the above-ide	entified patent applicat	ion.					
			Time is enclosed.						
	Terr 37 C.F.R. § 1	minal Disclain .20(d) are end	ner(s) and the 🔲 \$ 70 closed.	☐ \$ 140 fee pe	r Disclaimer due und	der			
$\boxtimes$	Also enclosed Statement, P	d is/are: <u>Sixth</u> TO Form-144	Information Disclosur 9, and (1) document.	<u>e Statement Trar</u>	smittal, Sixth Inform	ation Disclos	<u>ure</u>		
	Small entity s	tatus is hereb	y claimed.						
$\boxtimes$	No additional	claim fee is re	equired.						
	An additional	claim fee is re	equired, and is calcula	ted as shown bel	ow:				
			AMENDE	ED CLAIMS					
		No. of Claims	Highest No. of Claims Previously Paid For	Extra Claims	Rate	Additic	onal Fee		
Total C	laims	56	78	0	x \$ 52 (1202)	\$	0		
Indepe	ndent Claims	4	5	0	x \$ 220 (1201)		0		
☐ If A	mendment adds m	ultiple depende	nt claims, add \$ 390 (120	03)		\$	0		
Total (	Claim Amendmen	t Fee				\$	0		
☐ Sm	all Entity Status cla	aimed - subtract	50% of Total Claim Ame	endment Fee			0		
TOTAL	ADDITIONAL CL	AIM FEE DUE	FOR THIS AMENDMEN	T	<u> </u>	\$	0		
	Charge	t	o Deposit Account No	. 02-4800 for the	fee due.				
	Charge	t	o credit card for the fe	e due. Form PT0	D-2038 is attached.				
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			Respectfully	submitted,		•	1 1 9		
			Buchanan In-	GERSOLL & ROONE	EY PC	• •	- 1 m		
	December 16, 2	<u>2009</u>		Katherine Baume tration No. 2625		v			
	36 6620				•		4		

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P	PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875					A	Application or Docket Number Filing Date 11/14/2006			To be Mailed	
	APPLICATION AS FILED – PART I (Column 1) (Column 2)						SMALL	FNTITY $\Box$	OR		HER THAN
	FOR NUMBER FILED NUMBER EXTRA				1	RATE (\$)	FEE (\$)		RATE (\$)	FEE (\$)	
BASIC FEE (37 CFR 1.16(a), (b), or (c))			N/A		N/A		N/A		1	N/A	, ,
	SEARCH FEE (37 CFR 1.16(k), (i), (i)		N/A		N/A		N/A			N/A	
	EXAMINATION FE (37 CFR 1.16(o), (p),		N/A		N/A		N/A			N/A	
	FAL CLAIMS CFR 1.16(i))		mir	nus 20 = *			x \$ =		OR	x \$ =	
	EPENDENT CLAIM CFR 1.16(h))	S	m	inus 3 = *			x \$ =			x \$ =	
	APPLICATION SIZE (37 CFR 1.16(s))	shee is \$2 addit	ts of pap 50 (\$125 tional 50 s	er, the applica for small enti sheets or frac	wings exceed 100 ation size fee due ity) for each stion thereof. See 37 CFR 1.16(s).						
	MULTIPLE DEPEN	IDENT CLAIM PR	ESENT (3	7 CFR 1.16(j))							
* If	the difference in colu	umn 1 is less than	zero, ente	r "0" in column	2.		TOTAL			TOTAL	
	APPI	LICATION AS (Column 1)	AMEND	DED – PART (Column 2)		_	SMAL	L ENTITY	OR		ER THAN ALL ENTITY
AMENDMENT	12/16/2009	CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUSL PAID FOR	PRESENT Y EXTRA		RATE (\$)	ADDITIONAL FEE (\$)		RATE (\$)	ADDITIONAL FEE (\$)
ME	Total (37 CFR 1.16(i))	* 56	Minus	** 78	= 0		x \$ =		OR	X \$52=	0
١	Independent (37 CFR 1.16(h))	* 5	Minus	***6	= 0		x \$ =		OR	X \$220=	0
√ME	Application Si	ize Fee (37 CFR 1	.16(s))								
_	FIRST PRESEN	NTATION OF MULTIF	PLE DEPEN	DENT CLAIM (37	CFR 1.16(j))	1			OR		
							TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	0
		(Column 1)		(Column 2)	(Column 3)						
		CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUSL PAID FOR	Y EXTRA		RATE (\$)	ADDITIONAL FEE (\$)		RATE (\$)	ADDITIONAL FEE (\$)
N	Total (37 CFR 1.16(i))	*	Minus	**	=		x \$ =		OR	x \$ =	
AMENDMENT	Independent (37 CFR 1.16(h))	*	Minus	***	=		x \$ =		OR	x \$ =	
Ш	Application Si	ize Fee (37 CFR 1	.16(s))								
AM	FIRST PRESEN	NTATION OF MULTIF	PLE DEPEN	DENT CLAIM (37	CFR 1.16(j))	1			OR		
							TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	
** If	* If the entry in column 1 is less than the entry in column 2, write "0" in column 3.  ** If the "Highest Number Previously Paid For" IN THIS SPACE is less than 20, enter "20".  *** If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, enter "3".  The "Highest Number Previously Paid For" (Total or Independent) is the highest number found in the appropriate box in column 1.										

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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/551,205	11/14/2006	Nicholas S. Bodor	0056192-000024	4092	
	7590 03/30/201 INGERSOLL & ROOI		EXAM	INER	
POST OFFICE	BOX 1404	LAU, JONATHAN S			
ALEXANDRIA	A, VA 22313-1404	ART UNIT	PAPER NUMBER		
		1623			
			NOTIFICATION DATE	DELIVERY MODE	
			03/30/2010	ELECTRONIC	

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The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

ADIPFDD@bipc.com offserv@bipc.com

		Application No.	Applicant(s)						
	Office Action Comments	10/551,205	BODOR ET AL.						
	Office Action Summary	Examiner	Art Unit						
		Jonathan S. Lau	1623						
	The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply								
WHIC - Exter after - If NO - Failui Any r	A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.  - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).  Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).								
Status									
1) 又	Responsive to communication(s) filed on 16 De	ecember 2009.							
•		action is non-final.							
7—	Since this application is in condition for allowan		secution as to the merits is						
<i>/</i>	closed in accordance with the practice under <i>E</i> .								
D: ''	·	•							
-	on of Claims								
5)□ 6)⊠ 7)□	7) Claim(s) is/are objected to.								
Applicati	on Papers								
<ul> <li>9) The specification is objected to by the Examiner.</li> <li>10) The drawing(s) filed on is/are: a) accepted or b) objected to by the Examiner.  Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).</li> <li>11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.</li> </ul>									
Priority u	ınder 35 U.S.C. § 119								
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  a) All b) Some * c) None of:  1. Certified copies of the priority documents have been received.  2. Certified copies of the priority documents have been received in Application No.  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.									
Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date 1 pg / 16 Dec 2009.  4) Interview Summary (PTO-413)  Paper No(s)/Mail Date  5) Notice of Informal Patent Application  6) Other:									

#### **DETAILED ACTION**

This Office Action is responsive to Applicant's Amendment and Remarks, filed 16 Dec 2009.

This application is the national stage entry of PCT/US04/09387, filed 26 Mar 2004; and claims benefit of provisional application 60/458,922, filed 28 Mar 2003; and claims benefit of provisional application 60/484,756, filed 02 July 2003; and claims benefit of provisional application 60/541,247, filed 04 Feb 2004.

The filing date of the instant claims 12, 83, 85 and 89 are deemed to be the filing date of the instant application which is the filing date of PCT/US04/09387, 26 Mar 2004. The filing date of instant claims 1, 2, 8, 9, 11, 56, 57, 63, 64, 82, 84 and 86-98 are deemed to be the filing date of provisional application 60/541,247, filed 04 Feb 2004.

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 67-98 are pending in the current application. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81, drawn to non-elected inventions, are withdrawn. Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 are examined on the merits herein.

#### Rejections Withdrawn

Applicant's Remarks, filed 16 Dec 2009, with respect to claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Wrenn

Jr. (US Patent 6,174,873, issued 16 Jan 2001, cited in PTO-892) and in view of Loftsson et al. (US Patent 6,699,849, filed 16 Feb 1999, cited in PTO-892) has been fully considered and is persuasive, as Applicant is persuasive that one of ordinary skill in the art would not have a reasonable expectation of success in combining the teaching of Wrenn Jr. drawn to a an amorphous formulation using a polymer cross-linked technology with the teaching of Schultz et al. drawn to a solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin.

This rejection has been withdrawn.

The following are new grounds of rejection not necessitated by Applicant's Amendment.

#### Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Amended Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 are rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Pitha (US Patent 4,727,064, issued 23 Feb 1988, provided by Applicant in IDS mailed 4 Apr 2008) and in view of Loftsson J Pharm Sci 2002 (Journal of Pharmaceutical Sciences, 2002, 91(11), p2307-2316, cited in PTO-892).

Schultz et al. discloses a solid pharmaceutical oral dosage form of cladribine comprising cladribine and cyclodextrin (column 2, lines 31-39). Schultz et al. teaches βcyclodextrins are known to possess the ability to form inclusion complexes and to have concomitant solubilizing properties (column 2, lines 10-15). Schultz et al. discloses the use of β-cyclodextrins (column 2, lines 56-58) and derivatives wherein one or more cyclodextrin hydroxy groups are replaced with groups such as hydroxypropyl (column 3, lines 26-27). Schultz et al. discloses the solid oral dosage form in the form of a tablet (column 5, lines 37-38) including the excipients sorbitol and magnesium stearate (column 6, lines 2-7). Schultz et al. discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, which renders obvious to one of skill in the art the sub-range of a cladribine to cyclodextrin ratio ranging from 15 mg:100 mg to 15mg:500 mg, or 1:6.67 to 1:33.3 by weight (column 6, lines 23-31). Schultz et al. implicitly discloses an oral dosage form comprising up to 15 mg cladribine and cyclodextrin from 100 to 500 mg, or a cladribine to cyclodextrin ratio ranging from 1:6.67 to 1:33.3 by weight (column 6, lines 23-31).

Schultz et al. does not specifically disclose the composition comprising no significant amount of free crystalline cladribine therein (instant claims 1). Schultz et al. does not specifically disclose the composition corresponding to a point located on the curve of a phase solubility diagram for saturated complex cladribine-cyclodextrin complexes, said curve defining complex saturated complexes of cladribine in varying concentrations of the cyclodextrin (instant claim 11). Schultz et al. does not specifically disclose the complex consisting of (a) an amorphous inclusion complex of cladribine

with an amorphous cyclodextrin and (b) amorphous free cladribine associated with amorphous cyclodextrin as a non-inclusion complex (instant claim 56). Schultz et al. does not specifically disclose the composition comprising a cladribine to cyclodextrin ratio from about 1:10 to about 1:16 (instant claims 6, 7, 10, 61, 62 and 65), or a ratio of about 1:14 (instant claims 8 and 63) or about 1:11 (instant claims 9 and 64). Schultz et al. does not specifically disclose the complex wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b) (instant claims 12 and 66). Schultz et al. does not specifically disclose the product-by-process wherein 12.00 parts by weight of cladribine and 172.50 parts by weight of hydroxypropyl-βcyclodextrin are introduced in step (i) of the process (instant claim 91 and 93), to give a cladribine to cyclodextrin ratio of 1:14.38. Schultz et al. does not specifically disclose the product-by-process wherein 16.35 parts by weight of cladribine and 172.50 parts by weight of the hydroxypropyl-β-cyclodextrin are introduced in step (i) of the process (instant claim 92), to give a cladribine to cyclodextrin ratio of 1:10.55.

Pitha teaches a pharmaceutical combination of drug and amorphous cyclodextrin to give a stable amorphous state that improves dissolution properties of the drug and absorption by the body (column 1, lines 10-15) and that prevents crystallization processes within the pharmaceutical preparation (column 1, lines 20-25). Pitha teaches the embodiment wherein the amorphous cyclodextrin is hydroxypropylbeta-cyclodextrin (table 1 spanning columns 3 and 4). Pitha teaches the product made

by freeze-drying, or lyophilizing, a solution of cyclodextrin and drug (column 7, lines 5-40).

Loftsson J Pharm Sci 2002 teaches drug/cyclodextrin complexes self-associate to form water-soluble aggregates as non-inclusion complexes in addition to formation of the inclusion complex (abstract). Loftsson J Pharm Sci 2002 teaches the formation of only the drug/cyclodextrin inclusion complex is a general assumption (page 2307, section Introduction), and that drug/cyclodextrin complexes self-associate to form water-soluble aggregates as non-inclusion complexes (page 2315, section Conclusions).

It would have been obvious to one of ordinary skill in the art at the time of the invention to combine Schultz et al. in view of Pitha and in view of Loftsson J Pharm Sci 2002. Schultz et al. teaches solid formulations for oral administration of cladribine and cyclodextrin. One of ordinary skill in the art would have been motivated to combine Schultz et al. in view of Pitha because Schultz et al. teaches undesirable recrystallization of cladribine in tissue may occur and damage the surrounding tissue and that complexes with cyclodextrin are known to solubilize the compound and Pitha teaches the pharmaceutical combination of drug and amorphous cyclodextrin to give a stable amorphous state that improves dissolution properties of the drug and absorption by the body. One of ordinary skill in the art would have a reasonable expectation of success in combining Schultz et al. in view of Pitha because Pitha teaches a application of a wide variety of drugs in the complex taught by Pitha and Schultz et al. teaches the formation of the cladribine and cyclodextrin complex in solution. Schultz et al. in view of Pitha and in view of Loftsson J Pharm Sci 2002 does not teach the specific cladribine to

cyclodextrin ratios of 1:14.38 or 1:10.55, however these ratios are encompassed by the prior art and Schultz et al. teaches it is within the level of skill in the art to optimize the ratio of cyclodextrin relative to cladribine (column 4, lines 35-45). See also MPEP 2144.05 II.A, "Generally, differences in concentration or temperature will not support the patentability of subject matter encompassed by the prior art unless there is evidence indicating such concentration or temperature is critical." One of ordinary skill in the art would be motivated to optimize the cladribine to cyclodextrin ratio to give the composition comprising no significant amount of free crystalline cladribine therein because Schultz et al. teaches undesirable recrystallization of cladribine in tissue may occur and damage the surrounding tissue (Schultz et al. column 2, lines 1-15).

Loftsson J Pharm Sci 2002 provides evidence that the property of self-association of the drug/cyclodextrin complex is necessarily present in the drug/cyclodextrin composition taught by Schultz et al. in view of Pitha. Therefore there is reasonable evidence to conclude that the process of self-association of the drug/cyclodextrin complex at the ratio taught by Schultz et al. in view of Pitha would necessarily result in both inclusion complexes and non-inclusion complexes and wherein from about 30 to about 40 percent by weight of the cladribine is in the inclusion complex (a) and from about 70 to about 60 percent by weight of the cladribine is in the non-inclusion complex (b). See also MPEP 2112.

Claims 82-90 and 94-98 are drawn to a product-by-process. The disclosed product is substantially identical to the instantly claimed product-by-process, a pharmaceutical solid oral dosage form comprising an amorphous inclusion complex of

cladribine and cyclodextrin and a non-inclusion complex of an amorphous cladribine and an amorphous cyclodextrin as detailed above. "[E]ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same as or obvious from a product of the prior art, the claim is unpatentable even though the prior product was made by a different process." In re Thorpe, 777 F.2d 695, 698, 227 USPQ 964, 966 (Fed. Cir. 1985) (citations omitted) (Claim was directed to a novolac color developer. The process of making the developer was allowed. The difference between the inventive process and the prior art was the addition of metal oxide and carboxylic acid as separate ingredients instead of adding the more expensive pre-reacted metal carboxylate. The product-by-process claim was rejected because the end product, in both the prior art and the allowed process, ends up containing metal carboxylate. The fact that the metal carboxylate is not directly added, but is instead produced in-situ does not change the end product.). See MPEP 2113.

#### **Response to Applicant's Remarks:**

Applicant's Remarks, filed 16 Dec 2009, have been fully considered and not found to be persuasive in view of the new grounds of rejection.

Applicant's note that the invention disclosed by Schultz et al. via a melt-extrusion process results in the formation of the mixture and not a complex of cladribine and cyclodextrin in solid form, as provided by evidence in Van Axel Castelli et al. However, MPEP 2121.01 II. provides a non-enabling reference may qualify as prior art for the

purpose of determining obviousness under 35 U.S.C. 103. Schultz et al. discloses the melt-extrusion as one method of preparing solid oral dosage forms (col 5, line 50), thus the teaching of Schultz et al. does not teach away from the combination of Schultz et al. in view of Pitha, teaching a product made by freeze-drying, and in view of Loftsson J Pharm Sci 2002 to support a conclusion of obviousness.

#### Conclusion

No claim is found to be allowable.

This Office Action details new grounds of rejection not necessitated by Applicant's Amendment. Therefore this Office Action is Non-Final.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan S. Lau whose telephone number is 571-270-3531. The examiner can normally be reached on Monday - Thursday, 9 am - 4 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Shaojia Anna Jiang can be reached on 571-272-0627. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Art Unit: 1623

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jonathan Lau Patent Examiner Art Unit 1623 /Shaojia Anna Jiang/ Supervisory Patent Examiner Art Unit 1623

Notice of References Cited	Application/Control No. 10/551,205	Applicant(s)/Patent Under Reexamination BODOR ET AL.	
	Examiner	Art Unit	
	Jonathan S. Lau	1623	Page 1 of 1

#### U.S. PATENT DOCUMENTS

*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Name	Classification
	Α	US-			
	В	US-			
	С	US-			
	D	US-			
	Е	US-			
	F	US-			
	G	US-			
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#### FOREIGN PATENT DOCUMENTS

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#### **NON-PATENT DOCUMENTS**

	NON-FATENT DOCUMENTS							
*		Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)						
	U	Loftsson et al. Journal of Pharmaceutical Sciences, 2002, 91(11), p2307-2316.						
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\*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).)

Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

**Notice of References Cited** 

Part of Paper No. 20100325

# Index of Claims Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

✓	Rejected	•	Cancelled
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# Index of Claims Index of Claims 10551205 Examiner Jonathan S Lau Applicant(s)/Patent Under Reexamination BODOR ET AL. Art Unit 1623

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U.S. Patent and Trademark Office

	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	10551205	BODOR ET AL.
	Examiner	Art Unit
	Jonathan S Lau	1623

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	Α	Appeal
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U.S. Patent and Trademark Office Part of Paper No.: 20100325

# Search Notes

Application/Control No.	Applicant(s)/Patent Under Reexamination
10551205	BODOR ET AL.
Examiner	Art Unit
Jonathan S Lau	1623

SEARCHED							
Class	Subclass	Date	Examiner				

SEARCH NOTES								
Search Notes	Date	Examiner						
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker)	3/26/2008	JSL						
EAST - see attached notes	3/26/2008	JSL						
Google Scholar - see attached notes	3/26/2008	JSL						
EAST - see attached notes	9/10/2009	JSL						
Google Scholar - see attached notes	9/10/2009	JSL						
STN - CAPlus file - see attached notes	9/10/2009	JSL						
EAST - see attached notes	3/25/2010	JSL						
Google Scholar - see attached notes	3/25/2010	JSL						

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COORE SCHOLAR Cyclodextrin non-inclusion

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Results 1 - 10 of about 456. (0.23 sec)

# Self-association of **cyclodextrins** and **cyclodextrin** complexes

T Loftsson, M Másson, ME ... - Journal of ..., 2004 - interscience.wiley.com

... 107 Reykjavik, Iceland. Telephone: 354-525-4464; Fax: 354-525-4071. Keywords.

cyclodextrin • self-association • complexation • non-inclusion • aggregates •

phase-solubility • solubilization. Abstract. Cyclodextrins are useful ...

Cited by 76 - Related articles - Bt. Direct - All 4 versions

# ... on inclusion and non-inclusion phenomena between β-cyclodextrin and new anti- ...

ME Amato, KB Lipkowitz, GM Lombardo, GC ... - Journal of the Chemical ..., 1996 - rsc.org The formation of supramolecular complexes of -CD with new generation anti-Alzheimer's agents of the 1,2,3,4-tetrahydro-9-aminoacridine family (tacrine hydrochloride, 1, velnacrine maleate, 2 and suronacrine maleate, 3), has been studied using molecular dynamics ...

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# Cyclodextrins and their pharmaceutical applications

uniroma Lit (PDF)

T Loftsson, D Duchêne - International journal of pharmaceutics, 2007 - Elsevier ... period, 1970 to present 2.3.1. Production of **cyclodextrins** 2.3.2. **Cyclodextrin** derivatives 2.3.3. Industrial applications of **cyclodextrins** 2.3.4. Inclusion and **non-inclusion** complexes 2.3.5.

Methods to enhance the complexation efficiency 2.3.6. **Cyclodextrin** aggregates 2.3.7. Drug ...

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# Evaluation of **cyclodextrin** solubilization of drugs

T Loftsson, D Hreinsdótlir, M Másson - International journal of ..., 2005 - Elsevier

... However, **cyclodextrins** (the hosts) are also known to form **non-inclusion** complexes (Loftsson et al., 2002 and Loftsson et al., 2004b). Most lipophilic compounds (the guests) form apparent

1:1 guest/host complex although apparent higher order complexes are not uncommon. ...

Cited by 79 - Related articles - All 4 versions

# [PDF] Non-chromatographic analytical uses of cyclodextrins

L Szente, J Szeitli - The Analyst, 1998 - rsc.org

... From a mechanistic standpoint, the selectivity of the interaction seems to involve partially a **non**-

rsc.org [PDF]

**inclusion** process, because of the high ... detectors and indicators Numerous papers and patents have been dedicated to the use of **cyclodextrins** and **cyclodextrin** inclusion complexes ...

Cited by 28 - Related articles - View as HTML - BL Direct - All 4 versions

# The effects of organic salts on the cyclodextrin solubilization of drugs

T Loftsson, K Matthiasson, M Másson - International journal of ..., 2003 - Elsevier

... For example, Gabelica et al. have shown that  $\alpha$ -cyclodextrin ( $\alpha$ CD) forms both inclusion and **non-inclusion** complexes with dicarboxylic acids and that the two types of complexes coexist in aqueous solutions ( [Gabelica et al]). ...

Cited by 21 - Related articles - Ali 4 versions

# Influence of response factors on determining equilibrium association constants of ...

V Gabelica, N Galic, F Rosu, C ... - Journal of Mass ..., 2003 - interscience.wiley.com

... **cyclodextrin**. This may be due to the fact that **cyclodextrin** is neutral in solution, whereas the complex is charged, but it can also stem from the fact that a significant proportion of the complex is in a **non-inclusion** geometry. The ...

Cited by 48 - Related articles - BL Direct - All 7 versions

# Cyclodextrins as pharmaceutical solubilizers

ME Brewster, T Loftsson - Advanced drug delivery reviews, 2007 - Elsevier

... Other solubilizing attribute may include the ability to form **non-inclusion** based complexes, the formation of aggregates and related domains and the ability of **cyclodextrins** to form and stabilize supersaturated drug solutions. ...

Cited by 99 - Related articles - All 7 versions

# Cyclodextrin solubilization of the antibacterial agents triclosan and triclocarban: ...

MS Duan, N Zhao, ÍB Össurardóttir, T ... - International journal of ..., 2005 - Elsevier ... the **cyclodextrin** solubilization of drugs. It is thought that these additives enhance the **cyclodextrin** complexation of drugs by forming **non-inclusion** complexes with **cyclodextrins** and their complexes. For example, it has been shown ...

Cited by 22 - Related articles - All 4 versions

# Self-association and cyclodextrin solubilization of drugs

T Loftsson, A Magnúsdóttir, M Másson, ... - Journal of ..., 2002 - interscience.wiley.com ... studies. Furthermore, the results indicate that drug/cyclodextrin complexes can self-associate to form water-soluble aggregates, which then can further solubilize the drug through non-inclusion complexation. © 2002 Wiley-Liss, Inc. ...

Cited by 50 - Related articles - Bt. Direct - All 3 versions

uniromat.it (PDF)

Result Page:

cyclodextrin non-inclusion Search

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### **EAST Search History**

# **EAST Search History (Prior Art)**

Ref#	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
S1	1	"6194395".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:14
S2	1	"6699849".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:15
S3	1	"6174873".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:21
S4	1148	cladribine and cyclodextrin	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:23
S5	271	S4 and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:23
S6	75	S5 and (amorphous or noncrystal \$5)	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:24
S7	3	S6 and ((cladribine same cyclodextrin) or (cladribine near9 cyclodextrin))	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:25
S8	20	((adenosine same cyclodextrin) or (adenosine near9 cyclodextrin)) and (amorphous or noncrystal\$5 or non-crystal\$5) and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:38
S9	1	S2 and (amorphous or noncrystal \$5)	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:43
S10	1	S3 and (amorphous or noncrystal \$5)	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 14:48
S11	32	cladribine.ti,ab,bsum. and cyclodextrin and (amorphous or noncrystal\$5 or non-crystal\$5) and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 15:11

S12	О	cladribine.ti,ab,bsum. and "cyclodextrin.ti,ab,bsum" and (amorphous or noncrystal\$5 or non-crystal \$5) and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 15:12
S13	17	cladribine.ti,ab,bsum. and cyclodextrin.ti,ab,bsum. and (amorphous or noncrystal\$5 or non-crystal\$5) and @ad<="20040326"	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 15:12
S14	1	"4727064".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 15:46
S15	1	"7115586".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 16:20
S16	1	"20030109492".pn.	US-PGPUB; USPAT; USOCR	ADJ	ON	2010/03/25 16:22

# **EAST Search History (Interference)**

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3/25/2010 5:57:25 PM

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of

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Sheet

# SIXTH INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(use as many sheets as necessary)

Complete if Known					
Application Number	10/551,205				
Filing Date	November 14, 2006				
First Named Inventor	Nicholas Bodor et al.				
Examiner Name	JONATHAN S LAU				
Attorney Docket No.	0056192-000024				

U.S. PATENT DOCUMENTS Pages, Columns, Lines Name of Patentee or Applicant Examiner **Document Number-Publication Date** Where Relevant Passages Initials Kind Code MM-DD-YYYY of Cited Document or Figures Appear US-US-US-ÚŚ-ÚS-US-

**FOREIGN PATENT DOCUMENTS** STATUS Foreign Patent Document Partial Translation Eng. Lang. Summary Translation Name of Patentee or Search Report **Publication Date** Applicant of Cited PER Cited in Spec. / Examiner Country Code<sup>1</sup>, Number, (MM-DD-YYYY) Pg. No(s). initials Document Kind Code WO 99/42111 08-26-1999 CYCLOPS, EHF Enter Office that issued the document, by the two-letter code.

Examiner Initials Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.

Examiner Signature	/Jonathan Lau/	Date Considered	03/25/2010

<sup>\*</sup>EXAMINER: Initial if reference considered, whether or not citation is in conformance with M.P.E.P. § 609. Draw line through citation if not in conformance and not considered. Include copy of this form with next corruptication to Applicant.

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

MAIL STOP AMENDMENT
Group Art Unit: 1623
Examiner: JONATHAN S LAU
Confirmation No.: 4092

### **REPLY TO OFFICIAL ACTION**

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

In response to the Office Action dated March 30, 2010, the following remarks are offered:

### **REMARKS**

Applicants request reexamination and reconsideration of the subject application pursuant to and consistent with 37 C.F.R. § 1.112 in light of the following:

### STATUS OF CLAIMS

The status of the claims under examination has been misstated. Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 66-98 remain in this application. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81 have been withdrawn from consideration as drawn to non-elected subject matter. Claims 1, 2, 8, 9, 11, 12, 56, 57, 63, 64, 66 and 82-98 are under examination. It is again pointed out that Claims 12 and 66 were not included in the previous rejection and are not included in the current rejection. Clarification is requested.

### INFORMATION DISCLOSURE STATEMENT

Applicants appreciate the Examiner's consideration of the December 16, 2009 Information Disclosure Statement and the return of the initialed Form PTO-1449.

### **REJECTIONS WITHDRAWN**

Applicants appreciate the Examiner's withdrawal of the previous 35 U.S.C. § 103(a) rejection.

### CLAIM REJECTIONS - 35 U.S.C. § 103(a)

Claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 have been rejected as being unpatentable over Schultz et al. US 6194395 in view of Pitha US 4727064 and in view of Loftsson et al., <u>J. Pharm. Sci.</u>, 2002, 91(11), pp. 2307-2316. Applicants believe that all of the claims under examination, that is, all of Claims 1, 2, 8, 9, 11, 12, 56, 57, 63, 64, 66 and 82-98 are patentable over this combination of references.

Applicants have previously established that, while Schultz et al. describe inclusion complex formation in solution to form injectable solutions, Schultz et al. describe solid formulations which are mixtures, not complexes, of cladribine and cylodextrin. Applicants have already shown, particularly with reference to the experiments described in the Van Axel Castelli et al. paper previously submitted, that

contrary to Schultz et al., the instantly claimed cladribine/cyclodextrin complex is not a simple mixture of the ingredients and has different properties from a mixture of cladribine and cyclodextrin. Therefore, the ratios of cladribine and cyclodextrin in Schultz et al.'s solid are irrelevant to the ratios of cladribine and cylodextrin in the presently claimed complexes. The fact remains that Schultz et al. only discloses complexes in solutions for injection, not for solid oral dosage forms.

Moreover, in making his rejection, the Examiner has taken a teaching of Schultz et al. completely out of context and from this he has constructed a rejection which is improper. Specifically, the Examiner states that Schultz et al. teach undesirable recrystallization of cladribine in tissue may occur and damage surrounding tissue (col. 2, lines 1-2), and on this teaching he builds his position about solid oral formulations of cladribine/cyclodextrin complexes. However, the passage in Schultz et al. relied upon by the Examiner needs to be read together with the preceding passages in col. 1., which clearly refers only to injectable formulations of high osmolality when injected by the subcutaneous route (sentence bridging columns 1 and 2 of Schultz et al.). The passage quoted has therefore no relation to solid oral dosage forms. After oral absorption, no crystallization in any tissue could or would occur. Thus, it cannot provide any motivation to one of ordinary skill to combine Schultz et al with the other cited references.

Nevertheless, Pitha US 4727064 has been cited in combination with Schultz et al. Indeed, Pitha was prior art cited during the examination of the Schultz et al. patent, as is evident from the fact that it is listed on the face of the Schultz et al. patent. The Pitha patent issued eleven years before the Schultz et al. application was filed and was clearly available to Schultz et al. at the time of the Schultz et al. invention. Nevertheless, Schultz et al. turned to inclusion complex formulation only as a way of providing suitable injectable formulations of cladribine. In contrast, for solid dosage forms they suggested simple mixtures. Moreover, Pitha did not even address cladribine as such, much less the fact that it is acid-labile and unstable in the acidic environment of the gastrointestinal system (Schultz et al., col. 1, lines 47-51). It is not seen how one of ordinary skill would be motivated to try to combine Pitha with Schultz et al. to provide an alternate solution for solid oral use.

Still further, since there is no mention of cladribine as such in Pitha, preparation of inclusion complexes disclosed therein of, for example, sex hormones (Example 4 in column 7) clearly uses conditions different from those used herein for cladribine.

With respect to the Loftsson et al. <u>J. Pharm. Sci.</u> literature article, applicants believe that the conclusions of the article have been taken out of context. Loftsson et al. do indeed describe particular situations in which <u>self-association of cyclodextrin complexes</u> may explain some observed solubilization phenomena. However, Loftsson et al. first studied the solubility of ibuprofen sodium salt, diflumisal sodium salt, alprazolam,  $17\beta$ -estradiol and diethylstilbestrol in HP $\beta$ CD. Then, aqueous HP $\beta$ CD solutions, previously saturated with the sodium salts of either ibuprofen or diflunisal were saturated with a second drug ( $17\beta$ -estradiol, diethylstilbestrol or alprazolam). On page 2313, left column, first full paragraph, Loftsson et al. summarize their conclusions from these experiments:

If the solubilization of a given drug (the first drug) is solely attributable to inclusion complex formation and if the slope of the obtained phase-solubility diagram is greater than unity, then it can be assumed that almost all cyclodextrin molecules in the aqueous complexation medium will be forming inclusion complexes with the drug. In this case, the concentration of free cyclodextrin in a saturated drug solution will be very low and under such conditions there will be very little capacity to solubilize a second water-insoluble drug in the same medium. Introduction of a second drug will then always result in some precipitation of the drug that previously was used to saturate the solution. However, if the first drug is partially solubilized through non-inclusion association, then there could be some capacity in the solution to solubilize the second drug in a similar manner as drugs are solubilized in micelles. In other words, with inclusion complexation, we will expect to see a consistent competitive effect between the first drug and the second drug, but if solubilization through non-inclusion association exists in the solution, we can expect to observe a cooperative effect, especially if the complexation efficacy of the second drug is low.

Attorney Docket No. 0056192-000024 Application No. 10/551,205 Page 5

It is clear from the foregoing that Loftsson et al. were dealing with combinations of two drugs, one of which was negatively charged. Loftsson et al. do not mention cladribine or suggest both inclusion and non-inclusion association for it by itself in solid form. The presently claimed subject matter, a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex consisting of (a) an amorphous inclusion complex of cladribine with hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16, is in no way suggested by the combination of references. See especially Claim 1 and Claim 56, drawn to the complex. Again, we emphasize that the ratios disclosed by Schultz et al. are for simple mixtures, not complexes. There is no reasonable evidence that any of the features of applicants' invention not taught by Schultz et al. are taught by the cited combination of references. Withdrawal of the record rejection is earnestly solicited.

In the event that any issues remain, the Examiner is urged to telephone the undersigned so that such issues can be handled promptly. Further, favorable action in the form of a Notice of Allowance is believed to be next in order and is earnestly solicited. If the Examiner has no intention to allow the application, a personal interview with him and his supervisory Examiner is respectfully requested.

Respectfully submitted,

BUCHANAN INGERSOLL & ROONEY PC

July 30, 2010

Bv:

Mary Katherine Baumeister

Registration No. 26254

Customer No. 21839 703 836 6620

Electronic Patent	App	olication Fee	Transm	ittal		
Application Number:	10:	551205				
Filing Date:	14-Nov-2006					
Title of Invention:	Oral formulations of cladribine					
First Named Inventor/Applicant Name:	Nicholas S. Bodor					
Filer:	Mary Katherine Baumeister/Diana Francis					
Attorney Docket Number:	0056192-000024					
Filed as Large Entity						
U.S. National Stage under 35 USC 371 Filing	Fee	s				
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)	
Basic Filing:						
Pages:						
Claims:						
Miscellaneous-Filing:						
Petition:						
Patent-Appeals-and-Interference:						
Post-Allowance-and-Post-Issuance:						
Extension-of-Time:						
Extension - 1 month with \$0 paid		1251	1	130	130	

Description	Fee Code	Fee Code Quantity		Sub-Total in USD(\$)
Miscellaneous:				
	Tot	al in USD	(\$)	130

Electronic Ac	Electronic Acknowledgement Receipt				
EFS ID:	8126835				
Application Number:	10551205				
International Application Number:					
Confirmation Number:	4092				
Title of Invention:	Oral formulations of cladribine				
First Named Inventor/Applicant Name:	Nicholas S. Bodor				
Customer Number:	21839				
Filer:	Mary Katherine Baumeister/Diana Francis				
Filer Authorized By:	Mary Katherine Baumeister				
Attorney Docket Number:	0056192-000024				
Receipt Date:	30-JUL-2010				
Filing Date:	14-NOV-2006				
Time Stamp:	15:13:56				
Application Type:	U.S. National Stage under 35 USC 371				

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Information	:				
2	Extension of Time	005619224EOT.pdf	32539	no	1
	Extension of fille	003019224E01.pui	9849dce9867f1291950e618b9ae8a89f067 b5f7e	110	
Warnings:					
Information	:				
3	Amendment/Req. Reconsideration-After	0056192AMEND.pdf	240956	no	5
3	Non-Final Reject	0030132AMEND.pdi	7dab22c2229c397eb2e738bbba73f765ff25 f4eb	110	
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### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re F	Patent Application	n of		) MAIL STOP AMENDMENT				
Nicho	las Bodor et al.			) Group Art l	Group Art Unit: 1623			
Applic	ation No.: 10/55	1,205		Examiner:	Examiner: JONATHAN S LAU			
Filing	Date: Novembe	r 14, 2006		) Confirmation	on No.: 4092			
Title:	ORAL FORM	ULATIONS O	F CLADRIBINE	) ) )				
		AMI	ENDMENT/REPLY TR	RANSMITTAL LE	TTER			
P.O. E	nissioner for Pate 3ox 1450 ndria, VA 22313-							
Sir:								
Enclo	sed is a reply for	the above-ide	entified patent applicat	ion.				
$\boxtimes$	A Petition for	Extension of	Time is enclosed.					
	Terr 37 C.F.R. § 1	ninal Disclaim .20(d) are end	er(s) and the $\ \square\ $ \$ 70 closed.	☐ \$ 140 fee per	Disclaimer due und	er		
	Also enclosed	d is/are:						
	Small entity s	tatus is hereb	y claimed.					
$\boxtimes$	No additional	claim fee is re	equired.					
<u> </u>	An additional	claim fee is re	equired, and is calcula	ted as shown belo	ow:			
			AMENDE	ED CLAIMS				
		No. of Claims	Highest No. of Claims Previously Paid For	Extra Claims	Rate	Additional Fee		
Total C	Claims	56	78	0	x \$ 52 (1202)	\$		
Indepe	endent Claims	4	5	0	x \$ 220 (1201)			
☐ If A	mendment adds m	ultiple depende	nt claims, add \$ 390 (120	03)		\$		
Total (	Claim Amendmen	t Fee				\$		
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	-		o Deposit Account No.	. 02-4800 for the f	ee due.			
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$\boxtimes$	The Director is hereby authorized to charge any appropriate fees under 37 C.F.R. §§ 1.16, 1.17 and 1.20(d) and 1.21 that may be required by this paper, and to credit any overpayment, to Deposit Account No. 02-4800.							
	Respectfully submitted,							
			Buchanan Ind	GERSOLL & ROONE	Y PC			
Date	te July 30, 2010 By: Mary Kilheria Caunintee							
Customer No. 21839 Mary Katherine Baumeister Registration No. 26254								

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	) MAIL STOP AMENDMENT
Nicholas Bodor et al.	) Group Art Unit: 1623
Application No.: 10/551,205	) Examiner: JONATHAN S LAU
Filing Date: November 14, 2006	) Confirmation No.: 4092
Title: ORAL FORMULATIONS OF CLADRIBINE	) ) )
PETITION FOR	EXTENSION OF TIME
Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450	
Sir:	
The following extension of time is requ Action dated March 30, 2010 for	ested to: extend the period for response to the Office
One Month to July 30, 2010	
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☐ Charge <u>\$ 130</u> to credit card.	
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Res	spectfully submitted,
Bud	CHANAN INGERSOLL & ROONEY PC
Date: <u>July 30, 2010</u> By:	Mary Katherine Baumeister Registration No. 26254
<b>Customer No. 21839</b> 703 836 6620	

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P	ATENT APPL	Substitute for			ON RECORD	Α	Application or Docket Number 10/551,205 Filing Date 11/14/2006					
	AI	PPLICATION A	AS FILE (Column 1		(Column 2)		OTHER THAN  SMALL ENTITY OR SMALL ENTITY					
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	SEARCH FEE (37 CFR 1.16(k), (i), (i)		N/A		N/A		N/A		1	N/A		
	EXAMINATION FE (37 CFR 1.16(o), (p),	Ε	N/A		N/A		N/A		1	N/A		
	TAL CLAIMS CFR 1.16(i))		mir	us 20 = *		l	x \$ =		OR	x \$ =		
INDEPENDENT CLAIMS (37 CFR 1.16(h)) minus 3 = *							x \$ =		1	x \$ =		
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	MULTIPLE DEPEN	IDENT CLAIM PR	ESENT (3	7 CFR 1.16(j))								
* If 1	he difference in colu	umn 1 is less than	zero, ente	r "0" in column	2.		TOTAL			TOTAL		
	APP	LICATION AS (Column 1)	AMEND	DED — PART (Column 2)			SMAL	L ENTITY	OR		ER THAN ALL ENTITY	
AMENDMENT	07/30/2010	CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUSL PAID FOR	PRESENT Y EXTRA		RATE (\$)	ADDITIONAL FEE (\$)		RATE (\$)	ADDITIONAL FEE (\$)	
ME	Total (37 CFR 1.16(i))	* 56	Minus	** 78	= 0		x \$ =		OR	X \$52=	0	
I Z	Independent (37 CFR 1.16(h))	* 5	Minus	***6	= 0		x \$ =		OR	X \$220=	0	
ME	Application S	ize Fee (37 CFR 1	.16(s))									
1	FIRST PRESEN	NTATION OF MULTIF	LE DEPEN	DENT CLAIM (37	CFR 1.16(j))				OR			
							TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	0	
		(Column 1)		(Column 2)	) (Column 3)		'					
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Ä	Total (37 CFR 1.16(i))	*	Minus	**	=		x \$ =		OR	x \$ =		
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This collection of information is required by 37 CFR 1.16. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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BUCHANAN, INGERSOLL & ROONEY PC POST OFFICE BOX 1404 ALEXANDRIA, VA 22313-1404 EXAMINER

LAU, JONATHAN S

ART UNIT PAPER NUMBER

1623

DATE MAILED: 10/04/2010

APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 10/551,205 11/14/2006 Nicholas S. Bodor 0056192-000024 4092

TITLE OF INVENTION: ORAL FORMULATIONS OF CLADRIBINE

APPLN. TYPE	SMALL ENTITY	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSUE FEE	TOTAL FEE(S) DUE	DATE DUE
nonprovisional	NO	\$1510	\$300	\$0	\$1810	01/04/2011

THE APPLICATION IDENTIFIED ABOVE HAS BEEN EXAMINED AND IS ALLOWED FOR ISSUANCE AS A PATENT. PROSECUTION ON THE MERITS IS CLOSED. THIS NOTICE OF ALLOWANCE IS NOT A GRANT OF PATENT RIGHTS. THIS APPLICATION IS SUBJECT TO WITHDRAWAL FROM ISSUE AT THE INITIATIVE OF THE OFFICE OR UPON PETITION BY THE APPLICANT. SEE 37 CFR 1.313 AND MPEP 1308.

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10/551,205 TITLE OF INVENTION	11/14/2006 : ORAL FORMULATIC	ONS OF CLADRIBINE	Nicholas S. Bodor		0056192-000024	4092
APPLN. TYPE	SMALL ENTITY	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSUE F	EE TOTAL FEE(S) DUE	DATE DUE
nonprovisional	NO	\$1510	\$300	\$0	\$1810	01/04/2011
EXAM	INER	ART UNIT	CLASS-SUBCLASS	]		
LAU, JON	ATHAN S	1623	514-045000			
"Fee Address" ind PTO/SB/47; Rev 03-C Number is required.  3. ASSIGNEE NAME A PLEASE NOTE: Unl	ND RESIDENCE DATA	'Indication form ed. Use of a Customer A TO BE PRINTED ON 'ified below, no assignee	(1) the names of up to or agents OR, alternative (2) the name of a single registered attorney or a 2 registered patent attoristed, no name will be THE PATENT (print or type data will appear on the part of the p	vely, e firm (having as a m agent) and the names rneys or agents. If no printed.  be) atent. If an assignee	ember a 2 of up to name is 3	ocument has been filed for
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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/551,205	11/14/2006	Nicholas S. Bodor	0056192-000024	4092
21839 75	90 10/04/2010		EXAM	INER
BUCHANAN, IN	NGERSOLL & ROO	LAU, JON	ATHAN S	
POST OFFICE BC		ART UNIT	PAPER NUMBER	
ALEXANDRIA, V	'A 22313-1404		1623	
			DATE MAILED: 10/04/201	0

### **Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)**

(application filed on or after May 29, 2000)

The Patent Term Adjustment to date is 0 day(s). If the issue fee is paid on the date that is three months after the mailing date of this notice and the patent issues on the Tuesday before the date that is 28 weeks (six and a half months) after the mailing date of this notice, the Patent Term Adjustment will be 0 day(s).

If a Continued Prosecution Application (CPA) was filed in the above-identified application, the filing date that determines Patent Term Adjustment is the filing date of the most recent CPA.

Applicant will be able to obtain more detailed information by accessing the Patent Application Information Retrieval (PAIR) WEB site (http://pair.uspto.gov).

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Customer Service Center of the Office of Patent Publication at 1-(888)-786-0101 (571)-272-4200.

	Application No.	Applicant(s)					
Al-('	10/551,205	BODOR ET AL.					
Notice of Allowability	Examiner	Art Unit					
	Jonathan S. Lau	1623					
The MAILING DATE of this communication appe All claims being allowable, PROSECUTION ON THE MERITS IS herewith (or previously mailed), a Notice of Allowance (PTOL-85) NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATENT RI of the Office or upon petition by the applicant. See 37 CFR 1.313	(OR REMAINS) CLOSED in this app or other appropriate communication GHTS. This application is subject to	vilication. If not included will be mailed in due course. <b>THIS</b>					
1. 🔀 This communication is responsive to Applicant's Amendme	ent and Remarks, filed 30 Jul 2010.						
2. X The allowed claim(s) is/are 1,2,8,9,11-14,20,21,23-26,28,3	2,33,35,56,57,63,64 and 66-98.						
<ul> <li>3. ☐ Acknowledgment is made of a claim for foreign priority ur</li> <li>a) ☐ All b) ☐ Some* c) ☐ None of the:</li> <li>1. ☐ Certified copies of the priority documents have</li> </ul>							
2. Certified copies of the priority documents have							
3. Copies of the certified copies of the priority doc	cuments have been received in this r	national stage application from the					
International Bureau (PCT Rule 17.2(a)).							
* Certified copies not received:							
Applicant has THREE MONTHS FROM THE "MAILING DATE" noted below. Failure to timely comply will result in ABANDONM THIS THREE-MONTH PERIOD IS NOT EXTENDABLE.		complying with the requirements					
4. A SUBSTITUTE OATH OR DECLARATION must be subm INFORMAL PATENT APPLICATION (PTO-152) which give							
5. CORRECTED DRAWINGS ( as "replacement sheets") mus	t be submitted.						
(a) ☐ including changes required by the Notice of Draftspers	on's Patent Drawing Review ( PTO-9	948) attached					
1) ☐ hereto or 2) ☐ to Paper No./Mail Date							
<ul><li>(b) ☐ including changes required by the attached Examiner's Paper No./Mail Date</li></ul>	s Amendment / Comment or in the O	ffice action of					
Identifying indicia such as the application number (see 37 CFR 1 each sheet. Replacement sheet(s) should be labeled as such in the							
6. DEPOSIT OF and/or INFORMATION about the deposit attached Examiner's comment regarding REQUIREMENT							
Attachment(s)	5 D Nation of Information	alant Anni Carlan					
1. Notice of References Cited (PTO-892)	5. Notice of Informal Pa						
2. Notice of Draftperson's Patent Drawing Review (PTO-948)	6.  ☐ Interview Summary Paper No./Mail Date	ė .					
Information Disclosure Statements (PTO/SB/08),  7.   Examiner's Amendment/Comment  Paper No./Mail Date							
4. Examiner's Comment Regarding Requirement for Deposit	8. 🛛 Examiner's Stateme	nt of Reasons for Allowance					
of Biological Material	9.						
Jonathan Lau	/Shaojia Anna Jiang/						
Patent Examiner	Supervisory Patent Exa	miner, Art Unit 1623					
Art Unit 1623							

### **EXAMINER'S AMENDMENT**

An examiner's amendment to the record appears below. Should the changes and/or additions be unacceptable to applicant, an amendment may be filed as provided by 37 CFR 1.312. To ensure consideration of such an amendment, it MUST be submitted no later than the payment of the issue fee.

Authorization for this examiner's amendment was given in a telephone interview with Kathy Baumeister on 27 Sep 2010.

The application has been amended as follows:

### Amendment to the Claims

• Claims 25, 27 and 28 are amended as follows:

Claim 25. (Currently Amended) A method for the treatment of symptoms of a cladribine-responsive condition a condition selected from the group consisting of multiple sclerosis, rheumatoid arthtritis and leukemia in a subject suffering from said symptoms condition comprising orally administering to said subject a pharmaceutical composition comprising a complex cladribine-cyclodextrin complex which is an intimate amorphous admixture consisting of (a) an amorphous inclusion complex of cladribine with the amorphous cyclodextrin hydroxypropyl-β-cyclodextrin and (b) amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex, formulated into a solid oral dosage form, said composition comprising no significant amount of free crystalline cladribine therein, said composition having a weight ratio of cladribine to said amorphous cyclodextrin of from about 1:10 to about 1:16.

Claim 27. (Canceled)

Application/Control Number: 10/551,205 Page 3

Art Unit: 1623

Claim 28. (Currently Amended) The method according to Claim <del>27</del> <u>25</u>, wherein the <del>cladribine-responsive</del> condition is multiple sclerosis.

Application/Control Number: 10/551,205 Page 4

Art Unit: 1623

### **DETAILED ACTION**

This Office Action is responsive to Applicant's Amendment and Remarks, filed 30 Jul 2010.

This application is the national stage entry of PCT/US04/09387, filed 26 Mar 2004; and claims benefit of provisional application 60/458,922, filed 28 Mar 2003; and claims benefit of provisional application 60/484,756, filed 02 July 2003; and claims benefit of provisional application 60/541,247, filed 04 Feb 2004.

The filing date of the instant claims 12, 83, 85 and 89 are deemed to be the filing date of the instant application which is the filing date of PCT/US04/09387, 26 Mar 2004. The filing date of instant claims 1, 2, 8, 9, 11, 56, 57, 63, 64, 82, 84 and 86-98 are deemed to be the filing date of provisional application 60/541,247, filed 04 Feb 2004.

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-28, 32, 33, 35, 56, 57, 63, 64 and 66-98 are pending in the current application. Claims 13, 14, 20, 21, 23-28, 32, 33, 35 and 67-81, drawn to non-elected inventions, are rejoined herein. Claim 27 is canceled by Examiner's Amendment herein. Claims 1, 2, 8, 9, 11-14, 20, 21, 23-26, 28, 32, 33, 35, 56, 57, 63, 64 and 66-98 are allowed herein.

**Reasons for Allowance** 

Rejections Withdrawn

Applicant's Remarks, filed 30 Jul 2010, with respect to claims 1, 2, 8, 9, 11, 56, 57, 63, 64 and 82-98 rejected under 35 U.S.C. 103(a) as being unpatentable over Schultz et al. (US Patent 6,194,395, published 27 Feb 2001, of record) in view of Pitha (US Patent 4,727,064, issued 23 Feb 1988, provided by Applicant in IDS mailed 4 Apr 2008) and in view of Loftsson J Pharm Sci 2002 (Journal of Pharmaceutical Sciences, 2002, 91(11), p2307-2316, cited in PTO-892) has been fully considered and is persuasive, as Applicant's remarks are persuasive that Schultz et al. in view of Pitha and in view of Loftsson J Pharm Sci 2002 does not teach a composition comprising amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex. Applicant notes that dependent claims 12 and 66 were omitted from the statement of rejection, however this is moot in view of the withdrawal of this rejection in view of Applicant's Remarks.

This rejection has been withdrawn.

The following is an examiner's statement of reasons for allowance:

The closest prior art is Schultz et al. in view of Pitha and in view of Loftsson J Pharm Sci 2002.

Applicant's Remarks, filed 30 Jul 2010, have been fully considered and found to be persuasive that Schultz et al. in view of Pitha and in view of Loftsson J Pharm Sci 2002 does not teach all features of the instant invention as claimed. None of Schultz et al., Pitha or Loftsson J Pharm Sci 2002 teach or fairly suggest the instant composition comprising amorphous free cladribine associated with said amorphous cyclodextrin as a

non-inclusion complex, and no evidence teaches or fairly suggests that this is an inherent property in the composition that is necessarily present. For example, Pitha teaches crystalline drugs and cyclodextrins that have the ability to form inclusion complexes which are intrinsically amorphous (column 1, lines 25-35), however Pitha does not teach or fairly suggest amorphous free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex and neither Schultz et al. nor Loftsson J Pharm Sci 2002 remedy this. For example, Schultz et al. does not teach or fairly suggest that formation of free cladribine associated with said amorphous cyclodextrin as a non-inclusion complex is necessarily amorphous but rather suggests cladribine will crystallize out of solution (column 1, lines 60-65 and column 2, lines 1-15).

Therefore the prior art does not teach or fairly suggest the instant invention as claimed.

Any comments considered necessary by applicant must be submitted no later than the payment of the issue fee and, to avoid processing delays, should preferably accompany the issue fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allowance."

### Conclusion

Claims 1, 2, 8, 9, 11-14, 20, 21, 23-26, 28, 32, 33, 35, 56, 57, 63, 64 and 66-98 are allowed herein.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan S. Lau whose telephone number is 571-270-3531. The examiner can normally be reached on Monday - Thursday, 9 am - 4 pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Shaojia Anna Jiang can be reached on 571-272-0627. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jonathan Lau Patent Examiner Art Unit 1623 /Shaojia Anna Jiang/ Supervisory Patent Examiner Art Unit 1623

# Issue Classification



ı	Application/Control No.	Applicant(s)/Patent Under Reexamination
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ı	10551205	BODOR ET AL.
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ı	Jonathan S Lau	1623

		ORIG	NAL						INTERNATIONAL	CLAS	CLASSIFICATION			
	CLASS SUBCLASS					CLAIMED						NON-CLAIMED		
CROSS REFERENCE(S)				Α	6	1	К	31 / 7076 (2006.01.01)						
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	☐ Claims renumbered in the same order as presented by applicant ☐ CPA ☐ T.D. ☐ R.1.47														
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/Jonathan S Lau/ Examiner.Art Unit 1623	9/27/10		ns Allowed:	
(Assistant Examiner)	(Date)	55		
/Shaojia Anna Jiang/ Supervisory Patent Examiner.Art Unit 1623	09/27/2010	O.G. Print Claim(s)	O.G. Print Figure	
(Primary Examiner)	(Date)	1	none	

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# Search Notes

Application/Control No.	Applicant(s)/Patent Under Reexamination
10551205	BODOR ET AL.
Examiner	Art Unit
Jonathan S Lau	1623

	SEARCHED			
Class Subclass Date Examiner				
514	46, 58	9/27/2010	JSL	

SEARCH NOTES					
Search Notes	Date	Examiner			
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker)	3/26/2008	JSL			
EAST - see attached notes	3/26/2008	JSL			
Google Scholar - see attached notes	3/26/2008	JSL			
EAST - see attached notes	9/10/2009	JSL			
Google Scholar - see attached notes	9/10/2009	JSL			
STN - CAPlus file - see attached notes	9/10/2009	JSL			
EAST - see attached notes	3/25/2010	JSL			
Google Scholar - see attached notes	3/25/2010	JSL			
EAST - inventor name search (Nicholas Bodor; Yogesh Dandiker) updated	9/27/2010	JSL			

	INTERFERENCE SEARCH		
Class	Subclass	Date	Examiner
514	46, 58	9/27/2010	JSL

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### **AMENDMENTS TO THE SPECIFICATION:**

Please replace the paragraph at page 22, lines 8-16 of the specification with the following amended paragraph:

The compositions of the invention are particularly suitable as modalities for the treatment of any cladribine-responsive disease. Several disease states responsive to cladribine are well-documented in the literature (see *infra*). For any target disease state, an effective amount of the complex cladribine-cyclodextrin complex complex, *i.e.* the amorphous mixture of the optimized amorphous saturated cladribine-amorphous cyclodextrin complex with amorphous free cladribine as described above is used (e.g., an amount affective effective for the treatment of multiple sclerosis, rheumatoid arthritis, or leukemia).

Please replace the paragraph at page 23, lines 7-28, of the specification with the following amended paragraph:

11-10-12

Moreover, the route of administration for which the therapeutically effective dosages are taught in the literature should be taken into consideration. While the instant compositions optimize the bioavailability of cladribine following oral administration, it will be appreciated that even optimal bioavailability from oral dosage forms is not expected to approach bioavailability ebtain obtained after intravenous administration, particularly at early time points. Thus, it is often appropriate to increase a dosage suggested for intravenous administration to arrive at a suitable dosage for incorporation into a solid oral dosage form. At the present time, it is envisioned that, for the treatment of multiple sclerosis, 10 mg of cladribine in the instant complex cladribine-cyclodextrin complex in the instant solid dosage form would be administered once per day for a period of five to seven days in the first month, repeated for another period of five to seven days in the second month, followed by ten months of no treatment. Alternatively the patient would

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APPLICATION NO.	FILING DATE		FIRST NAMED INVENT	ГOR		ATTOR	NEY DOCKET NO.	CONFIRMATION	NO.
10/551,205	11/14/2006		Nicholas S. Bodor			005	6192-000024	4092	
TITLE OF INVENTION	: ORAL FORMULATIO	ONS OF CLADRIBINE							
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Please check the appropr	iate assignee category or	categories (will not be pr	inted on the patent):	☐ Inc	dividual 😾 Co	rporatio	n or other private gro	ap entity 🚨 Gove	rnment
4a. The following fee(s)  lssue Fee  Publication Fee (N  Advance Order -	No small entity discount p		o. Payment of Fee(s): (I  A check is enclose  Payment by credit  The Director is her overpayment, to De	ed. card. F	Form PTO-2038	is attacl	hed.		ıy form).
_ "	tus (from status indicated	,	b. Applicant is no	longer	claiming SMAL	L ENTI	TY status. See 37 CF	R 1 27(g)(2)	,
NOTE: The Issue Fee an	d Publication Fee (if req	uired) will not be accepted tes Patent and Trademark	d from anyone other tha						party in
Authorized Signature	, , , , ,	huin Bauneix	,	3	Date Janı	ıary	4, 2011		
Typed or printed nam	<sub>e</sub> Mary Kathe	erine Baumeist	er		Registration No	-	26254		
submitting the complete this form and/or suggest Box 1450, Alexandria, V Alexandria, Virginia 223	d application form to the ions for reducing this but irginia 22313-1450. DC i13-1450.	FR 1.311. The informatic U.S.C. 122 and 37 CFR USPTO. Time will vary den, should be sent to the NOT SEND FEES OR Copersons are required to respect to the sent of the sent sent of the sent sent sent sent sent sent sent sen	depending upon the in e Chief Information Of COMPLETED FORMS	ndividua ficer, U S TO TI	al case. Any cor J.S. Patent and T HIS ADDRESS.	nments Fradema . SEND	on the amount of timers. Office, U.S. Departor. Commissioner for the com	te you require to continue to	omolete

Electronic Patent A	Electronic Patent Application Fee Transmittal						
Application Number:	10:	551205					
Filing Date:	14	-Nov-2006					
Title of Invention:	ORAL FORMULATIONS OF CLADRIBINE						
First Named Inventor/Applicant Name:	Nicholas S. Bodor						
Filer:	Mary Katherine Baumeister/Diana Francis						
Attorney Docket Number:	0056192-000024						
Filed as Large Entity							
U.S. National Stage under 35 USC 371 Filing	Fee	s					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Basic Filing:							
Pages:							
Claims:							
Miscellaneous-Filing:							
Petition:							
Patent-Appeals-and-Interference:							
Post-Allowance-and-Post-Issuance:							
Utility Appl issue fee		1501	1	1510	1510		
Publ. Fee- early, voluntary, or normal		1504	1	300	300		

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Extension-of-Time:				
Miscellaneous:				
Printed copy of patent - no color	8001	6	3	18
	Tot	al in USD	(\$)	1828

Electronic Ack	knowledgement Receipt
EFS ID:	9157246
Application Number:	10551205
International Application Number:	
Confirmation Number:	4092
Title of Invention:	ORAL FORMULATIONS OF CLADRIBINE
First Named Inventor/Applicant Name:	Nicholas S. Bodor
Customer Number:	21839
Filer:	Mary Katherine Baumeister/Diana Francis
Filer Authorized By:	Mary Katherine Baumeister
Attorney Docket Number:	0056192-000024
Receipt Date:	04-JAN-2011
Filing Date:	14-NOV-2006
Time Stamp:	11:06:26
Application Type:	U.S. National Stage under 35 USC 371

# **Payment information:**

Submitted with Payment	yes
Payment Type	Credit Card
Payment was successfully received in RAM	\$1828
RAM confirmation Number	9174
Deposit Account	
Authorized User	

# File Listing:

Document Number	Document Description	le Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
		772	•		

_		IssuaFee pdf	118520		1
1	Issue Fee Payment (PTO-85B)	Issue Fee.pdf	1b54aae6db7cf01858924fed8ebc312ed72 ce8dd	no	1
Warnings:					
Information:					
2	Fee Worksheet (PTO-875)	fee-info.pdf	33596	no	2
_	ree wonsheet (110 075)	·	ce0d1bfcc1aba05d6eab1c57381ae34f5b94 ab1c		
Warnings:			•		-
Information:					
		Total Files Size (in bytes):	15	52116	

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### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



### UNITED STATES PATENT AND TRADEMARK OFFICE

01/26/2011

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450

Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO.	ISSUE DATE	PATENT NO.	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/551,205	02/15/2011	7888328	0056192-000024	4092

21839 7590

BUCHANAN, INGERSOLL & ROONEY PC POST OFFICE BOX 1404 ALEXANDRIA, VA 22313-1404

### ISSUE NOTIFICATION

The projected patent number and issue date are specified above.

### **Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)**

(application filed on or after May 29, 2000)

The Patent Term Adjustment is 16 day(s). Any patent to issue from the above-identified application will include an indication of the adjustment on the front page.

If a Continued Prosecution Application (CPA) was filed in the above-identified application, the filing date that determines Patent Term Adjustment is the filing date of the most recent CPA.

Applicant will be able to obtain more detailed information by accessing the Patent Application Information Retrieval (PAIR) WEB site (http://pair.uspto.gov).

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Application Assistance Unit (AAU) of the Office of Data Management (ODM) at (571)-272-4200.

APPLICANT(s) (Please see PAIR WEB site http://pair.uspto.gov for additional applicants):

Nicholas S. Bodor, Bal Harbour, FL; Yogesh Dandiker, Toronto, CANADA;

IR103 (Rev. 10/09)

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

### POWER OF ATTORNEY TO PROSECUTE APPLICATIONS BEFORE THE USPTO I hereby revoke all previous powers of attorney given in the application identified in the attached statement under 37 CFR 3.73(b) I hereby appoint: Practitioners associated with the Customer Number: 13974 Practitioner(s) named below (if more than ten patent practitioners are to be named, then a customer number must be used): Registration Name Registration Number Number as attorney(s) or agent(s) to represent the undersigned before the United States Patent and Trademark Office (USPTO) in connection with any and all patent applications assigned only to the undersigned according to the USPTO assignment records or assignment documents attached to this form in accordance with 37 CFR 3.73(b). Please change the correspondence address for the application Identified in the attached statement under 37 CFR 3.73(b) to: 13974 The address associated with Customer Number: ORFirm or Individual Name Address City State Zip Country Telephone Email Assignee Name and Address: ARES TRADING S.A. ZONE INDUSTRIELLE D L'OURIETTAZ CH-1170 AUBONNE, SWITZERLAND A copy of this form, together with a statement under 37 CFR 3.73(b) (Form PTO/SB/96 or equivalent) is required to be filed in each application in which this form is used. The statement under 37 CFR 3.73(b) may be completed by one of the practitioners appointed in this form if the appointed practitioner is authorized to act on behalf of the assignee, and must identify the application in which this Power of Attorney is to be filed. SIGNATURE of Assignee of Record idual whose signature and title is supplied below is authorized to act on behalf of the assignee 10.04.2012 Signature Date Bjoern Colin KAHRS Name Telephone **Authorized Representative**

Title

This collection of Information is required by 37 CFR 1.31, 1.32 and 1.33. The information is required by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by \$33.24251421365555711111112. This collection is estimated to take 3 minutes to complete, including gathering, preparing, and submitting the completed application to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

Title

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2.

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STATEMENT UNDE	R 37 CFR 3.73(b)
Applicant/Patent Owner: ARES TRADING S.A.	
Application No./Patent No.: 10/551,205/788328	Filed/Issue Date: 11-14-2006/02-15-2011
Titled: ORAL FORMULATIONS OF CLADRIBINE	
ARES TRADING S.A. , a Corpora	ation
(Name of Assignee) (Type of	Assignee, e.g., corporation, partnership, university, government agency, etc.
states that it is:	
1. X the assignee of the entire right, title, and interest in;	
an assignee of less than the entire right, title, and interest i     (The extent (by percentage) of its ownership interest is	
3. the assignee of an undivided interest in the entirety of (a co	omplete assignment from one of the joint inventors was made)
the patent application/patent identified above, by virtue of either:	
the United States Patent and Trademark Office at Reel copy therefore is attached.	n/patent identified above. The assignment was recorded in, Frame, or for which a
B. A chain of title from the inventor(s), of the patent application	n/patent identified above, to the current assignee as follows:
1. From: Nicholas S. Bodor et al	To: IVAX CORPORATION
The document was recorded in the United States  Reel 018337 , Frame 0636	s Patent and Trademark Office at
2. From: IVAX CORPORATION	To: ARES TRADING S.A.
The document was recorded in the United States	s Patent and Trademark Office at
Reel 018337 , Frame 0696	, or for which a copy thereof is attached.
3. From:	To:
The document was recorded in the United States	
Reel, Frame	or for which a copy thereof is attached.
Additional documents in the chain of title are listed on a su	upplemental sheet(s).
As required by 37 CFR 3.73(b)(1)(i), the documentary evidence or concurrently is being, submitted for recordation pursuant to 3	e of the chain of title from the original owner to the assignee was, 7 CFR 3.11.
[NOTE: A separate copy (i.e., a true copy of the original assign accordance with 37 CFR Part 3, to record the assignment in the	nment document(s)) must be submitted to Assignment Division in records of the USPTO. See MPEP 302.08]
The undersigned (whose title is supplied below) is authorized to act on	behalf of the assignee.
/Martin A. Bruehs/	MAY 24, 2013
Signature	Date
MARTIN A. BRUEHS	Attorney for Applicant(s)
Printed or Typed Name	Title

This collection of information is required by 37 CFR 3.73(b). The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

### Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- A record from this system of records may be disclosed, as a routine use, in the course of
  presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to
  opposing counsel in the course of settlement negotiations.
- A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Electronic Acknowledgement Receipt				
EFS ID:	15868660			
Application Number:	10551205			
International Application Number:				
Confirmation Number:	4092			
Title of Invention:	ORAL FORMULATIONS OF CLADRIBINE			
First Named Inventor/Applicant Name:	Nicholas S. Bodor			
Customer Number:	21839			
Filer:	Martin A. Bruehs/Louie Malloy			
Filer Authorized By:	Martin A. Bruehs			
Attorney Docket Number:	0056192-000024			
Receipt Date:	24-MAY-2013			
Filing Date:	14-NOV-2006			
Time Stamp:	15:44:33			
Application Type:	U.S. National Stage under 35 USC 371			

# **Payment information:**

Submitted with Payment	no
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# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1		DOC.PDF	91689	yes	3
			0ca5964082b7640fd59aa3cfd62ad8a26053 eb0a		

Multipart Description/PDF files in .zip description				
Document Description	Start	End		
Power of Attorney	1	1		
Assignee showing of ownership per 37 CFR 3.73.	2	3		

### Warnings:

#### Information:

Total Files Size (in bytes):	91689

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#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



21839

### United States Patent and Trademark Office

United States Patent and Trademark Office
Address: COMMISSIONER FOR PATENTS
PO. Box 1450
Alexandria, Virgina 22313-1450
www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER

BUCHANAN, INGERSOLL & ROONEY PC

10/551,205

POST OFFICE BOX 1404 ALEXANDRIA, VA 22313-1404

FILING OR 371(C) DATE 11/14/2006

FIRST NAMED APPLICANT Nicholas S. Bodor

ATTY. DOCKET NO./TITLE 0056192-000024

**CONFIRMATION NO. 4092 POWER OF ATTORNEY NOTICE** 

Date Mailed: 06/24/2013

# NOTICE REGARDING CHANGE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 06/21/2013.

 The Power of Attorney to you in this application has been revoked by the assignee who has intervened as provided by 37 CFR 3.71. Future correspondence will be mailed to the new address of record(37 CFR 1.33).

/dolipscomb/						
O(f) ( D ) M	Application Assistance Heit (F74)	070 1000	(574) 070	1000	4 000 704	

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101



13974

### United States Patent and Trademark Office

United States Patent and Trademark Office
Address: COMMISSIONER FOR PATENTS
PO. Box 1450
Alexandria, Virginia 22313-1450
www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER

FILING OR 371(C) DATE

FIRST NAMED APPLICANT

ATTY. DOCKET NO./TITLE 0056192-000024

10/551,205

**DENTONS US LLP** 

P.O. BOX 061080 Chicago, IL 60606-1080 11/14/2006

Nicholas S. Bodor

**CONFIRMATION NO. 4092** POA ACCEPTANCE LETTER

\*OC00000062151026\*

Date Mailed: 06/24/2013

### NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 06/21/2013.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

/dolipscomb/			

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101