Handbook of Pharmaceutical Excipients

FIFTH EDITION

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Mannitol

1 Nonproprietary Names

BP: Mannitol
JP: D-Mannitol
PhEur: Mannitolum
USP: Mannitol

2 Synonyms

Cordycepic acid; C*PharmMannidex*; E421; manna sugar; D-mannite; mannite; *Mannogem*; *Pearlitol*.

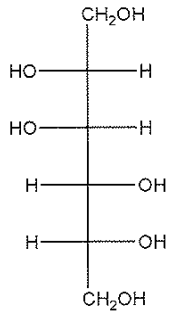
3 Chemical Name and CAS Registry Number

D-Mannitol [69-65-8]

4 Empirical Formula and Molecular Weight

$C_6H_{14}O_6$ 182.17

5 Structural Formula



6 Functional Category

Diluent; diluent for lyphilized preparations; sweetening agent; tablet and capsule diluent; tonicity agent.

7 Applications in Pharmaceutical Formulation or Technology

Mannitol is widely used in pharmaceutical formulations and food products. In pharmaceutical preparations it is primarily used as a diluent (10–90% w/w) in tablet formulations, where it is of particular value since it is not hygroscopic and may thus be used with moisture-sensitive active ingredients.[1,2]

Mannitol may be used in direct-compression tablet applications,[3–7] for which the granular and spray-dried forms are available, or in wet granulations.[8] Granulations containing mannitol have the advantage of being dried easily. Specific tablet applications include antacid preparations, glyceryl trinitrate tablets, and vitamin preparations. Mannitol is commonly used as an excipient in the manufacture of chewable tablet formulations because of its negative heat of solution, sweetness, and 'mouth feel'.[9,10]

In lyophilized preparations, mannitol (20–90% w/w) has been included as a carrier to produce a stiff, homogeneous cake that improves the appearance of the lyophilized plug in a vial.[11–20] A pyrogen-free form is available specifically for this use.

Mannitol has also been used to prevent thickening in aqueous antacid suspensions of aluminum hydroxide (<7% w/v). It has been suggested as a plasticizer in soft-gelatin capsules, as a component of sustained-release tablet formulations,[21] and as a carrier in dry powder inhalers.[22,23] It is also used as a diluent in rapidly dispersing oral dosage forms.[24,25] It is used in food applications as a bulking agent.

Therapeutically, mannitol administered parenterally is used as an osmotic diuretic, as a diagnostic agent for kidney function, as an adjunct in the treatment of acute renal failure, and as an agent to reduce intracranial pressure, treat cerebral edema, and reduce intraocular pressure. Given orally, mannitol is not absorbed significantly from the GI tract, but in large doses it can cause osmotic diarrhea; *see* Section 14.

8 Description

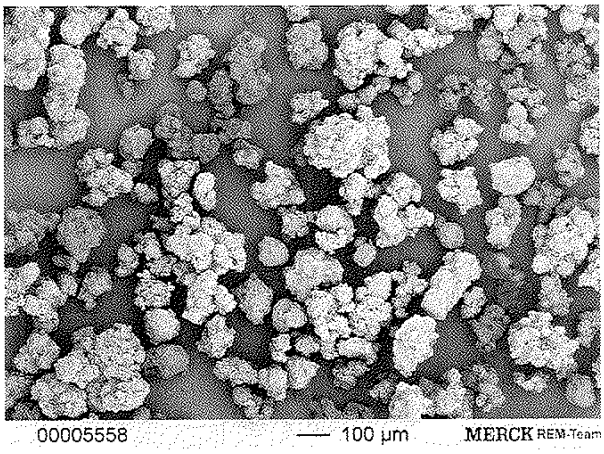
Mannitol is D-mannitol. It is a hexahydric alcohol related to mannose and is isomeric with sorbitol.

Mannitol occurs as a white, odorless, crystalline powder, or free-flowing granules. It has a sweet taste, approximately as sweet as glucose and half as sweet as sucrose, and imparts a cooling sensation in the mouth. Microscopically, it appears as orthorhombic needles when crystallized from alcohol. Mannitol shows polymorphism.[26]

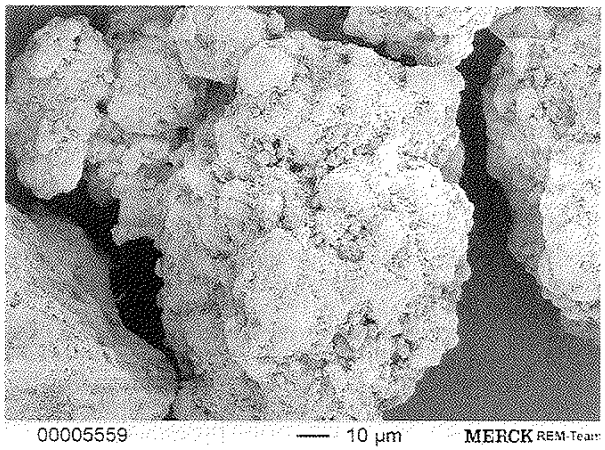
9 Pharmacopeial Specifications

See Table I.

SEM: 1
Excipient: Mannitol
Manufacturer: Merck
Magnification: 50×
Voltage: 3.5 kV

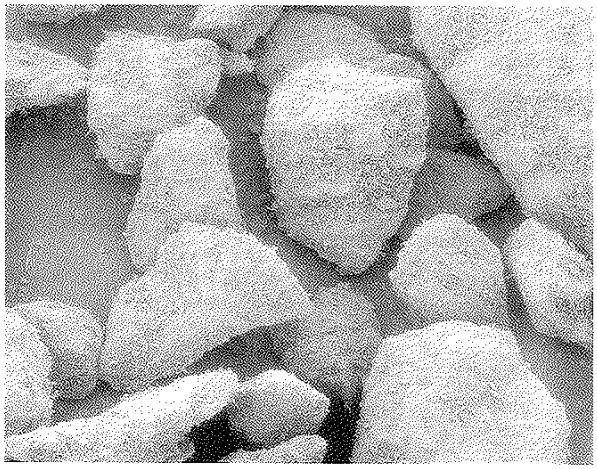


SEM: 2
Excipient: Mannitol
Manufacturer: Merck
Magnification: 500×
Voltage: 3.5 kV



SEM: 3
Excipient: Mannitol powder
Manufacturer: SPI Polyols Inc.
Lot No: 3140G8
Magnification: 100×

SEM: 4
Excipient: Mannitol granular
Manufacturer: SPI Polyols Inc.
Lot No: 2034F8
Magnification: 100×



10 Typical Properties

Compressibility: *see* Figure 1.
Density (bulk):
0.430 g/cm^3 for powder;
0.7 g/cm^3 for granules.
Density (tapped):
0.734 g/cm^3 for powder;
0.8 g/cm^3 for granules.
Density (true): 1.514 g/cm^3
Dissociation constant: pK_a = 13.5 at 18°C
Flash point: <150°C
Flowability: powder is cohesive, granules are free flowing.
Heat of combustion: 16.57 kJ/g (3.96 kcal/g)
Heat of solution: −120.9 J/g (−28.9 cal/g) at 25°C
Melting point: 166–168°C
Moisture content: *see* Figure 2.
Osmolarity: a 5.07% w/v aqueous solution is isoosmotic with serum.
Particle size distribution:
Pearlitol 300 DC: maximum of 0.1% greater than 500 µm and minimum of 90% greater than 200 µm in size;
Pearlitol 400 DC: maximum of 20% greater than 500 µm and minimum of 85% greater than 100 µm in size;
Pearlitol 500 DC: maximum of 0.5% greater than 841 µm and minimum of 90% greater than 150 µm in size.
Average particle diameter is 250 µm for *Pearlitol 300 DC*, 360 µm for *Pearlitol 400 DC* and 520 µm for *Pearlitol 500 DC*.[27] *See also* Figure 3.

Table I: Pharmacopeial specifications for mannitol.

Test	JP 2001	PhEur 2005	USP 28
Identification	+	+	+
Characters	—	+	—
Solution appearance	+	+	—
Melting range	166–169°C	165–170°C	164–169°C
Specific rotation	+137° to +145°	+23° to +25°	+137° to +145°
Conductivity	–	⩽20 µS·cm^{-1}	—
Acidity	+	—	+
Loss on drying	⩽0.3%	⩽0.5%	⩽0.3%
Chloride	⩽0.007%	—	⩽0.007%
Sulfate	⩽0.01%	—	⩽0.01%
Arsenic	⩽1.3 ppm	—	⩽1 ppm
Lead	–	⩽0.5 ppm	—
Nickel	+	⩽1 ppm	—
Heavy metals	⩽5 ppm	—	—
Reducing sugars	+	⩽0.2%	+
Residue on ignition	⩽0.10%	—	—
Related substances	—	⩽0.1%	—
Bacterial endotoxins	—	⩽4 IU/g[a]	—
Microbial contamination	—	⩽100/g	—
Assay (dried basis)	⩾ 98.0%	98.0–102.0%	96.0–101.5%

[a] Test applied only if the mannitol is to be used in the manufacture of parenteral dosage forms.

Table II: Solubility of mannitol.

Solvent	Solubility at 20°C
Alkalis	Soluble
Ethanol (95%)	1 in 83
Ether	Practically insoluble
Glycerin	1 in 18
Propan-2-ol	1 in 100
Water	1 in 5.5

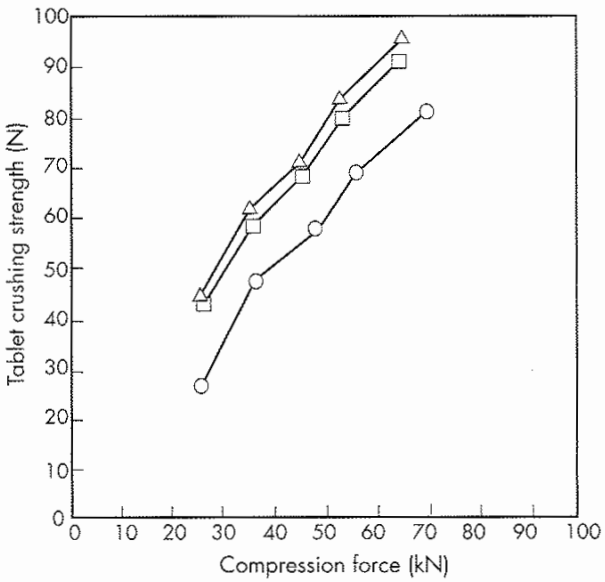


Figure 1: Compression characteristics of granular mannitol (*Pearlitol*, Roquette Frères).
○: *Pearlitol 300DC*
□: *Pearlitol 400DC*
△: *Pearlitol 500DC*
Tablet diameter: 20 mm
Lubricant: magnesium stearate 0.7% w/w for *Pearlitol 400DC* and *Pearlitol 500DC*; magnesium stearate 1% w/w for *Pearlitol 300DC*.

Refractive index: $n_D^{20} = 1.333$
Solubility: *see* Table II.
Specific surface area: 0.37–0.39 m^2/g

11 Stability and Storage Conditions

Mannitol is stable in the dry state and in aqueous solutions. Solutions may be sterilized by filtration or by autoclaving and if necessary may be autoclaved repeatedly with no adverse physical or chemical effects.[28] In solution, mannitol is not attacked by cold, dilute acids or alkalis, nor by atmospheric oxygen in the absence of catalysts. Mannitol does not undergo Maillard reactions.

The bulk material should be stored in a well-closed container in a cool, dry place.

12 Incompatibilities

Mannitol solutions, 20% w/v or stronger, may be salted out by potassium chloride or sodium chloride.[29] Precipitation has been reported to occur when a 25% w/v mannitol solution was allowed to contact plastic.[30] Sodium cephapirin at 2 mg/mL and 30 mg/mL concentration is incompatible with 20% w/v aqueous mannitol solution. Mannitol is incompatible with xylitol infusion and may form complexes with some metals such as aluminum, copper, and iron. Reducing sugar impurities in mannitol have been implicated in the oxidative degradation of a peptide in a lyophilized formation.[31] Mannitol was found to reduce the oral bioavailability of cimetidine compared to sucrose.[32]

13 Method of Manufacture

Mannitol may be extracted from the dried sap of manna and other natural sources by means of hot alcohol or other selective solvents. It is commercially produced by the catalytic or electrolytic reduction of monosaccharides such as mannose and glucose.

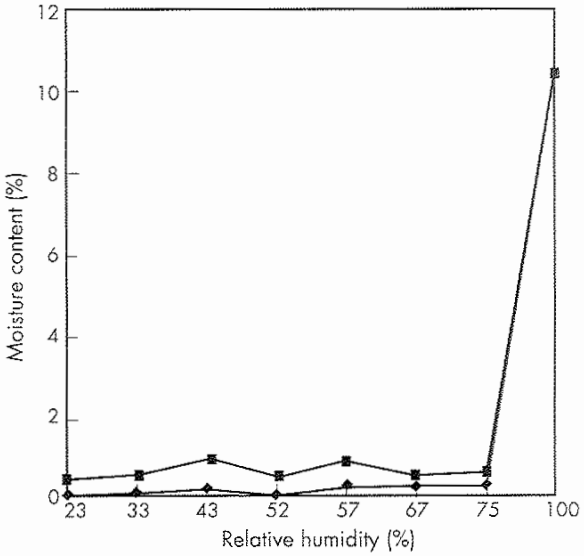


Figure 2: Sorption–desorption isotherm for mannitol.
◆: Sorption equilibrium moisture
■: Desorption equilibrium moisture

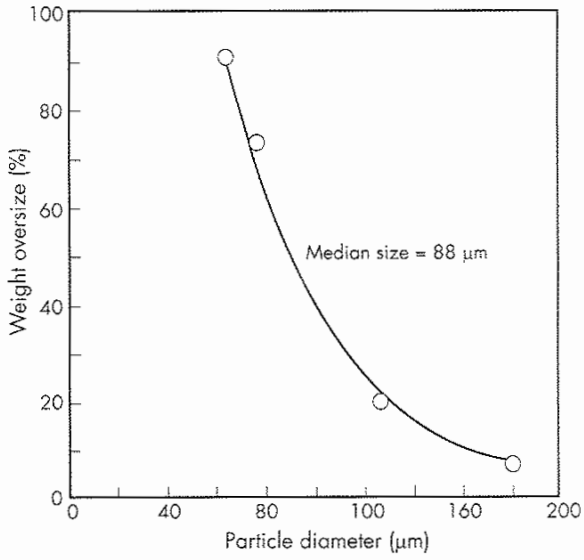


Figure 3: Particle size distribution of mannitol powder.

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