

ish medical bulletin.
46, no. 3 (June 1999)
neral Collection
1 DA545
ceived: 07-24-1999

Danish Medical Bulletin

JOURNAL OF THE HEALTH SCIENCES
No. 3. June 1999. Vol. 46. Pages 183-290

DOCTOR OF MEDICAL SCIENCE

Intraluminal pH of the human gastrointestinal tract. *Jan Fallingborg* 183

On the pathophysiology of late onset non-insulin dependent diabetes mellitus. Current controversies and new insights. *Allan Vaag* 197

Left ventricular systolic function after acute myocardial infarction: prognostic importance, relation to congestive heart failure and as a target for intervention. *Lars Kober* 235

ORIGINAL ARTICLES

Gastric mucosal interleukin-8 and IL-8 antibody concentrations related to prevailing *Helicobacter pylori* infections. A Danish-Albanian study. *Ole Haagen Nielsen, Thomas Horn, Skerdi Prifti, Peter Peichl, Jens Henrik Scheibel & Ivan James Dalton Lindley* 249

Cardiovascular effects of oesophageal dilation under general anaesthesia. *Charlotte Holm Jakobsen, Verner Rasmussen & Jacob Rosenberg* 252

Risk factors in wound infections after laparotomy in obstetrics and gynaecology. *Isabel de la Fuente Fonnest, Bjarne Stigsby & Lars Heisterberg* 254

Risk indicators of disability pension. A 15 year follow-up study. *Fin Biering Sorensen, Jens Lund, Ola J Høydalsmo, Erik M. Darre, Allan Deis, Peter Kryger & Cecilia Florvall Müller* 258

The Danish National Hospital Register A valuable source of data for modern health sciences. *Tavs Folmer Andersen, Mette Madsen, Jorgen Jørgensen, Lene Mellemkjær & Jørgen H. Olsen* 263

The Danish Prevention Register A comprehensive health and socio-economic, individual based register. *Annette Soberg Roed, Claus Juhl & Finn Kamper-Jørgensen* 269

ABSTRACTS OF PH.D. DISSERTATIONS. See cover page 2

MEDICAL THESES AND PH.D. DISSERTATIONS. 288

SUMMARIES

Bibliotek for Læger. A journal devoted to medical history, ethics, philosophy and clinical theory, founded in 1809. 289



PROPERTY OF THE
NATIONAL
LIBRARY OF
MEDICINE

Published by the
Danish Medical Association and Danish Medical Society
with support from the Faculties of Health Sciences
of the Universities of Copenhagen,
Aarhus, and Odense, and the Danish National Board of Health
This material was copied

Intraluminal pH of the human gastrointestinal tract

Jan Fallingborg

This review has been accepted as a thesis together with five previously published papers, by the University of Aarhus, December 14, 1998, and defended on April 23, 1999.

Department of Medical Gastroenterology, Aalborg Sygehus.

Correspondence: Jan Fallingborg, Duebrodrevvej 23, DK-9000 Aalborg.

Official opponents: Per Brobech Mortensen, MD, Peter M. Funch Jensen, MD, and Steen Lindkær Jensen, professor, MD.

1. INTRODUCTION

The recent and increasing interest in the pH of the contents of the gastrointestinal tract is well founded. One reason is the development of several oral preparations which release the drug at a site and/or at a rate determined by the pH of the surrounding fluid (1-10). Precise knowledge of the pH profile of the gut, including the interindividual variability, is a prerequisite for designing optimal delivery systems based on this concept.

From a theoretical point of view the intraluminal pH of the gut primarily is determined by three factors: the absorption and secretion of acid and base by the intestinal epithelium, the bacterial degradation of ingested food, and the rate of transport of the intestinal contents through the gut.

Determinations of the intraluminal pH of the gastrointestinal tract have been performed for several decades. Early studies were based on aspiration of gastric or intestinal fluids, or measurements on faeces. However, the development of small pH-sensitive, radiotransmitting capsules has provided a method which allows pH-determination under almost physiological conditions.

The aims of the investigations (11-15) presented in this review were as follows: to evaluate a method of determining local intraluminal pH of the gut based upon the combined use of pH-capsules and fluoroscopy; to use this method in studying the pH-profile of the gastrointestinal tract in healthy human subjects and in children; to study the effect of various factors on this pH-profile.

The present review discusses current knowledge of the intraluminal intestinal pH with special reference to data generated with pH-capsules.

2. METHODS

pH of the gut is a very sensitive parameter, influenced by many kinds of outside impacts. A major problem in measuring gastrointestinal pH is, therefore, that it involves the introduction of a tube or a transducer into the gut of the subject studied, and the procedure in itself might induce changes in the pH it measures. Various methods have been developed with the intent to minimize the physical trauma to the subject, thereby minimizing the possible effect of the procedure on the recorded pH.

2.1 INTUBATION TECHNIQUES

Naso-intestinal intubation techniques are easily performed, but pose a number of limitations that can provide less than accurate results. For example, hypersalivation caused by the intubation may dilute the aspirate and thereby raise the gastric pH due to the alkaline pH of saliva. Reflux through the pyloric sphincter due to nausea caused by the procedure may also falsely raise the gastric pH. Conversely, gastric fluid transported along the tube may falsely decrease the pH in the small intestine.

2.1.1 Aspiration technique

Aspiration of gastric fluids through catheters has been employed for decades, and it has the advantage of direct measurement on the fluid

nique, but Barbero *et al* (16) studied the entire intestinal tract – from the stomach to the rectum – of infants (aged two weeks to three months) by an aspiration technique using an infant Miller-Abbott tube with a balloon inflated with 5-8 ml air. The balloon was inflated when the tube was placed in the duodenum. The tube then passed rapidly to the terminal ileum, and more slowly, within 24-36 hours, through the colon to the rectum. The ethical aspects of performing this potentially hazardous procedure in infants are mentioned in their article.

2.1.2 Dialysis bags

A method introduced by Marner (17), later described and utilized by Rume and Koster (18) involved the use of a suitable membrane filled with distilled water and placed in the stomach via a connecting tube. After a time the contents of the bag were withdrawn, and the hydrogen-ion content was determined. This method is useful in determining an average pH level in the stomach, but does not reflect rapid changes in gastric pH.

2.1.3 Electrodes

Intubated glass electrodes or antimony electrodes may be used to determine the intraluminal pH of the upper gastrointestinal tract. Electrodes have the advantage over dialysis bags and aspiration techniques that they are capable of reflecting pH fluctuations occurring in the gut lumen. Furthermore, registrations can be made continuously on ambulant patients when the electrode is connected to an ambulatory recording system. Savarino *et al* (19) found an excellent correlation between values obtained with an intragastric pH monitoring equipment (Digitrapper 6000, Synetics, Sweden) and simultaneous gastric aspiration.

2.2 TUBELESS TECHNIQUES

2.2.1 Chemical methods

Indirect methods of determining the presence or absence of gastric acidity have been suggested. One method involved the use of cationic exchange resins to determine the absence or presence of free hydrochloric acid. The procedure depends on the dissociation of the resin by the hydrochloric acid in the stomach. The cation is absorbed and excreted in the urine in the presence of free acid only (20).

Another technique involves the use of pills containing methylene blue wrapped in an indigestible sac tied by a catgut suture. Free acid, if present in the stomach, dissolves the suture and allows the release of the dye. The dye is then absorbed and excreted in the urine. Lack of free acid prevents the release of the dye. This procedure (Desmoid pill technique) was first devised by Sahli in 1905 and reintroduced by Levere and Palmer in 1960 (21).

Obviously, these methods are indirect, nonspecific, time consuming, and consequently of limited value.

2.2.2 Radiotelemetry capsules

In 1957 the first radiotelemetry capsule for measuring pH was invented by Jacobsen and MacKay (22). A copolymer resin, which changed its dimensions with changes in pH, was used as the transducer. The change in dimensions was transferred to an iron core moving inside a coil, and this caused a shift in the frequency of the oscillator in the capsule. The very slow response time of the transducer decreased the clinical value of the capsule, and there are no reports of its use for clinical investigations. In 1959 a telemetry capsule using an antimony electrode as transducer was developed by Nöller (23), and this pH-capsule was later referred to as the Heidelberg capsule. The pH measuring cell in the Heidelberg capsule is made up of an annular external antimony electrode which is in contact with the surrounding fluid, and an internal AgCl electrode separated by a semi-permeable membrane (24). pH changes the potential difference between the electrodes, which in turn control the frequency of the radio transmitter. The main disadvantages with the early type of this capsule were the short life and serious drift problems, caused by an oxidation of the antimony electrode in the presence of intestinal fluids (25, 26). The limitations of this capsule made it suitable only for studies of the stomach, and this is still the case for the Heidelberg

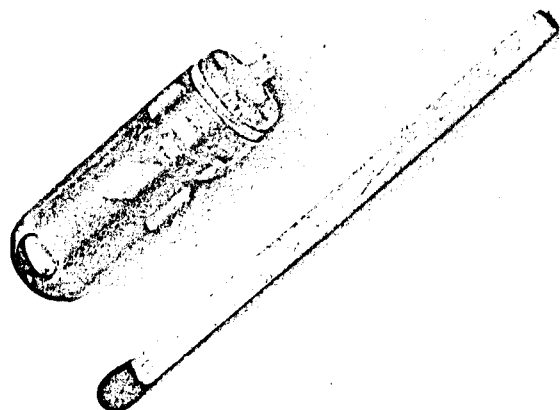
unteers and patients. The study subjects initially passed a tube throughout the gastrointestinal canal and then the capsule was pulled through. Later improvement of the capsule has made it a more reliable tool of pH measurement, but the accuracy of the capsule is still only about 0.5 pH units (28, 29), and the lifetime is only 22 h (29). In 1965-6 two pH capsules using glass electrodes were designed: a Japanese capsule (30), which, however, has never later been mentioned in clinical studies, and a capsule manufactured by Remote Control Systems Ltd, London, England, referred to in the present paper as the RCS capsule (31). The main problem with the pH glass electrode is the high impedance. The electrode must also be sealed to prevent development of leakage paths, which shunt the potential produced by the transducer. The early RCS capsule depended on epoxy resins to form the seal, but moisture caused a gradual deterioration in the pH response and a short lifetime for these capsules. Later a covering layer of glass was added which increased the lifetime of the capsule to more than one month (32). The Japanese capsule and the RCS capsules respond to alterations in pH by changing the transmission frequency of the transmitter in the capsule. The RCS capsule uses a FM oscillator with a transmission frequency of approximately 400 kHz, whereas the transmission frequency of the Japanese capsule is 1.9 MHz. The electrical circuits of the two capsules share many aspects. A glass pH-electrode is in contact with the surrounding fluid and an Ag/AgCl reference electrode is situated in the battery cap of the capsule. Variation in potential difference between the pH-electrode and the reference electrode causes a change in the capacitance of a reverse biased diode (varicap), with a corresponding change in output frequency (32, 33). The response in transmission frequency of the RCS capsule is linear in the interval between pH 1 and pH 9 (33, 34).

A small number of other types of radiotelemetry pH-capsules have been developed but the Heidelberg capsule and the RCS capsule (Fig. 1) are those most frequently used in published clinical studies.

2.2.2.1 Methodological problems using the pH-capsule

a. Localization of the capsule

When an untethered capsule is ingested, it travels freely throughout the gastrointestinal canal, and a crucial point is how to determine the precise location of the capsule in the gut at the time of pH measurement. The location which poses most problems is the ileo-caecal region. By using a radio directional antenna to determine the position at which the maximum signal strength of transmission is received, the regional localization of the capsule can be made with an accuracy of 5 cm (35, 36), but it is not possible to determine by this method whether it is located in the caecum or in the terminal ileum. Fluoroscopy enables a more precise localization, but if the gas content of the colon is sparse, the distinction between the terminal ileum



and the caecum may be difficult. The use of radiopaque contrast (15) or gamma scintigraphy technique (35) to visualize the region may be an advantage. Repeated fluoroscopic investigations with short intervals in between can be of great help, especially when sudden changes in pH occur. It is often observed that a capsule located in the ileocaecal region has dropped 5 to 10 cm downwards (i.e. into the caecum) between two fluoroscopic investigations, and usually this is associated with a sharp decrease in pH. However, it must be recognized that even with the use of frequent fluoroscopic determinations of the location of the capsule, and even with the assistance of a skilled radiologist, a small number of the determinations may be incorrect. The major problem with repeated fluoroscopic investigations is the radiation exposure. The length of radiation must be kept as short as possible, and this may best be achieved with the collaboration of a radiologist.

An improved method of localisation of the pH-capsule in the caecum was used by *Sasaki et al* (37). A contrast colonogram was obtained before the investigation. The receiver was connected to a computer-assisted analysing system, and when pH sharply decreased by 1 pH unit/min or more (i.e. when the capsule entered the caecum) the system generated a beeping sound. A plain abdominal x-ray was performed and superimposed on the previously performed contrast colonogram, and the presence of the capsule in the caecum could be verified. Thereafter, the position of the capsule in the colon was determined at 2-h intervals with a radio directional antenna. In two patients a plain x-ray film was taken after a measurement of pH by the antenna, and the position of the capsule judged by the antenna method was correct. The study demonstrates, that if the configuration of the colon is known, the localization of the capsule can be reasonably well determined by the use of a radio directional antenna. The method is elegant and the computer-assisted analysing system could be of great help in monitoring pH in the colon of patients with chronic inflammatory bowel disease, where very low pH values have been reported (14). However, a contrast colonogram may not be available in all patients, and the topographic position of the colon may be slightly influenced by the position (upright or succumb) of the patient.

b. Gastrointestinal transit of the capsule

When the capsule is ingested, it is freely mobile and its localization in the gut will be determined by the propulsion movements of the gastrointestinal myometrium. The gastric emptying of large (>1 mm) particles is dependent on the interdigestive migrating myoelectric complex (IMMC) (38, 39). When solid food is present in the stomach it contracts 3-4 times per minute, and the pylorus is partially opened, allowing liquid and small particles to pass. When the stomach is empty of food, several phases of myoelectric activity occur, ending with the phase III, the IMMC, which consists of an opening of the pylorus and of 3-4 peristaltic contractions from the stomach to the caecum, allowing emptying of the stomach of undigested material ("the housekeeper wave") (40, 41). This cyclic pattern of events occurs on average every two hours in fasting humans, but it is interrupted when food is ingested. The gastric residence time (GRT) of the pH-capsule therefore depends on the dietary state of the subject. In a fasting subject the average GRT of the capsule is 1.1 to 1.9 hours (11-13, 29, 42). A small liquid meal prolongs the mean GRT to 2.6 hours, and frequent intakes of food increases it to more than 14.5 hours (42). This important aspect concerning the gastric emptying must be taken into account when sustained release tablets with coatings resistant to acid are prescribed. If such tablets are taken together with meals, they will remain in the stomach until it is empty of food, and if the subject eats frequent meals during the day the stomach will not be empty before sometime during the night. Consequently, it is of no relevance to take such tablets two or more times daily, while all or most of the tablets will remain in the stomach the entire day and later all are emptied into the duodenum sometime during the following night (43, 44).

The small intestinal transit time (SITT) of the capsule in adults

the duodenum to the distal part of the small intestine, and about two-thirds of the SITT is spent in the part of the small intestine that is located in the lower right abdominal quadrant (11). This slow transit through the distal small intestine is in accordance with the observation of *Kerlin & Phillips* (46), who found that the IMMC travels through the small intestine with a velocity that decreases from 4.7 cm/min in the jejunum to 0.9 cm/min in the terminal ileum. Ingestion of food also interrupts the IMMC of the small intestine (46). In children, the median SITT was almost identical to that of adults (7.5 hours), and for three quarters of that time the capsule is located in the distal small intestine (13). The rapid transit through the jejunum makes it difficult to study the pH of this part of the gut with a freely moving pH capsule. In a study using capsules tethered with a 2 metres long nylon line it was possible to obtain pH measurements from the proximal part of the jejunum (34). However, the line slowed the transit of the capsule considerably, so it was not possible to measure pH in the more distal parts of the small intestine in that study.

The colonic transit time of the capsule is about 17.5 hours (equal in adults and children), but it varies widely from less than 10 hours to 112 hours, and the capsule is located in the caecum for about half of this time (11, 13). The colonic transit time tends to be longer in females than in males (11).

The day-to-day variation of the regional transit times is considerable, and in a study of the day-to-day variation in 13 healthy subjects the variation coefficients of the GRT and colonic transit time were about one (11). The SITT was more consistent and the variation coefficient was 0.40 (11).

c. Linearity, frequency drift, precision and accuracy

The response of the transmission frequency to alterations in pH was tested by *Meldrum et al* (33) and by *Fallingborg et al* (34) concerning the RCS capsule, and was found to be linear within the interval between pH 1 and pH 9. The response of the Heidelberg capsule has also been found to be linear in the interval between pH 2 and 7, but above and below this interval the frequency response decreased (47). Calibrations of the capsules before and after a study will secure their accuracy at the calibration levels at the beginning and at the end of the study. As previously mentioned, the main problem with the early types of radiotelemetry capsules, and especially the Heidelberg capsule, was the frequency drift. This was mainly due to an oxidation of the antimony electrode in the presence of intestinal fluids. The problem was less with capsules using glass electrodes, and in the study of *Fallingborg et al* (11), using the RCS capsule, a maximal drift of 0.5 pH units was registered, with the exception of one subject in whom the increased drift was due to a defective reference cap. The frequency drift may affect both the accuracy (zero-point drift) and the precision (change of the slope of the response-curve) of the pH capsule, but, assuming that the drift develops with a constant rate during the study, a correction of the frequency drift can be performed after recalibration of the recovered capsule (14):

$$pH_{Tx} = 1 - 8 \times \frac{\dot{O}_{Tx} - \dot{O}_{pH1(T_o)} - \frac{T_x - T_o}{T_{end} - T_o} \times (\dot{O}_{pH1(T_o)} - \dot{O}_{pH1(T_{end})})}{\dot{O}_{pH1(T_o)} - \dot{O}_{pH9(T_o)} - \frac{T_x - T_o}{T_{end} - T_o} \times (\dot{O}_{pH1(T_o)} - \dot{O}_{pH1(T_{end})}) - \dot{O}_{pH9(T_o)} - \dot{O}_{pH9(T_{end})}}$$

pH_{Tx} : Corrected pH value, measured at T_x
 T_o : Time of pre-calibration
 T_{end} : Time of post-calibration
 T_x : Time of pH measurement
 \dot{O}_{Tx} : Transmission frequency measured at time T_x
 $\dot{O}_{pH1(T_o)}$: Transmission frequency of the capsule at pH1, 37°C, measured at pre-calibration
 $\dot{O}_{pH1(T_{end})}$: Transmission frequency of the capsule at pH1, 37°C, measured at post-calibration
 $\dot{O}_{pH9(T_o)}$: Transmission frequency of the capsule at pH9, 37°C, measured at pre-calibration
 $\dot{O}_{pH9(T_{end})}$: Transmission frequency of the capsule at pH9, 37°C, measured at post-calibration

The accuracy of the RCS capsule has been evaluated by *Meldrum et al* (33) and by *Fallingborg et al* (12) by comparing pH in faeces and ileostomy output, measured by the RCS capsules and with a pH-meter (Radiometer, Copenhagen), respectively. The maximal difference between pH values measured with the two methods was 0.2 and 0.3 pH units, respectively, in the two studies.

d. Frequency of measurements

In some studies pH was automatically registered 6 or 60 times per minute and stored in a recording system, whereas in other studies including those made by the author of this thesis the measurements were performed manually with intervals of 10 minutes or more. The frequent, automatic recording method has several advantages: it can be used in outpatients, it is able to demonstrate pH-changes of short duration, and measurements can also be performed while the person sleep. The major disadvantages of this system are that only one subject can be studied at a time for each recording system, and that measurements obtained during periods with low signal or signal loss may be incorrect. The importance of a high signal quality was emphasized by *Press et al* (48) who observed that artificial low pH values could be registered when the recorder indicated poor signal quality. The main advantage with manual recording is that an optimal signal quality can be secured at each recording, and that more than one subject can be studied at the same time. The disadvantages are that pH-changes that occur between measurements will not be discovered, and that the method cannot be used in outpatients.

3. GASTROINTESTINAL pH IN NORMAL HUMAN SUBJECTS

Since 1964 a number of studies on human gastrointestinal pH using pH-sensitive capsules have been published. The majority of these studies deal with measurements of gastric residence time or measurements of the acid secretory capacity of the stomach, and, therefore, only contain pH measurements from the stomach and the duodenum (Table 1). However, a number of studies also report pH measurements from more distal parts of the gut (Table 2). Many papers contain data obtained from normal human subjects, but in several of

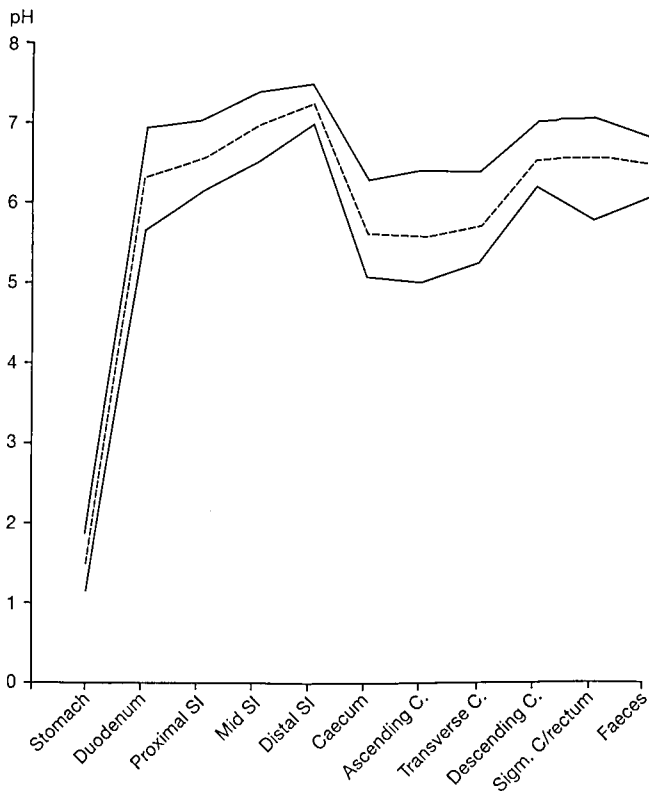


Table 1. Published studies on the use of pH-sensitive, radiotransmitting capsules. In vitro studies and human studies confined to measurements in the stomach and/or in the duodenum.

| References | Type of capsule | Subjects |
|-----------------------------------|------------------------------------|--|
| Jacobson & MacKay, 1957 (22) | prototype, 400 kHz | – |
| Nöller, 1959 (23) | prototype, 1.9 MHz (Heidelberg) | – |
| Connel & Waters, 1964 (25) | Heidelberg | 30 duodenal ulcer patients |
| Steinberg et al, 1965 (47) | Heidelberg | – |
| Watson & Patton, 1965 (49) | Heidelberg | 30 normal subjects and patients |
| Ranft & Nöller, 1965 (50) | Heidelberg | 50 normal subjects (+ antacids) |
| Watson & Kay, 1965 (31) | prototype, 400 kHz (RCS) | – |
| Stavney et al, 1966 (51) | Heidelberg | 20 duodenal ulcer patients |
| Kuhn et al, 1966 (52) | Heidelberg | 18 patients with dystrophia myotonica |
| Nöller, 1967 (53) | Heidelberg | 46 dyspeptic patients (+ Spasmo-Nervogastrol®) |
| Stilliger, 1967 (54) | Heidelberg | 12 patients (+ metoclopramide) |
| Maiwald et al, 1967 (55) | Heidelberg | Cases |
| Kütter, 1967 (56) | Heidelberg | Cases |
| Aynacıyan & Bingham, 1969 (26) | Heidelberg | 24 normal subjects, 22 duodenal ulcer patients |
| Yarbrough et al, 1969 (24) | Heidelberg | 26 patients |
| Deyhle et al, 1969 (57) | Heidelberg | 29 patients |
| Williamson et al, 1969 (58) | Heidelberg | 45 patients with anaemia (21 with pernicious anaemia) |
| Stack, 1969 (59) | Heidelberg | 1 normal subject and 29 patients |
| Russell & Goldberg, 1970 (60) | Heidelberg | 23 dyspeptic patients (+ acetylsalicylic acid) |
| Goldstein & Packman, 1970 (61) | Heidelberg | 13 normal subjects (+ antacids) |
| Andres & Bingham, 1970 (62) | Heidelberg | 27 dyspeptic patients |
| Kuntz et al, 1971 (63) | prototype, 1.9 MHz Heidelberg | Cases |
| Jóhannesson et al, 1973, (64) | Heidelberg | 10 duodenal ulcer patients |
| Sadehikova, 1973 (65) | Heidelberg | 37 healthy children |
| Madee & Müller-Wieland, 1975 (66) | Heidelberg | 16 normal subjects (+ antacids) |
| Kotter et al, 1975 (67) | Heidelberg | 23 normal subjects (transcranial electrotherapy) |
| Truchaud et al, 1975 (68) | Heidelberg | 118 normal subjects (+ antacids) |
| Ekenved & Walan, 1975 (69) | Heidelberg | 10 normal subjects (+ antacids) |
| Stacher & Starker, 1975 (70) | Heidelberg | 5 normal subjects (+ insulin ± bromazepam) |
| Misaki & Kawai, 1976 (71) | 2 prototypes | 4 normal subjects + 19 peptic ulcer patients |
| Ritschel & Erni, 1977 (72) | Heidelberg | 12 normal subjects |
| Heinkel, 1980 (73) | Heidelberg | 147 patients |
| McGraw et al, 1981 (74) | Heidelberg | 11 normal subjects (+ sucralfate / + antacids) |
| Uusitalo & Keyrilainen, 1983 (75) | Heidelberg | 24 normal subjects, 31 duodenal ulcer patients. |
| Puch et al, 1984 (76) | Heidelberg | 10 normal subjects (+ antacids) |
| Mojaverian et al, 1985 (42) | Heidelberg | 16 normal subjects |
| Faegenburg et al, 1985 (77) | Heidelberg | 1 patient with Crohn's disease |
| Reynolds et al, 1986 (78) | RCS | 4 duodenal ulcer patients |
| Rocci et al, 1987 (79) | Heidelberg | 8 normal men (+ procainamide) |
| Mojaverian et al, 1987 (80) | Heidelberg | 8 normal subjects (+ aspirin) |
| Mojaverian et al, 1988 (81) | Heidelberg | 45 normal subjects |
| Ewe et al, 1989 (82) | Heidelberg | 10 normal subjects |
| Mojaverian et al, 1990 (83) | Heidelberg | 12 normal men (± ranitidine) |
| Chan et al, 1990 (84) | Heidelberg | 8 healthy men (+ diclofenac sodium) |
| Mojaverian et al, 1991 (29) | Heidelberg | 4 normal subjects |
| Piscitelli et al, 1991 (85) | Heidelberg | 6 normal males (+ ketoconazole ± cimetidine ± sucralfate) |
| Knapp et al, 1991 (86) | Heidelberg | 6 normal subjects (+ glutamic acid ± ranitidine) |
| Betlach et al, 1991 (87) | Heidelberg | 12 normal subjects (+ theophylline ± ranitidine) |
| Lebsach et al, 1992 (88) | Heidelberg | 12 normal subjects (+ enoxacin ± ranitidine/pentagastrin) |
| Russell et al, 1993 (89) | Heidelberg | 79 healthy elderly subjects |
| Meyer et al, 1993 (90) | Heidelberg | 8 normal subjects, 7 achlorhydric subjects. (+ theophylline) |
| Alioth et al, 1993 (91) | Heidelberg | 18 normal subjects (+ diclofenac ± ranitidine) |
| Zimmermann et al, 1994 (92, 93) | Heidelberg | 24 normal subjects (+ fluconazole or itraconazole) |
| Russell et al, 1994 (94) | Heidelberg | 11 healthy subjects (> 64 years, + dipyridamole ± famotidine) |
| Zimmermann et al, 1994 (95) | Heidelberg | 12 normal subjects (+ fluconazole ± omeprazole) |
| Henderson et al, 1995 (96) | Heidelberg | 10 normal subjects (+ zinc acetate or zinc oxide ± famotidine) |
| Groning & Berntgen, 1996 (97) | Heidelberg | 5 normal subjects |
| Lewis & Heaton, 1997 (98) | ? | 13 normal subjects |

these articles the results from the healthy subjects and those obtained from patients are not separated. Therefore, only a limited number of these study results can be used to describe the pH-profile of normal humans (Table 3). On the basis of these selected studies, supplemented with results from studies using other methods, a picture of the pH profile of the normal gastrointestinal tract can be made.

However, it should be kept in mind that a normal colonic pH-profile may vary in different ethnic and cultural groups and in different parts of the world, because dietary habits may affect the intraluminal pH (see Chapter 7.1).

The pH-profile of the normal gastrointestinal tract, based on

3.1 STOMACH

The acid environment in the lumen of the stomach is attained by proton secretion from the parietal cells in the corporal mucosa. In a fasting subject, the pH of the gastric fluid ranges from 1 to 3.5 (11, 27, 30, 89), but ingestion of food, milk, or antacids may shortly increase pH to about 7 (24, 33, 89, 110).

3.2 DUODENUM

Due to the rapid transit of the untethered pH capsule through the duodenum, only a few recordings can be performed in this region using this method. Thus, results obtained with tubes, glass electrodes, or

Explore Litigation Insights

Docket Alarm provides insights to develop a more informed litigation strategy and the peace of mind of knowing you're on top of things.

Real-Time Litigation Alerts



Keep your litigation team up-to-date with **real-time alerts** and advanced team management tools built for the enterprise, all while greatly reducing PACER spend.

Our comprehensive service means we can handle Federal, State, and Administrative courts across the country.

Advanced Docket Research



With over 230 million records, Docket Alarm's cloud-native docket research platform finds what other services can't. Coverage includes Federal, State, plus PTAB, TTAB, ITC and NLRB decisions, all in one place.

Identify arguments that have been successful in the past with full text, pinpoint searching. Link to case law cited within any court document via Fastcase.

Analytics At Your Fingertips



Learn what happened the last time a particular judge, opposing counsel or company faced cases similar to yours.

Advanced out-of-the-box PTAB and TTAB analytics are always at your fingertips.

API

Docket Alarm offers a powerful API (application programming interface) to developers that want to integrate case filings into their apps.

LAW FIRMS

Build custom dashboards for your attorneys and clients with live data direct from the court.

Automate many repetitive legal tasks like conflict checks, document management, and marketing.

FINANCIAL INSTITUTIONS

Litigation and bankruptcy checks for companies and debtors.

E-DISCOVERY AND LEGAL VENDORS

Sync your system to PACER to automate legal marketing.