Súbclass 10N CLASSIFIC/ ass 5711849 Щ MA SS PATENT DATE UTILITY PATENT JAN 27 1998 SERIAL NUMBER NUMBER EXAMINER A BILLE GOLA GROUP ART UNIT SUBCLASS 643,) FILING DATE CLASS SERIAL NUMBER DANIEL L. FLAMM, WALNUT CREEK, CA; JOHN CHET! 18 VERBONCOEUR, HAYWARD, ₽ **CONTINUING DATA*********** VERIFIED work **FOREIGN/PCT APPLICATIONS************ VERIFIED WORK Mt ł, FOREIGN FILING LICENSE GRANTED 08/08/95 ***** SMALL ENTITY ***** INDEP. CLAIMS ATTORNEY'S TOTAL FILING FEE STATE OR SHEETS Foreign priority claimed ges 35 USC 119 conditions met ges ⊠ no Z no AS FILED DOCKET NO. RECEIVED COUNTRY DRWGS. \$Ś01,60 16655-000109 Verified and Acknowledged Examiner's initials CEANA RICHARD T OGAWA TOWNSEND AND TOWNSEND KHOURTE AND CREW LLP STELART STREET TONER CHE MARKET PLAZA SOTH FLOOP TWO EMBATCA DETO CENTET, 876F/ ADDRES SAN FRANCISCO CA 9410 94111-3834 PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING U.S. DEPT. OF COMM./ PAT. & TM-PTO-436L (Rev.12 PARTS OF APPLICATION Ample FILED SEPARATELY Applications Examiner CLAIMS ALLOWED NOTICE OF ALLOWANCE MAILED Total Claims Print Claim 29 20 Assistant Examiner ISSUE FEE DRAWING MAX Date Paid /// Sheets Drwg. Figs. Drwg. Print Fig. Amount Due 645.er Ż M 13 14 0 MARTIN ANGEBRANNDT PRIMARY EXAMINER. ISSUE **GROUP 1100** BATCH Primary Examiner NUMBER / Label PREPARED FOR ISSUE Area WARNING: The information disclosed herein may be restricted. Unauthorized disclosure may be prohibited by the United States Code Title 35, Sections 122, 181 and 368. Possession outside the U.S. Patent & Trademark Office is restricted to authorized employees and contractors only. Form PTO-436A (Rev. 8/92) (FACE)

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INDEX OF CLAIMS

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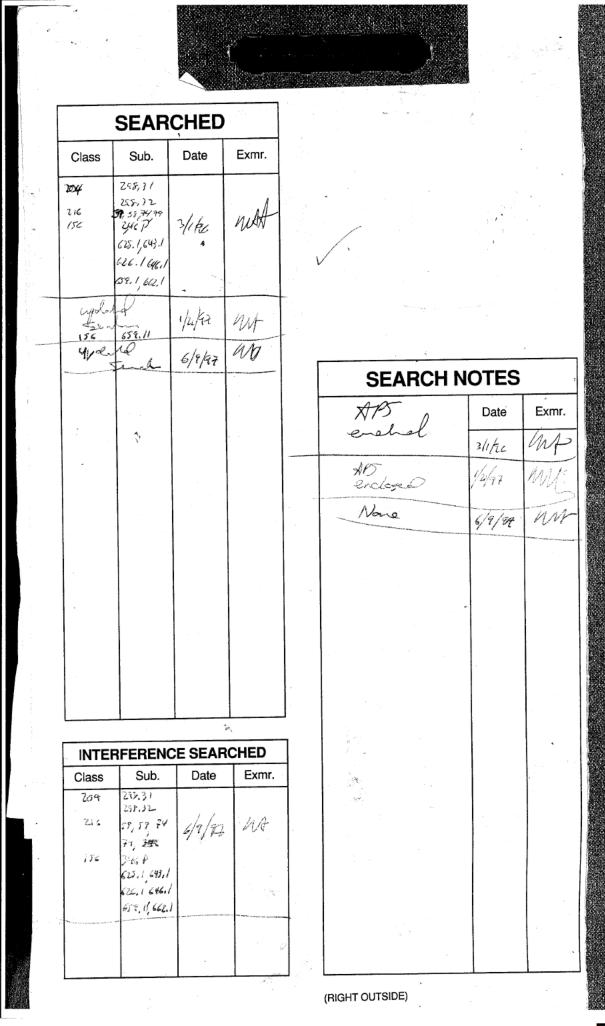
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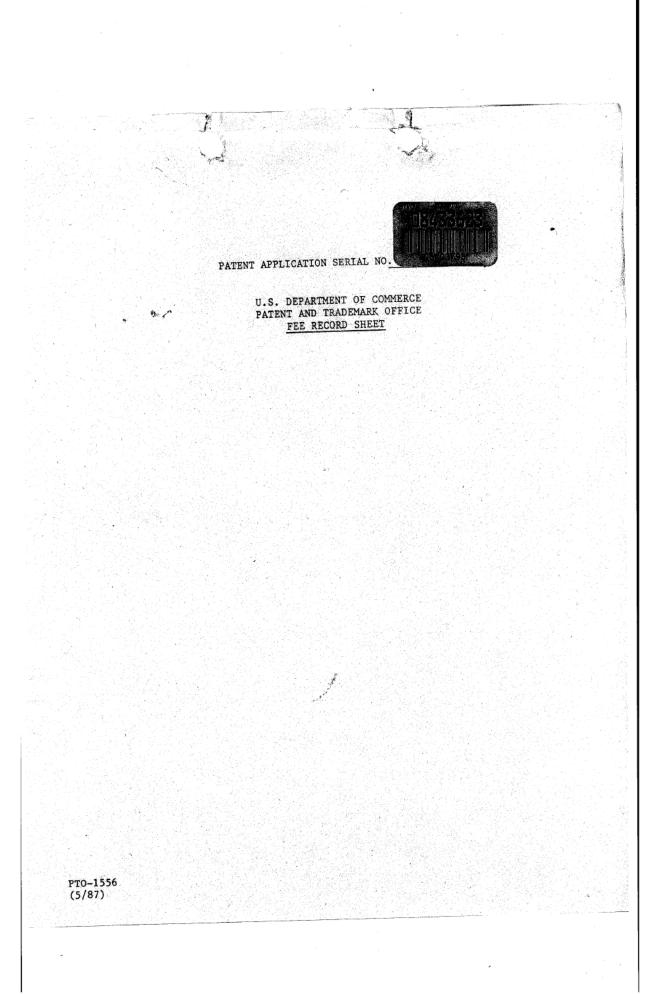
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Date of Deposit MAY 3, 1995

I hereby certify that this is being deposited with the United States Postal Service "Express Mail Post Office to Addressee" service under 37 CFR 1.10 on the date indicated above and is addressed to the Commissioner of Patents and Trademarks, Washington, D. C. 20231

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For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

Enclosed are:

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Inventor: DANIEL I

Sir:

- [x] 13 sheet(s) of [] formal [x] informal drawing(s).
- [] An assignment of the invention to _____
- [x] A [] signed [x] unsigned Declaration & Power of Attorney.
- [] A [] signed [] unsigned Declaration.
- [] A Power of Attorney.

PATENT APPLICATION

Washington, D. C. 20231

Transmitted herewith for filing is the

[x] patent application of[] design patent application of

- [] A verified statement to establish small entity status under 37 CFR 1.9 and 37 CFR 1.27.
- [] A certified copy of a ______ application.

FLAMM and JOHN VERBONCOEUR

[] Information Disclosure Statement under 37 CFR 1.97.

In view of the Unsigned Declaration as filed with this application and pursuant to 37 CFR §1.53(d), Applicant requests deferral of the filing fee until submission of the Missing Parts of Application.

DO NOT CHARGE THE FILING FEE AT THIS TIME.

nard T. Ogawa

Reg. No.: 37,692

Telephone: (415) 543-9600 APPNOFEE.TRN 12/92



Attorney Docket No. 016655-0001

PATENT APPLICATION



PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

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John Verboncoeur, a U.S. citizen, residing at 3350 Oakes Drive Hayward, California 94542.

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DWGTT

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Attorney Docket No. 016655-0001

PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

BACKGROUND OF THE INVENTION

The present invention relates to integrated circuits and their manufacture. The present invention is illustrated in an example with regard to plasma etching, and more particularly to plasma etching in resist strippers in semiconductor processing. But it will be recognized that the invention has a wider range of applicability in other technologies such as flat panel displays, large area substrate processing, and the like. Merely by way of example, the invention may be applied in plasma etching of materials such as silicon, silicon dioxide, silicon nitride, polysilicon, photoresist, polyimide, tungsten, among others.

Industry utilizes or has proposed several techniques for plasma etching. One such method is conventional chemical gas phase dry etching. Conventional chemical gas phase dry etching relies upon a reaction between a neutral gas phase species and a surface material layer, typically for removal. The reaction generally forms volatile products with the surface material layer for its removal. In such method, the neutral gas phase species may be formed by way of a plasma discharge.

A limitation with the conventional plasma etching technique is obtaining and maintaining etching uniformity within selected predetermined limits. In fact, the conventional technique for obtaining and maintaining uniform etching relies upon a "trial and error" process. The trial and error process often cannot anticipate the effects of parameter changes for actual wafer production. Accordingly, the conventional technique for obtaining and maintaining etching uniformity is often costly, laborious, and difficult to achieve.

Another limitation with the conventional plasma etching technique is reaction rates between the etching species and the etched material are often not available. Accordingly, it is often impossible to anticipate actual etch

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rates from reaction rate constants since no accurate reaction rate constants are available. In fact, conventional techniques require the actual construction and operation of an etching apparatus to obtain accurate etch rates. Full scale prototype equipment and the use of actual semiconductor wafers are often required, thereby being an expensive and time consuming process.

From the above it is seen that a method and apparatus of etching semiconductor wafers that is easy, 10 reliable, faster, predictable, and cost effective is often desired.

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SUMMARY OF THE INVENTION

According to the present invention, a plasma etching method that includes determining a reaction rate coefficient based upon etch profile data is provided. The present plasma etching method provides for an easy and cost effective way to select appropriate etching parameters such as reactor dimensions, temperature, pressure, radio frequency (rf) power, flow rate and the like by way of the etch profile data.

In a specific embodiment, the present invention provides an integrated circuit fabrication method. The present method includes steps of providing a plasma etching apparatus having a substrate therein. The substrate includes

a top surface and a film overlying the top surface. The film includes a top film surface. The present method also includes chemically etching the top film surface to define an etching profile on the film, and defining etch rate data which includes an etch rate and a spatial coordinate from the

etching profile. A step of extracting a reaction rate constant from the etch rate data, and using the reaction rate constant in adjusting a plasma etching apparatus is also included.

In an alternative specific embodiment, the present invention also provides a method of designing a reactor. The present method includes providing a first plasma etching apparatus having a substrate therein. The substrate has a top surface and a film overlying the top surface. The film has a

top film surface. The present method also includes chemically etching the top film surface to define an etching profile on the film, and defining etch rate data which has an etch rate and a spatial coordinate from the etching profile. A step of extracting a reaction rate constant from the etch rate data, and using the reaction rate constant in designing a second plasma etching apparatus is also included.

A further alternative embodiment provides another method of fabricating an integrated circuit device. The present method includes steps of providing a uniformity value for an etching reaction. The etching reaction includes a substrate and etchant species. The present method also includes defining etching parameters ranges providing the uniformity value. A step of adjusting at least one of the etching parameters to produce a selected etching rate is also included. The etching rate provides an etching condition for fabrication of an integrated circuit device.

The present invention achieves these benefits in the context of known process technology. However, a further understanding of the nature and advantages of the present invention may be realized by reference to the latter portions of the specification and attached drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1/is a simplified diagram of a plasma etching apparatus according to the present invention;

Fig. 1A is a simplified cross-sectional view of a wafer profile according of the plasma etching apparatus of Fig. 1;

Fig. 2/is a simplified diagram of an alternative embodiment of a plasma etching apparatus according to the present invention; / ,

Figs. 345 are simplified flow diagrams of plasma etching methods according to the present invention;

Fig. 5X is a plot of uniformity, temperature, pressure, and gap for an etching process according to the present invention;

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Fig. 6 is a simplified plot of 1/ash rate vs. LCD plate number according to the present invention;

Figs. 7-9 illustrate an example with regard to circular substrates according to the present invention; and Figs. 10-12 illustrate an example with regard to

rectangular substrates according to the present invention.

DESCRIPTION OF THE SPECIFIC EMBODIMENT Plasma Etching Apparatus

Fig. 1 is a simplified diagram of a plasma etching apparatus 10 according to the present invention. The plasma etching apparatus also known as a co-axial reactor includes at least three processing zones. The three processing zones are defined as a plasma generating zone (PG) 13, a transport zone (TZ) 15, a plate stack zone (PS) 17, and others. Also shown are a chemical feed F and exhaust E. The plasma generating zone provides for reactant species in plasma form and others. Excitation is often derived from a 13.56 MHz rf discharge 8 and may use either capacitor plates or a wrapped coil, but can also be derived from other sources. The co-axial reactor 10 also includes a chemical flow source 14 and a temperature and pressure control 12, among other features.

Chemical effects are often enhanced over ion induced effects and other effects by way of perforated metal shields 18 to confine the discharge to a region between an outer wall 16 and shields 18. The co-axial reactor relies substantially upon diffusion to obtain the desired etching uniformity. The co-axial reactor also relies upon a chemical etch rate which is diffusion to also relies upon a chemical etch rate which is diffusion to also relies upon a chemical etch rate the is generally defined as a chemical reaction rate of etchant species plus at least a diffusion rate of etchant species. When the diffusion rate of etchant species is much greater than the chemical reaction rate, the chemical etch rate is often determined by the diffusion rate. A more detailed analysis of such chemical etch rate will be described by way of the subsequent embodiments.

Etchant species from the plasma generating zone diffuse through the transport zone 15 of the reaction chamber,

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and enter the plate stack zone space over surfaces of substrates 21. A concentration of etchant in the transport zone, which is often annular, between the plasma generating zone and the plate stack zone is defined as n_{o0} . As etchant diffuses radially from the transport zone into the plate stack zone and over surfaces of the substrates, it is consumed by an etching reaction. A reactant concentration above the substrate can be defined as $n_o(r,z)$, where r is the distance from the center of the substrate and z is the distance above the substrate. A diffusive velocity v_o of etchant species in the plate stack zone is characterized by Fick's law.

 $v_o = -D \frac{\nabla n_o}{n}$

In a specific embodiment, a gap d_{gap} above the substrate is much less than the lateral extent $d_{gap} << r$ and gas phase mass transfer resistance across the small axial distance is negligible so that the axial (z-direction) term of the concentration gradient can be ignored. The embodiment can be applied without this restriction; however, numerical mesh computer solutions are then required to evaluate the reaction rate constant and uniformity. In the embodiment, the surface etching reaction bears a first order form:

 $O+S \rightarrow SO$

where

S is a substrate atom (e.g., resist unit "mer"); and O is the gas-phase etchant (for example oxygen atoms) with certain etching kinetics. The first order etching reaction can be defined as follows:

 $R_{os} = n_o A \sqrt{T} e^{-B_{ACT}/RT}$

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	R _{os} defines a reaction rate;
	n_o defines a concentration;
	A defines a reaction rate constant;
	T defines a temperature;

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25 761X E_{ACT} defines an activation energy; and R defines a gas constant.

An example of the first reaction is described in D.L. Flamm and D.M. Manos "Plasma Etching," (1989), which is hereby incorporated by reference for all purposes. Of course, other order reactions, reaction relations, and models may be applied depending upon the particular application.

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An example of an etched substrate 21 from the plate stack zone is illustrated by Fig. 1A. The substrate 21 is defined in spatial coordinates such as z and r. The substrate includes a bottom surface 23, sides 25, and a top surface film 27. As can be seen, the top surface film includes a convex region, or etching profile. The etching profile occurs by way of different etch rates along the r-direction of the substrate corresponding to different etchant species concentrations. A concentration profile $n_o(r,z)$ is also shown where the greatest concentration of reactant species exists at the outer periphery of the top surface film. In the present invention, an etch rate constant may be obtained by correlation to the etching profile. By way of the etch rate constant, other etching parameters such as certain reactor dimensions

including a distance between substrates, pressure, temperature, and the like are easily calculated.

Fig. 2 illustrates an alternative example of an etching apparatus 50 according to the present invention. The etching apparatus 50 is a single wafer etching apparatus with elements such as a chamber 53, a top electrode 55, a bottom electrode 57, a power source 59, a platen 64, and others. The bottom electrode 57 is at a ground potential, and the top

- 30 electrode is operably coupled to the power source 59 at a high voltage potential. A plasma exists in a region 54 between the top electrode 55 and the bottom electrode 57, which is often a grid configuration or the like. Reactant species are directed via power source from a plasma source to a wafer substrate 61
- 35 by diffusion. A temperature and pressure controller 67 and a flow controller 69 are also shown. The etching apparatus also includes a chemical source feed F and a exhaust E. Of course,

other elements may also be available based upon the particular application.

By way of a plate 63 interposed between the wafer substrate 61 and the bottom electrode 57, the reactant species do not directly bombard the wafer substrate. The plate is preferably made of an inert material appropriate for the particular etching such as pyrex or glass for resist ashing, alumina for fluorine atom etching of silicon, silicon nitride, or silicon dioxide, and the like. In an ashing reaction, the

10 plate is placed at a distance ranging from about 5 mm to 50 mm and less from the wafer substrate 61. Of course, other dimensions may be used depending upon the particular application. The reactant species are transported via diffusion from the plasma source to the wafer substrate around 15 the periphery of the plate 63. Accordingly, the reaction rate at the wafer substrate is controlled by a balance between chemical reaction and diffusion effects, rather than

directional bombardment.

By way of the diffusion effects, an etching rate constant may be obtained for the etching apparatus 50 of Fig. 2. In particular, the etching rate constant derives from a etching profile 65, which can be measured by conventional techniques. The present invention uses the etching rate constant to select other etching rate parameters such as reactor dimensions, such as spacing between the substrate and its adjacent surface, temperatures, pressures, and the like. But the present invention can be used with other reactor types

where etching may not be controlled by diffusion. For

example, the present invention provides a reaction rate which can be used in the design of reactors where diffusion does not control such as a directional etcher and the like. The reaction rate constant may also be used in the directional etcher to predict an extent of, for example, undercutting of unprotected sidewalls while ion bombardment drives reaction in

a vertical direction. Of course, the invention may be applied to other reactors such as large batch, high pressure, chemical, single wafer, and others. The invention can also be applied to different substrate materials, and the like.

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Plasma Etching Method

Figs. 3-5 illustrate simplified flow diagrams of plasma etching methods according to the present invention. The present methods provide for improved etching conditions by way, a reaction rate constant derived from, for example, an etching profile. It should be noted that the present methods as illustrated should not be construed as limiting the invention as defined in the claims. One of ordinary skill in the art would easily recognize other applications to the inventions described here.

In a specific embodiment, a method of extracting a rate constant 100 for a plasma etching step according to the present invention is illustrated by the flow diagram of Fig. 3. A substrate with an overlying film is placed into a plasma etching apparatus or the like. The overlying film is defined as an etching film. In the present embodiment, the overlying film is a photoresist film, but can also be other films such as a silicon film, a polysilicon film, silicon nitride, silicon oxide, polyimide, and the like.

A step of plasma etching the film is performed by step 101. The plasma etching step occurs at constant pressure and preferably constant plasma source characteristics. $\not\!\!\!/$ More preferably, the plasma etching step occurs isothermally at temperature T_1 , but can also be performed with changing temperatures where temperature and time histories can be monitored. Plasma etching of the film stops before the endpoint (or etch stop). Alternatively, plasma etching stops at a first sign of the endpoint (or etch stop). The plasma etching step preferably stops before etching into an etch stop layer underlying the film to define a "clean" etching profile.

The substrate including etched film is removed from the chamber of the plasma etching apparatus. The etched film includes an etching profile (step 103) made by way of plasma etching (step 101). The etching profile converts into a relative etch rate, relative concentration ratio, a relative etch depth, and the like at selected spatial coordinates. The

etch depth, and the like at selected spatial coordinates. The relative etch rate is defined as an etch rate at a selected spatial coordinate over an etch rate at the substrate edge.

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The relative concentration ratio is defined as a concentration of etchant species at a selected spatial coordinate over a concentration of etchant at the substrate edge.

In x-y-z coordinates, the relative etch rate in the z-direction, and the spatial coordinates are defined in the x-y coordinates. The etching profile is thereby characterized as a relative etch rate u, a x-location, and a y-location $u_{u,v}(x, y)$. In cylindrical coordinates, the relative etch rate is also in the z-direction, and the spatial coordinates are defined in the r and θ coordinates. The etching profile is characterized as a relative etch rate u, a r-location, and a θ -location (u, r, θ) . An array of data points in either the x-y coordinates or r- θ coordinates define the etching profile. The array of data points can be defined as an n x 3 array, where n represents the number of points sampled and 3 represents the etch rate and two spatial dimensions. Of course, the choice of coordinates depends upon the particular application.

Optionally, in a non-isothermal condition, an average etch rate is measured. By approximate integration of a time dependent etch rate, suitable starting point approximations for an etching rate constant pre-exponential and activation energy can be selected. The etch rate is integrated over time (and temperature) using measured

temperature-time data (or history). An etched depth profile and the etching rate from the integration can then be compared with actual data. A rate constant is appropriately readjusted and the aforementioned method is repeated as necessary.

An etch constant (or a reaction rate constant) over 30 diffusivity (k_{vo}/D) and an etch rate at an edge is calculated at step 105. The etch constant over diffusivity correlates with data points representing the etch rate profile. In x-y coordinates, the relationship between k_{vo}/D and the relative ρ_{i}^{i} etch rate u(x,y) is defined as follows:

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where

a and b define substrate lengths in, respectively, an x-direction and a y-direction.

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 $\cosh \sqrt{k_{vo}/D + (m\pi/b)^2}$ $\cosh\left[\sqrt{k_{vo}/D + (m\pi/b)^2} \frac{a}{2} \frac{\cosh\left[\sqrt{k_{vo}/D + (m\pi/a)^2}y\right]}{\cosh\left[\sqrt{k_{vo}/D + (m\pi/a)^2}y\right]}\right]$ $u(x,y) = \sum_{m=1}^{\infty} \frac{4}{m\pi} \sin \frac{m\pi}{2}$ $\cos \frac{m\pi x}{a}$

In cylindrical coordinates, the relationship between the etch constant over diffusivity k_{vo}/D and the relative etch rate u(r) is defined as follows:

u(r) =	$I_0\left(\sqrt{\frac{k_{vo}}{D}}r\right)$
4,17	$I_0\left(\sqrt{\frac{k_{vo}}{D}}a\right)$

where *a* is an outer radius (or edge) of the substrate. In step 106, a diffusivity is calculated for the particular etchants. The binary diffusivity D_{AB} may be calculated based upon the well known Chapman-Enskog kinetic theory equation:

$$D_{AB} = 2.2646 \ x \ 10^{-5} \frac{\sqrt{T\left(\frac{1}{M_A} + \frac{1}{M_B}\right)}}{\sigma_{AB} 2 \Omega_{D,AB} C}$$

where

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T is a temperature; C is a total molar concentration; M_A and M_B are molecular weights; D_{AB} is a binary diffusivity; σ_{AB} is a collision diameter; and $\Omega_{D,AB}$ is a collision integral.

The Chapman-Enskog kinetic theory equation is described in detail in part III of R.B. Bird, W.E. Stewart, and E.N. Lightfoot, "Transport Phenomena," Wiley (1960) which is hereby incorporated by reference for all purposes. Of course, other techniques for calculating a diffusivity may also be used. The equivalent volumetric reaction rate constant k_{vo} is derived from the diffusivity as follows.

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$$k_{vo} = \left(\frac{k_{vo}}{D}\right) D_{AB}$$

Once the reaction rate constant k_{vo} is extracted, the surface reaction rate constant k_s may be isolated from the previous equation as follows.

$$k_{g} = (k_{vo}) d_{gap}$$

Repeat steps 101-106 at different temperatures T_2 , $T_3...T_n$ to calculate additional reaction rate constants $k(T_2)$, $k(T_3)...k(T_n)$. The steps are repeated at least two times and more, and preferably at least three times and more. Each temperature is at least 5 °C greater than the previous temperature. Of course, the selection of temperatures and trial numbers depend upon the particular application.

Extract an activation energy E_{act} for a first order reaction from the data $k(T_2)$, $k(T_3)$... $k(T_n)$ at T_2 , $T_3...T_n$ collected via step 109 by way of the follow equation:

The activation energy is preferably calculated by a least square fit of data collected at step 109 or any other suitable statistical technique. By way of the same equation, the present method calculates surface reaction rate constant k_s at any temperature.

 $k_s(T) = A\sqrt{T} e^{\frac{-E_{act}}{RT}}$

In step 111, a concentration n_o at the substrate edge is calculated. The concentration n_o deduces from the following relationship:

 h_{is} $n_o = R_o / k_{ves}$

where

R_{os}is an etch rate.

From the concentration and the surface reaction rate, the particular etching step can be improved by way of adjusting selected etching parameters. In an alternative specific embodiment, a method to "tune" a plasma source using a loading effect relationship (or equation) is illustrated by the simplified flow diagram 200 of Fig. 4. The method includes a step 201 of measuring an etch rate against an effective etchable area A_w . The effective etchable area changes by varying the number *m* of wafers in the reactor, varying the size of the wafer, or the like. The effective area can be changed 209 by altering a gap between a wafer and its above surface 211, changing wafer quantity in the reactor 213, and varying substrate support member dimensions 215. The method preferably occurs at constant

temperature and pressure. However, the effective etchable area may also be varied by way of changing a temperature and/or a pressure.

The method calculates a uniformity value (step 217) from the measured values of etch rate vs. effective area in steps 211, 213, and 215. The uniformity is calculated by, for example:

 $uniformity=100 \frac{\int_{-\frac{1}{2}}^{1} R_{MAX} - R_{MIN}}{2\sum_{i}^{m} \frac{R_{i}}{m}}$

where

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 R_{MAX} is a maximum etch rate; R_{MIN} is a minimum etch rate; m is a sample number;

 R_i is a general etch rate for an *ith* sample; uniformity is a planarity measurement in percentage.

In a specific embodiment, a uniformity of about 90% and greater or preferably 95% and greater indicates that the effective area of the substrate is substantially equal to the actual substrate area (step 221) via branch 216. Of course, other methods of calculating a uniformity from etch rates and effective areas may also be used depending upon the particular application. Alternatively, an etching profile is measured and the effective area A_{eff} is calculated (step 219) by way of, e.g., the loading effect relationship.

At least two and more different effective etchable areas (step 223) are measured, or preferably at least three and more different etchable areas are measured.

Alternatively, the flow diagram returns via branch 224 to step 209, and takes another etch rate measurement at a different effective area. The flow diagram then turns to step 203.

In step 203, a supply of etchant S^T in the reactor is calculated. Based upon the different etchable areas a slope mA_{eff} deduces from the loading effect relationship as follows.

 $\frac{1}{R_{of}(m)} = \frac{1}{k_{s}n_{o}} = \frac{k_{r}A_{r}+F}{k_{o}S^{T}} + \frac{mA_{eff}}{S^{T}}$

 \mathcal{R}_{05} where $\mathcal{R}_{0}(m)$ is the etching rate at the boundary between the plate stack zone and transport zone when m substrates are present in the reactor. The first term includes a recombination term proportional to the total effective area A_r which acts to catalyze loss of etchant on reactor surfaces in the reactor plus a convection term F. The second term is the loading effect relation, where the reciprocal etch rate is proportional to the amount of effective etchable substrate area A_{eff} times the number of substrates m. When the etching

20 across a substrate is uniform, A_{eff} is the geometrical substrate area A_w . When etching is nonuniform, on the other hand, A_{eff} is a function of k_{vo}/D and geometrical reactor dimensions. The supply of etchant S^T may be calculated for a different plasma source or plasma source parameters such as 25 temperature, pressure, or the like by repetition 207 of steps 201 and 203. By way of the supply of etchant to the reactor, other plasma source parameters may be varied to obtain desired

etching rates and uniformity for the particular reactor. Step 205 provides for the modification of chamber materials and the like to reduce slope numerator $(k_r, A_r + F)$ in selecting the desired etching conditions. The chamber materials can be modified to reduce, for example, the recombination rate in the reactor. The recombination rate is directly related to the effective reactor recombination area

Ar. In step 205, the recombination rate can be adjusted by

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changing A_r via changing chamber material, coating chamber surfaces with, for example, a product sold under the trademark TEFLON^m or KALREZ^m and the like, among others. Alternatively, the slope numerator flow term F is reduced when F contributes as a substantial loss term. Of course, the particular materials used depend upon the application.

In step 207, the method changes plasma source parameters such as rf power, flow rate, and the like to select desired etching conditions. Once one of the aforementioned parameters is adjusted, the method returns to step 201 via

parameters is adjusted, the method returns to step 201 via branch 208. At step 201, an etch rate vs. effective etchable area is measured and the method continues through the steps until desired etching condition are achieved. Of course, other sequences of the aforementioned steps for tuning the plasma source may also exist depending upon the particular application.

Fig. 5 is a simplified flow diagram for a method of selecting a desired uniformity and desired etching parameters within selected ranges to provide a desired etch rate for a particular etching process. The etching parameters include process variables such as reactor dimensions, a pressure, a temperature, and the like for a particular substrate and reactants. Other etching parameters may also be used depending upon the particular application.

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In step 301, select a uniformity for the selected substrate and the reactants. The selected uniformity becomes an upper operating limit for the reaction according to the present method. The upper operating limit ensures a "worst case" uniformity value for an etched substrate according this method. Uniformity can be defined by, for example:

can be dering $\begin{bmatrix}
1 - \\
uniformity=100 \\
2\sum_{i=1}^{m} \frac{R_i}{m} \\
\land$

where

 R_{MAX} is a maximum etch rate; R_{MIN} is a minimum etch rate; m is a sample number;

 R_i is a general etch rate for an *ith* sample; uniformity is a planarity measurement in percentage.

In certain embodiments, the selected uniformity ranges from about 90 % and greater or more preferably 95 % and greater. Of course, other uniformity values may be selected based upon the particular application.

Based upon the selected uniformity, use the selected uniformity as a stating point to extract a plurality of reaction rate constants k_s . The reaction rate constants may be also be obtained by an input activation energy for the etching process, among other techniques (step 303). Alternatively, calculate k_s at one or more temperatures, and preferably two or more temperatures (step 303) from a

15 plurality of uniformity values. The uniformity values can be within the selected uniformity or outside the selected uniformity.

In step 307, prepare an array of etching parameters including a temperature T, a pressure P, a characteristic reactor dimension, and a uniformity value. In an embodiment, the characteristic reactor dimension can be a gap d_{gap} between the substrate and its adjacent surface. The array of etching parameters can be illustrated by way of a three dimensional plot.

An example of such array is illustrated by way of a

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three dimensional plot 500 in Fig. 5A. It should be noted that the illustration is merely an example of one application of the specific embodiment, and other examples can readily be

of the specific embodiment, and other examples can readily be determined by one of ordinary skill in the art. The plot includes a temperature axis, a pressure axis, and a gap axis. Each square region 501 represents a point defined by a specific temperature, pressure, and gap. Each square region 501 also includes a gray scale. Each different gray scale corresponds to a different uniformity value. In this example,

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the darker gray scale values 505 represent lower uniformity values than the lighter gray scale values 507.

Values than the lighter gray scale values 507. Based upon the array, compute locus of highest $T_j = \frac{VLTSUS}{VS_U}$ P and d_{gap} , and of highest $P_j = \frac{VLTSUS}{VS_U}$ T and d_{gap} 511 where

uniformity meets the specification, e.g., the selected uniformity from step 301. All points bounded within the highest $T_j \lor S_{\mathcal{H}} P$ and d_{gap} , and the highest $P_j \lor S_{\mathcal{H}} T$ and d_{gap} fall within the uniformity specification. Points outside the $v_{\mathcal{H}} r_{\mathcal{H}} S_{\mathcal{H}} P$ and d_{gap} and the highest $P_j \lor S_{\mathcal{H}} T$ and d_{gap} fall outside the uniformity specification. The points that fall within the uniformity specification defines the calculated uniformity limit manifold having outer boundaries at P_o and T_o .

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In the calculated uniformity limit manifold, select a gap d_{gap} , and adjust a locus of P and T below the calculated uniformity limit manifold by a predetermined amount to allow for statistical and experimental error and process drift. This step defines a new uniformity limit manifold, and ensures that points defined by a temperature, a pressure, and a gap, selected during subsequent steps fall within the selected uniformity (step 301) despite any error or process drift from the calculation. The new uniformity limit manifold includes outer boundaries at P_i and T_i which are respectively less than P_o and T_o .

In step 311, a maximum edge etch rate R_{os} and supply of etchant from a plasma source (S) for a selected rf power, a reactant flow, a pressure, a temperature, and a gap within the new uniformity limit manifold is determined. The maximum edge etch rate can be used in defining a desired flow rate of

source chemicals. Once the desired flow rate is determined, it should be held constant during subsequent steps in the embodiment.

A step (step 313) of locating an intersection space 30 of $P < P_i$, $T < T_i$, and a maximum etch rate (or an etchant supply) at selected rf power values is included. The intersection of space defines a maximum etch rate for the selected pressure P, temperature T, and gap d. Of course, other etching parameters may be adjusted depending upon the 35 particular application.

The method provides a resulting etch rate from the etching reaction using the aforementioned parameters which is compared with a desired etch rate. If the resulting etch rate

is too low (or high), change power and/or reduce the effective etchable area, e.g., increase d_{gap} , decrease number of substrates, use smaller substrates, and the like. Of course, other sequences of steps may be used in selecting a desired temperature, pressure, gap, and other parameters to provide the desired etch rate. The embodiment provides for a desired etch rate with a selected uniformity based upon a range of temperatures, pressures, and gap values, all within the selected uniformity specification.

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Theoretical Model of Apparatus

1. Plasma Generating Zone

In the specific embodiment, the plasma generating zone can be modeled as a "black box" where etchant flow of reactant species from the plasma generating source is determined from an etching rate at the plate stack zone. In particular, the etching rate is proportional to a product $n_o k_s$ of etchant concentration no above an etchable material film surface and an etching reaction rate constant k_s . The etching reaction rate constant k_s can be independently determined from uniformity data previously noted. Since the relative change in $n_o k_s$ and the absolute value of k_r (the surface recombination rate per unit reactor area) can be determined, n_o is easily extracted and used to study the effects of discharge and surface parameters on production of etchant species in the plasma generating zone. Accordingly, the efficiency of radical production by the plasma generating zone (the source term in a mass the mass balance of n_{o}) as a function of various parameters (pressure, power, temperature, etc.) can be extracted from indirect measurements.

2. Transport Zone

In the specific embodiment, etchant species concentrations in the transport space zone are approximated as "well-mixed". In the well-mixed embodiment, substantially all etchant species in the transport space zone are supplied by the plasma generating zone and are removed by at least: 1) etching reactions in the plate stack zone;

2) recombination; or 3) convection by flow out of the reactor. A supply S^T of etchant from the plasma generating zone is equated to the three aforementioned loss terms as follows:

$S^{T} = k_{r}A_{r}n_{o} + mA_{eff}k_{s}n_{o} + Fn_{o}$

where $k_r A_r$ is an effective loss term with regard to recombination effects, k_s is an etching reaction rate constant, A_{eff} is an effective etchable area of a substrate, n_o is the etchant concentration, m is the number of substrates and F is the gas flow rate out of the reactor. The equation may be rewritten in the form of a canonical loading effect relationship:

$$\frac{1}{R_o(m)} = \frac{1}{k_s n_o} = \frac{k_r A_r + F}{k_s S^T} + \frac{m A_{off}}{S^T}$$

where $R_o(m)$ is the etching rate at the boundary between the plate stack zone and transport zone when *m* substrates are present in the reactor, and the first term includes a recombination term proportional to the total effective area A_r which acts to catalyze loss of etchant on reactor surfaces plus convection *F*. The second term is the loading relation, wherein the reciprocal etch rate is proportional to the amount of effective etchable substrate area A_{wL} times the number of substrates *m*. When the etching across a substrate is uniform, A_{eff} is the geometrical substrate area A_w . When etching is nonuniform, on the other hand, A_{eff} is a function of k_{vo}/D and geometrical reactor dimensions. Accordingly, A_{eff} becomes a function of parameters such as temperature, pressure, reactor configuration, and the like.

Fig. 6 shows etch rate data vs. the number of substrates in a reactor along with a line corresponding to the loading effect relationship in the form

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$$\frac{1}{Ash \; Rate} = C_o + C_1 m A_w$$

where

 $C_{o} = 0.00030171936426 \text{ min/Å};$

and $C_I A_w = 2.3003912550 \times 10^{-5}$ are best fit constants for the conditions in Fig. 6. The equation gives an etching rate at the edge of the plate stack zone as a function of the number of substrates *m* and etchable exposed effective surface of a substrate A_{eff} . Other variables such as the temperature, etchant generation rate, flow rate, and reactor size parameters were held constant. While $R_o(m)$ as written strictly applies to the etch rate at the edge of a substrate, when storing uniformity is high the retching rates at any other fixed relative position on the substrates are related to $R_o(m)$ by a constant factor of proportionality, and so they will also conform to the form of these relations.

In the general case where etching is nonuniform across a substrate, the equivalent area A_w is smaller than the geometrical substrate area by a constant factor as a function of k_{vo}/D . It turns out that k_s can be independently deduced from the profile of the etching rate in the stack zone, and in turn permits the absolute value of n_o to be computed from the etching rate $R_o(m)$ at the edge of a substrate. If the slope of the isothermal loading effect curve

 $\frac{\partial \left[\frac{1}{k_{g}n_{o}}\right]}{\partial m} = \frac{A_{W}eff}{S^{T}}$

is measured along with etching uniformity, the rate of etchant supplied by the source S^T can be found by for substituting $A_{eff}(k_{vo}/D)$ evaluated on the basis of etching uniformity measurements.

3. Stack Zone

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For etchant mass transport from the transport zone into the plate stack zone, the distance between stacked wafers d_{gap} is small compared to the lineal dimensions of a substrate in the embodiment. Consequently, it will be assumed that the concentration is substantially uniform in the axial z direction and there is equi-molal, isothermal, and isobaric counter-diffusion (e.g., no net flux, $\Sigma n_1 = 0$) x and y directions. Since the ashing reaction is proportional to n_o , and O-atom consumption is proportional to the ashing rate, the continuity equation for O-atoms in two dimensions becomes:

$$\frac{\partial n_o}{\partial t} + \nabla \cdot n_o v = -k_{vo} n_o$$

where k_{vo} is the volume equivalent surface reaction rate constant, and v is the diffusive velocity of oxygen atoms. Inserting Fick's law

 $n_o v = D \nabla n_o$

the diffusion equation is obtained

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 $\frac{\partial n_o}{\partial t} - \nabla (D \nabla n_o) = -k_{vo} n_o$

And at steady-state in two dimensions and where D is not a function of spatial coordinate(s), it is rewritten as

 $\frac{\partial^2 u}{\partial x^2} + \frac{\partial^2 u}{\partial x^2} = \frac{k_{vo}}{D} u$

where $u(x, y) = n_0(x, y) / n_{00}$ and n_{00} is the etchant concentration 10 at the outer edges of the substrates. The boundary conditions are therefore u = 1. The equation is cast in dimensionless form as

$$\frac{\partial^2 u}{\partial^2 (x/L_x)} + \frac{\partial^2 u}{\partial^2 (y/L_y)} = \frac{k_{vo}}{D} u$$

where L_x and L_y are characteristic independent lengths and widths of substrates. From this equation, it is clear that $u(x/L_x, y/L_y)$ is a function solely of k_{vo}/D and the boundary conditions. Consequently, if experimental values of

 $\int_{L}^{H} (x/L_x, y/L_y)$ are measured at two positions on the substrate (i.e., at the center and edge), two algebraic equations based on this measurement can be used to eliminate $\frac{n_s}{N_h}$ and solve for k_{vo}/D . The diffusivity D can be calculated to good accuracy with the Hirshfelder equation; hence, k_{vo} is measured with this procedure.

For circular substrates, there is only one independent dimension (e.g., where r=a is the substrate

radius). At steady state in one dimensional cylindrical coordinates the equation can be written:

$$\frac{1}{r}\frac{\partial \left(r\frac{\partial u}{\partial r}\right)}{\partial r} + \frac{\partial u}{\partial z} = \frac{k_{vo}}{D}u$$

where $u(r) = n_o(r) / n_{o0}$ and the boundary condition is u(a) = 1 at the substrate (wafer) edge.

In the subsequent sections, analytic solutions to these relationships are developed for rectangular and circular substrates (e.g., for flat panel display substrates and semiconductor wafers). The framework is used to derive uniformity relationships for flat panel resist stripping equipment.

Examples

1. Circular Substrate (Wafer) Stacked Etcher To prove the principles of the aforementioned

embodiments, the present method and apparatus was applied to

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etching of circular substrates in a stacked etcher. Of course, the present method and apparatus can be applied to other geometries and etcher types. The present example is therefore not intended to be limiting in any way. The present method and apparatus is applied to the circular substrates as illustrated by way of Fig. 7. The present method relies upon etching of substrate material S by way of oxygen using a reaction which is substantially chemical etching.

An illustration of a circular substrate according to the present invention is shown in Fig. 8. Assume that the 25 distance d_{gap} between stacked wafers is relatively small compared to the wafer radius a such that $d_{qap} << a$. Based upon the assumption, the oxygen concentration will be substantially uniform in the axial direction z. Accordingly, only radial diffusion in the r-direction needs consideration.

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Assuming an equi-molal counterdiffusion

 $\sum_{i=1}^{m} n_i v_i = 0$

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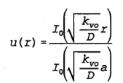
and an isobaric and isothermal stack zone, the problem reduces to two dimensions and becomes

$\frac{1}{r}\frac{\partial(r\frac{\partial u(r)}{\partial r})}{\partial r}=\frac{k_{vo}}{D}u(r)$

where $u(r) = n_o(r)/n_{o0}$. The boundary condition is u(a) = 1 at a wafer edge, and the solution of the equation becomes

$$u(r) = c_1 I_0 \left(\sqrt{\frac{k_{vo}}{D}} r \right) + c_2 K_0 \left(\sqrt{\frac{k_{vo}}{D}} r \right)$$

where I_0 and K_0 are modified Bessel functions of the first and second kind, respectively, and c_1 and c_2 are constants. For a finite, normalized oxygen concentration u(0) at the center of the wafer, the equation requires $c_2=0$. The remaining boundary condition of u(a)=1, sets the solution:



Note that the functional form u(r) describes both the relative etch rate profile $R_o(r)/R_o(a)$ and the relative oxygen atom etchant concentration $n_o(r)/n_o(a)$. The relative etch rate profile can easily be obtained by measuring an etching rate profile on a circular substrate made by way of the present method.

Fig. 9 is a simplified plot of a normalized stripping rate vs. radial distance from a wafer center for the circular substrate example. The plot shows a profile of u(r) for $k_{vo}/D=0.1$, and an a=150 mm. As can be seen, the

normalized stripping rate is lower at a center region of the wafer, and increases to 1 at the wafer edge. Based upon a slope of the plot, a reaction rate coefficient can be extracted by way of a diffusivity.

2. Rectangular Substrate Stack Asher

To further provide the principle and operation of the present method and apparatus, the present method and apparatus is applied to a rectangular substrate configuration in a stack asher. Again, the present example should not be taken as limiting the scope of the claims described herein, but is merely an example. An analytical solution for etching profiles in the stack zone are derived for etching/ashing a stack of rectangular substrates as illustrated in Fig. 9. The rectangular substrate can be a flat panel display such as a liquid crystal display (LCD) plate and the like in the coordinate system of Fig. 10. To solve an equation for the present rectangular configuration where D is not dependent upon spatial coordinates, write the solution as:

$$u = u_1 + u_2$$

15 where u_1 is satisfied by the following equation 7240% ∂n

$$\frac{\partial n_o}{\partial t} - D\nabla^2 n_o = -k_{vo} n_o$$

where $u_1 = 0$ at $y = \pm b/2$;

and u_2 is a solution that is 0 at $x = \pm a/2$. The solution for $u_1(x, y) = X(x)Y(y)$ is obtained by a separation of variables as follows.

$$\frac{\partial \left[\frac{\partial X}{\partial x}\right]}{X} + \frac{\partial \left[\frac{\partial Y}{\partial y}\right]}{Y} = \lambda^2 = \lambda_x^2 - \lambda_y^2 = \frac{k}{L}$$

The sign of the sum decomposing λ^2 is chosen so that X(x) and Y(y) both have real values, as shown below. Since the boundary conditions on Y(y) are:

$$Y(-b/2) = Y(b/2) = 0$$

the solution is,

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d

$$Y = c_y \cos \lambda_y y.$$

From the boundary conditions, $c_y = m/b$ where m = 1, 3, 5, ...Similarly, the solution for X is

where λ_{x} is given by:

where
$$\lambda_x$$
 is given by
 $\lambda_x = \sqrt{k_{vo}/D + (m\pi/b)^2}$.
The

the general solution is the sum:

u

$$c_{1} = \sum_{m}^{\infty} c_{m} \cosh \left[\sqrt{k_{vo} / D + (m\pi / b)^{2}} x \right] \cos \frac{m\pi y}{b}$$

where $c_m = 0$ for $m = 0, 2, 4, \dots$ to satisfy the 19. d boundary conditions. Setting $u_1(a/2, y) = f(y)$, where f(y) is the even-function square wave of magnitude 1, the Fourier series is obtained,

$$c_{\rm m} \cosh\left[\sqrt{k_{\rm vo}/D + m\pi/b}\right]^2 \frac{a}{2} = \frac{2}{b} \int_0^b f(y) \cos\frac{m\pi y}{b} dy$$

and after the integration

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$$c_{\rm m} \cosh\left[\sqrt{k_{\rm vo}/D + m\pi/b}\right]^2 \frac{a}{2} = \frac{4}{m\pi} \sin\frac{m\pi}{2}$$

$$C_m \cosh\left[\sqrt{K_{vo}}/D + m\pi/D\right]^2 \frac{1}{2} = \frac{1}{m\pi} \sin \frac{1}{2}$$

which is zero when m is even, as required. Thus, the u_1 part

of the solution can be written

$$u_1(x,y) = \sum_{m=1}^{\infty} \frac{4}{m\pi} \sin \frac{m\pi}{2} \frac{\cosh[\sqrt{k_{vo}/D + (m\pi/b)^2 x}]}{\cosh[\sqrt{k_{vo}/D + (m\pi/b)^2 \frac{a}{2}}]} \cos \frac{m\pi y}{b}$$

m=1,3,5,...

Note that $u_1(a/2, y) = 1$ for (-b/2 < y < b/2). The solution for u_2 can be obtained in a similar way. The solution is then

$$u(x, y) = \sum_{m=1}^{\infty} \frac{4}{m\pi} \sin \frac{m\pi}{2} \begin{cases} \frac{\cosh[\sqrt{k_{vo}/D + (m\pi/b)^2 x}]}{\cosh[\sqrt{k_{vo}/D + (m\pi/b)^2 \frac{a}{2}}]} \cos \frac{m\pi y}{b} \\ \frac{\cosh[\sqrt{k_{vo}/D + (m\pi/a)^2 y}]}{\cosh[\sqrt{k_{vo}/D + (m\pi/a)^2 \frac{b}{2}}]} \cos \frac{m\pi x}{a} \end{cases}$$

where m is odd. As $b \rightarrow \infty$, this approaches the solution for 1dimensional diffusion (corresponding to an infinitely long strip):

 $u(\mathbf{x}) = \frac{\cosh\left[\sqrt{k_{vo}/D}\mathbf{x}\right]}{\cosh\left[\sqrt{k_{vo}/D}\frac{a}{2}\right]}$

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two-dimensional

The previous, equation is now applied to interpret ashing uniformity data and predict uniformity and the atomic oxygen concentration profile n_o along the surface of a substrate for selected operating conditions. To use the relationship, values of k_{vo} and D are required. For atomic oxygen diffusing through O2, diffusivity was computed as $D(cm^2/s) = 0.044T^{3/2}$ (T is in K) from J.O. Hirschfelder, C.F. Curtiss, R.B. Byrd, "Molecular Theory of Gases and Liquids," pp. 538-541 and 578-582, John Wiley & Sons, 2nd Printing (1963), which is hereby incorporated by reference for all purposes. Of course, other techniques for calculating the diffusivity also exist.

In general, k_{vo} will be a function of at least gap, resist composition, temperature, and other parameters. In an example, k_{vo} is unknown, although the activation energy for resist ashing is conventionally reported to be in the 11-12 kCal range from industry literature. However, the solutions for u(x, y) depend only on k_{vo}/D and geometrical chamber dimensions such as gap (as incorporated into k_{vo}), a, b, and the like. Accordingly, k_{vo}/D is deduced from the etching rate profile, as previously described.

In particular, k_{vo}/D can be obtained from measurements of the amount of resist removed at two independent points (points where the theoretically predicted etch depth ratios $u(x_1, y_1)$, $u(x_2, y_2)$ are unequal by solving the appropriate equation for k_{vo}/D and substitute for D(T, P). But the present example used a more robust procedure: determine k_{vo}/D from a least squares fit to the entire experimental etch profile data set taken by a conventional stylus profilometer.

Fig. If shows an experimental etching profile data taken on a 30x30 cm resist-covered substrate spin-coated with 2.1 microns of MCPR 200 resist (Mitsubishi Chemical Corp., equivalent to Tokyo Ohka Kogyo Co. OFPR 800). A vertical axis

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1201 defines an ashing rate R_o with respect to an x-direction 1203 and a y-direction 1205. A grid pattern 1207 represents a "fitted" surface region via aforementioned equation representing ashing rates. Actual data points for each ashing rate are defined as the circular points 1211, and plots representing the fitted surface region are defined as cross points 1209. Ashing rate is greater around the periphery of the substrate, than substrate center regions.

The reactor held 1.1 mm thick substrates with a 28.9 mm gap (d_{gap}) above the wafer. A 4 kW rf power source drove a plasma with pure oxygen gas flowing into the reactor at 3 liters/min. Thermocouple sensors and heaters kept the reactor chamber and substrates at T=220 °C during the etch process, and a throttle valve maintained pressure at P=1.2 Torr. Etching occurred for 5 min. Resist thickness was measured before and after etching using a Nanometrics Model 210 Nanospec Auto Film Thickness Monitor. The surface of Fig. $\frac{12}{L_{L_{c}}}$ represents a least squares fit the aforementioned equation for u(x,y) with k_{vo}/D as the only adjustable parameter. The least

squares fit gives $k_{vo}/D=0.047$. At P=1.27 Torr and T=493 K into $D(cm^2/s)=0.044T^{3/2}$ yields D=400 cm^2/s . By way of the relationship $k_{vo}/D=0.047$, the etch rate constant is now $k_{vo}=19.5$ sec.⁻¹. In the manner, e.g., by fitting profile data to the solution for given substrate geometry, k_{vo} can be measured under various process conditions. By way of k_{vo} , other parameters such as n_o , k_s , and the like may also be

calculated. Once k_{vo} is known as a function of temperature, ashing rate and uniformity can be calculated as a function of reactor size parameters (a,b) and process variables (p, T, and $n_o)$. While the etching rate is proportional to n_{o0} , n_{o0} does not affect the etch depth profile and need not be known to compute k_{vo} . However, after k_{vo} is obtained, n_{o0} can be computed from the experimentally measured etching rate per $k_{vo}^* = k_s n_{o0}$. The procedure applies up to endpoint (endpoint is the time at which resist has been "stripped" and is no longer covering the region of the substrate where etching was

fastest). At endpoint, resist begins to be cleared from the

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substrate so that etchable area changes. Hence, n_{o0} will start to change (increase) after endpoint. The magnitude of n_{o0} during the steady-state period when resist is etching controlled by the plasma source, the number of substrates loaded into the reactor and (possibly) convective loss.

Predicting Etch Rate

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D

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The effect of profile uniformity on loading can be explicitly accounted for by defining a profile-average substrate area $A_{\mathbf{w}, eff}$

 $R_{off} A_{m,off} = \iint A_{substr} k_{vo} n_o(x, y) \, dx dy = n_{o0} k_{vo} \iint_{substr} u(x, y) \, dx dy = n_{o0} k_{vo} A_{m,off}$

so that $n_{o0}k_{vo}A_{\mathbf{K},eff}$ is the per substrate etchant consumption with nonuniformity resulting from effects of diffusion and reaction taken into account. Then for given plasma source (etchant supply), the etch rate/loading effect equation becomes:

 $\frac{1}{R_o(m)} = \frac{1}{k_s n_o} = \frac{k_r A_r + F}{k_s S^T} + \frac{m A_{eff}}{S^T}$

All of the terms can be computed explicitly from etch rate profile data, except for the rate of etchant production by the source. The etchant production rate can be computed from two measurements of etching rate when changing $k_{vo}A_{w_veff}$. A_{w_veff} can be changed either by changing the number of substrates or changing the etch rate profile (with constant etchant supply).

The present invention provides a method of selecting uniformity in chemical plasma etching as a function of processing parameters. The present invention also provides for a method of measuring absolute gas-surface reaction rates in commercial processing equipment without the benefit of sophisticated diagnostic equipment.

Gas-surface radical reaction rates are often needed for the design of plasma processing equipment and for

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selection of desired reaction conditions. Unfortunately, few data are available on absolute reaction rates in systems of practical interest in the prior art. Most experimental data have been taken in difficult flow tube experiments, or by related techniques which require reactant concentrations to be quantified using sophisticated methods such as gas-phase titration, laser fluorescence or mass spectrometry. These measurements require great care and specialized instrumentation. In contrast, the present invention describes a technique for measuring etching rate constants. It can be carried out in commercial processing equipment and the like, and it does not require sophisticated instrumentation, direct radical measurements, or the like. An isothermal reaction rate constant may be derived from a single measurement of etching uniformity. From this information, the etching rate uniformity as a function of substrate spacing and pressure can be computed. If experimental data on uniformity are taken at several temperatures, an intrinsic activation energy can be derived and the effects of temperature can be expressed

analytically.

While the above is a full description of the specific embodiments, various modifications, alternative constructions and equivalents may be used. For example, while the description above is in terms of a plasma etching method, it would be possible to implement the present invention with other etching methods or the like.

Therefore, the above description and illustrations should not be taken as limiting the scope of the present invention which is defined by the appended claims.

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WHAT IS CLAIMED IS:

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1. An integrated circuit device fabrication method comprising the steps of:

providing à plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate constant from said etch rate data, and using said reaction rate constant in adjusting said plasma etching apparatus.

2. The method of claim 1 wherein said chemical etching step is diffusion limiting.

3. The method of claim 1 wherein said spatial coordinates include a radius and an angle.

4. The method of claim 1 wherein said spatial coordinates include an x-direction and a y-direction.

5. The method of claim 1 wherein said extracting step correlates said reaction rate constant over a diffusivity with said an etching rate, said etching rate being defined by said etching profile.

5. The method of claim 1 wherein said etching rate is defined by said etching profile at selected spatial coordinates over a time.

7. The method of claim 1 wherein said chemical etching is an ashing method.

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8. The method of claim 1 wherein said ashing method comprises reactants including an oxygen and a photoresist.

9. A method of designing a reactor comprising the

steps of:

providing a first plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate constant from said etch rate data, and using said reaction rate constant in designing a second plasma etching apparatus.

10. The method of claim 9 wherein said chemical etching step is diffusion limiting.

11. The method of claim 9 wherein said spatial coordinates include a radius and an angle.

12. The method of claim 9 wherein said spatial coordinates include an x-direction and a y-direction.

3. The method of claim 9 wherein said extracting step correlates said reaction rate constant over a diffusivity with said an etching rate, said etching rate being defined by said etching profile.

14. The method of claim 9 wherein said etching rate is defined by said otching profile at selected spatial coordinates over a time.

15. The method of claim 9 wherein said chemical etching is an ashing method.



16. The method of claim 9 wherein said ashing method comprises reactants including an oxygen and a photoresist.

 \mathcal{X} . The method of claim \mathcal{Y} wherein said second plasma etching apparatus is a co-axial reactor.

17. The method of claim \mathscr{S} wherein said second plasma etching apparatus is a plasma etching apparatus.

19. A substrate fabrication method comprising: providing a substrate selected from a group consisting of a semiconductor wafer, a plate, and a flat panel display, said substrate comprising a top surface;

forming a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define a profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate constant from said etch rate data, and using said reaction rate constant in adjusting said method.

20. A method of fabricating an integrated circuit device, said method comprising:

providing a uniformity value for an etching reaction, said etching reaction including a substrate and etchant species;

defining etching parameters ranges providing said uniformity value; and

adjusting at least one of said etching parameters to produce a selected etching rate;

wherein said etching rate providing an etching condition for fabrication of an integrated circuit device.

23 21. The method of claim 20 wherein said etching parameters can be selected from a group consisting of a temperature, a pressure, a power, a gap, and a flow rate.

The method of claim 20 wherein said uniformity 22. ranges from 90% and greater.

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The method of claim 20 wherein said uniformity 23. ranges from 95% and greater.

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Sub B1

PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

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ABSTRACT OF THE DISCLOSURE

A method of designing a reactor 10. The present reactor design method includes steps of providing a first . plasma etching apparatus 10 having a substrate 21 therein. The substrate includes a top surface and a film overlying the top surface, and the film having a top film surface. The present reactor design method also includes chemical etching the top film surface to define a profile 27 on the film, and defining etch rate data from the profile region. A step of extracting a reaction rate constant from the etch rate data, and a step of using the reaction rate constant in designing a second plasma etching apparatus is also included.

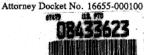
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DECLARATION AND POWER OF ATTORNEY



As a below named inventor, I declare that:

My residence, post office address and citizenship are as stated below next to my name; I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural inventors are named below) of the subject matter which is claimed and for which a patent is sought on the invention entitled: **PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING** the specification of which <u>X</u> is attached hereto or <u>was filed on</u> as Application Serial No. <u>and was amended on</u> (if applicable).

I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above. I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, Section 1.56. I claim foreign priority benefits under Title 35, United States Code, Section 119 of any foreign applications(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)

Country	Application No.	Date of Filing	Priority Claimed Under 35 USC 119
N/A			Yes No
			Yes No

I claim the benefit under Title 35, United States Code, Section 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, Section 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, Section 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

Application Serial No.	Date of Filing	Status
N/A		Patented Pending Abandoned
	,	Patented Pending Abandoned

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith.

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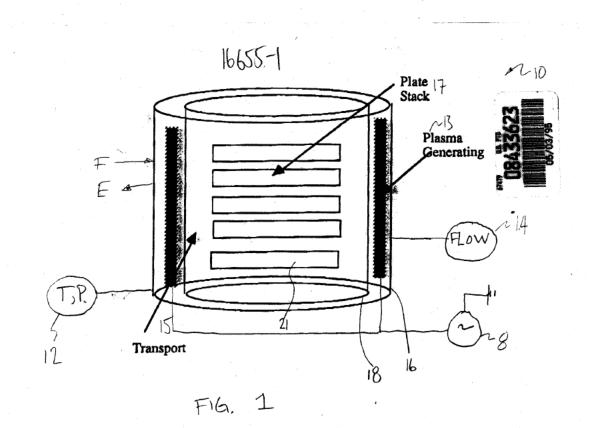
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I further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Signature of Inventor 1	Signature of Inventor 2
DANIEL L. FLAMM	JOHN VERBONCOEUR
Date	Date

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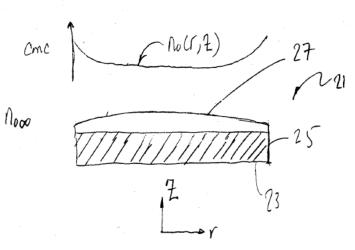
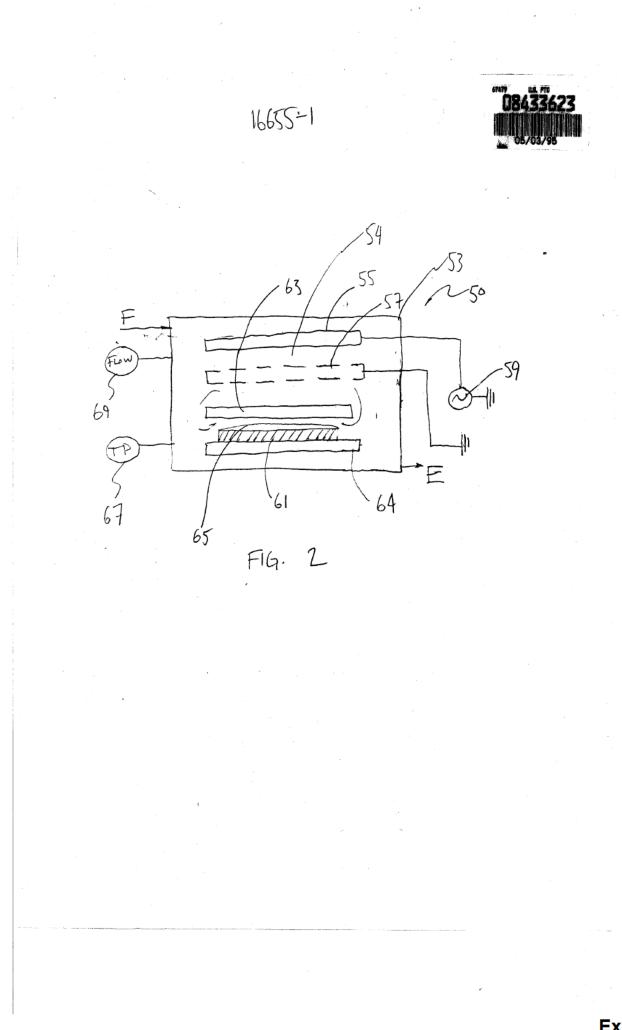
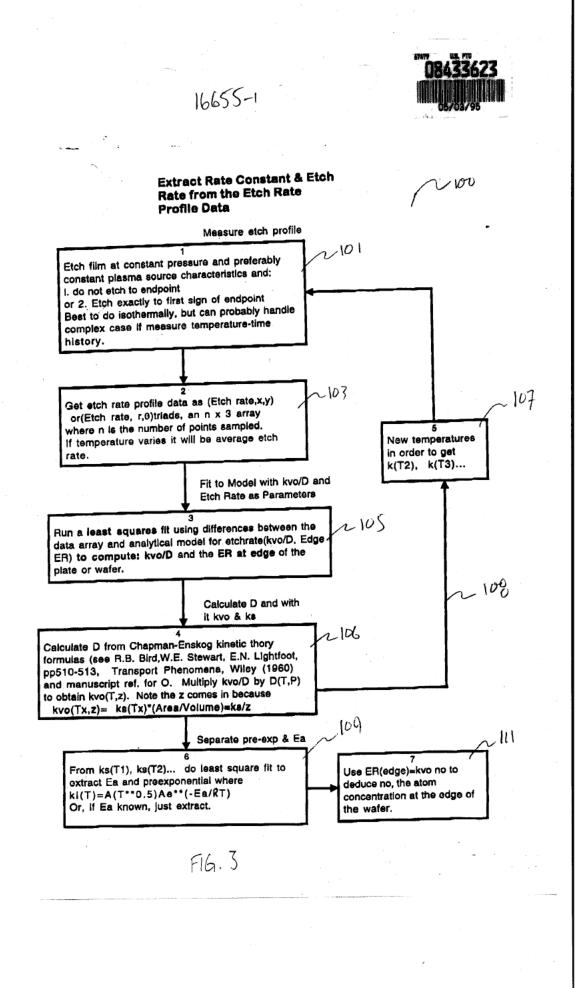
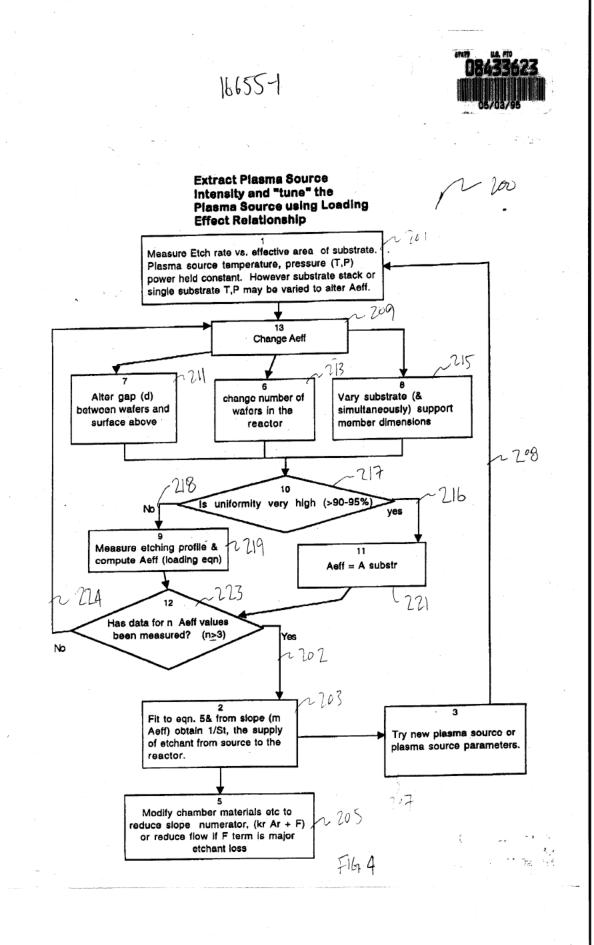


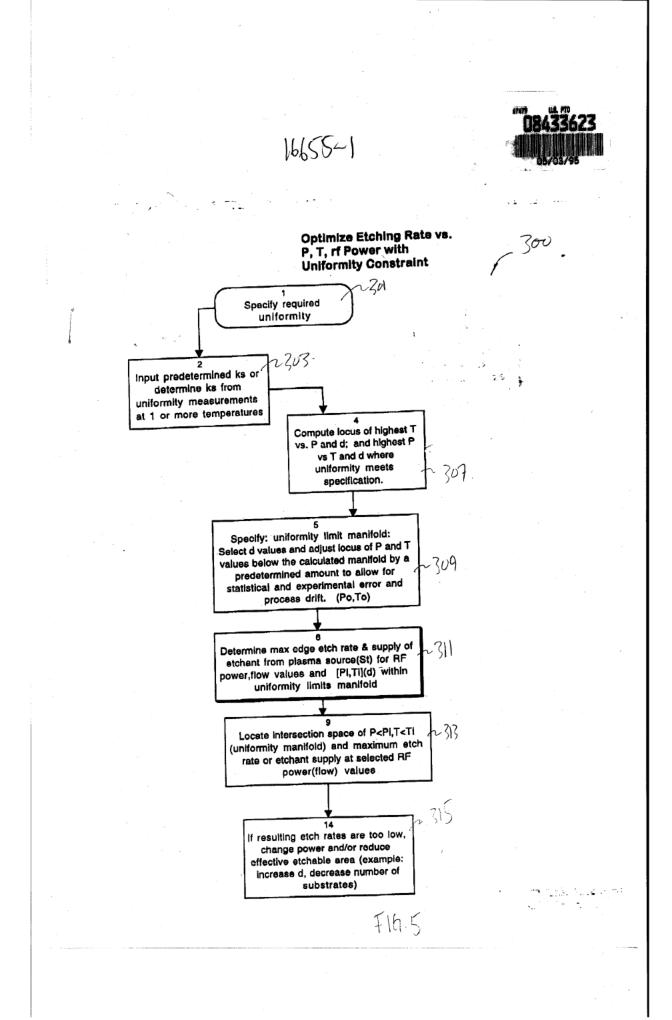
FIG. 1A

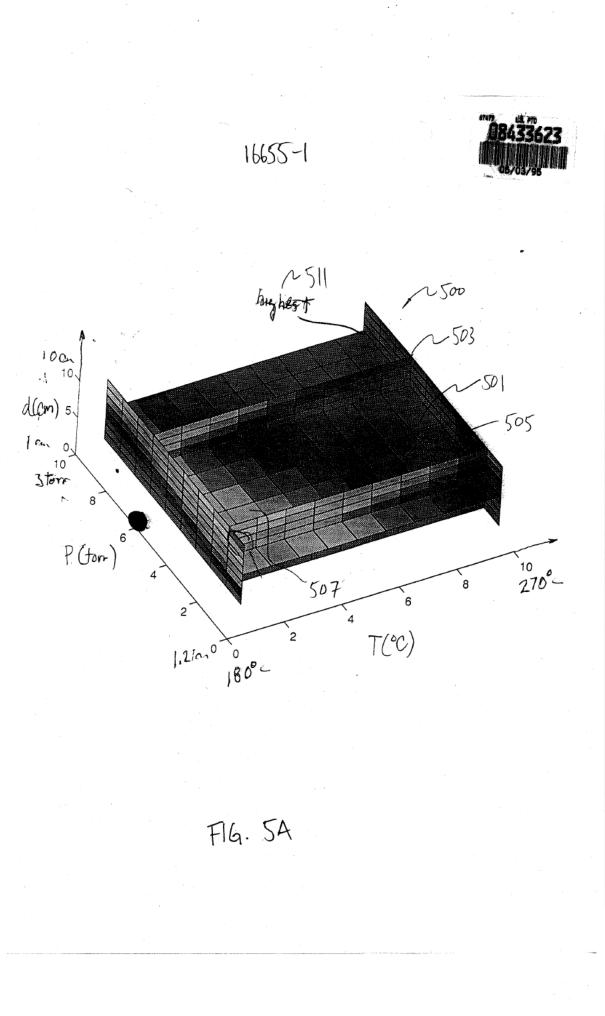


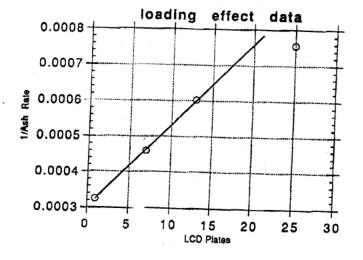




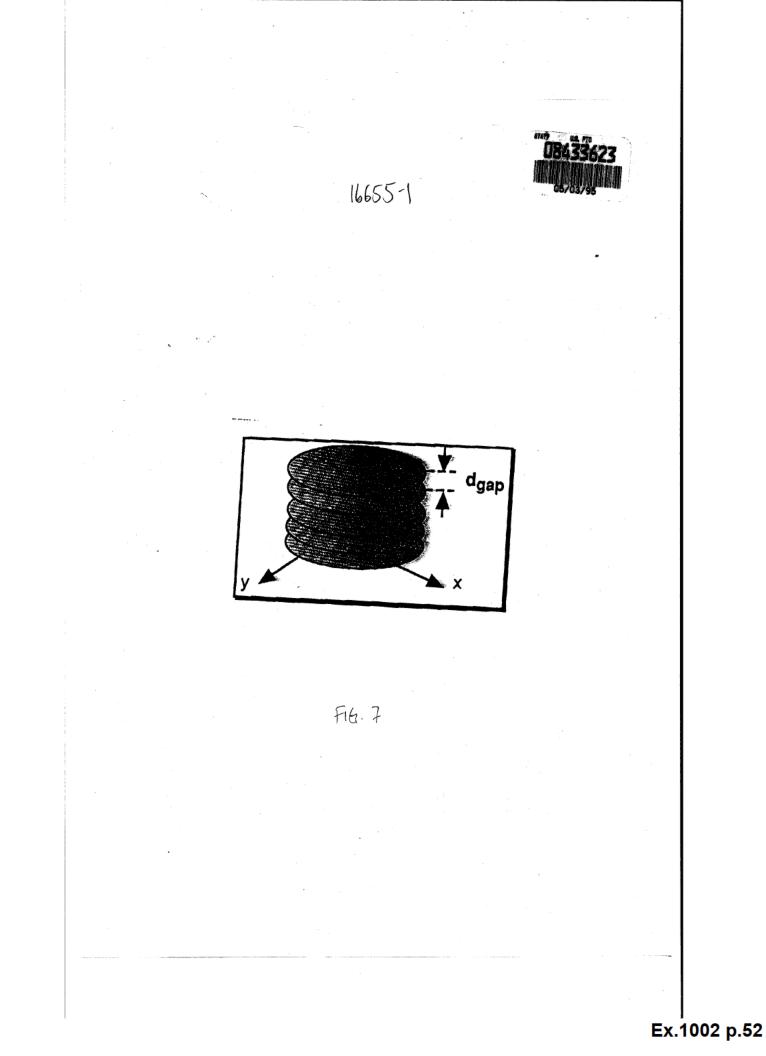
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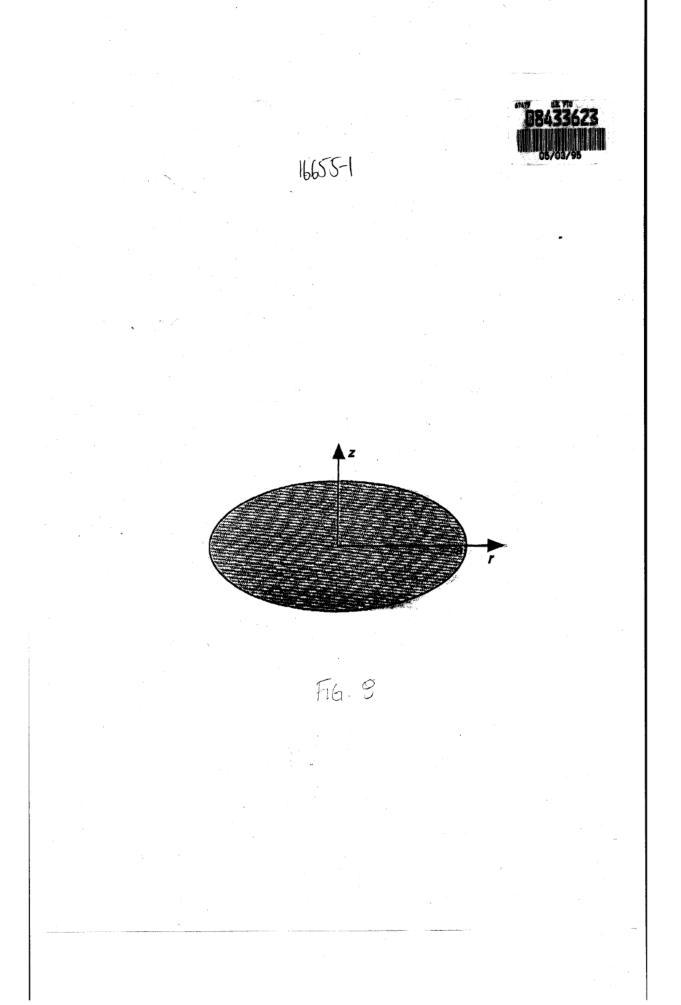


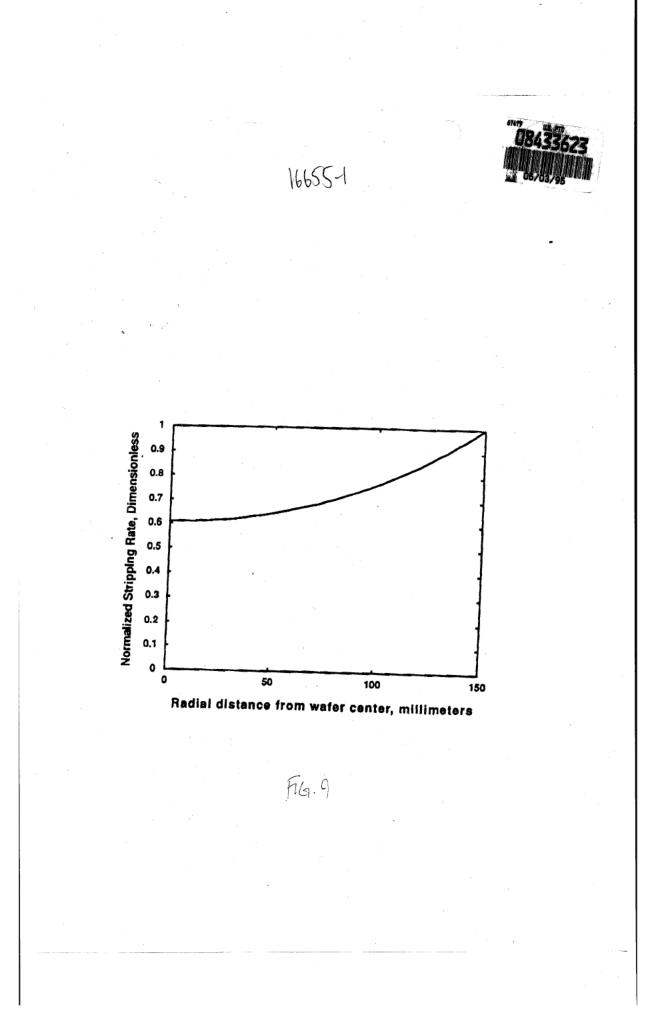


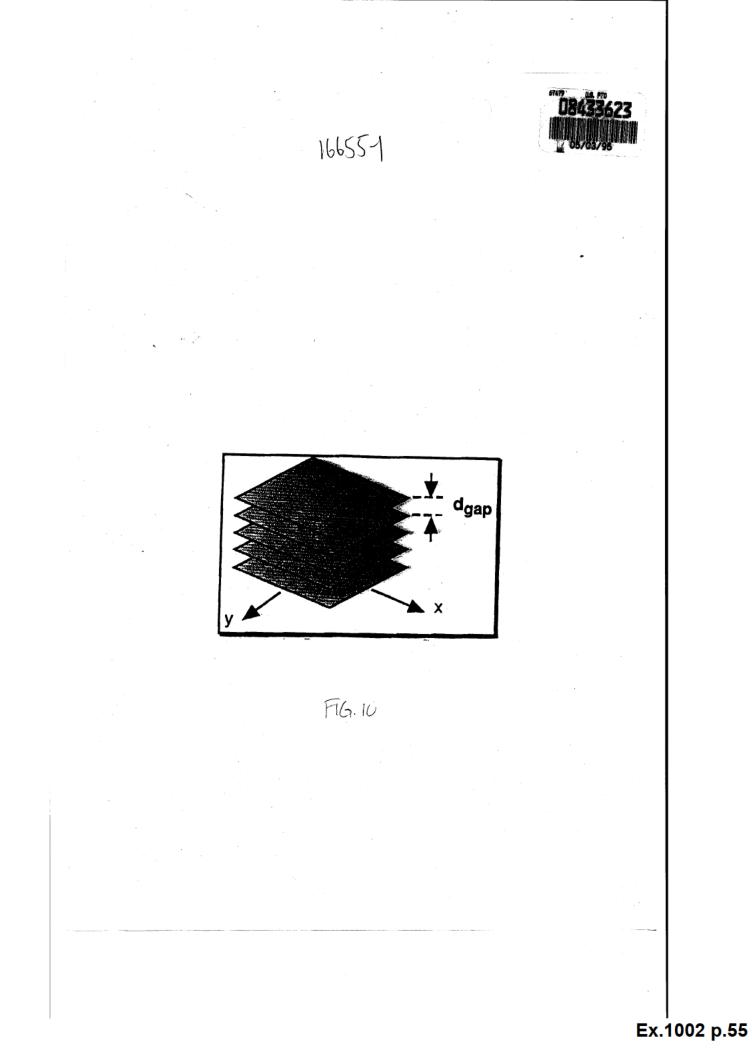


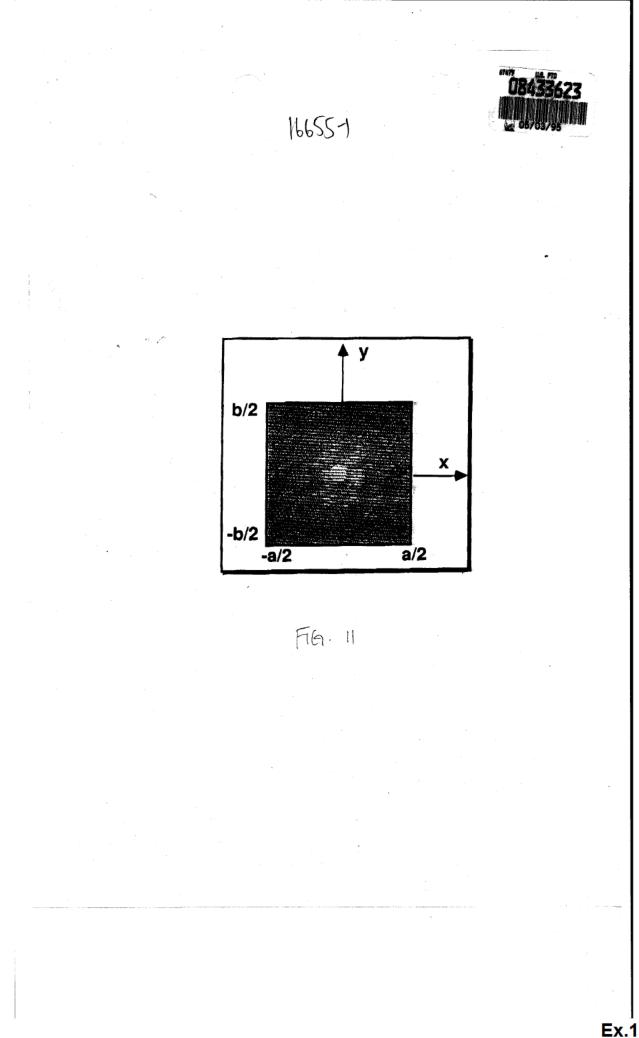




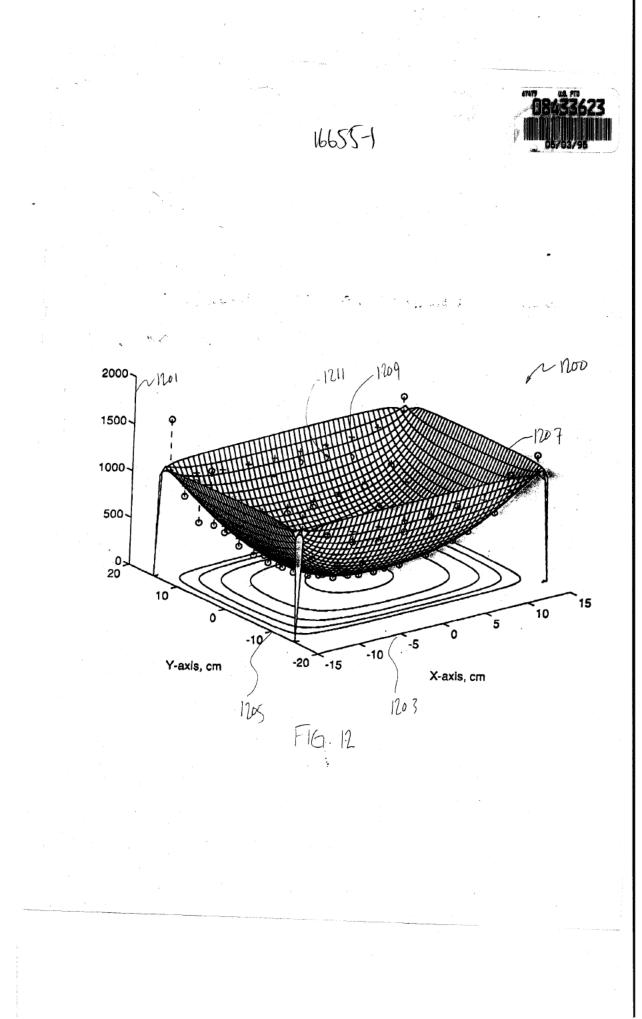


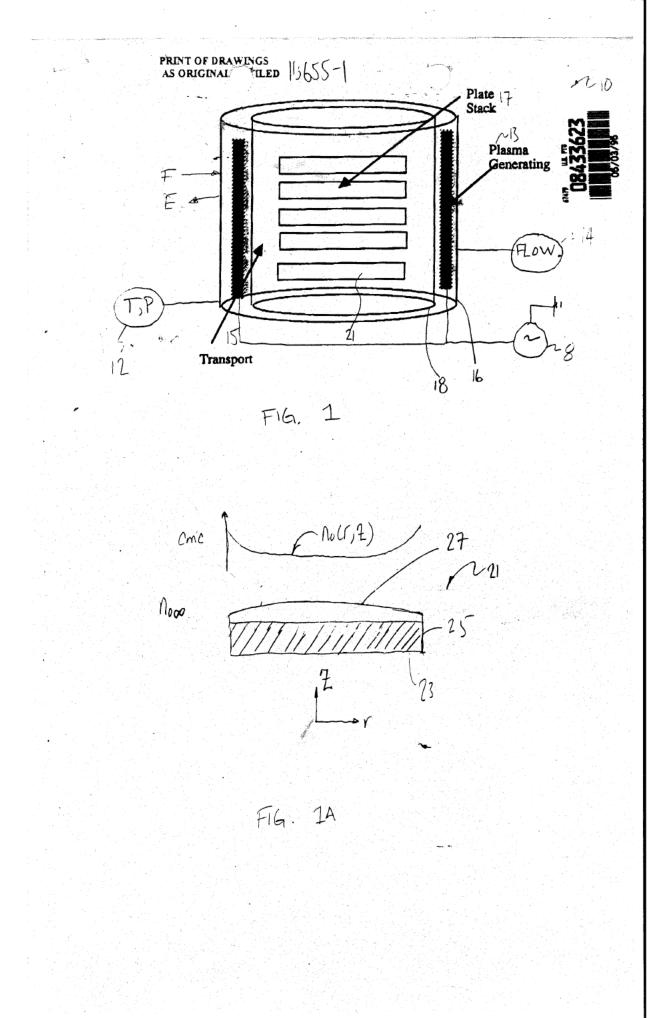




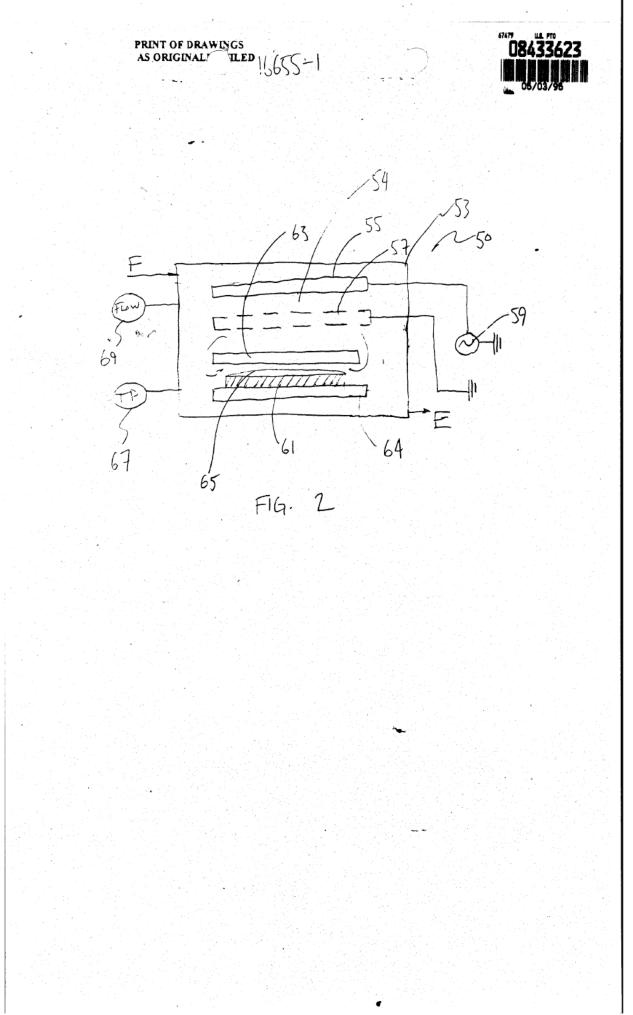


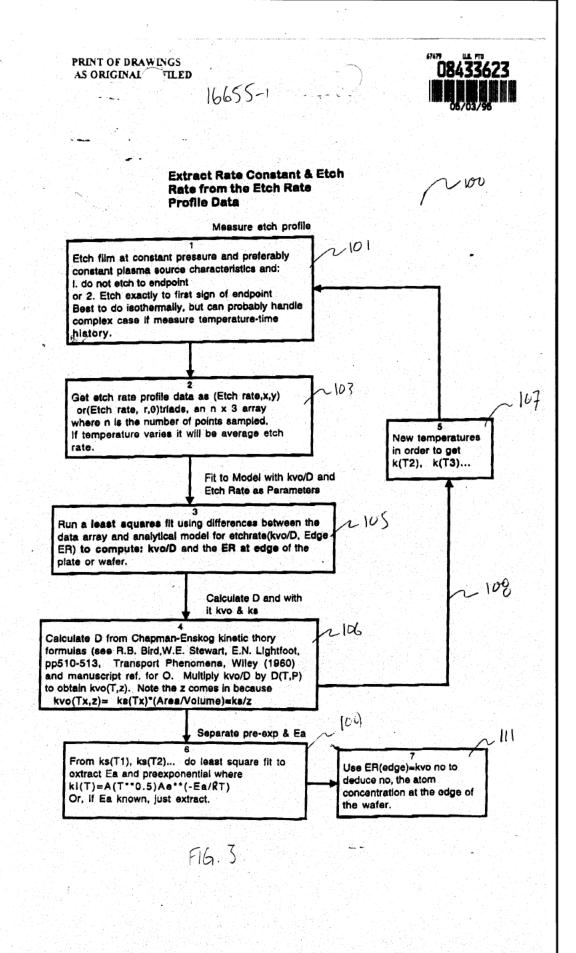
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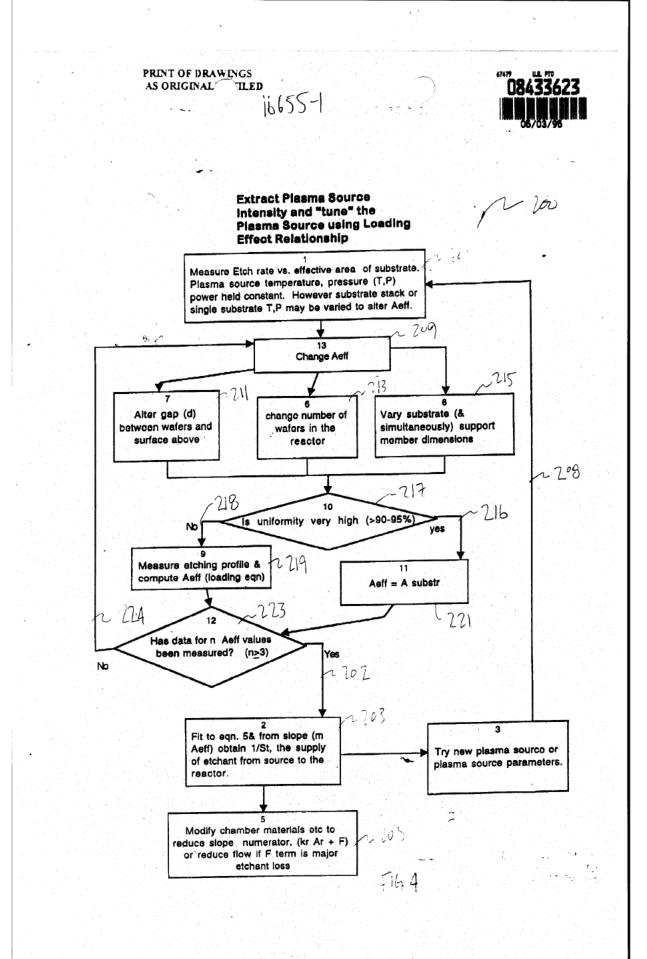


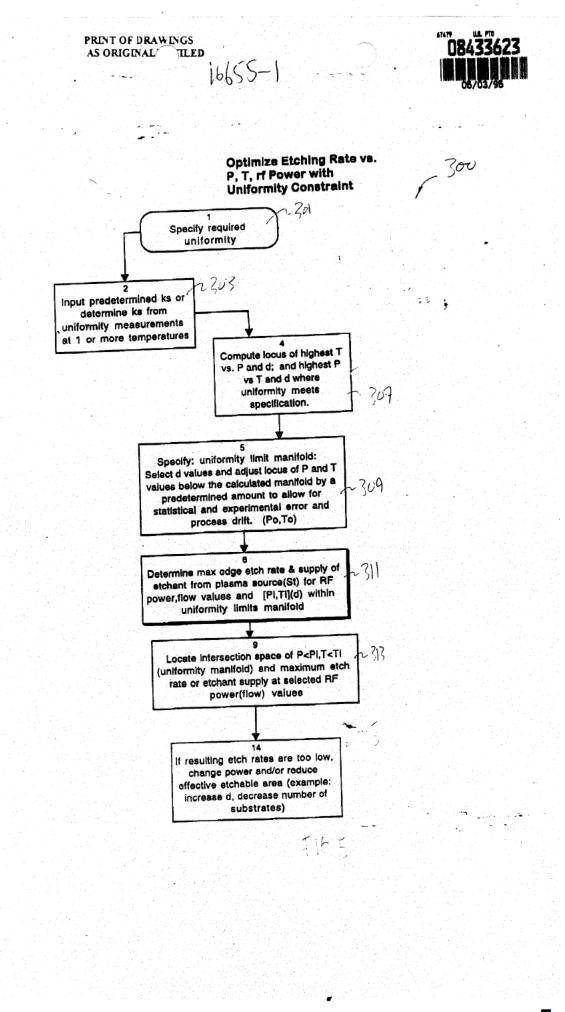


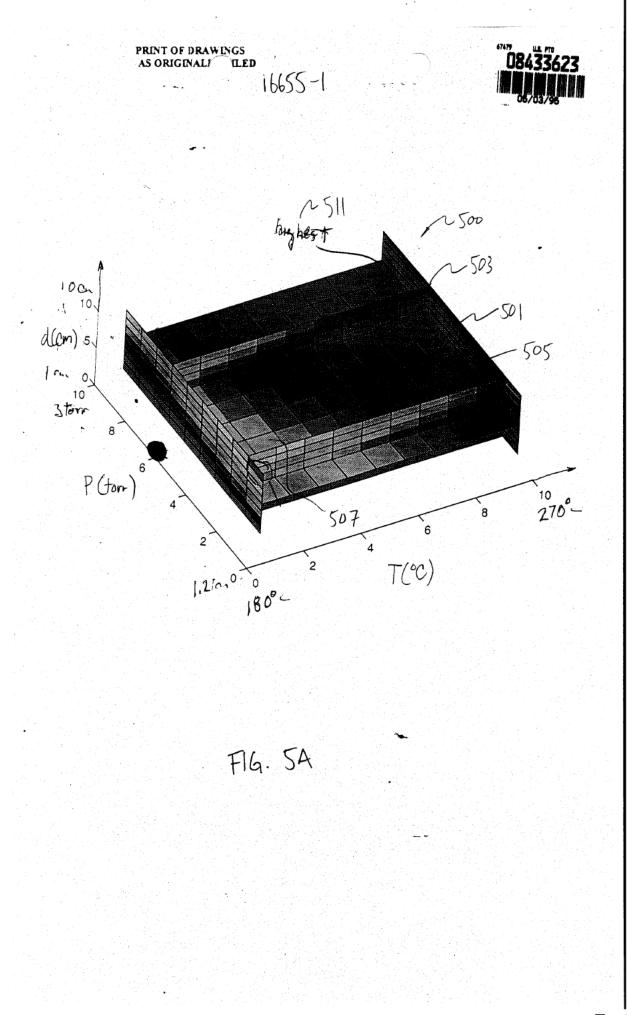
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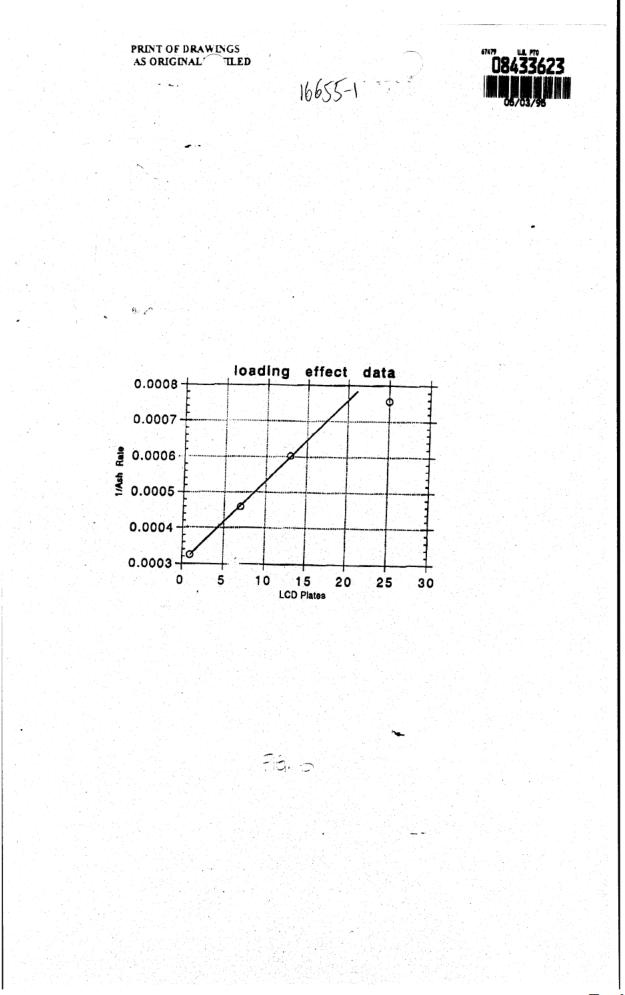


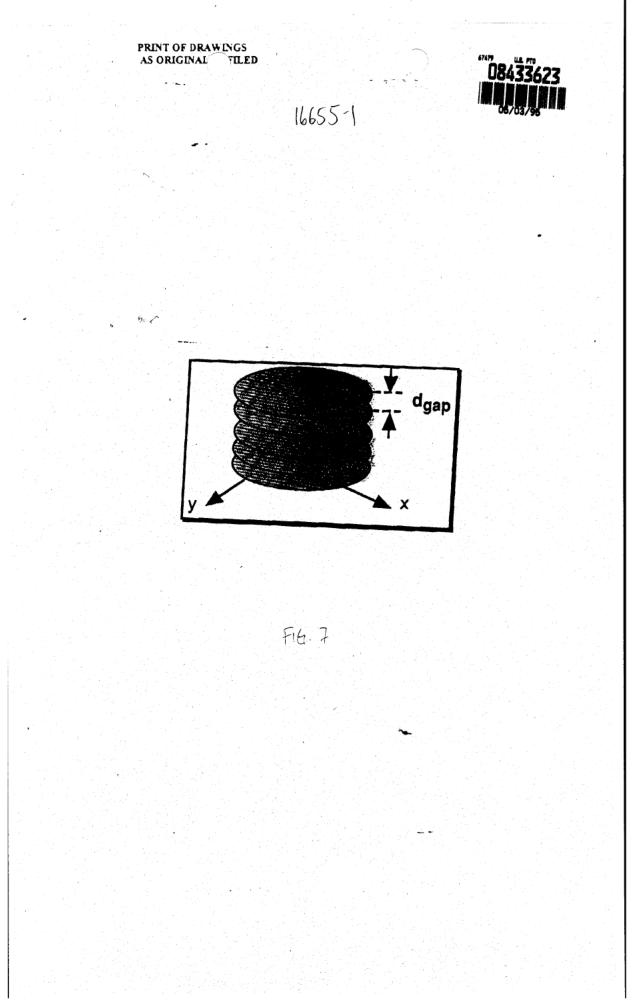


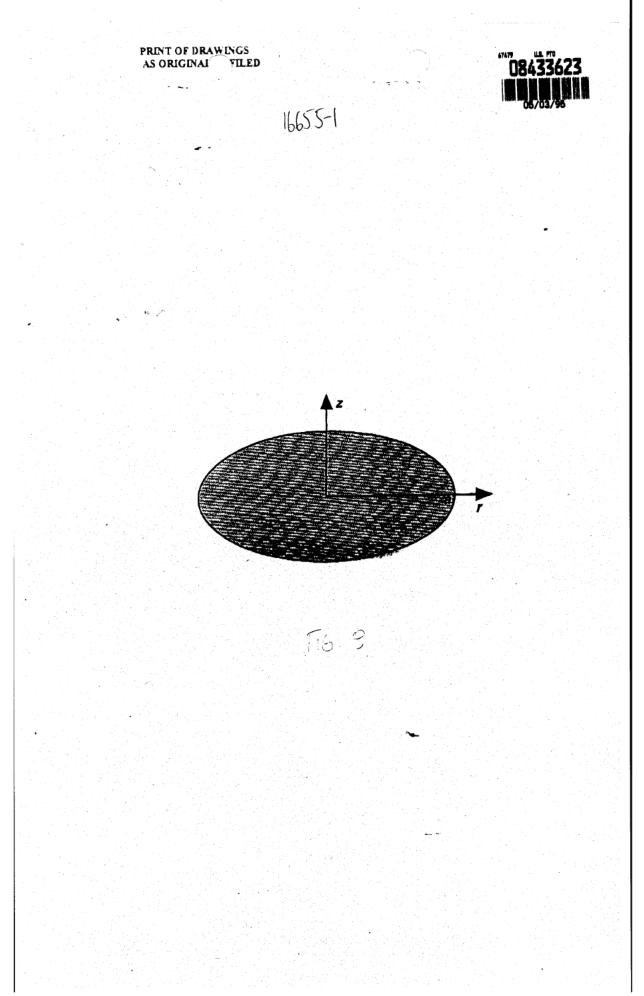


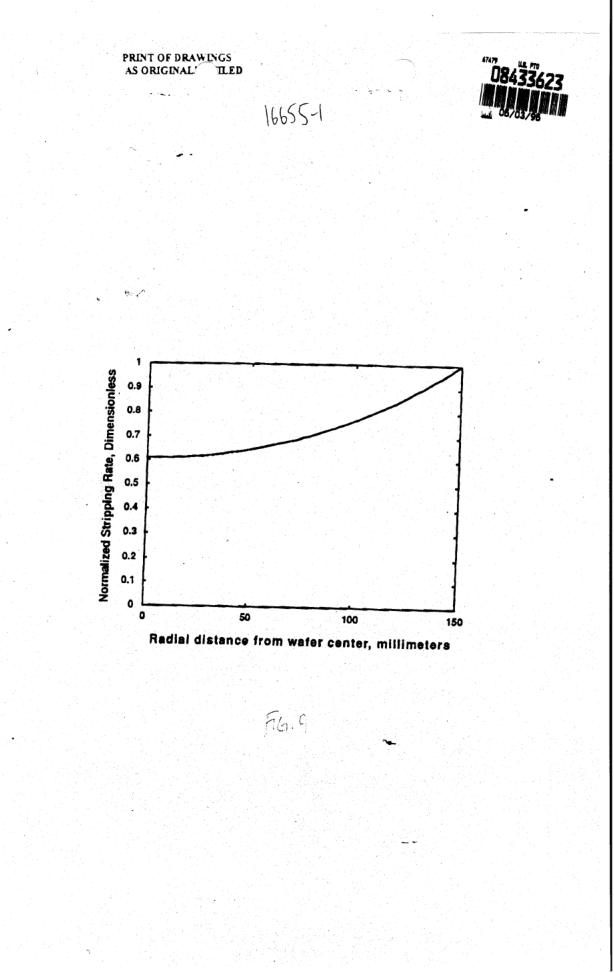


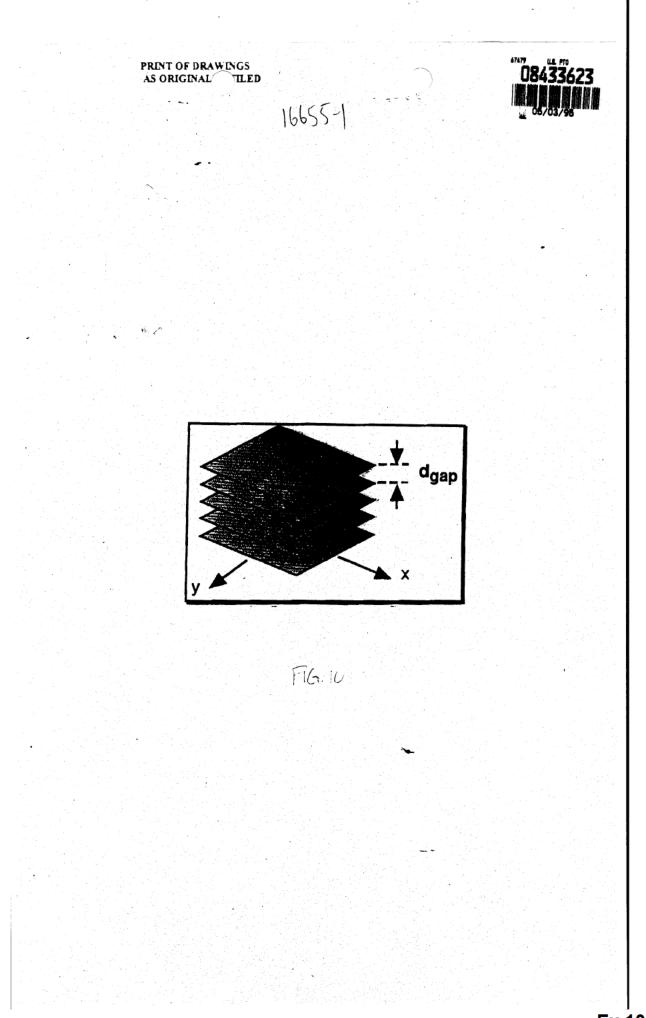


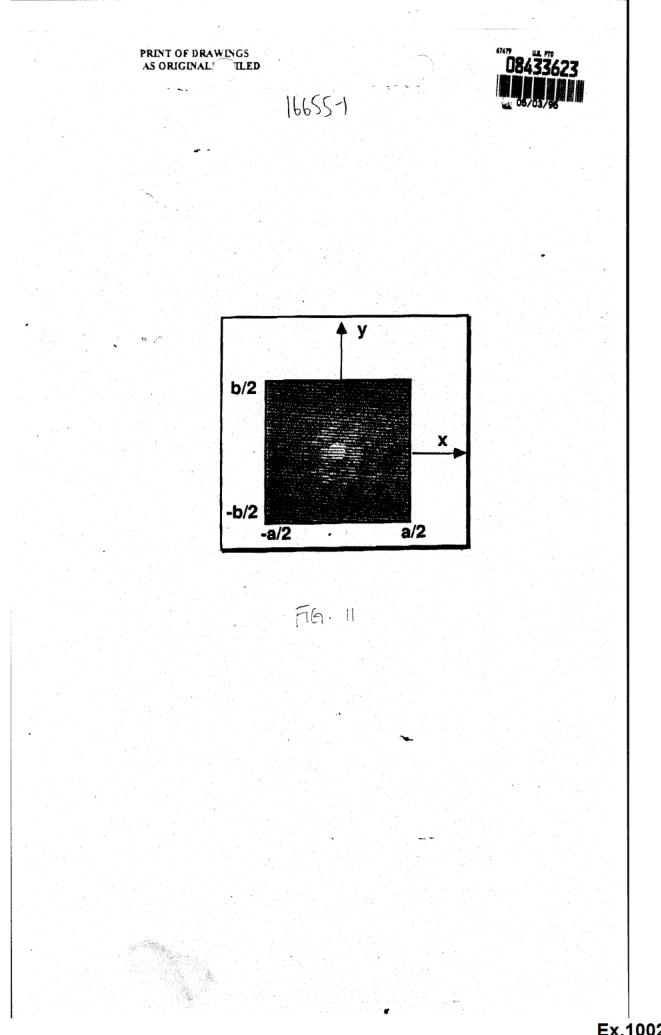


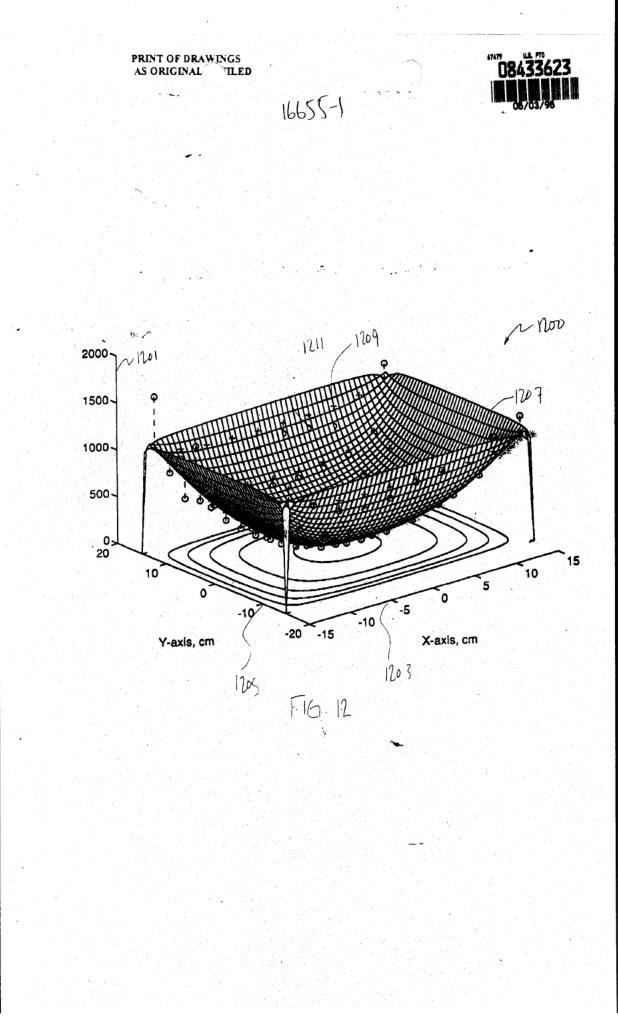














UNITED STATES DEPARTMENT OF COMMERCE Patent and Trademark Office Address: COMMISSIONER OF PATENTS AND

Washington, D.C. 20231

APPLICATION NUMBER FILING DATE FIRST NAMED APPLICANT ATTY, DOCKET NO/TITLE

05/03/95 FLAMM 08/433,623

16655000100

05/22/95

0212/0522

TOWNSEND AND TOWNSEND KHOURIE AND CREW STEUART STREET TOWER ONE MARKET PLAZA SAN FRANCISCO CA 94105

0000

DATE MAILED:

D

NOTICE TO FILE MISSING PARTS OF APPLICATION **FILING DATE GRANTED**

An Application Number and Filing Date have been assigned to this application. However, the items indicated below are missing. The required items and fees identified below must be timely submitted ALONG WITH THE PAYMENT OF A SURCHARGE for items 1 and 3-6 only of $\frac{500}{1000}$ for large entities or ____for large entities or 65 for small entities who have filed a verified statement claiming such status. The surcharge is set forth in 37 CFR 1.16(e).

If all required items on this form are filed within the period set below, the total amount owed by applicant as a \square large entity, \square small entity (verified statement filed), is $\$/2 - \Im^2$.

Applicant is given ONE MONTH FROM THE DATE OF THIS LETTER, OR TWO MONTHS FROM THE FILING DATE of this application, WHICHEVER IS LATER, within which to file all required items and pay any fees required above to avoid abandonment. Extensions of time may be obtained by filing a petition accompanied by the extension fee under the provisions of 37 CFR 1.136(a).

- 1. 🖉 The statutory basic filing fee is: 🖉 missing 🗆 insufficient. Applicant as a 🖉 large entity 🗆 small entity, must submit \$_ 30 _to complete the basic filing fee.
- M2 as a \mathbb{B} large entity, \Box small entity, including any 2. 🗷 Additional claim fees of \$ required multiple dependent claim fee, are required. Applicant must submit the additional claim fees or cancel the additional claims for which fees are due.

3. The oath or declaration:

🗆 is missing.

does not cover the newly submitted items.

An oath or declaration in compliance with 37 CFR 1.63, identifying the application by the above Application Number and Filing Date is required.

- 4. 🗆 The oath or declaration does not identify the application to which it applies. An oath or declaration in compliance with 37 CFR 1.63, identifying the application by the above Application Number and Filing Date, is required.
- 5. \mathbb{Z} The signature(s) to the oath or declaration is/are: \mathbb{Z} missing; \Box by a person other than the inventor or a person qualified under 37 CFR 1.42, 1.43, or 1.47. A properly signed oath or declaration in compliance with 37 CFR 1.63, identifying the application by the above Application Number and Filing Date, is required.
- 6. The signature of the following joint inventor(s) is missing from the oath or declaration:

An oath or declaration listing the names of all inventors and signed by the omitted inventor(s), identifying this application by the above Application Number and Filing Date, is required.

- 7. 🗆 The application was filed in a language other than English. Applicant must file a verified English under 37 CFR 1.17(k), unless this fee has translation of the application and a fee of \$_ already been paid.
- processing fee is required since your check was returned without payment. 8. 🗆 A \$ (37 CFR 1.21(m)).
- 9. 🗇 Your filing receipt was mailed in error because your check was returned without payment.

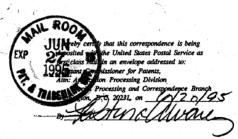
11. [] Other.

Direct the response to Box Missing Part and refer any questions to the Customer Service Center at (703) 308-1202.

A copy of this notice <u>MUST</u> be returned with the response.

OFFICE CORV

^{10.} The application does not comply with the Sequence Rules. See attached Notice to Comply with Sequence Rules 37 CFR 1.821-1.825.



-PATENT

Attorney Docket No. 16655-1

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Daniel L. Flamm et al.

Serial No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING Examiner: Unassigned Art Unit: Unassigned TRANSMITTAL LETTER

Attn: Application Processing Division Special Processing and Correspondence Branch Assistant Commissioner for Patents Washington, D.C. 20231

sir:

Pursuant to the Notice to File Missing Parts of Application - Filing Date Granted dated May 3, 1995, enclosed are the following to be made of record in the above-identified application:

Executed Declaration and Power of Attorney;

2) Verified Statement Claiming Small Entity Status;

- Copy of Notice of Missing Parts;
- Assignment; and

Recordation Form PTO-1595.

Please charge the statutory basic filing fee of \$436.00 (total claims, 23 and 4 independent claims), \$40.00 for Recordation of the Assignment, and the surcharge of \$65.00 -Total \$541.00 to Deposit Account No. 20-1430 of the undersigned. The Assistant Commissioner is hereby authorized to charge any additional fees associated with this paper or during the pendency of this application, or credit any overpayment to Deposit Account Daniel L. Fla...d Serial No.: 08/433,623 Page 2

No. 20-1430 for this paper and during the prosecution of this application. This Transmittal Letter is submitted in triplicate.

Respectfully submitted,

Richard T. Ogawa Reg. No. 37,692

TOWNSEND and TOWNSEND and CREW One Market Plaza Steuart Street Tower, 20th Floor San Francisco, California 94105 (415) 326-2400

RTO/ka

1.1

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PATENT



DECLARATION AND POWER OF ATTORNEY

Attorney Docket No. 16655-1

As a below named inventor, I declare that:

 $b_{2} \geq 0$

My residence, post office address and citizenship are as stated below next to my name; I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural inventors are named below) of the subject matter which is claimed and for which a patent is sought on the invention entitled: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING the specification of which ______ is attached hereto or _X__ was filed on ______ May 3, 1995_ as Application Serial No. <u>08/433,623</u> and was amended on ______ (if applicable).

I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above. I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, Section 1.56. I claim foreign priority benefits under Title 35, United States Code, Section 119 of any foreign applications(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)

Country	Application No.	Date of Filing	Priority Claimed Under 35 USC 119
			Yes No
			Yes No

I claim the benefit under Title 35, United States Code, Section 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, Section 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, Section 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

Application Serial No.	Date of Filing	Status
		Patented Pending Abandoned
		Patented Pending Abandoned

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith.

Richard T. Ogawa, Reg. No. 37,692 William J. Bohler, Reg. No. 31,487 Kenneth R. Allen, Reg. No. 27,301

Send Correspondence to:	Direct Telephone Calls to:
Richard T. Ogawa	(Name, Reg. No., Telephone No.)
TOWNSEND and TOWNSEND KHOURIE and CREW	
Steuart Street Tower	Name: Richard T. Ogawa
One Market Plaza, 20th Floor	Reg. No. 37,692
San Francisco, CA 94105	Telephone: 415 326-2400

(Page 1 of 2)

Full Name	Last Name)-00	First Name	Middle Name or Initial	
of Inventor 1	Flamm	Daniel		
Residence &	City	State/Foreign Country	Country of Citizenship	
Citizenship	Walnut Creek CA	California	U.S.A.	
Post Office	Post Office Address	City	State/Country Zip Code	
Address	476 Green View Drive	Walnut Creek	California 94596	
Full Name	Last Name	First Name	Middle Name or Initial	
of Inventor 2	Verboncoeur 2-00	John	P	
Residence &	City	State/Foreign Country	Country of Citizen	nship
Citizenship	Hayward CA	California	U.S.A.	
Post Office	Post Office Address	City	State/Country	Zip Code
Address	3350 Oakes Drive	Hayward	California	94542
Full Name of Inventor 3	Last Name	First Name	Middle Name or Initial	
Residence & Citizenship	City	State/Foreign Country	Country of Citizenship	
Post Office Address	Post Office Address	City	State/Country Zip Code	

I further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Signature of Inventor 1	Signature of Investor 2	Signature of Inventor 3	
Quie L. Hamm	(John P. Unbonwew		
Daniel L. Flamm	John P. Verboncoeur		
Date June 9, 1995	Date 06-12-95	Date	
			1

DP.MRG 1/93

(Page 2 of 2)

Attorney Docket No. 16655-1

DECLARATION AND POWER OF ATTORNEY

 ${}^{\oslash}$

As a below named inventor, I declare that:

My residence, post office address and citizenship are as stated below next to my name; I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural inventors are named below) of the subject matter which is claimed and for which a patent is sought on the invention entitled: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING the specification of which ______ is attached hereto or __X_ was filed on __May 3, 1995_ as Application Serial No. 08/433,623 and was amended on ______ (if applicable).

I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above. I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, Section 1.56. I claim foreign priority benefits under Title 35, United States Code, Section 119 of any foreign applications(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)

16. 1

Country	Application No.	Date of Filing	Priority Claimed Under 35 USC 119
			Yes No
			Yes No

I claim the benefit under Title 35, United States Code, Section 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, Section 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, Section 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

Application Serial No.	Date of Filing	Status
		PatentedPendingAbandoned
		PatentedPendingAbandoned

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith.

> Richard T. Ogawa, Reg. No. 37,692 William J. Bohler, Reg. No. 31,487 Kenneth R. Allen, Reg. No. 27,301

Send Correspondence to:	Direct Telephone Calls to:
Richard T. Ogawa	(Name, Reg. No., Telephone No.)
TOWNSEND and TOWNSEND KHOURIE and CREW	
Steuart Street Tower	Name: Richard T. Ogawa
One Market Plaza, 20th Floor	Reg. No. 37,692
San Francisco, CA 94105	Telephone: 415 326-2400

(Page 1 of 2)

Full Name	Last Name	First Name	Middle Name or Initial	
of Inventor 1	Flamm	Daniel	L.	
Residence &	City	State/Foreign Country	Country of Citizenship	
Citizenship	Walnut Creek	California	U.S.A.	
Post Office	Post Office Address	City	State/Country Zip Code	
Address	476 Green View Drive	Walnut Creek	California 945%	
Full Name	Last Name	First Name	Middle Name or Initial	
of Inventor 2	Verboncoeur	John	P.	
Residence &	City	State/Foreign Country	Country of Citizenship	
Citizenship	Hayward	California	U.S.A.	
Post Office .	Post Office Address	City	State/Country Zip Code	
Address	3350 Oakes Drive	Hayward	California 94542	
Full Name of Inventor 3	Last Name	First Name	Middle Name or Initial	
Residence & Citizenship	City	State/Foreign Country	Country of Citizenship	
Post Office Address	Post Office Address	City	State/Country Zip Code	

I further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Signature of Inventor	Signature of Inventor 2	Signature of Inventor 3
Jame Hamm		
Daniel L. Flamm	John P. Verboncoeur	
Date June 9, 1995	Date	Date

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(Page 2 of 2)

ANL A		1		Atty. Docket No. 16655
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De JUN	VER	RIFIED STATEMENT (DECLARATION)	CLAIMING SMALL ENTIT	Z
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Serial or Patent N		init and John P. Verboneoeur		
Filed or Issued: N				the contraction of the
Title: PROCESS	OPTIMIZATION IN	GAS PHASE DRY ETCHING		
		declare that I qualify as an independent invo ice regarding the invention entitled <u>PROCE</u>		
described m.	[] the spec	cification herewith.		
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in the invention to	any person who wo	f or licensed and am under no obligation un uld not qualify as an independent inventor as a small business concern under 37 CFR	under 37 CFR 1.9(c) if that p	erson had made the invention, or
		to which I have assigned, granted, conveye ights in the invention is listed below:*	d, or licensed or am under an	obligation under contract or law
	[X] No such	h person, concern, or organization		
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		NAME OF INVENTOR:	NAME OF	INVENTOR:
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-485.81			
Applicant or Pate	ntee: Dan	iel L. Flamm and John P. Verboncoeur	
Serial or Patent N	No.: 08/433	,623	
Filed or Issued:	May 3, 199	5	
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		I hereby declare that I qualify as an independent inventor as defined in 37 CFR 1.9(c) f mark Office regarding the invention entitled <u>PROCESS OPTIMIZATION IN GAS PH</u>	
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	[X]	application Serial No. 08/433,623 , filed May 3, 1995	and the second second second second
	[]	Patent No, issued	والمرجاة فتعرج ججا الزدادي

I have not assigned, granted, conveyed or licensed and am under no obligation under contract or law to assign, grant, convey or license, any rights in the invention to any person who would not qualify as an independent inventor under 37 CFR 1.9(c) if that person had made the invention, or to any concern which would not qualify as a small business concern under 37 CFR 1.9(d) or a nonprofit organization under 37 CFR 1.9(e).

Each person, concern or organization to which I have assigned, granted, conveyed, or licensed or am under an obligation under contract or law to assign, grant, convey or license any rights in the invention is listed below:*

- [] No such person, concern, or organization
- [X] Persons, concerns or organizations listed below*

*NOTE: Separate verified statements are required from each named person, concern or organization having rights to the invention averring to their status as small entities. (37 CFR 1.27)

	[X] INDIVIDUAL []	SMALL BUSINESS CONCERN	[] NONPROFIT ORGANIZATION
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I acknowledge the duty to file, in this application or patent, notification of any change in status resulting in loss of entitlement to small entity status prior to paying, or at the time of paying, the earliest of the issue fee or any maintenance fee due after the date on which status as a small entity is no longer appropriate. (37 CFR 1.28(b)).

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application, any patent issuing thereon, or any patent to which this verified statement is directed.

NAME OF INVENTOR:	NAME OF INVENTOR:	NAME OF INVENTOR:
JOHN P. VERBONCOEUR		
Aphn P. Dentonwew		
Signature of Inventor	Signature of Inventor	Signature of Inventor
Date 06-12-95	Date:	Date:

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re appli	cation of Daniel L	. Flamm, e	et al.		I here	by certify	that this is	being o	leposited v	with the Unit
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1995 E	I hereby certify that this correspondence is being deposited to the United States Postal Service as first class mail in an envel addressed to: Assistant Commissioner for Patents Washington, D.C. 20231, on <u>November 9</u> , 199
	Date: 11/9/95 By: Neine Elzing
	PATE 16655-000100
IN THE UNITED STATES PA	ATENT AND TRADEMARK OFFICE
In re application of:	
Daniel L, Flamm, et al.)) Examiner: Unassigned
Serial No. 08/433,623) Art Unit: Unassigned
Filed: May 3, 1995	
For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING) <u>PRELIMINARY AMENDMENT</u>
ASSISTANT COMMISSIONER FOR PAT Washington, D.C. 20231	ENTS
Sir:	
follows:	please amend the above-identified application as
IOHOWS.	
IN THE CLAIMS:	
IN THE CLAIMS:	and add new claims 24 and 25 as follows. The
IN THE CLAIMS: Please amend claims 1 and 5	and add new claims 24 and 25 as follows. The ended claims, are presented in Appendix A for
IN THE CLAIMS: Please amend claims 1 and 5	
IN THE CLAIMS: Please amend claims 1 and 5 pending claims, including the presently ame ease of reference: 1. (Amended) [An integrate	
IN THE CLAIMS: Please amend claims 1 and 5 pending claims, including the presently ame ease of reference: 1. (Amended) [An integrate Phile steps of:	ended claims, are presented in Appendix A for ad circuit] <u>A</u> device fabrication method comprisir
IN THE CLAIMS: Please amend claims 1 and 5 pending claims, including the presently ame ease of reference: 1. (Amended) [An integrate Pthe steps of: providing a plasma etching ap	ended claims, are presented in Appendix A for ad circuit] \underline{A} device fabrication method comprising pparatus comprising a substrate therein, said
IN THE CLAIMS: Please amend claims 1 and 5 pending claims, including the presently ame ease of reference: 1. (Amended) [An integrate Phile steps of: providing a plasma etching and substrate comprising a top surface and a film	ended claims, are presented in Appendix A for ad circuit] <u>A</u> device fabrication method comprising pparatus comprising a substrate therein, said m overlying said toposurfaces said film comprising $\frac{1}{200}$ Sc $\frac{1}{20}$ 1
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IN THE CLAIMS: Please amend claims 1 and 5 pending claims, including the presently ame ease of reference: 1. (Amended) [An integrate Phile steps of: providing a plasma etching and substrate comprising a top surface and a film	ended claims, are presented in Appendix A for ad circuit] <u>A</u> device fabrication method comprising paratus comprising a substrate therein, said m overlying said top surface said film comprising 300 SC 20-1 39,00CH 30079 202 39,00CH 30080 203 22,00CH

chemical etching said top film surface to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate [constant] from said etch rate data[, and using said reaction rate constant in adjusting said plasma etching apparatus].

5. (Amended) The method of claim 1 wherein said extracting step correlates said reaction rate [constant] over a diffusivity with said etching rate, said etching rate being defined by said etching profile.....

 $9_{24.}$ (New) The method of claim 1 further comprising a step of using said reaction rate constant in adjusting said plasma etch apparatus.

25 (New) A process for fabricating a device, said device being fabricated by use of a reaction rate constant, said reaction rate constant being derived from a method comprising:

providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

etching said top surface at a temperature to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting from said etching rate data a reaction rate for said temperature:

REMARKS

Applicant adds new claims 24 and 25 to the subject application for

examination. No new matter has been introduced thereby.

S

If the Examiner believes a telephone conference would expedite prosecution of

this application, please telephone the undersigned at (415) 326-2400. 2ξ

Respectfully submitted,

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TOWNSEND and TOWNSEND and CREW

Date: 119195

By: Richard T. Ogawa Reg. No. 37,692

RTO:de rio\work\16655\1-prem.amd

APPENDIX A

1. (Amended) A device fabrication method comprising the steps of: providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate [constant] from said etch rate data[, and using said reaction rate constant in adjusting said plasma etching apparatus].

2. The method of claim 1 wherein said chemical etching step is diffusion limiting.

3. The method of claim 1 wherein said spatial coordinates include a radius and an angle.

4. The method of claim 1 wherein said spatial coordinates include an xdirection and a y-direction.

5. (Amended) The method of claim 1 wherein said extracting step correlates said reaction rate over a diffusivity with said etching rate, said etching rate being defined by said etching profile.

6. The method of claim 1 wherein said etching rate is defined by said etching profile at selected spatial coordinates over a time.

7. The method of claim 1 wherein said chemical etching is an ashing method.

8. The method of claim 1 wherein said ashing method comprises reactants including an oxygen and a photoresist.

9. A method of designing a reactor comprising the steps of:

providing a first plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate constant from said etch rate data, and using said reaction rate constant in designing a second plasma etching apparatus.

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10. The method of claim 9 wherein said chemical etching step is diffusion

11. The method of claim 9 wherein said spatial coordinates include a radius and an angle.

12. The method of claim 9 wherein said spatial coordinates include an xdirection and a y-direction.

13. The method of claim 9 wherein said extracting step correlates said reaction rate constant over a diffusivity with said an etching rate, said etching rate being defined by said etching profile.

14. The method of claim 9 wherein said etching rate is defined by said etching profile at selected spatial coordinates over a time.

15. The method of claim 9 wherein said chemical etching is an ashing method.

16. The method of claim 9 wherein said ashing method comprises reactants including an oxygen and a photoresist.

17. The method of claim 9 wherein said second plasma etching apparatus is a co-axial reactor.

18. The method of claim 9 wherein said second plasma etching apparatus is a plasma etching apparatus.

19. A substrate fabrication method comprising:

providing a substrate selected from a group consisting of a semiconductor wafer, a plate, and a flat panel display, said substrate comprising a top surface;

forming a film overlying said top surface, said film comprising a top film surface;

chemical etching said top film surface to define a profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting a reaction rate constant from said etch rate data, and using said reaction rate constant in adjusting said method.

20. A method of fabricating an integrated circuit device, said method

providing a uniformity value for an etching reaction, said etching reaction including a substrate and etchant species;

defining etching parameters ranges providing said uniformity value; and adjusting at least one of said etching parameters to produce a selected etching

rate;

comprising:

limiting.

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wherein said etching rate providing an etching condition for fabrication of an integrated circuit device.

21. The method of claim 20 wherein said etching parameters can be selected from a group consisting of a temperature, a pressure, a power, a gap, and a flow rate.

22. The method of claim 20 wherein said uniformity ranges from 90% and greater.

23. The method of claim 20 wherein said uniformity ranges from 95% and greater.

24. (New) The method of claim 1 further comprising a step of using said reaction rate constant in adjusting said plasma etch apparatus.

25. (New) A process for fabricating a device, said device being fabricated by use of a reaction rate constant, said reaction rate constant being derived from a method comprising:

providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

etching said top surface at a temperature to define an etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile; and

extracting from said etching rate data a reaction rate for said temperature.

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art I THE FOLLOWING	ATTACHMENT(S) AR	E PART OF THIS ACTION:			
	nces Cited by Examine				ent Drawing Review, PTO-9 Application, PTO-152,
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rt II SUMMARY OF AC	CTION				
	1-25				are pending in the applicat
Of the above,	claims			are	withdrawn from consideratio
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Art Unit: 1113

15 Claims 1-25 are rejected under 35 U.S.C. § 112, second paragraph, as being indefinite_ for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

The claims should clearly indicate that this is a localized etching rate constant, since it combines the effects and rates of diffusion from the etching source to the surface of the thin film, the diffusion of the resulting products away from the surface of the thin film, adsorption and desorption as well as the rate of the actual etching process to make it clear what parameter is being used to evaluate and optimize uniformity.

Also please note that the applicant optimizes the same parameters (ie "temperature pressure, reactor configuration, and the like" (page 18 lines 21-24) and changing the chamber material or coating the chamber surfaces ... rf power, flow rate and the like (page 14/lines 1-16 and 2/14-20) which are commonly optimized as part of the process. Some of these parameters are not disclosed as affecting the rate constants.

16 The following is a quotation of the appropriate paragraphs of 35 U.S.C. § 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless --

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

(e) the invention was described in a patent granted on an application for patent by another filed in the United States before the invention thereof by the applicant for patent, or on an international application by another who has fulfilled the requirements of paragraphs (1), (2), and (4) of section 371(c) of this title before the invention thereof by the applicant for patent.

-2-

Art Unit: 1113

17 The following is a quotation of 35 U.S.C. § 103 which forms the basis for all obviousness rejections set forth in this Office action:

A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Subject matter developed by another person, which qualifies as prior art only under subsection (f) or (g) of section 102 of this title, shall not preclude patentability under this section where the subject matter and the claimed invention were, at the time the invention was made, owned by the same person or subject to an obligation of assignment to the same person.

18 Claims 1-6,9-14 and 18-25 are rejected under 35 U.S.C. § 102(b) as anticipated by or,

in the alternative, under 35 U.S.C. § 103 as obvious over Bobbio '520.

Bobbio '520 describes a prior art apparatus in which the barriers are placed parallel to each other the inherent etch rates as a function of position are determined in figures 4a and 4b. These are corrected to produce an etch which is relatively uniform over the entire wafer. Typically no more than three positioning steps are required to achieve uniformity deviations of less than 2% (col 8/lines 43-50, hereinafter 8/43-50).

The examiner holds that the relative etch rates disclosed are equivalent to the combined etch rate constant and that the observations and varying of the position of the barriers results in a new etching apparatus. Alternatively, it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more

Art Unit: 1113

uniform etch in the course of routine optimization. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

19 Claims 1-25 are rejected under 35 U.S.C. § 102(b) as anticipated by or, in the alternative, under 35 U.S.C. § 103 as obvious over Horiike '706.

The examiner holds that the relative etch rates as a function of distance disclosed are equivalent to the combined etch rate constant and that the observations and varying of the position of the etch target results in a new etching apparatus (see figures 9 and 10). Alternatively, it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

20 Claims 1-6,9-14 and 18-25 are rejected under 35 U.S.C. § 102(b) as anticipated by or, in the alternative, under 35 U.S.C. § 103 as obvious over Mogab '665.

Mogab '665 teaches that loading is known in the art to result in inter or intra wafer non-uniformity. Reactor design is known to increase etch uniformity. (2/15-22) The routine optimization to decrease the effects of loading on the system is disclosed. (2/63+) These changes are disclosed as reducing non-uniformity of the etch. (3/11-17) The use of a radial

-4-

Art Unit: 1113

flow apparatus to determine experimentally the optimum conditions for an etch process is disclosed (5/8-14).

The examiner holds that the relative etch rates as a function of loading disclosed are equivalent to the combined etch rate constant and that the observations and varying of the etch conditions results in a new etching apparatus, since the composition of the etchant, which makes it an etching apparatus is changed. Alternatively, it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

21 Claims 1-6,9-14 and 18-25 are rejected under 35 U.S.C. § 102(b) as anticipated by or, in the alternative, under 35 U.S.C. § 103 as obvious over Mundt et al. '162.

Mundt et al. '162 teaches the relative etch rates as a function of position for both flat and convex electrodes in figures 2 and 3.

The examiner holds that the relative etch rates disclosed are equivalent to the combined etch rate constant and that the observations and changing the electrode results in a new etching apparatus. Alternatively, it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the

-5-

Art Unit: 1113

course of routine optimization. Please note that one skilled in the art would also be expectedto either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

Claims 1-6,9-14 and 18-25 are rejected under 35 U.S.C. § 102(b) as anticipated by or, in the alternative, under 35 U.S.C. § 103 as obvious over Stefani et al. '229.

Stefani et al. '229 teaches the use of ellisometry to determine the localized etch rates over the wafer surface and/or over a plurality of wafer surfaces in the same etch chamber. (2/34-38) am model is developed to describe the relative non-uniformity of the etch. (4/49-56) This data allows the operator to change parameters and evaluate the effects of doing so. (6/39-60)

The examiner holds that the relative etch rates as a function of position as equivalent to the combined etch rate constant recited and that the observations and varying of the etch conditions results in a new etching apparatus, since the composition of the etchant, which makes it an etching apparatus is changed. Alternatively, it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

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Art Unit: 1113

23 Claims 1-25 are rejected under 35 U.S.C. § 103 as obvious over Thompson "Introduction to Microlithography".

Thompson "Introduction to Microlithography" teaches the relationship between the loading effect and the etch rate, including the etching rate constant. (page 234) The use of "large volume reactors" to reduce loading effects is disclosed. (page 234) The cases of diffusion limited processes is disclosed. (page 234) The reduction of localized depletion (non-uniformity of the etch) of the etchant may be pressure and flow rates. (page 235) Several examples of etching chambers appear on page 230..

The examiner holds that it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization in known etching processes including ashing processes which are commonly used to remove resists. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

24 Claims 1-25 are rejected under 35 U.S.C. § 103 as obvious over Mogab '665 and Thompson "Introduction to Microlithography".

It would have been obvious to one skilled in the art to preform the calculations disclosed by Thompson "Introduction to Microlithography" in the process disclosed by Mogab

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Art Unit: 1113

'665 for the different etching apparatus/conditions to optimize the parameters for a particular etch process with a reasonable expectation of success.

Claims 1-25 are rejected under 35 U.S.C. § 103 as being unpatentable over either Stefani et al. '229, Mundt et al. '162, Horiike '706 or Bobbio '520, in view of Babanov et al. Plasma Chemistry and Plasma Processing and Thompson "Introduction to Microlithography".

It would have been obvious to one skilled in the art to determine the parameters including the etch rate constants in a manner similar to that disclosed by Babanov et al. Plasma Chemistry and Plasma Processing and Thompson "Introduction to Microlithography with each of the etch processes taught by either Stefani et al. '229, Mundt et al. '162, Horiike '706 or Bobbio '520 to further optimize known parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization in known etching processes based upon the teachings of Thompson "Introduction to Microlithography" to do so to improve the uniformity of the etch process. This includes ashing processes which are commonly used to remove resists. 26 The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

Elliott "Integrated Circuit Fabrication Technology" teaches types of etching apparatus and processes for their use.

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Art Unit: 1113

Giapis et al. Appl. Phys. Lett., Ha et al. Plasma Chemistry and Plasma Processing and Gregus et al. Plasma Chemistry and Plasma Processing teach the effects of temperature of etch uniformity.

Hendricks et al. '461 and Ikeda et al. '506 teach the use of a baffle plate to increase

etch uniformity. (see figure 6)

Kubota et al. '606 teaches the use of rf fields to control the movement of the

electrons in the plasma and increase etch uniformity. (see figures 7c and 8b)

Kojima et al. '709 teaches cooling the support and using a baffle plate to increase the

uniformity of the etch process across the wafer.

Ryan et al. Plasma Chemistry and Plasma Processing teach the process of determining

etch rates

27 The applicants are reminded of thier duty of disclosure..

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Martin Angebranndt whose telephone number is (703) 308-4397.

I am normally available between 7:30 AM and 5:00 PM, Monday through Thursday and 7:30 AM and 4:00 PM on alternate Fridays.

If repeated attempts to reach me are unsuccessful, my supervisor may be reached at (703) 308-2417.

Facsimile correspondence should be directed to (703) 305-3599.

Art Unit: 1113

Any inquiry of a general nature or relating to the status of this application should be directed to the Group receptionist whose telephone number is (703) 308-0661.

Martin J. Angebranndt Patent Examiner, Group 1100 March 4, 1996 -10-

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c	5,330,606	7/1994	Kubot	a et al.		156	34	5P	10/19	91
D	4,340,461	7/1982	Hendricks et al.			156	34	5P		
Е	5,399,229	3/1995	Stefani et al.			156	62	6.1	5/19	93
F	4,297,162	10/1981	Mundt et al.			156	34	5P		
G	4,226,665	10/1980	Mogab			156	64	3.1		
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Transmittal Letter



In re application of: DANIEL L. FLAMM et al.

Serial No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN A GAS PHASE DRY ETCHING

BOX IDS ASSISTANT COMMISSIONER FOR PATENTS Washington, D.C. 20231

Sir:

Transmitted herewith are the following documents:

1) Information Disclosure Statement Under 37 CFR §1.97 and §1.98;

- 2) Form PTO-1449 (including 37 references);
- 3) Postcard.

Please charge Deposit Account No. 20-1430 as follows:

[X] Any additional fees associated with this paper or during the pendency of this application.

20231.

2 copies of this sheet are enclosed.

TOWNSEND and TOWNSEND and CREW

Atty. Docket No. 16655-000100

I hereby certify that this is being deposited with

the United States Postal Service as first class

mail in an envelope addressed to: Assistant Commissioner for Patents, Washington, D. C.

Date February 16, 1996

Date: February 16, 1996

Ogawa

Reg. No.: 37,692 Attorneys for Applicant

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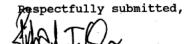
I hereby certify that this correspondence is being eposited with the United States Postal Service irst class mail in an envelope addressed to: histant Commissioner for Patents, gton, D.C. 20231, 96 Attorney Docket No. 16655-000100 16 bran nd TOWNSEND and CREW ROUP michael A. Mirande IN THE UNITED STATES PATENT AND TRADEMARK OFFICE In re application of: DANIEL L. FLAMM et al. Examiner: Application No.: 08/433,623 Art Unit: Filed: May 3, 1995 INFORMATION DISCLOSURE STATEMENT UNDER For: PROCESS OPTIMIZATION IN A 37 CFR §1.97 and §1.98 GAS PHASE DRY ETCHING Assistant Commissioner for Patents MAR Washington, D.C. 20231 Sir: GROUP 1100 The references cited on attached form PTO-1449 are

being called to the attention of the Examiner. A copy of each is enclosed.

It is respectfully requested that the cited information be expressly considered during the prosecution of this application, and the references be made of record therein and appear among the "references cited" on any patent to issue therefrom.

Applicant believes that <u>no fee is required</u> for submission of this statement, since it is being submitted prior to the first Office Action. However, if a fee is required, the Commissioner is authorized to charge such fee to Deposit Account DANIEL L. FLAM. et al. Application No.: 08/433,623 Page 2

No. 20-1430. Please charge any additional fees or credit any overpayment to the above-noted Deposit Account.



Richard T. Ogawa Reg. No. 37,692

TOWNSEND and TOWNSEND and CREW One Market Plaza Steuart Street Tower, 20th Floor San Francisco, California 94105

(415) 326-2400 Fax (415) 326-2422

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EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609; Draw line through citation if not in conformance and not considered. Include copy of this form with next communication to applicant.

1449.WP 4/95

		Amendment Tra	nsmittal	
TOWNSEND and TOWNSEND and			Atty. Docket No	0
Two Embarcadero Center, 8th Floor San Francisco, CA 94111-3834 (415) 326-2400	WALL ROOD	\ \	Date	
In re application of Daniel L. Flamme	eral. 10 3	8/	I hereby certify that this is bein States Postal Service as first	0 1
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Filed: May 3, 1995	- Children		Assistant Commissioner for Pat Washington, D. C. 20231.	tents -
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ASSISTANT COMMISSIONER FOR I Washington, D.C. 20231	PATENTS			ICT 8 1996
Sir:			GH	OUP 1100
Transmitted herewith is an amendment [X] Amendment.	in the above-ident	tified application	•	L
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** If the "Highest Number Previously Paid For" IN THIS SPACE is less than 20, write "20" in this space.

If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, write "3" in this space. The "Highest Number Previously Paid For" (Total or Independent) is the highest number found from the equivalent *** box in Col. 1 of a prior amendment or the number of claims originally filed.

[] No fee is due.

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Please charge Deposit Account No. 20-1430 as follows:

\$230.00 [X] Claims fee Any additional fees associated with this paper or during the pendency of this application. 1. 20-1430 10/03/96 08433623 copies of this sheet are enclosed. [X] _ extra copies of this sheet are enclosed. 2 39.00CH 25.00CH Richard T. Ogawa Reg. No.: 37,692 Attorneys for Applicant ø

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Attorney Docket No. 16655-000100

M. WA

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Daniel L. Flamm, et al.

Application No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING RECEIVED Examiner: OC Angebranner 1996 Art UnGROUP 1100 PETITION TO EXTEND TIME UNDER 37 CFR §1.136(a)

Assistant Commissioner for Patents Washington, D.C. 20231

Sir:

Applicants petition the Assistant Commissioner of Patents to extend the time for response to the Office Action, dated March 6, 1996 for three months, from June 6, 1996 to September 6, 1996. An appropriate response to the Office Action in the form of an Amendment is enclosed herewith.

Please charge \$450.00, pursuant to 37 CFR §1.17, to the Deposit Account No. 20-

1430. Please charge any additional fees or credit overpayment to the above Deposit Account. This Petition is submitted in triplicate.

Dated: 9696

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (415) 326-2400 RTO:de rto\work\16655\1-ext.tme Respectfully submitted,

Ogaw Reg. No. 37,692

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PATENT

Attorney Docket No. 16655-000100

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Daniel L. Flamm

Application No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING Examiner: M. Angebranndt

Art Unit: 1113

AMENDMENT UNDER 37 CFR §1.115

Assistant Commissioner for Patents Washington, D.C. 20231

Dear Sirs:

In response to the Office Action mailed March 6, 1996, the period for response being extended as a result of the enclosed Petition for Extension of Time and requisite fee, please amend the above-cited application as follows.

IN THE CLAIMS:

Please amend claims 1, 2, 5-10, 13-16, 19, 20, 22, 23, and 25; and add new claims 27-29 as follows. For the convenience of the Examiner, all claims subject to examination are shown, even if not being amended.

1. (Twice Amended) A device fabrication method comprising the steps of: providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film ov 2003 ng 2023 aid top surface 2024 said film comprising

a top film surface;

Daniel L. Flamm PATENT Application No.: 08/433,623 Page 2 [chemical] etching said top film surface to define [an] a relatively non-uniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said relatively hon-uniform etching profile, said etching comprising a reaction between a gas phase etchant and said film; and extracting a surface reaction rate constant [reaction rate] from said etch rate data, and using said surface reaction rate constant in the fabrication of a device. 2. (Amended) The method of claim 1 wherein said [chemical] etching step is diffusion limiting. 3. The method of claim 1 wherein said spatial coordinates include a radius and an angle. 4. The method of claim 1 wherein said spatial coordinates include an xdirection and a y-direction. 5. (Amended) The method of claim 1 wherein said extracting step correlates said surface reaction rate constant [reaction rate] over a diffusivity with said etch[ing] rate, said etch[ing] rate being defined by said relatively non-uniform etching profile. 6. (Amended) The method of claim 1 wherein said etch[ing] rate is defined by said relatively non-uniform etching profile at selected spatial coordinates over a time. 7. (Amended) The method of claim 1 wherein said [chemical] etching is an ashing method. 8. (Amended) The method of claim [1] 7 wherein said ashing method comprises reactants including an oxygen and a photoresist.

Daniel L. Flamm Application No.: 08/433,623 Page 3

9.

(Amended) A method of designing a reactor comprising the steps of: providing a first plasma etching apparatus comprising a substrate therein, said

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substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

[chemical] etching said top film surface to define [an] a relatively non-uniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile, said etching comprising a reaction between a gas phase etchant and said film; and

extracting a surface reaction rate constant [reaction rate constant] from said etch rate data, and using said [reaction rate constant] surface reaction rate constant in designing a second plasma etching apparatus

10. (Amended) The method of claim 9 wherein said [chemical] etching step is diffusion limiting

11. The method of claim 9 wherein said spatial coordinates include a radius and an angle.

12. The method of claim 9 wherein said spatial coordinates include an xdirection and a y-direction.

(Amended) The method of claim 9 wherein said extracting step ¥. correlates said surface reaction rate constant [reaction rate constant] over a diffusivity with said [an] etch[ing] rate, said etch[ing] rate being defined by said relatively non-uniform etching profile.

15 14. (Amended) The method of claim $\not B$ wherein said etch[ing] rate is defined by said relatively non-uniform etching profile at selected spatial coordinates over a time.

(Amended) The method of claim 9 wherein said [chemical] etching is an ashing method.

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	Daniel L. Flamm Application No.: 08/433,623
25	Page 4 $1/9$ 16. (Amended) The method of claim [9] 15 wherein said ashing method
9	comprises reactants including an oxygen and a photoresist.
	·
	17. The method of claim 9 wherein said second plasma etching apparatus is a co-axial reactor.
	18. The method of claim 9 wherein said second plasma etching apparatus is
	a plasma etching apparatus.
CH	19. (Amended) A substrate fabrication method comprising:
Mr.	providing a substrate selected from a group consisting of a semiconductor
UU_{i}	wafer, a plate, and a flat panel display, said substrate comprising a top surface;
	forming a film overlying said top surface, said film comprising a top film
Le	surface;
PI	[chemical] etching said top film surface to define a relatively non-uniform
-	profile on said film, and defining etch rate data comprising an etch rate and a spatial
Priferon Andre	coordinate from said etching profile, said etching comprising a reaction between a gas phase
and a second	etchant and said film; and
COULT DATE:	extracting a surface reaction rate constant [reaction rate constant] from said
1. Book and	etch rate data, and using said [reaction rate constant] surface reaction rate constant in
	adjusting said method.
	20. (Amended) A method of fabricating an integrated circuit device, said
(Section)	method comprising: providing a uniformity value and a surface reaction rate constant for an etching
	reaction, said etching reaction including a substrate and etchant species;
	defining etching parameters [ranges] providing said uniformity value; and
	adjusting at least one of said etching parameters using said surface reaction
	rate constant to produce a selected etching rate;
15 XX	wherein said etching rate providing an etching condition for fabrication of an
and the second	integrated circuit device.
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Daniel L. Flamm PATENT Application No.: 08/433,623 Page 5 21. The method of claim 20 wherein said etching parameters can be selected from a group consisting of a temperature, a pressure, a power, a gap, and a flow rate. 24/22 (Amended) The method of claim 20 wherein said uniformity value ranges from 90% and greater. (Amended) The method of claim 20 wherein said uniformity value ranges from 95% and greater. 24. The method of claim 1 further comprising a step of using said reaction rate constant in adjusting said plasma etch apparatus. 25. (Amended) A process for fabricating a device, said device being fabricated by use of a surface reaction rate constant [reaction rate constant], said surface reaction rate constant [reaction rate constant] being derived from a method comprising: providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface; etching said top surface at a temperature to define [an] a relatively nonuniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate from said etching profile, said etching comprising a reaction between a gas phase etchant and said film; and extracting from said etching rate data a surface reaction rate constant [reaction rate] for said temperature. The process of claim 25 wherein said surface reaction rate 26. (New) constant is derived from at least a diffusivity. 27. (New) The process of claim 25 wherein said etching is provided whereupon chemical effects are enhanced over ion bombardment effects.

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10 28. (New) The method of claims 1, 9, or 19 wherein said etching is provided whereupon chemical effects are enhanced over ion bombardment effects.

20

29. (New) A method of fabricating a device or designing a reactor using a surface reaction rate constant, said surface reaction rate constant is provided by at least a diffusivity value.

REMARKS

Reconsideration of these claims, as amended, is respectfully requested. Claims 1-29 are now pending in this application.

35 U.S.C. §112

Claims 1-25 were rejected under 35 U.S.C. §112, second paragraph, for failing to particularly point out and distinctly claim the subject matter which Applicants regard as the invention. In particular, the Examiner indicates that the "claims should clearly indicate that this is a localized etching rate constant." Office Action mailed 3/6/96, page 2, first paragraph. As described by the Examiner, the localized etching rate constant combines the effects and rates of diffusion, adsorption and desorption, and the actual etching rate.

Applicants, however, intended the "reaction rate constant" as recited by the claims to be a "surface reaction rate constant," which is described throughout the present patent specification and notably at page 17, lines 15-20 as k_s. This surface reaction rate constant, commonly termed ks, is not the so-called localized etching rate constant according to the Examiner's description. As indicated by the Examiner, the localized etching rate constant is believed to be what is commonly termed as a phenomenological or an overall etching rate constant, which includes influences of substantially all the etching variables.

The surface reaction rate constant, however, depends predominantly upon temperature, as defined throughout the present patent specification, but most notably by the equation at page 11 line 16. The surface reaction rate constant is also generally for the chemical reaction between a gas phase etchant and the film, which is not the overall etching process. Unlike the phenomenological reaction rate constant, the surface reaction rate constant can be used to derive process certain variables for device fabrication and reactor design, for example. These variables include temperature, flow rate, gap, reactor

dimensions, and others. This allows a user of such surface reaction rate to adjust these process or reactor variables without undue "trial and error" as disclosed by the patent specification.

Based upon the above description of the various types of reaction rate constants, it is believed that any claim indefiniteness perceived by the Examiner has been resolved. This should also clarify those variables which could be calculated by the use of the surface reaction rate constant, which are clearly defined throughout the specification. Accordingly, these rejections to claims 1-25 under 35 U.S.C. §112, second paragraph, are now moot.

35 U.S.C. §§102/103

The Examiner has rejected claims 1-6, 9-14, and 18-25 under 35 U.S.C. §102(b) as being anticipated by, or in alternative, under 35 U.S.C. §103 as being obvious over Bobbio. In particular, it is asserted that Bobbio discloses an apparatus and relative etch rates, which were believed similar to those in the claims.

Bobbio, however, fails to show or suggest the combination of etching and extracting a surface reaction rate constant, as recited by claims 1, 9, 19, and 25. In particular, Bobbio fails to show or suggest etching a top film surface to define a relatively non-uniform etching profile on the film and defining etch rate data comprising an etch rate and a spatial coordinate from the relatively non-uniform etching profile. Bobbio also fails to show or suggest a further combination of extracting a <u>surface reaction rate constant</u> from the etch rate data, which is provided by a non-uniform etching profile from etching using a gas phase etchant and the film. This surface reaction rate constant can be used to derive other etching parameters such as a temperature, a gap, a flow rate, reactor dimensions, and others, as shown by the present specification.

In fact, Bobbio does not even suggest any surface reaction rate constant to define these parameters, but merely uses the same "trial and error" process described in the background of invention section of the present patent specification. Clearly, Bobbio "found that typically no more than three positioning steps, including the initial positioning, are <u>required</u> to achieve acceptable uniformity for the split cathode magnetron," which is simply the conventional technique of 'trial and error.' [Emphasis added.] Bobbio, col. 8 lines 43-50. Bobbio also fails to show or suggest a later step of using the surface reaction rate

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constant for the fabrication of a device such as those recited by claims 1, 19, or 25, or for designing a reactor as recited by claim 9. Accordingly, claims 1-6, 9-14, and 18-25 are patentable over Bobbio.

The Examiner also rejected claims 1-25 under 35 U.S.C. §102(b) as being anticipated by or, in the alternative, under 35 U.S.C. §103 as being obvious over Horiike. Horiike supposedly disclosed etch rates as a function of distance. Horiike, however, also fails to show or suggest the combination of etching and extracting a surface reaction rate constant, as claimed.

In particular, Horiike does not show or suggest a step of etching a top film surface to define a relatively non-uniform etching profile on the film and defining etch rate data comprising an etch rate and a spatial coordinate from the relatively non-uniform etching profile. Horiike also fails to show or suggest any further step of extracting a <u>surface</u> <u>reaction rate constant</u> from the etch rate data. As noted above, this surface reaction rate constant can be used to derive other etching parameters such as a temperature, a gap, a flow rate, reactor dimensions, and others, as shown by the present specification. At best, the etch rate data of Fig. 9 in Horiike may provide a phenomenological etch rate constant, which appears to vary with position. In contrast, the surface reaction rate constant is generally independent of most variables including position, but is dependent predominantly on temperature, as noted above and described throughout the present patent specification. Furthermore, Horiike does not show or suggest using the surface reaction rate constant for fabricating a device such as those in claims 1, 19, and 25, or for designing the reactor of claim 9. Accordingly, claims 1-25 are patentable over Horiike.

Claims 1-6, 9-14, and 18-25 were rejected under 35 U.S.C. §102(b) as being anticipated by or, in the alternative, under 35 U.S.C. §103 as being obvious over Mogab. Mogab generally taught adjusting the reactor design to reduce the influences of loading. Mogab, however, clearly fails to show or suggest the further combination of etching a top film surface to define a relatively non-uniform etching profile on the film and defining etch rate data comprising an etch rate and a spatial coordinate from the relatively non-uniform etching profile, as claimed.

Furthermore, Mogab fails to show or suggest any further step of extracting a <u>surface reaction rate constant</u> from the etch rate data, which can be used to derive other etching parameters such as a temperature, a gap, a flow rate, reactor dimensions, and others.

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Daniel L. Flamm Application No.: 08/433,623 Page 9

At best, the etching taught by Mogab is an overall etching rate, where a phenomenological etch rate constant might be derived. As noted above and further emphasized herein, this phenomenological etch rate constant is not a surface reaction rate constant, which is predominately dependent upon temperature. Moreover, Mogab does not show or suggest using the surface reaction rate constant for device fabrication or using the reaction rate constant for designing a reactor, as claimed. Accordingly, claims 1-6, 9-14, and 18-25 are patentable over Mogab.

Claims 1-6, 9-14, and 18-25 were rejected under 35 U.S.C. §102(b) as being anticipated by or, in the alternative, under 35 U.S.C. §103 as being obvious over Mundt <u>et</u> <u>al</u>. This references supposedly taught relative etch rates as a function of position for both flat and convex electrodes in Figs. 2 and 3. Mundt <u>et al</u>., however, clearly fails to show or suggest the further combination of etching a top film surface to define a relatively nonuniform etching profile on the film and defining etch rate data comprising an etch rate and a spatial coordinate from the relatively non-uniform etching profile, as claimed.

Mundt <u>et al</u>. also fails to show or suggest any further step of extracting a <u>surface reaction rate constant</u> from the etch rate data, which can be used to derive other etching parameters such as a temperature, a gap, a flow rate, reactor dimensions, and others. Similar to those previous references, Mundt <u>et al</u>. merely discloses relative etching rates, which may produce a phenomenological rate constant. This phenomenological rate constant is not the same as the present surface reaction rate constant. Thus, Mundt <u>et al</u>. is no more relevant than any of the above references and clearly does not show or suggest using such surface reaction rate constant in fabricating a device or designing a reactor, as claimed. Accordingly, claims 1-6, 9-14, and 18-25 are patentable over Mundt <u>et al</u>.

Claims 1-6, 9-14, and 18-25 were rejected under 35 U.S.C. 102(b) as being anticipated by or, in the alternative, under 35 U.S.C. §103 as being obvious over Stefani. In particular, Stefani supposedly taught an ellipsometer to determine the localized etch rates over the wafer surface, which generally has nothing to do with the invention recited by the pending claims.

Clearly, Stefani did not show or suggest a combination of etching a top film surface to define a relatively non-uniform etching profile on the film, which defines an etch rate and a spatial coordinate, as recited by the claims. Stefani also fails to further show or suggest a step of extracting a <u>surface reaction rate constant</u> from the etch rate data. This

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surface reaction rate constant can be used to derive other etching parameters such as those noted above. Accordingly, claims 1-6, 9-14, and 18-25 are patentable over Stefani.

The Examiner rejected claims 1-25 under 35 U.S.C. §103 as being obvious over Thompson, "Introduction to Microlithography." In particular, the Examiner indicated that Thompson taught a relationship between the loading effect and the etch rate, including the etching rate constant.

Thompson, however, fails to show or suggest the claimed combination of etching a top film surface to define a relatively non-uniform etching profile on the film, which defines an etch rate and a spatial coordinate and extracting a surface reaction rate constant from the etch rate data. At best, Thompson merely taught the same "trial and error" technique of adjusting a reactor as disclosed by the present patent specification. Accordingly, claims 1-25 are patentable over Thompson.

Claims 1-25 were also rejected under 35 U.S.C. §103 as obvious over Mogab and Thompson, "Introduction to Microlithography". As shown above, neither of these references show or suggest the combination of etching a film to provide etch rate data and extracting a surface reaction rate constant. This surface reaction rate constant can be used in adjusting variables for reactor design or fabriating a device, as claimed. Accordingly, these claims should be patentable over the combination of these references.

Claims 1-25 were rejected under 35 U.S.C. §103 as being unpatentable over either Stefani <u>et al.</u>, Mundt <u>et al.</u>, Horiike, or Bobbio, in view of Babanov <u>et al.</u>, Plasma Chemistry and Plasma Processing and Thompson, "Introduction to Microlithography". Neither of these references show or suggest the further combination of etching a film to provide etch rate data and extracting a surface reaction rate constant from the data, as recited by the above claims. Furthermore, these references, alone or in combination, further fail to show or suggest the claimed step of using the surface reaction rate constant in fabricating a device or designing a reactor. Clearly, claims 1-25 are patentable over these references under 35 U.S.C. §102 and §103.

The above-cited references, alone or in combination, also fail to show or suggest the method of fabricating an integrated circuit device of claim 20, as amended. In particular, they fail to show or suggest steps of providing a uniformity value and a surface reaction rate constant for an etching reaction, which was discussed in detail above. These references also fail to show or suggest defining etching parameters, which provide the

uniformity value, and adjusting at least one of the etching parameters using the surface reaction rate constant to produce a selected etching rate. This etching rate provides an etching condition for fabrication of an integrated circuit device. By way of the surface reaction rate constant and the etching parameters, a user can adjust at least one of the etching rate parameters and determine its influence on the other parameters by way of this claim. Otherwise, the user would need to resort to "trial and error" as disclosed in the background of invention section of the present patent specification. Accordingly, claim 20 is patentable over the cited references under 35 U.S.C. §102 and §103.

Dependent claims 2-8, 10-18, and 21-24 are also patentable over the cited references for at least the same reasons noted above. In addition, these claims provide further patentable features to their dependent claims. In fact, since the Examiner does not appear to specifically point out the non-patentability of these claims directly, they must be patentable. Accordingly, the dependent claims should be patentable under 35 U.S.C. §102 and §103.

Applicants have also added new claims 27-29. No new matter has been introduced thereby. Accordingly, claims 27-28 should be entered by way of this amendment for examination.

Applicants have reviewed the other art cited by the Examiner and believe that they are no more relevant. In particular, Elliott, Giapis <u>et al.</u>, Hendricks <u>et al.</u>, Kubota et al., Kojima <u>et al.</u>, and Ryan <u>et al</u>. appear to be no more relevant than any of the other references cited above. These references seem to disclose the same concepts as disclosed in the background of invention section of the present patent specification. Accordingly, claims 1-25 are clearly patentable over these references.

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CONCLUSION

Therefore, in view of the remarks above, Applicant respectfully requests that the rejections be removed, that claims 1-29 be allowed, and the case passed to issue. The issuance of a formal Notice of Allowance at an early date is respectfully requested.

If the Examiner believes a telephone conference would expedite prosecution of this application, please telephone the undersigned at (415) 326-2400.

Respectfully submitted,

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Date:___

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (415) 326-2400 Fax (415) 326-2422

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Richard T. Reg. No. 37,692

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Art Unit: 1113

The reponses provided by the applicant has been read and given careful consideration. The applicant generally argues that the surface reaction rate constant can be used to determine or optimize a host of parameters of the etching conditions, including temperature, the gap (between the source and the surface), the flow rate, reactor dimensions, etc. Of these only the plasma temperature could reasonably be expected to have a small impact on the transport rate constants as it is related most closely with the energy of the etchant species. Additionally, the applicant appears to neglect in the arguments that many of these parameters are not disclosed as affecting the rate constants. Also please note that the applicant optimizes the same parameters (ie "temperature pressure, reactor configuration, and the like" (page 18 lines 21-24) and changing the chamber material or coating the chamber surfaces ... rf power, flow rate and the like (page 14/lines 1-16 and 2/14-20) which are commonly optimized as part of the process and represent the only parameters availible for optimization.

The examiner also notes that the applicant sent only a single page of the Bird et al. reference and the table of contents of the Manos and Flamm reference and hopes that the applicant recognizes that this means that only those portions of the references have been considered in determining patentability and that this cannot in any way be interpreted in mean that any other portion of these references has been considered.

16 Claims 26 and 29 are objected to under 37 CFR 1.75(c), as being of improper dependent form for failing to further limit the subject matter of a previous claim. Applicant

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is required to cancel the claim(s), or amend the claim(s) to place the claim(s) in proper dependent form, or rewrite the claim(s) in independent form.

The applicant argues that the surface reaction rate constant is the etching rate constant without contributions from transport phenomena. The claim seems to indicate otherwise. The diffusivity measurement is used to isolate or determine the actual surface reaction rate constant from the volumetric reaction rate constant which includes transport phenomena, but it cannot be said that the surface reaction rate is **derived** from the diffusivity as the former specifically excludes contributions from the latter.

17 Claim 29 is rejected under 35 U.S.C. 112, second paragraph, as failing to set forth the subject matter which applicant(s) regard as their invention.

The preamble of the claim recites two disimilar processes rendering the scope interminable. If the applicant intends to present separate new claims at this point in the prosecution directed purely to reactor design, these will be restricted and withdrawn from consideration.

Claim 19-23 and 29 are rejected under 35 U.S.C. § 112, first paragraph, as the disclosure is enabling only for claims limited to plasma etching processes. See M.P.E.P. §§ 706.03(n) and 706.03(z).

19 The following is a quotation of 35 U.S.C. § 103 which forms the basis for all obviousness rejections set forth in this Office action:

A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a

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whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Subject matter developed by another person, which qualifies as prior art only under subsection (f) or (g) of section 102 of this title, shall not preclude patentability under this section where the subject matter and the claimed invention were, at the time the invention was made, owned by the same person or subject to an obligation of assignment to the same person.

20 Claims 1-29 are rejected under 35 U.S.C. § 103 as obvious over Thompson

"Introduction to Microlithography".

Thompson "Introduction to Microlithography" teaches the relationship between the loading effect and the etch rate, including the etching rate constant. (page 234) The use of "large volume reactors" to reduce loading effects is disclosed. (page 234) The cases of diffusion limited processes is disclosed. (page 234) The reduction of localized depletion (non-uniformity of the etch) of the etchant may be pressure and flow rates. (page 235) Several examples of etching chambers appear on page 230.

The examiner holds that it would have been obvious to vary known etch parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization in known etching processes including ashing processes which are commonly used to remove resists. Please note that one skilled in the art would also be expected to either know or be able to find references teaching in general what conditions are suitable for etching the desired material.

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In response to the arguments raised by the applicant, the equations shown on page 234 of the reference would allow one skilled in the art to determine if the etching conditions were affected by the loading effect, thereby presenting useful information for diagnostic purposes. Note that without accounting for the transport phenomena, no optimization of the uniformity of the etch process can be performed. The rejection is maintained. 21 Claims 1-29 are rejected under 35 U.S.C. § 103 as obvious over Mogab '665 and Thompson "Introduction to Microlithography".

Mogab '665 teaches that loading is known in the art to result in inter or intra wafer non-uniformity. Reactor design is known to increase etch uniformity. (2/15-22) The routine optimization to decrease the effects of loading on the system is disclosed. (2/63+) These changes are disclosed as reducing non-uniformity of the etch. (3/11-17) The use of a radial flow apparatus to determine experimentally the optimum conditions for an etch process is disclosed (5/8-14).

It would have been obvious to one skilled in the art to preform the calculations disclosed by Thompson "Introduction to Microlithography" in the process disclosed by Mogab '665 for the different etching apparatus/conditions to preform diagnistics which describe the physical processes occuring within the plasma etching apparatus thereby allowing for optimization of the parameters for a particular etch process with a reasonable expectation of success.

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Serial Number: 08/433623 Art Unit: 1113

The response provided above is relied upon here without further comment. The rejection is maintained.

22 Claims 1-29 are rejected under 35 U.S.C. § 103 as being unpatentable over either Stefani et al. '229, Mundt et al. '162, Horiike '706 or Bobbio '520, in view of Babanov et al. Plasma Chemistry and Plasma Processing and Thompson "Introduction to Microlithography".

Stefani et al. '229 teaches the use of ellisometry to determine the localized etch rates over the wafer surface and/or over a plurality of wafer surfaces in the same etch chamber. (2/34-38) am model is developed to describe the relative non-uniformity of the etch. (4/49-56) This data allows the operator to change parameters and evaluate the effects of doing so. (6/39-60)

Mundt et al. '162 teaches the relative etch rates as a function of position for both flat and convex electrodes in figures 2 and 3.

Horiike '706 teaches that the relative etch rates as a function of distance disclosed are equivalent to the combined etch rate constant and that the observations and varying of the position of the etch target results in a new etching apparatus (see figures 9 and 10).

Bobbio '520 describes a prior art apparatus in which the barriers are placed parallel to each other the inherent etch rates as a function of position are determined in figures 4a and 4b. These are corrected to produce an etch which is relatively uniform over the entire wafer. Typically no more than three positioning steps are required to achieve uniformity deviations of less than 2% (col 8/lines 43-50, hereinafter 8/43-50).

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It would have been obvious to one skilled in the art to determine the parameters _ including the etch rate constants in a manner similar to that disclosed by Babanov et al. Plasma Chemistry and Plasma Processing and Thompson "Introduction to Microlithography with each of the etch processes taught by either Stefani et al. '229, Mundt et al. '162, Horiike '706 or Bobbio '520 to further optimize known parameters including reactive gas composition, temperature, pressure, reactor configuration, rf power, flow rate and the like as is known in the art to produce a more uniform etch in the course of routine optimization in known etching processes based upon the teachings of Thompson "Introduction to Microlithography" to do so to improve the uniformity of the etch process. This includes ashing processes which are commonly used to remove resists.

The examiner notes that all of the primary reference teach increasing unifomity and therefore would be held to be optimizing the "uniformity value" irrespective of whether this is actually calculated or not. The response provided above is relied upon here as well without further comment.

23 Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, THIS ACTION IS MADE FINAL. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for response to this final action is set to expire THREE MONTHS from the date of this action. In the event a first response is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until

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Art Unit: 1113

after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event will the statutory period for response expire later than SIX MONTHS from the date of this final action. New $C_{la, ne}$

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Martin Angebranndt whose telephone number is (703) 308-4397.

I am normally available between 7:30 AM and 5:00 PM, Monday through Thursday and 7:30 AM and 4:00 PM on alternate Fridays.

If repeated attempts to reach me are unsuccessful, my supervisor may be reached at (703) 308-2303.

Facsimile correspondence should be directed to (703) 305-3599.

Any inquiry of a general nature or relating to the status of this application should be directed to the Group receptionist whose telephone number is (703) 308-0661.

into

Martin J. Angebranndt Primary Examiner, Group 1100 January 21, 1997

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Amendment Transmittal

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, CA 94111-3834 (415) 326-2400

In re application of Daniel L et al. Serial No: 08/433,623 Filed: May 3, 1995

PROCESS OPTIMIZATION IN GAS For: PHASE DRY ETCHING

BOX FEE AMENDMENT ASSISTANT COMMISSIONER FOR PATENTS Washington, D.C. 20231

Sir:

Transmitted herewith are the following documents in response to the Final Office Action dated January 24399011P 1100

Amendment Under 37 CFR §1.116; 1)

2) Petition for Extension of Time (1-mo); and

3) Postcard.

Group Art Unit: 1113

The filing fee has been calculated as shown below:

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INDEP.	*5	MINUS	***6	=0		x39=	\$		x78=	\$
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* If the entry in Col. 1 is less than the entry in Col. 2,

write "0" in Col. 3. **

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If the "Highest Number Previously Paid For" IN THIS SPACE is less than 20, write "20" in this space. If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, write "3" in this space. The "Highest Number Previously Paid For" (Total or Independent) is the highest number found from the equivalent box in Col. 1 of a prior amendment or the number of claims originally filed.

Please charge Deposit Account No. 20-1430 as follows:

Petition to Extend Time (1-mo) [X]

[X] Any additional fees associated with this paper or during the pendency of this application.

_ extra copies of this sheet are enclosed.

TOWNSEND and TOWNSEND and CREW LLP

\$55.00

Richard T. Ogawa Reg. No.: 37,692 Attorneys for Applicant

Ex.1002 p.126

Atty. Docket No. 16655-000100

Date May 13, 1997

I hereby certify that this is being deposited with the United States Postal Service as first class mail in an envelope addressed to:

Assistant Commissioner for Patents Washington, D. C. 20231.

Date Kristina Alvarez

I hereby certify that this correspondence is being deposited with the United States Postal Service first class mail in an envelope addressed to Assistant Commissioner for Patents, CREW II TOWN IN THE UNITED STATES PATENT AND TRADEMARK OFFICE In re application of: Daniel L. Flamm, et al. Examiner: M. Angebranndt Application No.: 08/433,623 Art Unit: 1113

10.1 Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

BOX FEE AMENDMENT Assistant Commissioner for Patents Washington, D.C. 20231

Sir:

Applicants petition the Assistant Commissioner of Patents to extend the time for

response to the Office Action, dated January 24, 1997 for one month, from April 24, 1997 to May 24, 1997. An appropriate response to the Office Action in the form of an Amendment Under 37 CFR §1.116 is enclosed herewith.

Please charge \$55.00, pursuant to 37 CFR §1.17, to the Deposit Account No. 20-1430. Please charge any additional fees or credit overpayment to the above Deposit Account. This Petition is submitted in triplicate.

Respectfully submitted,

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Attorney Docket No. 16655-000100

PETITION TO EXTEND TIME UNDER

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JUN 4 1997

GROUP 1100

37 CFR §1.136(a)

Dated:

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (415) 326-2400 RTO:ka rto\work\16655\1-ext.tm2

Richard T. Ogawa Reg. No. 37,692

I hereby certify that this correspondence is being deposited with the United States Postal Service as first class mail in an envelope addressed to: Assistant Commissioner for Patents, PATENT Washington, D.C. 20231 on Attorney Docket No. 16655-000100 3 197 CREW LLF TOWNSEND IN THE UNITED STATES PATENT AND TRADEMARK OFFICE In re application of: Daniel L. Flamm Examiner: M. Angebranndt Application No.: 08/433,623 Art Unit: 1113 Filed: May 3, 1995 AMENDMENT UNDER 37 CFR §1.116 For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING RECEIVED Assistant Commissioner for Patents 4 1997 Washington, D.C. 20231 GROUP 1100 Dear Sirs: In response to the Office Action mailed January 24, 1997, please amend the above-cited application as follows. IN THE CLAIMS: Please amend claims 1-4, 9-12, 19, 20, 25, 26, and 29 as follows. For the convenience of the Examiner, all claims subject to examination are shown, even if not being amended. 1. (Amended) A device fabrication method comprising the steps of: providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface; etching said top film surface to define a relatively non-uniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate

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includes a radius and an angle.

which defines a position within [from] said relatively non-uniform etching profile on said substrate, said etching comprising a reaction between a gas phase etchant and said film; and

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extracting a surface reaction rate constant from said etch rate data, and using said surface reaction rate constant in the fabrication of a device.

(Amended) The method of claim 1 wherein said etching step is

(Amended) The method of claim 1 wherein said spatial coordinate[s]

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diffusion <u>limited</u> [limiting].

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4. (Amended) The method of claim 1 wherein said spatial coordinate[s]

includes an x-direction and a y-direction.

5. The method of claim 1 wherein said extracting step correlates said surface reaction rate constant over a diffusivity with said etch rate, said etch rate being defined by said relatively non-uniform etching profile.

6. The method of claim 1 wherein said etch rate is defined by said relatively non-uniform etching profile at selected spatial coordinates over a time.

7. The method of claim 1 wherein said etching is an ashing method.

8. The method of claim 7 wherein said ashing method comprises reactants including an oxygen and a photoresist.

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(Amended) A method of designing a reactor comprising the steps of: providing a first plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

etching said top film surface to define a relatively non-uniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate which defines a position within [from] said relatively non-uniform etching profile on said film of said substrate, said etching comprising a reaction between a gas phase etchant and said film; and

extracting a surface reaction rate constant from said etch rate data, and using said surface reaction rate constant in designing a second plasma etching apparatus.

postion on said relatively non-uniform etching profile is diffusion limited [limiting].

 \mathcal{O} (Amended) The method of claim \mathcal{S} wherein said spatial coordinate[s] which defines said position along said relatively non-uniform etching profile includes a radius and an angle.

12. (Amended) The method of claim \mathscr{G} wherein said spatial coordinate[s] which defines said position within said relatively non-uniform etching profile includes an x-direction and a y-direction.

13. The method of claim 9 wherein said extracting step correlates said surface reaction rate constant over a diffusivity with said etch rate, said etch rate being defined by said relatively non-uniform etching profile.

14. The method of claim 9 wherein said etch rate is defined by said relatively non-uniform etching profile at selected spatial coordinates over a time.

15. The method of claim 9 wherein said etching is an ashing method.

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16. The method of claim 15 wherein said ashing method comprises reactants including an oxygen and a photoresist.

17. The method of claim 9 wherein said second plasma etching apparatus is a co-axial reactor.

 The method of claim 9 wherein said second plasma etching apparatus is a plasma etching apparatus.

 \mathcal{H} (Amended) A substrate fabrication <u>method</u>, using a plasma etching apparatus, said method comprising:

providing a substrate selected from a group consisting of a semiconductor wafer, a plate, and a flat panel display, said substrate comprising a top surface;

forming a film overlying said top surface, said film comprising a top film surface;

etching said top film surface to define a relatively non-uniform profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate <u>which</u> <u>defines a position within</u> [from] said <u>relatively non-uniform</u> etching profile <u>of said film on</u> <u>said substrate</u>, said etching comprising a reaction between a gas phase etchant and said film; and

extracting a surface reaction rate constant from said etch rate data, and using said surface reaction rate constant [in adjusting said method].

20. (Amended) A method of fabricating an integrated circuit device, using a plasma etching apparatus, said method comprising:

providing a uniformity value and a surface reaction rate constant for an etching reaction, said etching reaction including a substrate and etchant species;

defining etching parameters providing said uniformity value; and

adjusting at least one of said etching parameters using said surface reaction rate constant to produce a selected etching rate;

wherein said etching rate providing an etching condition for fabrication of an integrated circuit device.

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21. The method of claim 20 wherein said etching parameters can be selected from a group consisting of a temperature, a pressure, a power, a gap, and a flow rate.

The method of claim 20 wherein said uniformity value ranges from
 90% and greater.

23. The method of claim 20 wherein said uniformity value ranges from 95% and greater.

24. The method of claim 1 further comprising a step of using said reaction rate constant in adjusting said plasma etch apparatus.

Amended) A process for fabricating a device <u>using a plasma etching</u> <u>apparatus</u>, said device being fabricated by use of a surface reaction rate constant, said surface reaction rate constant being derived from a method comprising:

providing a plasma etching apparatus comprising a substrate therein, said substrate comprising a top surface and a film overlying said top surface, said film comprising a top film surface;

etching said top surface at a temperature to define a relatively non-uniform etching profile on said film, and defining etch rate data comprising an etch rate and a spatial coordinate <u>which defined a position</u> from said <u>relatively non-uniform</u> etching profile <u>on said</u> <u>film of said substrate</u>, said etching comprising a reaction between a gas phase etchant and said film; and

extracting from said etching rate data a surface reaction rate constant for said temperature.

26. (Amended) The process of claim 25 wherein said surface reaction rate constant is derived [from] using at least a diffusivity value that is determined by an equation.

27. The process of claim 25 wherein said etching is provided whereupon chemical effects are enhanced over ion bombardment effects.

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28. The method of claims 1, 9, or 19 wherein said etching is provided whereupon chemical effects are enhanced over ion bombardment effects.

29. (Amended) <u>The method of claim 25 further comprising</u> [A method of fabricating a device or designing a reactor] using [a] <u>said</u> surface reaction rate constant <u>in</u> a method selected from a group consisting of a fabrication of a device or of designing a reactor, said surface reaction rate constant [is] <u>being</u> provided by at least a diffusivity value.

<u>REMARKS</u>

Applicant would like to thank Examiner Angebranndt for his time and cooperation in interviewing this subject patent application. Reconsideration of these claims, as amended, is respectfully requested. Claims 1-29 are now pending in this application.

35 U.S.C. §112

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Claim 29 was rejected under 35 U.S.C. §112, second paragraph as failing to set forth the subject matter which applicants regard as their invention. As shown, claim 29 has been amended to be dependent upon independent claim 25. Accordingly, the rejection should now be moot.

Claims 19-23 and 29 were rejected under 35 U.S.C. §112, first paragraph, since the disclosure is indicated as enabling only for claims limited to plasma etching processes. As shown above and explained to the Examiner, claims 19-23 and 29 generally use, for example, directly or indirectly, an apparatus for plasma etching and have been amended, as shown above. Accordingly, claims 19-23 and 29 are patentable under 35 U.S.C. §112, first paragraph.

35 U.S.C. §103

The Examiner has rejected claims 1-29 under 35 U.S.C. §103(b) as being obvious over Thompson "Introduction to Microlithography". In particular, the Examiner indicated that Thompson taught a relationship between the loading effect and the etch rate, including the etching rate constant. Thompson generally discloses an equation for loading effect. Thompson, however, fails to show or suggest the invention of claims 1-29, as

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amendment, which was suggested by the Examiner. These claims should be patentable over Thompson for the reasons discussed herein using claim 1 as merely an example.

In particular, claim 1 recites a device fabrication method which includes novel aspects of extracting a chemical reaction rate constant from an etching profile. The method includes providing a plasma etching apparatus having a substrate therein. The substrate has a top surface and a film overlying the top surface. The top film surface is etched to define a relatively non-uniform etching profile on the film, which defines etch rate data comprising an etch rate and a spatial coordinate. The spatial coordinate defines a position within the relatively non-uniform etching profile on the substrate. A step of extracting a surface reaction rate constant from the etch rate data is included. The surface reaction rate constant is used in the fabrication of a device. Accordingly, claim 1 provides a sequence of steps for extracting a chemical reaction rate constant from an etching profile in a relatively easy manner. At best, Thompson suggested a loading effect equation, which cannot derive the claimed surface reaction rate constant from an etching profile. Additionally, the loading effect equation includes a term "G" defined as "the generation rate of the active species" which is not available. Furthermore, the term kloss is generally not know, and is likely to be difficult to obtain. Moreover, the etching rate constant Ketch is also not know. Accordingly, the loading effect equation has at least three variables including k_{etch}, k_{loss}, and G, which are not known and cannot easily be solved for.

Thompson also suggests away from the invention of claim 1, as amended. Thompson taught a loading effect equation that <u>requires</u> a uniform reactant species density, i.e., a uniform etching profile. The uniform etching profile is required or is an assumed condition for applying the loading effect equation. Now, Thompson does not mention this assumption with the loading effect equation, but accordingly to Applicant Dr. Daniel Flamm, the loading effect equation uses the underlying assumption that the reactant species overlying the film to be etched is uniform, which leads to a constant etching rate over all portions of the film to be etched. This constant etching rate provides a <u>uniform</u> etching profile. In contrast, the present invention takes advantage of a relatively <u>non-uniform etching profile</u> to extract a surface reaction rate constant as recited by claim 1, for example. Accordingly, Thompson clearly suggests away from these novel aspects of the present invention, which are recited by claim 1.

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Claims 1-29 were also rejected under 35 U.S.C. §103 as obvious over Mogab and Thompson, "Introduction to Microlithography". Mogab may have taught certain aspects of the loading effect but still fails to show or suggest extracting a chemical reaction rate constant from an etching profile. Accordingly, claims 1-29 are patentable over Mogab and Thompson.

Claims 1-25 were rejected under 35 U.S.C. §103 as being unpatentable over either Stefani et al., Mundt et al., Horiike or Bobbio, in view of Babanov et al., Plasma Chemistry and Plasma Processing and Thompson, "Introduction to Microlithography". As noted above, it is clear that claim 1 is patentable and non-obvious over Thompson. Accordingly, claims 1-29 are at least patentable under 35 U.S.C. §103 for the reasons noted above. Additionally, any combination of Stefani et al., Mundt et al., Horiike or Bobbio, and even Babanov et al., further fail to show or suggest the claimed combination including extracting a chemical reaction rate constant from an etching profile. In particular, Stefani et al. appeared to suggest a system for monitoring and evaluating uniformity of a semiconductor wafer, which has nothing to do with extracting a chemical reaction rate constant from an etching profile within a wafer surface. Mundt et al. seems to suggest using curved electrodes for the manufacturing of semiconductor devices. Thus, Mundt et al. does not show or suggest extracting a chemical reaction rate constant from an etching profile. Horiike generally taught a reaction vessel for plasma etching, but fails to show or suggest the claimed combination including extracting a chemical reaction rate constant from an etching profile within a wafer surface. Bobbio seems to suggest a magnetron plasma processing apparatus for producing "a uniform processing rate over an entire substrate surface." Bobbio, Col. 2, lines 54-58. In contrast, the present invention takes advantage of a non-uniform processing rate or etching profile to extract a chemical reaction rate constant therefrom. As for Babanov et al., it merely describes some etching mechanisms, which do not have a thing to do with the present invention, which includes extracting a chemical reaction rate constant from an etching profile. Accordingly, claims 1-29 are clearly non-obvious and patentable over these references under 35 U.S.C. §103.

Numerous advantages are achieved by way of the invention of claim 1, for example. In the semiconductor industry, the surface reaction rate constant can be used to design a reactor such as a plasma etching reactor or even a chemical vapor deposition reactor. Additionally, the invention of claim 1 can be used for designing other types of

reactors which are generally know. Furthermore, the invention of claim 1 will generally provide more uniform films from the selection of plasma etching parameters or operating conditions without undue experimentation or "trial and error," which is often required using pre-existing or conventional techniques, i.e., Stefani <u>et al.</u>, Mundt <u>et al.</u>, Horiike, Bobbio, Babanov <u>et al.</u>, Thompson, and others. Applicant can provide further details and examples of these advantages, and others, by way of a declaration by Dr. Daniel L. Flamm, as the request of the Examiner.

CONCLUSION

Therefore, in view of the remarks above, Applicant respectfully requests that the rejections be removed, that claims 1-29 be allowed, and the case passed to issue. The issuance of a formal Notice of Allowance at an early date is respectfully requested.

If the Examiner believes a telephone conference would expedite prosecution of this application, please telephone the undersigned at (415) 326-2400.

Respectfully submitted,

Date:

Richard

Reg. No. 37,692

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (415) 326-2400 Fax (415) 326-2422

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MARTIN ANGEBRANNDT PRIMARY EXAMINER GROUP 1100

PTOL-37 (REV. 4-89) *

USCOMM-DC 89-3789

Art Unit: 1113

3 The following is an Examiner's Statement of Reasons for Allowance: The arguments forwarded by the applicant have been found persuasive when combined with the amendments to the claims tying the data concerning the non-uniformity of the etch directly to an individual substrate rather than merely a location within the reactor which was more addressable by the loading description of Thompson.

-2-

Any comments considered necessary by applicant must be submitted no later than the payment of the Issue Fee and, to avoid processing delays, should preferably **accompany** the Issue Fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allowance."

4 Any inquiry concerning this communication or earlier communications from the examiner should be directed to Martin Angebranndt whose telephone number is (703) 308-4397.

I am normally available between 7:30 AM and 5:00 PM, Monday through Thursday and 7:30 AM and 4:00 PM on alternate Fridays.

If repeated attempts to reach me are unsuccessful, my supervisor may be reached at (703) 308-2303.

Facsimile correspondence should be directed to (703) 305-3599.

Any inquiry of a general nature or relating to the status of this application should be directed to the Group receptionist whose telephone number is (703) 308-0661.

Martin J. Angebranndt Primary Examiner, Group 1100 June 9, 1997 The drawings submitted with this application were declared informal by the applicant. Accordingly they have not been reviewed by a draftsperson at this time. When formal drawings are submitted, the draftsperson will perform a review.

S.

Direct any inquires concerning drawing review to the Drawing Review Branch (703) 305-8404.

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The drawings submitted with this application were declared informal by the applicant. Accordingly they have not been reviewed by a draftsperson at this time. When formal drawings are submitted, the draftsperson will perform a review.

Direct any inquires concerning drawing review to the Drawing Review Branch (703) 305-8404.

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SUBSTITUTE PTO-948



UNITED STATES DEPARTMENT OF COMMERCE Patent and Trademark Office

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Address: Box ISSUE FEE ASSISTANT COMMISSIONER FOR PATENTS WASHINGTON, D.C. 20231

NOTICE OF ALLOWANCE AND ISSUE FEE DUE

11M1/0610 RICHARD T OGAWA TOWNSEND AND TOWNSEND KHOURIE AND CREW STEUART STREET TOWER ONE MARKET PLAZA 20TH FLOOR SAN FRANCISCO CA 94105

APPLICATION NO.	FILING DATE	TOTAL CLAIMS	EXAMINER AND GROUP AR	TUNIT	DATE MAILED
08/433,623	05/03/95	029	ANGEBRANNDT, M	1113	06/10/97
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TITLE OF PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

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-[ATTY	S DOCKET NO.	CLASS-SUBCLASS	BATCH NO.	APPLN. TYPE	SMALL ENTITY	FEE DUE	DATE DUE
ſ	1	16655-000	0100 156-643	.100 1	N95 UTIL	ITY YES	\$645.00	09/10/97
T								

THE APPLICATION IDENTIFIED ABOVE HAS BEEN EXAMINED AND IS ALLOWED FOR ISSUANCE AS A PATENT. PROSECUTION ON THE MERITS IS CLOSED.

THE ISSUE FEE MUST BE PAID WITHIN <u>THREE MONTHS</u> FROM THE MAILING DATE OF THIS NOTICE OR THIS APPLICATION SHALL BE REGARDED AS ABANDONED. <u>THIS STATUTORY PERIOD CANNOT BE EXTENDED.</u>

HOW TO RESPOND TO THIS NOTICE:

I. Review the SMALL ENTITY status shown above. If the SMALL ENTITY is shown as yes, verify your current SMALL ENTITY status:	If the SMALL ENTITY is shown as NO:
A. If the status is changed, pay twice the amount of the FEE DUE shown and notify the Patent and Trademark Office of the change in status, or	A. Pay FEE DUE shown above, or
B. If the status is the same, pay the FEE DUE shown above.	B. File verified statementof Small Entity Status before, or with, payment of 1/2 the FEE DUE shown above.

II. Part B of this notice should be completed and returned to the Patent and Trademark Office (PTO) with your ISSUE FEE. Even if the ISSUE FEE has already been paid by charge to deposit account, Part B should be completed and returned. If you are charging the ISSUE FEE to your deposit account, section "6b" of Part B should be completed.

III. All communications regarding this application must give application number and batch number. Please direct all communication prior to issuance to Box ISSUE FEE unless advised to the contrary.

IMPORTANT REMINDER: Patents issuing on applications filed on or after Dec. 12, 1980 may require payment of maintenance fees. It is patentee's responsibility to ensure timely payment of maintenance fees when due.

	3. PATENT AND TRADEMARK OFFICE COPY	period and service and	and have a trick away
PTOL-85 (REV. 05-96)(0651-0033)		*U.S. GP	O: 1996-404-496/40511
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TRANSMITTAL LETTER

TOWNSEND and TOWNSEND and CREW LLP
RECEIVED
San Francisco, CA 94105(650) 326-2400Publishing Division
SEP 0 9 1997

In re application of: Daniel L. Flamm, et al. Serial No.: 08/433,623 Filed: May 3, 1995 Group Art Unit: 1113 For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

BOX ISSUE FEE ASSISTANT COMMISSIONER FOR PATENTS Washington, D.C. 20231

Sir:

Transmitted herewith are the documents in response to the Notice of Allowance and Issue Fee Transmittal dated June 10, 1997:

- 1) Transmittal Letter (in trip.);
- 2) Issue Fee Transmittal (Part B);
- 3) Letter to Official Draftsman;
- 4) Formal Drawings (8 sheets); and
- 5) Postcard.
 - [X] A fee in the amount of \$645.00 is due.

Please charge Deposit Account No. 20-1430 as follows:

[X] Issue Fee

\$<u>645.00</u>

[X] Any additional fees associated with this paper or during the pendency of this application.

2 copies of this sheet are enclosed.

TOWNSEND and TOWNSEND and CREW LLP

Richard T. Ogawa

Reg. No.: 37,692

16655\1-issfee.trn

Atty. Docket No. <u>16655-000100</u> Date <u>September 9, 1997</u>

I hereby certify that this is being deposited with the United States Postal Service as first class mail in an envelope addressed to: Box Issue Fee, Assistant Commissioner for Patents, Washington, D. C. 20231.

September 1, 1977

I hereby certify that t \Rightarrow correspondence is being deposited with the United States Postal Service as first class mail in an envelope addressed to:

Box Issue Fee Assistant Commissioner of Patents and Trademarks Washington, D.C. 20231, on <u>Justin Just</u>

PATENT

Attorney Docket No. 16655-000100

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

 $\left(\right)$

Daniel L. Flamm, et al.

Serial No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS) PHASE DRY ETCHING) Examiner: M. Angebranndt

Art Unit: 1113

RECEIVED Publishing Division

Batch No.: N95

SEP 09 1997

LETTER TO OFFICIAL 09

Assistant Commissioner of Patents Washington, D.C. 20231

Sir:

Pursuant to the Notice of Allowability dated June 10, 1997, applicants submit

eight sheets of formal drawings to be made of record in the above-identified case.

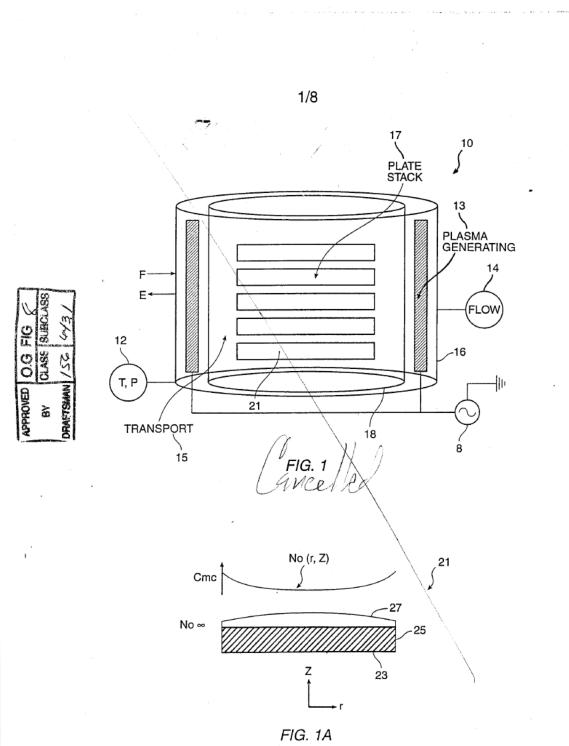
Respectfully submitted,

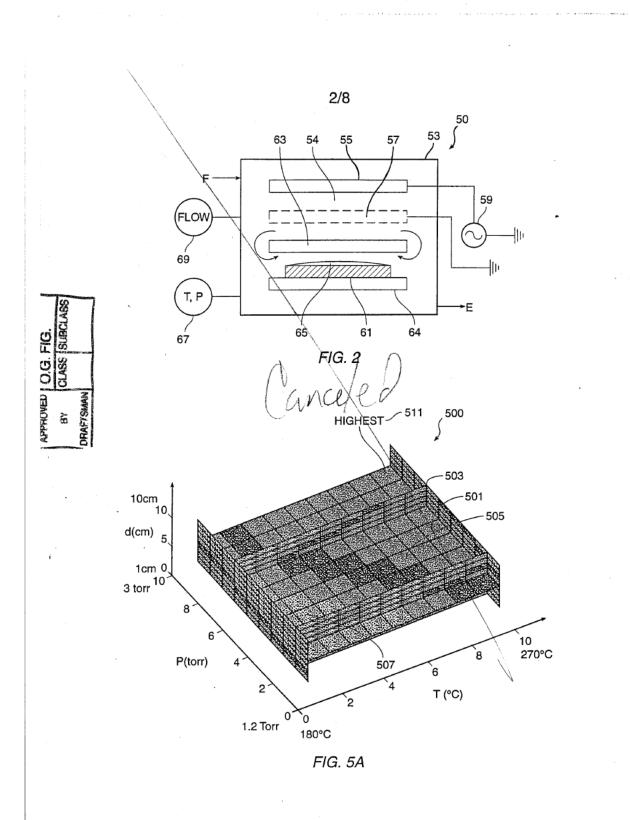
Richard T Ogawa Reg. No. 37,692

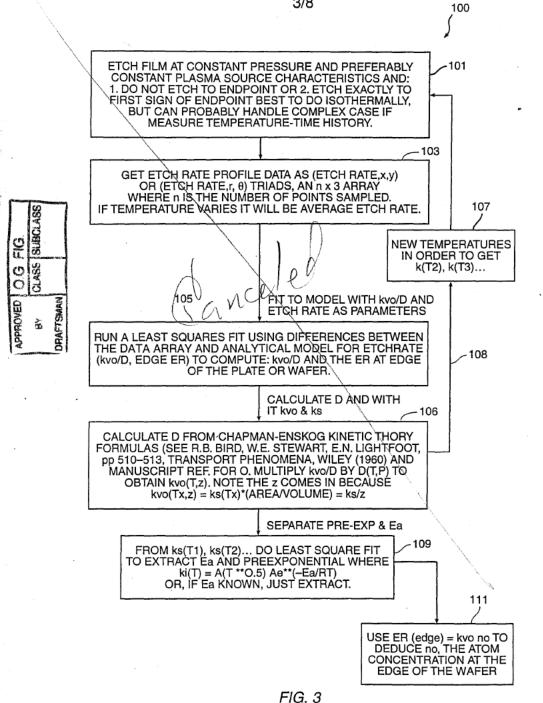
TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834

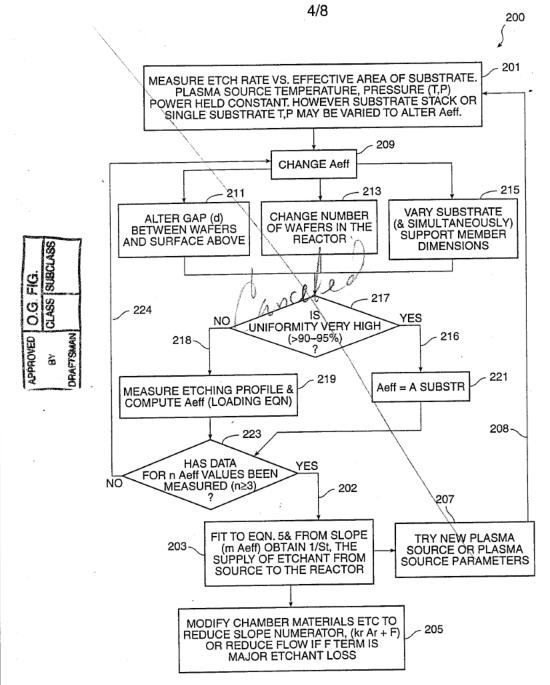
(650) 326-2400

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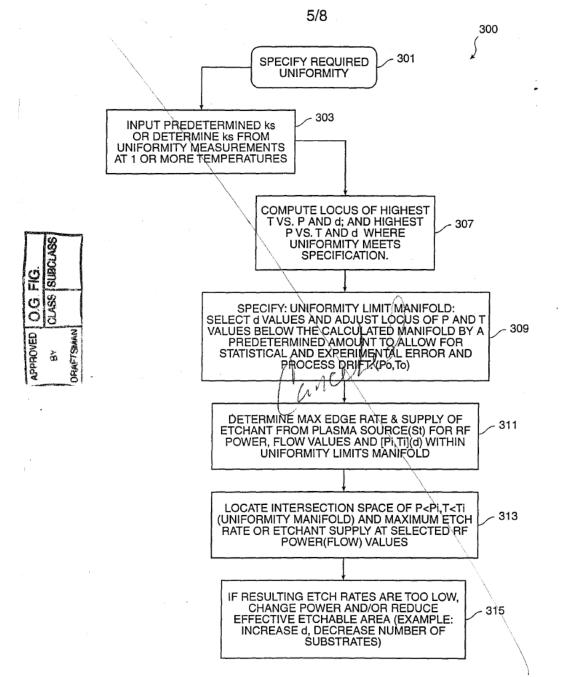
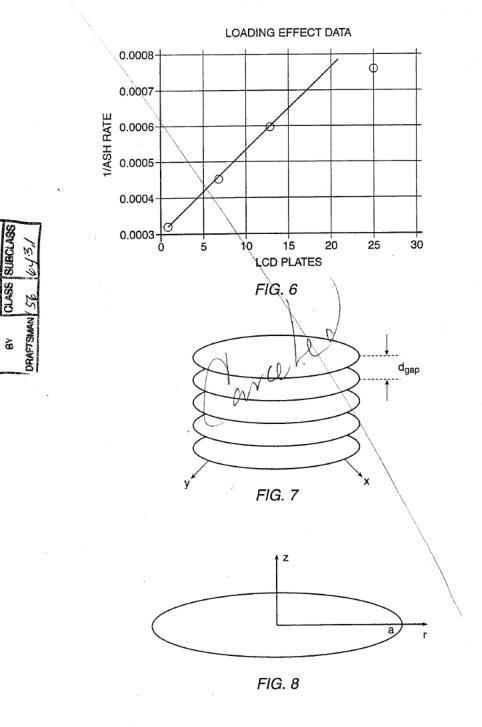
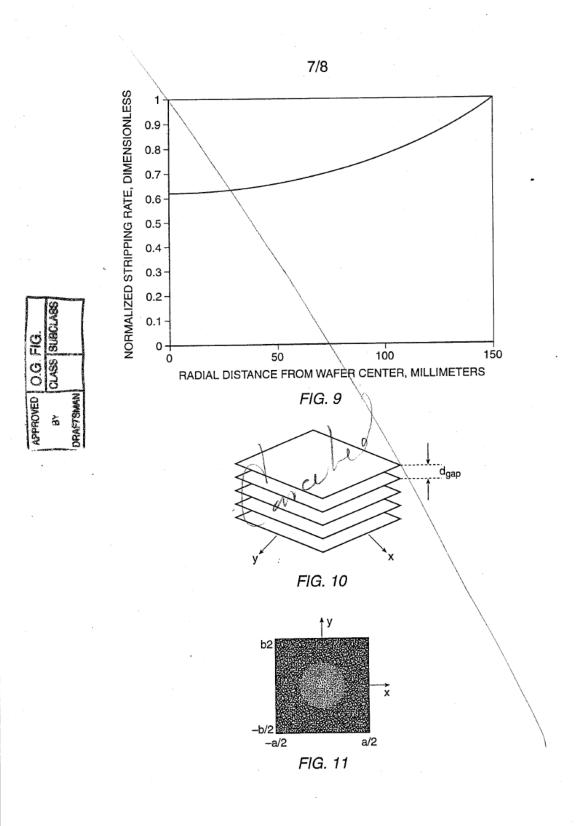


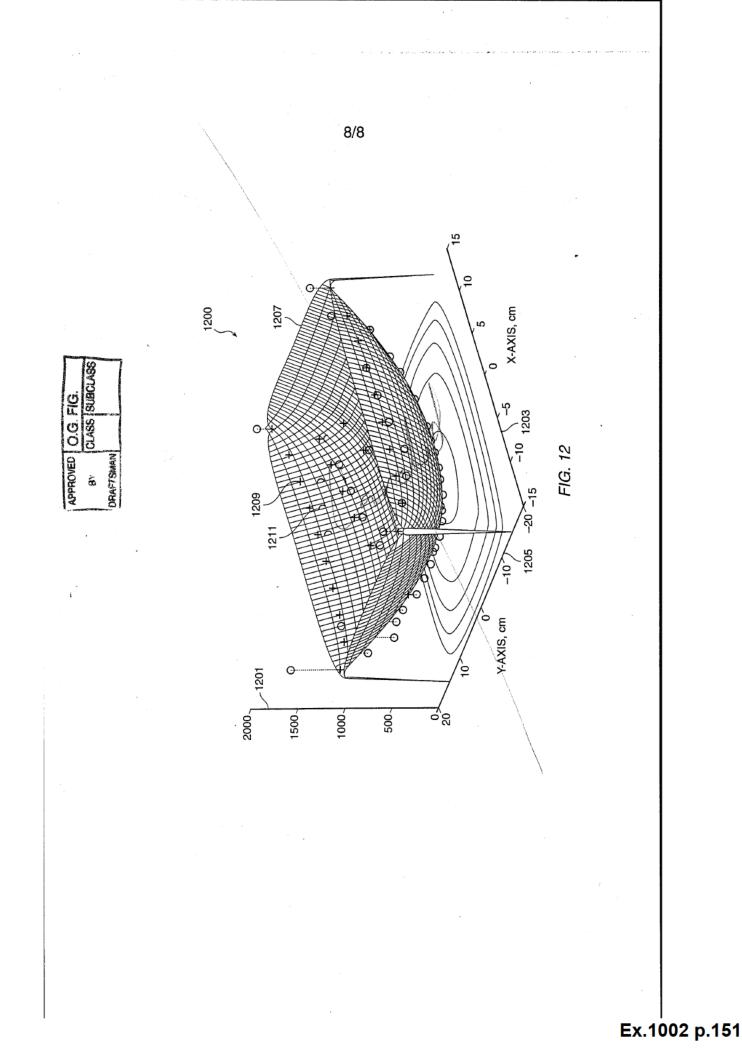
FIG. 5



0.6. FIC

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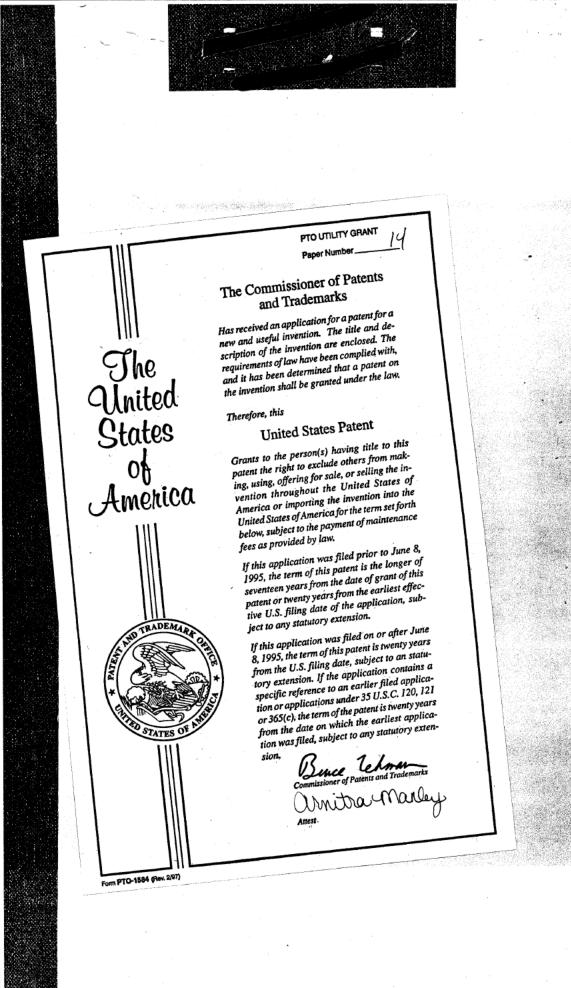




PART B-ISSUE FEE TRANSMITT/

MAILING INSTRUCTIONS: This form should be used for transmitting the ISSUE FEE. Blocks 2 through 6 sholud be completed where appropriate. All further correspondence including the issue Fee Receipt, the Patent, advance orders and notification of maintenance fees will be mailed to addresses entered in Block 1 unless you direct otherwise, by: (a) spectryinga new correspondence address in Block 3 below; or (b) providing the PTO with a separate "FEE ADDRESS" for maintenance fee notifications with the payment of issue Fee or thereafter. See reverse for Certificate of Mailing, below.

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	AND TOWNSEND	KHOURIE AN	RECEIVE	Street Address		•				
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APPLICATION NO.	FILING DATE	TOTALCLAIMS	00	EXAMINER AND GROUP	ART UNIT	DATE MAILED				
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San Francisco, CA	94111-3834			or agent. If no name is listed will be printed.	3					
ASSIGNMENT DATA TO BE PRINTED	ON THE PATENT (print or type)									
(1) NAME OF ASSIGNEE: Danie	1 L. Flamm	,		6a. The following fees are e		_				
(2) ADDRESS: (CITY & STATE OR COL	INTRY)			6b. The following fees shou	Advance Order - # of Copies id be charged to:	·				
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Assignment previously submitted to	the Patent and Trademark Office			Any Deficiencies in		0 10 100				
Assignment is being submitted under directed to Box ASSIGNMENTS.				The COMMISSIONER OF I requisited to apply the isso	Fee to the application identi	fied acon () ()				
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Amendment Transmittal TOWNSEND and TOWNSEND and (**WLLP** Atty. Docket No. 55-000100 Two Embarcadero Center, 8th Floor San Francisco, CA 94111-3834 226-2400 (650)Express Mail Label No .: EM140585555US ١P I hereby certify that this is being deposited with the United In re applicati States Postal Service "Express Mail Post Office to Addressee, of Daniel L. Flamm, et al. service under 37 CFR 1.10 on the date indicated below and is 06Jer217N199708 3 623 addressed to: Group 1100 Director - Theodore Morris Assistant Commissioner for Patents Filed: Mayo 1995 Washington, D. C. 20231 Date: 27 Group Art Unit: 1113 PROCESS OPTIMIZATION IN GAS For: PHASE DRY ETCHING and I also hereby certify that this correspondence is being sent by facsimile transmission to: Group Director - Theodore Morris Fax No: 1-703-305-359 Assistant Commissioner for Patents, Washington, D.C. 20231, on 27/97 Elyine Bv ASSISTANT COMMISSIONER FOR PATENTS Washington, D.C. 20231 Sir: Transmitted herewith is an amendment in the above-identified application. [X] Amendment After Payment of Issue Fee Under 37 CFR 1.312(b). [X] Petition for Amendment After Payment of Issue Fee Under 37 CFR 1.312(b). [X] Postcard (in Express Mail copy). The filing fee has been calculated as shown below: OTHER THAN A (Col 1) (Col. 2) (Col. 3) SMALL ENTITY SMALL ENTITY

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write "0" in Col. 3.

If the "Highest Number Previously Paid For" IN THIS SPACE is less than 20, write "20" in this space.

If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, write "3" in this space. The "Highest Number Previously Paid For" (Total or Independent) is the highest number found from the equivalent box in Col. 1 of a prior amendment or the number of claims originally filed.

[X] No fee is due.

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_ extra copies of this sheet are enclosed.

TOWNSEND and TOWNSEND and CREW LLP

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Gary T. Ak Reg. No(:/29,038

Attorneys for Applicant

16655\1-312.tm

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1P Ó OCT 2 7 1997 TRADEMAR

I hereby certify that this correspondence is being deposited with the United States Postal Service "Express Mail Post Office to Addressee," service under 37 C.F.R. 1.10 on the date indicated below and is addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231,

on 197 2 Express Mail EM140585555US TOWNSEND and TOWNSEND and CREW ¢ J. hierellu

and I also hereby certify that this correspondence is being sent by facsimile transmission to: Group Director - Theodore Morris Fax No: 1-703-305-3599 Assistant Commissioner for Patents, Washington, D.C. 20231, on <u>10,27,97</u>

PATENT

Attorney Docket No. 16655-000100

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:

Daniel L. Flamm

Application No.: 08/433,623

Filed: May 3, 1995

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING Examiner: M. Angebranndt

Art Unit: 1113

PETITION FOR AMENDMENT AFTER PAYMENT OF ISSUE FEE UNDER 37 C.F.R. § 1.312(b)

Assistant Commissioner for Patents Washington, D.C. 20231

Dear Sirs:



Pursuant to 37 C.F.R. § 1.312(b), Applicant petitions for entry of the changes in the accompanying amendment. As indicated in the amendment, Applicant believes the changes are necessary for the correction of various errors. In view of the many technical changes, the amendment was not earlier presented.

Please charge the petition fee of \$130.00 to Deposit Account 20-1430. This

petition is submitted in triplicate. Please charge any other fees or credit any overpayment to

Deposit Account 20-1430.

Respectfully submitted,

Date: 10/27/97

Gary T Aka Reg. No. 29,038

TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (650) 326-2400 Fax (650) 326-2422 GTARTO RAde H:WORK/16655000100/312VI.PET PATENT

I hereby certify that this correspondence is being deposited with the United States Postal Service "Express Mail Post Office to Addressee," service order 37 C.F.R. 1.10 on the date indicated below റ PATENT d is addressed to: istant Commissioner for Patents, OCT 2 7 1997 Attorney Docket No. 16655-000100 hington, D.C. 20231 127 10 TRADEN ress Mail Label No. EM140585555US TOWNSEND and TOWNSEND and CREW LLF ON and I also hereby certify that this correspondence is being sent by facsimile transmission to: Group Director - Theodore Morris Fax No: 1-703-305-3599 Assistant Commissioner for Patents, Washington, D.C. 20231. on ノルルフリラフ D.C. 20231, on IN THE UNITED STATES PATENT AND TRADEMARK OFFICE In re application of: Daniel L. Flamm Examiner: M. Angebranndt Application No.: 08/433,623 Art Unit: 1113 Filed: May 3, 1995 PROCESS OPTIMIZATION IN GAS) AMENDMENT AFTER PAYMENT OF For: PHASE DRY ETCHING ISSUE FEE UNDER 37 C.F.R. §1.312(b) Assistant Commissioner for Patents Washington, D.C. 20231 Dear Sirs: Pursuant to 37 C.F.R. §1.312(b), please amend the above-identified application as follows: IN THE SPECIFICATION: At page 3, line 13, please delete "parameters" and insert -- parameter --. At page 3, line 28, please delete "according of" and insert -- according to --. At page 4, line 21, delete "flow source" and insert -- controller --. At page 4, line 29, please delete "limiting" and insert -- limited --. At page 7, line 25, please delete "such as" and insert --, --.

	Daniel L. Flamm Application No.: 08/433,623 Page 2	
	At page 8, line 5, please delete "way" and insertway of	
	At page 8, line 9, please delete "applications to" and insert applications of	
	At page 8, line 22, please delete "".	
	At page 9, lines 7-8, please delete "(u,x,y)" and insertu(x,y)	
	At page 9, line 8, please delete "relative etch rate" and insertrelative etch rate is-	
	At page 9, line 34, please delete "is defined as follows" and insert is often	
	defined as follows	
	At page 10, in the first equation on the page, please insert+ after the term	
	$\cos \frac{m\pi y}{2}$.	
The strength of the strength of the	At page 10, line 5, after substrate, please delete "." and insert $-$; and I_0 is a	
10	modified Bessel function of the first kind	
a a fa a gan ga a gan gan gan gan gan gan gan	At page 10, line 16, please delete "collusion" and insertcollision	
	At page 11, line 15, please delete "follow equation." and insert following	
	equation:	
	At page 11, in the equation after line 23, please delete " R_0 " and insert R_{0s} ;	
	please delete " k_{ν_0} " and insert k_s	
Andrew Parks	At page 11, line 25, please delete " R_0 " and insert R_{os}	
	At page 12, in the equation after line 18, please delete " $\frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}$ " and please	
	insert $\left[1 - \frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}\right]$	
	At page 13, in the equation after line 10 and at line 11, please delete " R_0 " and insert πR_0 "	
	insert R_{os} At page 13, line 11, please delete " R_o " and insert R_{os}	
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Daniel L. Flamm Application No.: 08/433,623 Page 3

 $\left[1 - \frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}\right] -$

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At page 14," " line 30, please delete " $\frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}$ " and please insert ---

At page 15, line 37, please insert --,-- after T.

At page 15, line 37, please delete "vs." and insert --vørsus--.

At page 15, line 38, please insert --, -- after highest P.

At page 15, line 38, please delete "vs." and insert --versus---

At page 16, line 3, please insert --,-- after highest T.

At page 16, line 3, after T, please delete "vs." and insert --versus--.

At page 16, line 3, please insert --,-- after highest P.

At page 16, line 3, after P, please delete "vs." and insert --versus--.

At page 16, line 5, please insert --, -- after highest T.

At page 16, line 5, after T, please delete "vs." and insert --versus--.

At page 16, line 5, please insert --,-- after highest P.

At page 16, line 5, after P, please delete "vs." and insert --versus--.

At page 16, line 21, " R_0 " and insert -- R_{0s} --.

At page 17, lines 22-23, please delete "surface recombination rate" and insert --

effective surface recombination rate ---.

At page 18, line 18, please delete " A_w " and insert -- A_{eff} -.

At page 19, line 9, before etching rates, please insert --when etching uniformity is

high the --.

At page 19, line 16, please delete " k_e " and insert -- k_s --.

At page 19, in the equation after line 20, please delete " k_o " and insert -- k_s --.

At page 19, in the equation after line 20, please delete " A_w " and insert -- A_{eff} --.

At page 20, in the equation after line 5, please insert -- -- before D.

At page 20, in the equation after line 6, please delete "*" and insert --(-- before $D\nabla n_a$ and insert --)-- after $D\nabla n_a$.

PATENT

At page 20, line 17, please delete " $n(x/L_x y/L_y)$ " and insert $-u(x/L_x y/L_y)$ --.

At page 20, line 19, please delete " n_0 " and insert -- n_0/n_{00} --.

At page 23, line 9, please delete "9" and insert --10--.

At page 23, line 26, please delete "where λ_x is given by".

At page 24, between the first and second equation on the page, please insert -where λ_x is given by--.

> At page 24, line 1, please delete "the general" and insert -- The general--. At page 24, in the last equation on the page, please insert --+-- after the term

 $\cos \frac{m\pi y}{b}$

At page 25, line 1, please insert -- two-dimensional-- after previous.

At page 25, line 7, please delete "from" and insert --using relations in---

At page 25, line 26, please delete "substitute" and insert -- substituting--.

At page 25, line 31, please delete "11" and insert --12--.

At page 26, line 10, please delete "drove" and insert --sustained--.

At page 26, line 17, please delete "11" and insert --12---

At page 26, line 18, please insert --to-- after fit.

At page 26, line 35, please delete "R" and insert -- R_{os} --

At page 27, line 10, please delete " $A_{w,eff}$ " and insert -- A_{eff} --.

At page 27, in the equation after line 10, please replace all occurrences of " A_{weff} "

with --Aeff--.

At page 27, in the equation after line 10, please replace " R_e " with $-R_{os}$ --. At page 27, line 11, please delete " $A_{w,eff}$ " and insert $-A_{eff}$ -. At page 27, line 19, please replace both occurrences of " $A_{w,eff}$ " with $-A_{eff}$ -. At page 28, line 6, please delete "describe" and insert --describes--.

IN THE DRAWINGS:

Applicant submits new figures 1-12, consisting of eight sheets of formal drawings. Newly submitted figures correct some inadvertent errors in the originally submitted formal drawings. Amendments to the originally submitted informal figures have been highlighted in red ink.

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In Fig. 1, chemical flow source 14 was detached from the chemical feed F. In the new formal drawings chemical feed F is shown passing through the chemical flow source 14. Also, the ground connections have been amended to more clearly show the complete circuit for the rf discharge 8. Furthermore, the arrow indicating the exhaust E is moved to the chamber . bottom.

In Fig. 1A, "Conc" on the y-axis is replaced by "Concentration" for clarity. Also n_{ab} has been aligned with the y-axis.

In Fig. 2, as discussed with respect to Fig. 1, the chemical flow source F is moved to pass through the flow controller 69.

In Fig. 3, as highlighted in red ink, some editorial changes are made in boxes 105, 106, 107, and 109. For example, in box 109 $k_i(T)$ has been replaced with $k_s(T)$. In box 106, "z" has been replaced with " d_{gap} ". Similarly in box 107, "k" has been replaced with " k_s ". In box 109, k_i has been replaced with k_s and "preexponential" has been added at the end.

In Fig. 4, some editorial changes are made in boxes 203 and 205. In box 203, "St" is replaced by " $S^{T"}$. In box 205, "of" is added after slope.

In Fig. 5, some editorial changes are made in boxes 301, 303, and 309. In box 301, "1" is replace with "one". As discussed with respect to Fig. 4, in box 309, "St" is replaced by " S^{7*} ". In box 309, ":" is replaced with a period.

In Fig. 5A, scaling numbers 0, 2, 4, 6, 8, and 10 on both axis are replaced with temperature and pressure values. Also, reference number 503 is deleted.

In Fig. 7, the two Cartesian axes at the bottom of the informal drawing are deleted.

In Fig. 8, the label "S" has been added.

In Fig. 10, the x-y-z axes are added and new labels are added to show the relative gap between d=0 and $d=d_{gap}$.

In Fig. 12, the units on the *y*-axis are changed. Also, the downward spikes at the four corners of the informal drawing are deleted.

REMARKS

This amendment does not affect the merits and no new matter is involved.

The Specification has been amended to make some editorial changes for consistency. For example, the amendment made at page 10, line 5, is editorial for consistency with I_0 's definition at page 22, line 5. Similarly, the amendments made to the equations at pages 12 and 14 are editorial to make the equations consistent with the text. See, e.g., the Specification, at page 12, lines 26-29. Those with ordinary skill in the art would understand that the confirmaty equation would include the term "[1- ...]". Accordingly, this error has been corrected in accordance with the text. Other amendments represent other minor typographical corrections to the text.

The newly submitted formal figures correct some inadvertent typographical errors in the originally submitted formal figures. Amendments to the originally filed informal figures have been highlighted in red ink.

With respect to Fig. 1, in the new formal drawings, chemical feed F is correctly shown passing through the chemical flow source 14. See, e.g., Specification, page 4, lines 15-22. Also, the ground connections now more clearly show the complete circuit for the rf discharge 8. Furthermore, the arrow indicating the exhaust E is moved to the chamber bottom for clarity because no opening is shown on the chamber side.

With respect to Fig. 2, as discussed with respect to Fig. 1, the chemical flow source F is moved to more properly pass through the flow controller 69. See, e.g., Specification, page 6, lines 35-37.

With respect to Fig. 3, some editorial changes are made in boxes 105, 106, 107, and 109 for consistency with the Specification or to correct some typographical errors. For example, in box 109 $k_i(T)$ is replaced with $k_s(T)$ to maintain consistency with the text. See, e.g., Specification, page 11, the equation after line 15. In box 106, "z" is replaced with " d_{gap} " for consistency with the Specification, for example, at page 11, the equation after line 5. Similarly in box 107, "k" is replaced with " k_s " for consistency with the Specification.

With respect to Figs. 4 and 5, some editorial changes are made. For example, in box 203 of Fig. 4 and box 309 of Fig. 5, "St" is replaced by " S^{T} " for consistency with the Specification, for example, at page 13, line 7.

With respect to Fig. 5A, scaling numbers on both axis are replaced with temperature and pressure values. Also, reference number 503 is deleted because the text does not refer to it.

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With respect to Fig. 7, the two Cartesian axes at the bottom of the informal drawing are deleted because they are not referenced or used in the radial coordinate system of this figure.

With respect to Fig. 8, the label "S" is added to conform to the reference to a substrate as "S" in the Specification, at page 21, line 22.

In Fig. 10, all three *x-y-z* axes are show to more clearly demonstrate the relative gap between d=0 and $d=d_{gap}$.

In Fig. 12, the units shown on the *y*-axis of the newly submitted formal drawing are twice as great as the units shown in the original informal drawing. This amendment corrects a typographical error made when making the original drawing. While Applicant believes that those skilled in the art would understand the invention and the error in Fig. 12, this correction is made to avoid any possible confusion on the part of the reader. Also, the downward spikes at the four corners of the informal drawing are absent in the formal drawing. These spikes were the result of utilizing a limited number of mathematical terms to generate the plotted surface. The gridded pattern shown in the formal drawing has no spikes, for example, in accordance with the complete equation in the Specification at page 24, after line 10, and the text at page 26, lines 1-4.

This amendment is submitted in triplicate. Please charge any other fees or credit any overpayment to Deposit Account 20-1430.

If the Examiner believes a telephone conference would expedite issuance of this application, please telephone the undersigned at (650) 326-2400.

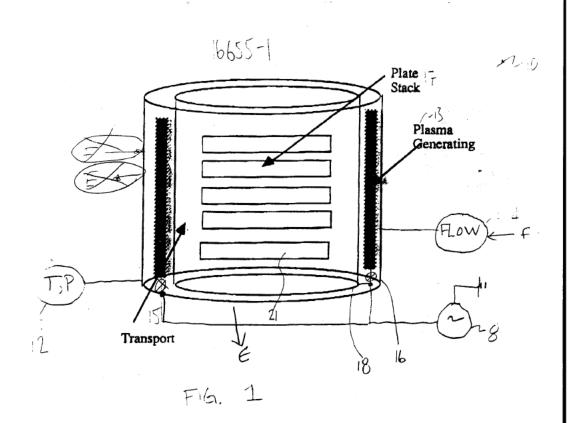
Respectfully submitted,

Date: 10/27/97

Vary J. alea Gary T. Aka

Gary 17 Aka Reg. No. 29,038

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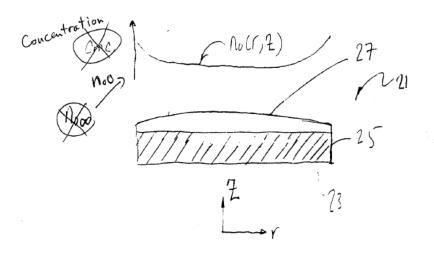
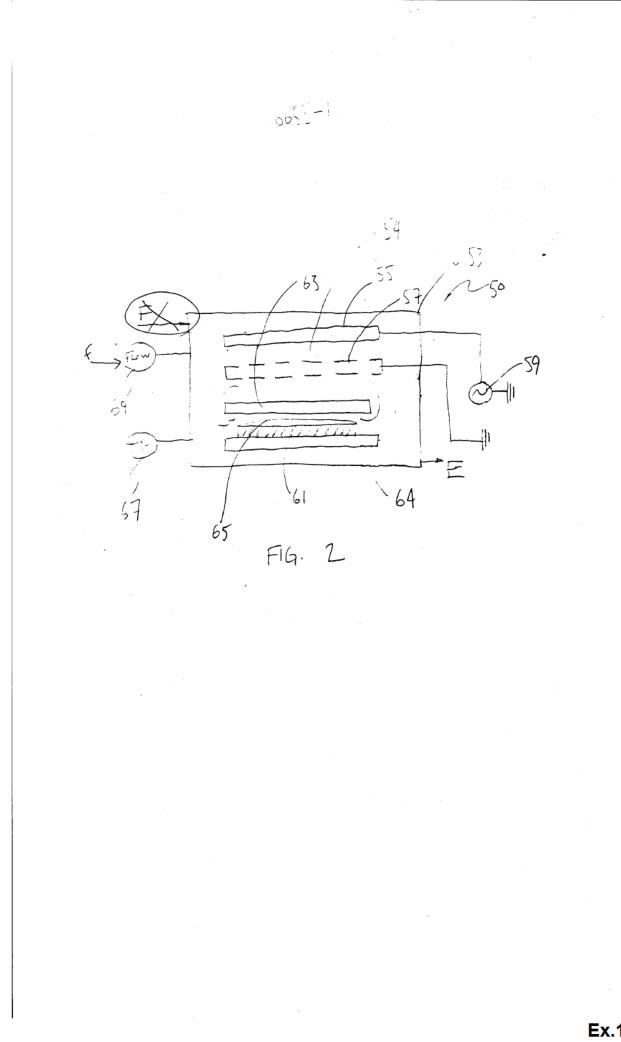
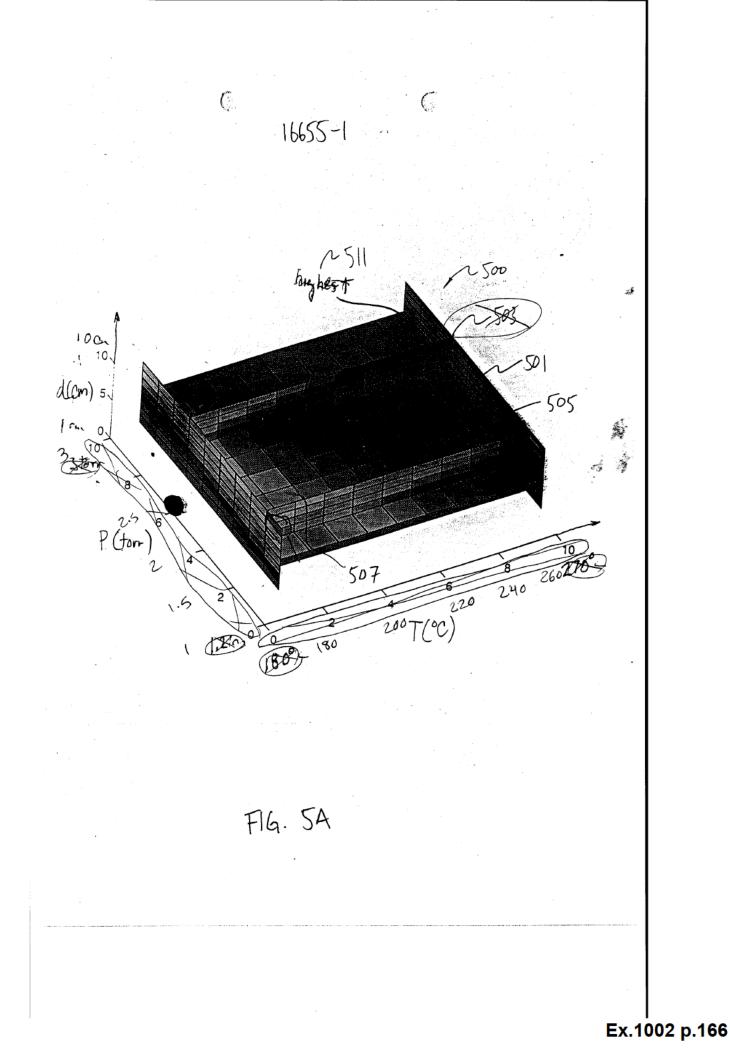
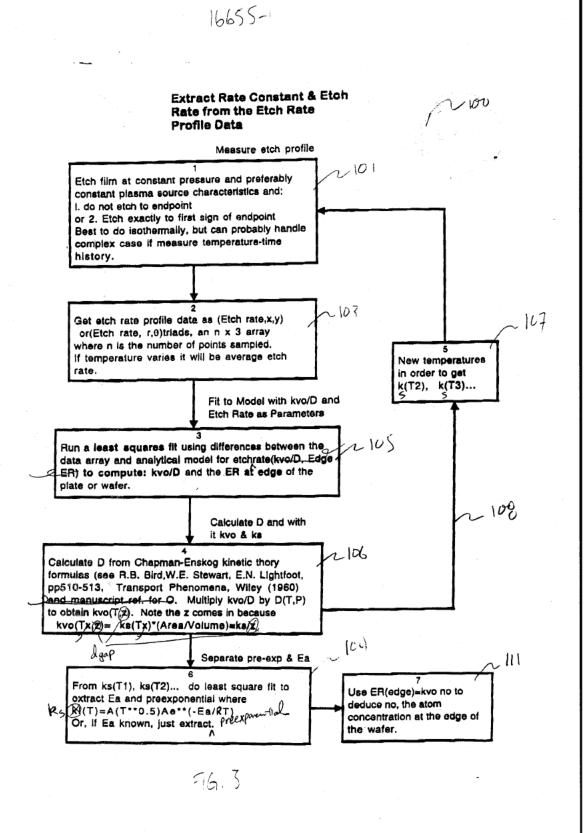
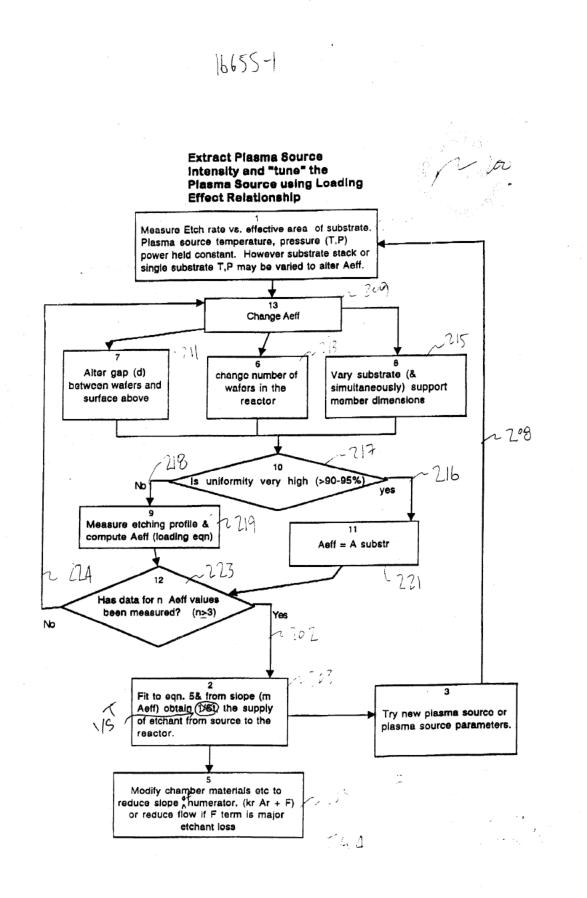


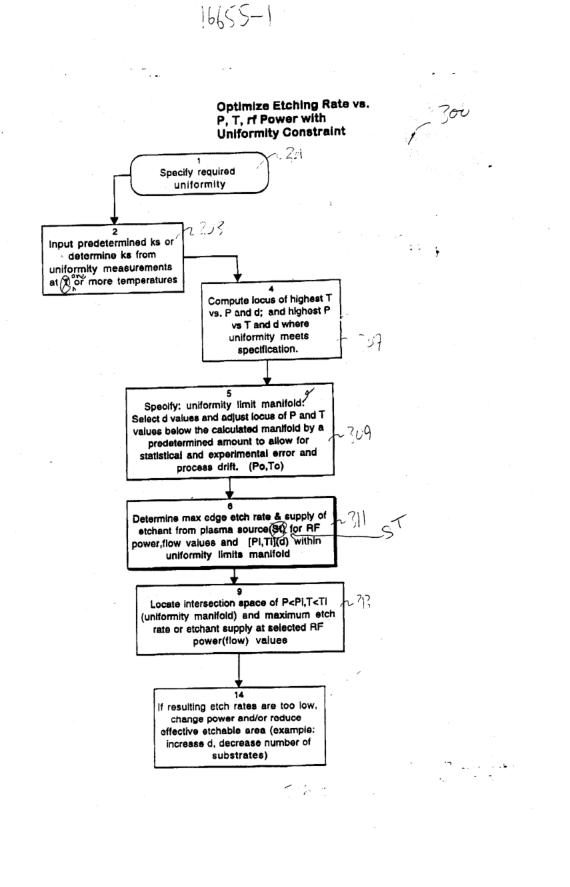
FIG. 1A

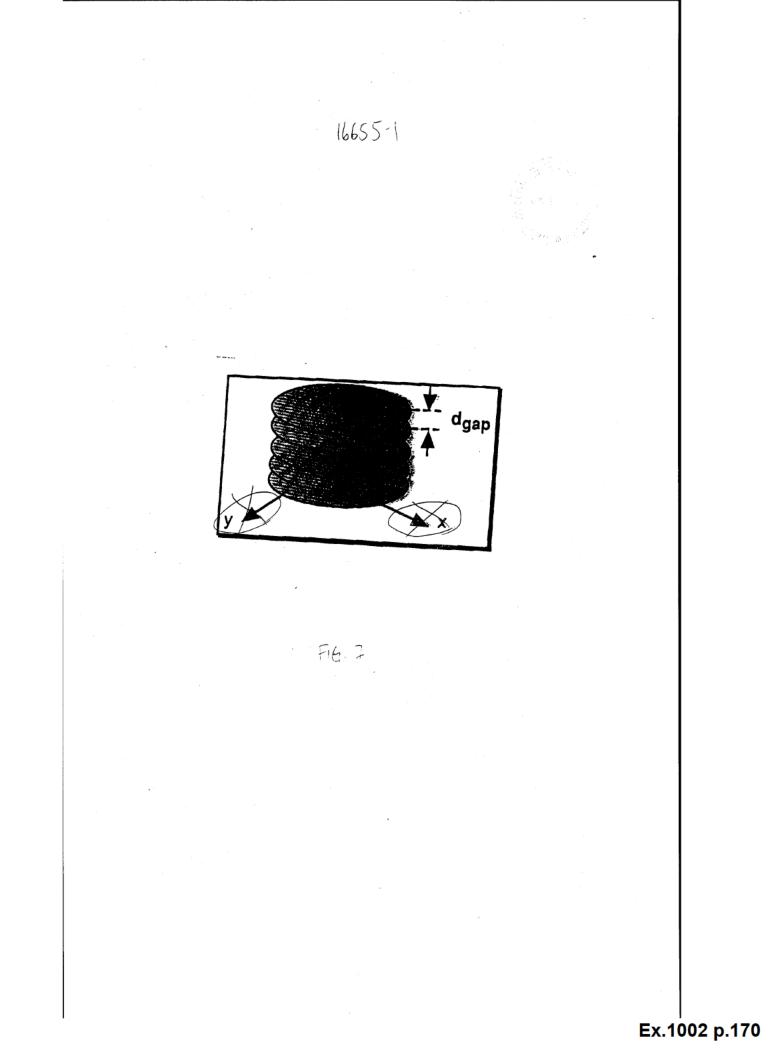












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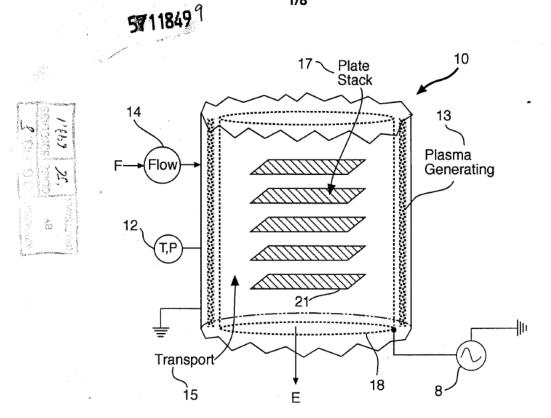
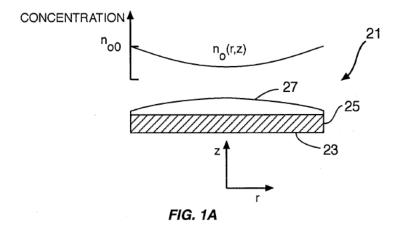
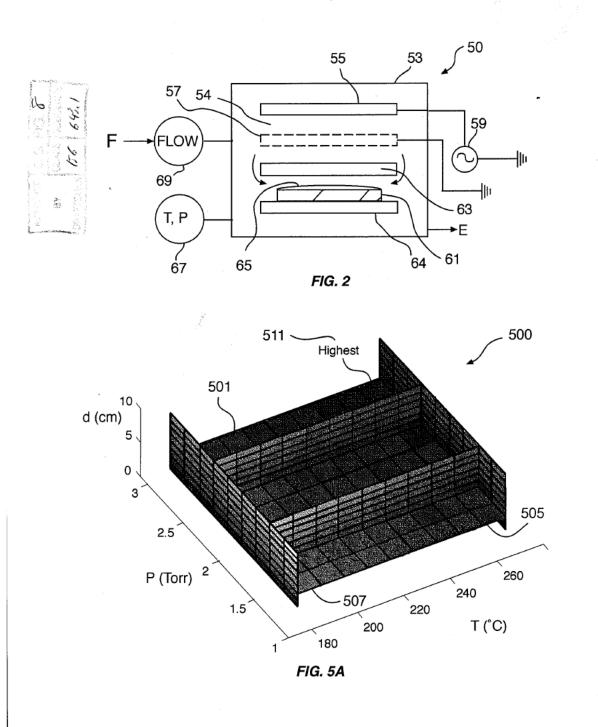
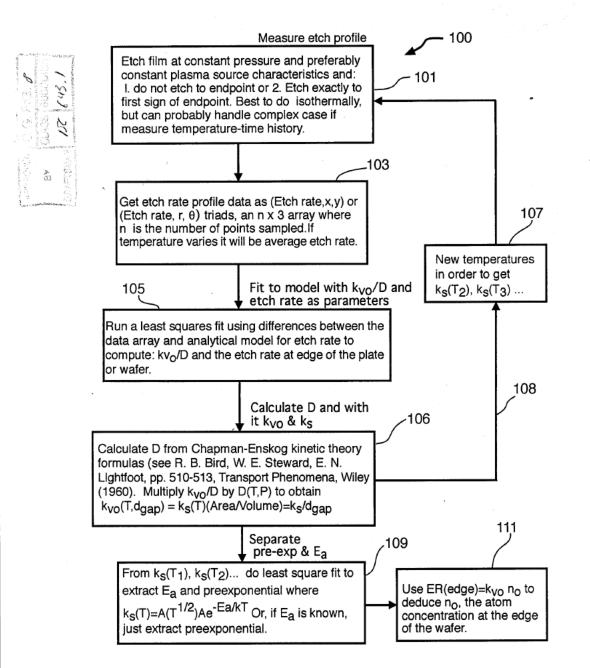


FIG. 1









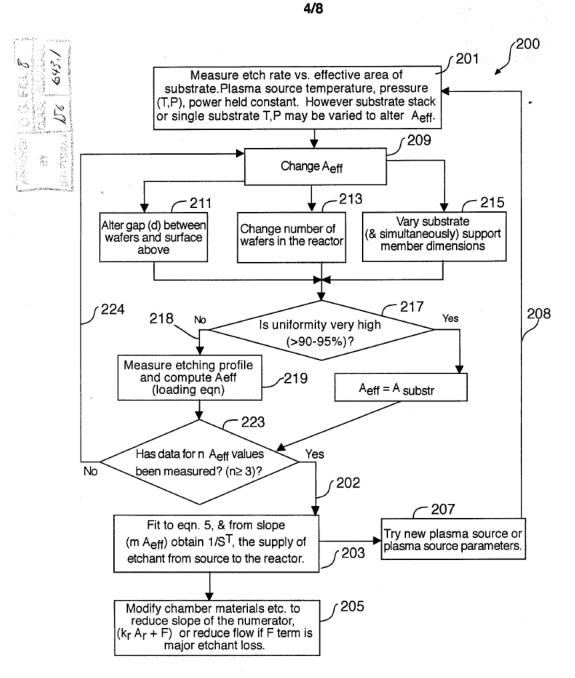
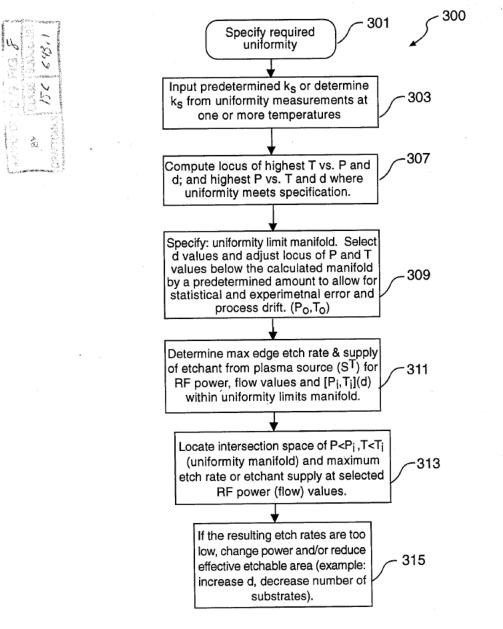
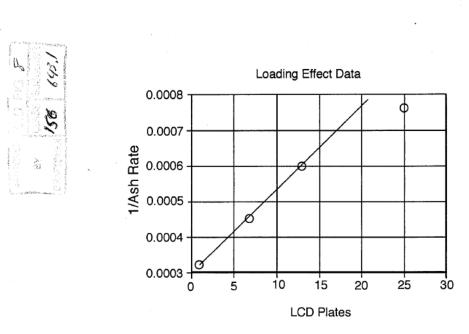


FIG. 4









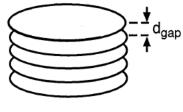


FIG. 7

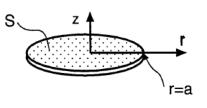
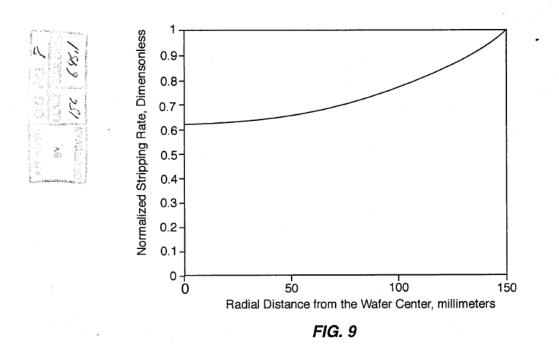
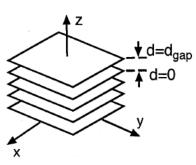


FIG. 8







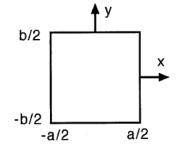
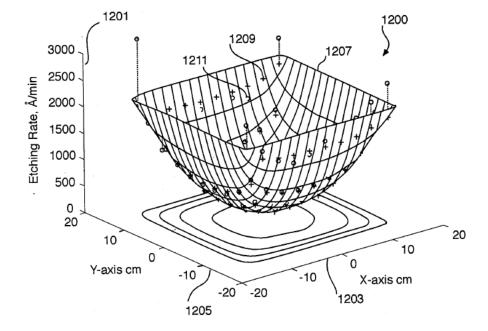


FIG. 11





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I hereby certify that the following Communication and copies of Amendment Transmittal; Amendment After Payment of Issue Fee Under 37 CFR 1.312(b); Petition for Amendment After Payment of Issue Fee Under 37 CFR 1.312(b); Serial No 08/433,623, Filed: May 3, 1995 for PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING is being facsimile transmitted to the Patent and Trademark Office on the date shown below.

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In re application of Daniel L. Flamm, et al.

Serial No: 08/433,623

Filed: May 3, 1995

Group Art Unit: 1113

For: PROCESS OPTIMIZATION IN GAS PHASE DRY ETCHING

Express Mail Label No.: EM140585555US I hereby certify that this is being deposited with the United States Postal Service "Express Mail Post Office to Addressee," service under 37 CFR 1.10 on the date indicated below and is addressed to: Group 1100 Director - Theodore Morris -Assistant Commissioner for Patents Washington, D. C. 20231,

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By same

ASSISTANT COMMISSIONER FOR PATENTS Washington, D.C. 20231

Sir:

Transmitted herewith is an amendment in the above-identified application.

[X] Amendment After Payment of Issue Fee Under 37 CFR 1.312(b).

[X] Petition for Amendment After Payment of Issue Fee Under 37 CFR 1.312(b).

[X] Postcard (in Express Mail copy).

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Attorneys for Applicant

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	In re application of: Daniel L. Flamm Application No.: 08/433,623 Filed: May 3, 1995 For: PROCESS OPTIMIZATION IN GA PHASE DRY ETCHING Assistant Commissioner for Patents Washington, D.C. 20231 Dear Sirs: Pursuant to 37 C.F.R. § 1.31 the accompanying amendment. As indicate	Attorney Docket No. 16655-000100 Attorney Docket No. 16655-000100 COPY PATENT AND TRADEMARK OFFICE

ь:09 PM Date: 10/27/97 Time: 415 326 2422 29 Sender: Pages: Subject: Fax Number: Company: 0CT 27 '97 06:11PM TTC PALO ALTO 415 326 2422 Type: P.4 Fax Daniel L. Flamm PATENT Application No.: 08/433,623 Page 2 Please charge the petition fee of \$130.00 to Deposit Account 20-1430. Thispetition is submitted in triplicate. Please charge any other fees or credit any overpayment to Deposit Account 20-1430. Respectfully submitted, Date: 10/27/97 Gary T Aka Reg. No. 29,038 TOWNSEND and TOWNSEND and CREW LLP Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (650) 326-2400 Fax (650) 326-2422 GTA:RTO:RA:de H:\WORK\16655\000100\312V1.PET 4

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insert [1-	$\frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}]$			
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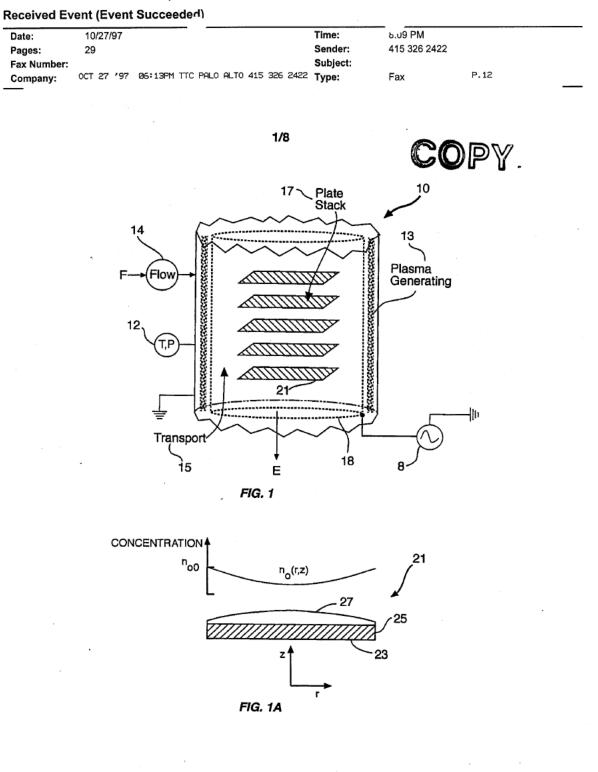
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• •	1		1
	Daniel L. Flamm Application No.: 08, Page 3	/433,623	PATENT
		R	-R _{mar}
	At page 14," "	line 30, please delete " $\frac{R_{MAX}}{2\sum}$	R_{i} and please insert
		2/	rí m
	$[1 - \frac{R_{MAX} - R_{MIN}}{2\sum_{i=1}^{m} \frac{R_i}{m}}]$		
	$2\sum_{i=1}^{n}\frac{K_i}{m}$		
2			
	At page 15, lir	ne 37, please insert, after T	:
	At page 15, lir	ne 37, please delete "vs." and in	nsertversus
	At page 15, lir	ne 38, please insert, after h	ighest P.
	At page 15, lir	e 38, please delete "vs." and in	nsertversus
	At page 16, lir	ne 3, please insert, after hig	zhest T.
		ne 3, after T, please delete "vs."	
		ne 3, please insert, after his	
		ne 3, after P, please delete "vs.	
		ie 5, please insert, after hig	
		ne 5, after T, please delete "vs."	
		ie 5, please insert, after his	
		ne 5, after P, please delete "vs.	
	· · · · ·	the 21, " R_0 " and insert R_{00}	
			ce recombination rate" and insert
	effective surface recombination	-	
		the 18, please delete " A_{*} " and in	nsert A
			se insertwhen etching uniformity is
	high the	,, <u>-</u> , <u>-</u>	
		te 16, please delete " k_e " and in	sertk
•			ase delete " k_o " and insert k_s
			ase delete " A_{μ} " and insert A_{eff}
	1	the equation after line 5, pleas	-
			se delete "*" and insert(before
	$D \nabla n_o$ and insert) after $D \nabla$		

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		Daniel L.	Flamm on No.: 08/433,623			PATENT
			At page 20, line 17, please d	elete " $n(x/L_n)$	/L.)" and insert	u(x/L, v/L.)
			At page 20, line 19, please d			*
			At page 23, line 9, please de			
			At page 23, line 26, please d			•
			At page 24, between the firs			e, please insert
		where λ_x is gi	ven by			
			At page 24, line 1, please de	lete "the gener	al" and insert The	e general
			At page 24, in the last equat	•		2
		$\cos \frac{m\pi y}{2}$.		10		
		b	At page 25, line 1, please in	serttwo-dim	ensional after pre	vious.
			At page 25, line 7, please de	lete "from" an	d insertusing rela	ations in
			At page 25, line 26, please d	lelete "substitu	te" and insert sub	stituting
			At page 25, line 31, please d	elete "11" and	insert 12	
			At page 26, line 10, please d	elete "drove" :	and insertsustain	ed
			At page 26, line 17, please d	elete "11" and	insert12	
	·		At page 26, line 18, please in	nsertto afte	er fit.	
			At page 26, line 35, please d	lelete "R" and	insert Ros	
			At page 27, line 10, please d	lelete "A _{weff} " as	nd insert Aer	
			At page 27, in the equation a	after line 10, p	lease replace all oc	currences of "A _{weff} "
		with <i>A_{eff}</i>				
		-	At page 27, in the equation	after line 10, p	lease replace "Re" v	with <i>R</i> _{os}
			At page 27, line 11, please d	lelete " $A_{w,eff}$ " a	nd insertA _{eff}	
			At page 27, line 19, please r	eplace both oc	currences of "A _{weff} "	" with <i>A_{eff}</i> .
			At page 28, line 6, please de	elete "describe"	and insert descr	ibes
		IN THE DRA	WINGS:			
		A COLORADO AND	Applicant submits new figu	res 1-12, consi	sting of eight sheet	ts of formal
		drawings. Ne	wly submitted figures correct	-	• •	
		•	gs. Amendments to the original			
		highlighted in			- monimum montos	

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ompany.			.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
		Daniel L. Flamm Application No.: 08/433,6: Page 5	23		PATENT
		In Fig. 1, chemical flow	w source 14 was deta	ached from the chem	ical feed F . In the
	. (new formal drawings chemical feed F	is shown passing th	rough the chemical f	low source 14.
		Also, the ground connections have been	en amended to more	clearly show the con	nplete circuit for
		the rf discharge 8. Furthermore, the a	rrow indicating the e	exhaust E is moved t	o the chamber
		bottom.			
	ì	In Fig. 1A, "Conc" on	the y-axis is replaced	d by "Concentration"	for clarity. Also
		n_{o0} has been aligned with the y-axis.			
		In Fig. 2, as discussed	with respect to Fig.	1, the chemical flow	source F is moved
		to pass through the flow controller 69.			
		In Fig. 3, as highlighte	d in red ink, some ed	litorial changes are r	nade in boxes 105,
		106, 107, and 109. For example, in be	$109 k_i(T)$ has been	n replaced with $k_s(T)$. In box 106, "z"
		has been replaced with " d_{gap} ". Similar	rly in box 107, " <i>k</i> " h	as been replaced wit	h "k,". In box 109,
		k_i has been replaced with k_i and "prees	cponential" has been	added at the end.	
		In Fig. 4, some editoria	al changes are made	in boxes 203 and 20	5. In box 203, "St"
		is replaced by "S". In box 205, "of" i	s added after slope.		
		In Fig. 5, some editoria	al changes are made	in boxes 301, 303, a	nd 309. In box
		301, "1" is replace with "one". As dis	cussed with respect	to Fig. 4, in box 309	, "St" is replaced
		by " S^{r} ". In box 309, ":" is replaced w	ith a period.		
		In Fig. 5A, scaling num	nbers 0, 2, 4, 6, 8, ar	nd 10 on both axis ar	e replaced with
		temperature and pressure values. Also	o, reference number	503 is deleted.	
		In Fig. 7, the two Carte	esian axes at the bott	tom of the informal of	lrawing are
		deleted.			
		In Fig. 8, the label "S"	has been added.		
		In Fig. 10, the x-y-z ax	es are added and nev	w labels are added to	show the relative
		gap between $d=0$ and $d=d_{gap}$.			
	···	In Fig. 12, the units on		ged. Also, the down	ward spikes at the
		four corners of the informal drawing a	are deleted.		
			REMARKS		
		This amendment does		and no new matter i	s involved.
			9		

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		Daniel L. Applicatic Page 6	Flamm n No.: 08/433,62	3		PATENT
		•	m			
			The Specification has b			e
		-	or example, the amendr			-
		-	tion at page 22, line 5.			
			ditorial to make the equ		-	
	٠.	•	at page 12, lines 26-29.			
			equation would include			
		corrected in ac	cordance with the text.	Other amendments r	epresent other mind	or typographical
		corrections to t				
			The newly submitted for	-	-	
		in the original	y submitted formal figu	res. Amendments to	the originally filed	informal figures
		have been high	lighted in red ink.			
			With respect to Fig. 1, i	n the new formal dra	wings, chemical fe	ed F is correctly
		shown passing	through the chemical fl	ow source 14. See, e	e.g., Specification,	page 4, lines 15-22.
		Also, the groun	nd connections now more	e clearly show the co	mplete circuit for	the rf discharge 8.
		Furthermore, t	he arrow indicating the	exhaust E is moved to	o the chamber botto	om for clarity
	•	because no ope	ening is shown on the ch	amber side.		
			With respect to Fig. 2,	as discussed with resp	pect to Fig. 1, the c	hemical flow
		source F is mo	ved to more properly pa	ss through the flow o	controller 69. See,	e.g., Specification,
		page 6, lines 3	5-37.			
			With respect to Fig. 3,	some editorial change	es are made in boxe	es 105, 106, 107,
		and 109 for co	nsistency with the Speci	fication or to correct	some typographics	al errors. For
		example, in bo	x 109 $k_i(T)$ is replaced v	with $k_s(T)$ to maintain	consistency with t	he text. See, e.g.,
		-	page 11, the equation af		-	
		_	th the Specification, for			
			' is replaced with "k," for		•	-
			With respect to Figs. 4	-	-	. For example, in
		box 203 of Fig	. 4 and box 309 of Fig.		•	
		-	for example, at page 13			-
		-r	With respect to Fig. 5A		both axis are repla	aced with
		temperature ar	d pressure values. Also		-	
		refer to it.		,		
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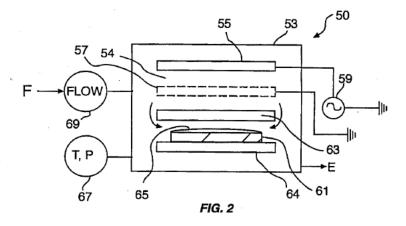
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		Daniel L. Flamm Application No.: 08/433,623 Page 7			PATENT
		With respect to Fig. 7, the t	wo Cartesian a	xes at the bottom of	f the informal
		drawing are deleted because they are not re			
		this figure.			difface System of
		With respect to Fig. 8, the 1	abel "S" is add	ed to conform to the	e reference to o
		substrate as "S" in the Specification, at page			c reference to a
		In Fig. 10, all three x-y-z ax		more clearly down	extends the relation
		gap between $d=0$ and $d=d_{eas}$.	us are show to	more clearly demor	istrate the relative
				64	
		In Fig. 12, the units shown			-
		are twice as great as the units shown in the			
		a typographical error made when making t			
		those skilled in the art would understand th			
		made to avoid any possible confusion on the			
		four corners of the informal drawing are ab			
		result of utilizing a limited number of math			
		gridded pattern shown in the formal drawin	ng has no spike	s, for example, in a	ccordance with the
		complete equation in the Specification at p	age 24, after lin	ne 10, and the text a	t page 26, lines 1-4.
	·	This amendment is submitted	ed in triplicate.	Please charge any	other fees or credit
	- 11	any overpayment to Deposit Account 20-1	430.		
		If the Examiner believes a t	elephone confe	rence would expedi	te issuance of this
		application, please telephone the undersign	ned at (650) 326	5-2400.	
			Resp	ectfully submitted,	
		Date: 10/27/97	Gary Reg.	an 7. de. T. Aka No. 29,038	2 <u>e</u>
		TOWNSEND and TOWNSEND and CRE Two Embarcadero Center, 8th Floor San Francisco, California 94111-3834 (650) 326-2400 Fax (650) 326-2422 GTARTO:RA:de H:WORK:166550000100312V2.AMD	WLLP		
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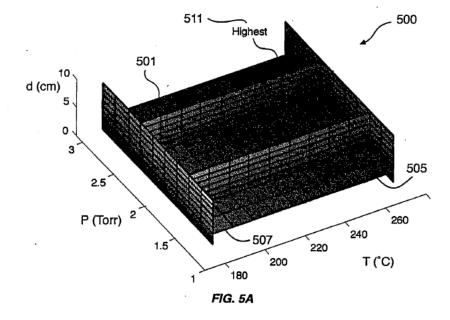


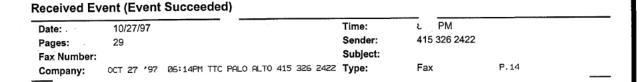
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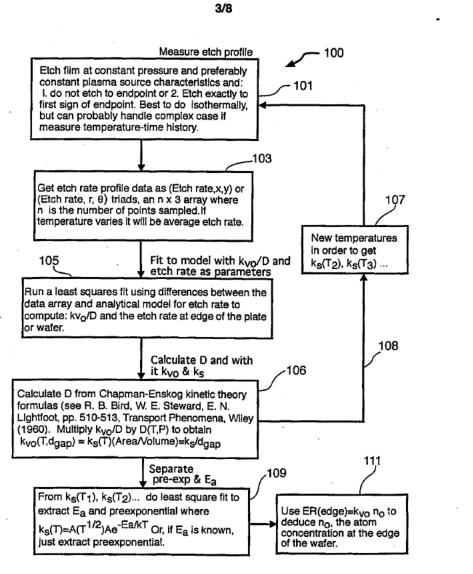
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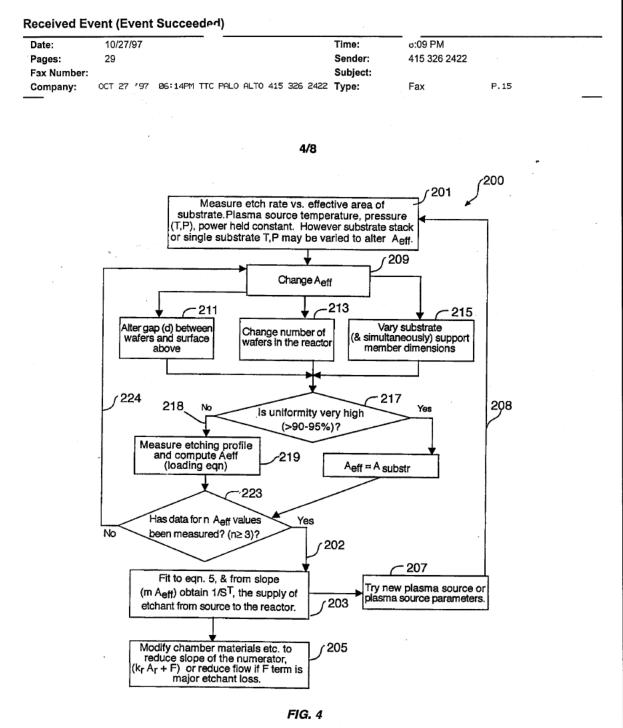


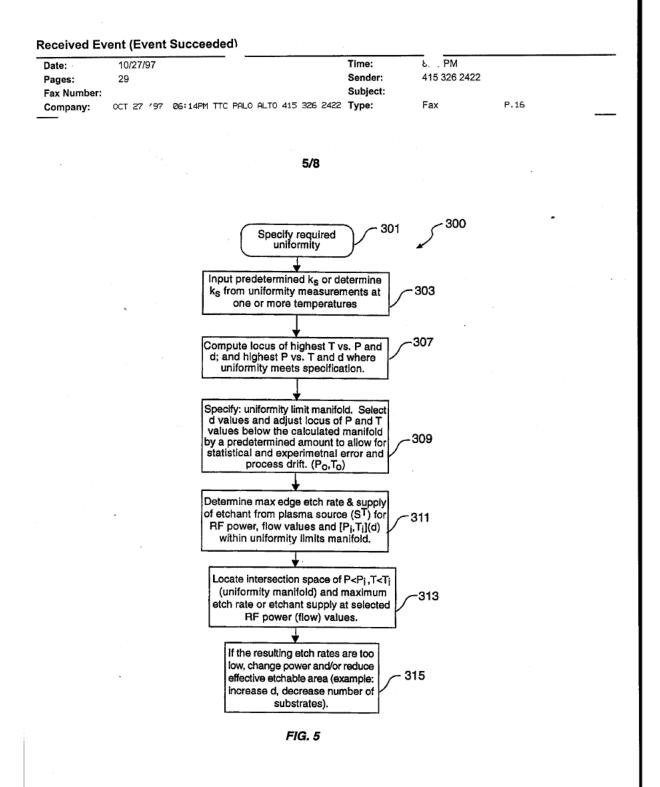












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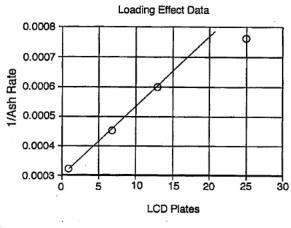


FIG. 6

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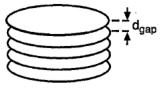


FIG. 7

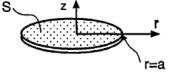
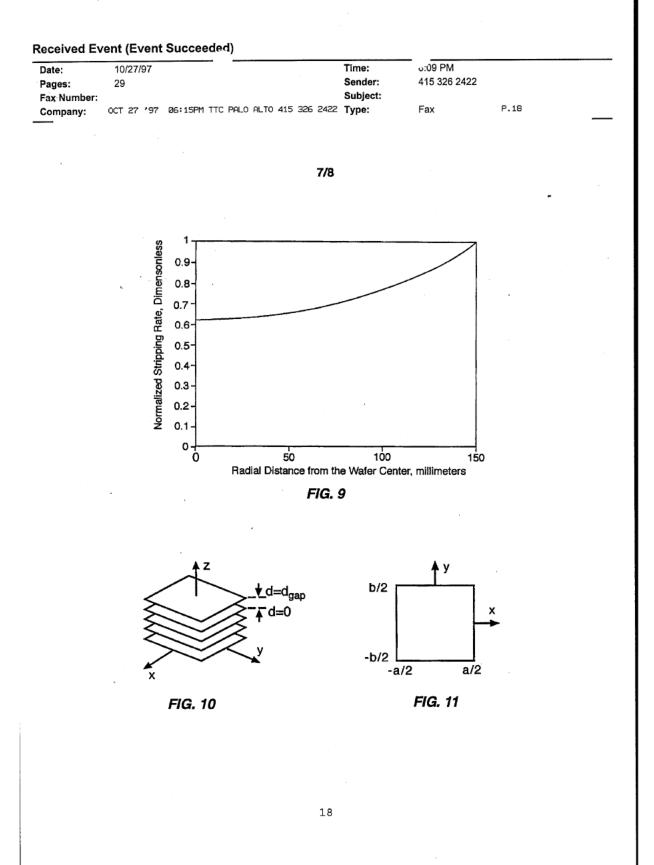
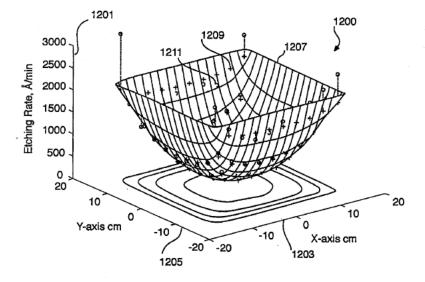


FIG. 8

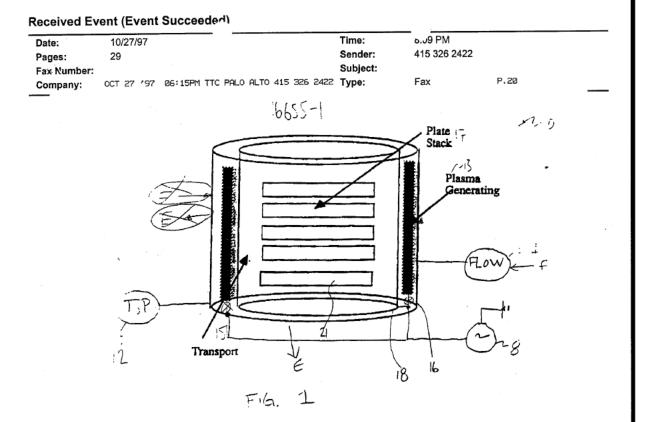


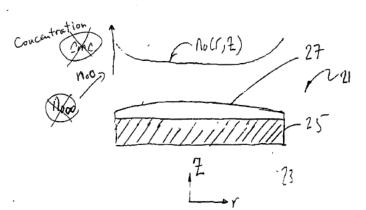
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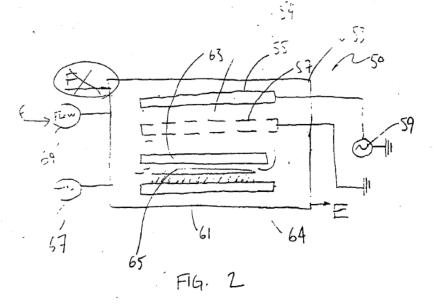


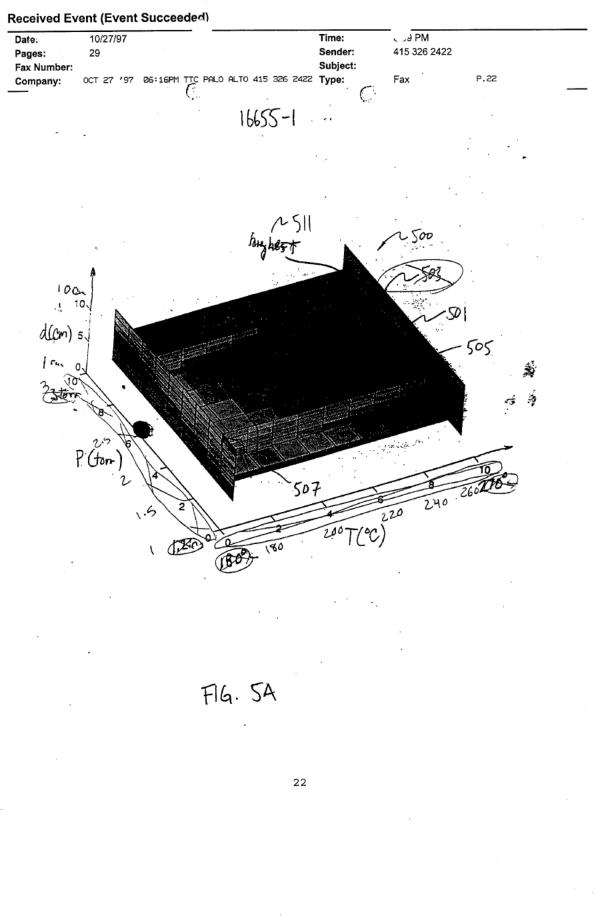
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APPLICATION NO.	FILING DATE	FIRST NAM	ED INVENTOR		ATTORNEY DOCKET NO.
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				DATE MAILED:	12/08/97

Please find below and/or attached an Office communication concerning this application or proceeding.

Commissioner of Patents and Trademarks

PTO-90C (Rev. 2/95)

Application No. Applicant(s) 08/433,623 Flamm Response to Rule 312 Examiner Group Art Unit Communication Martin J. Angebranndt 1113 🕅 The petition filed on <u>Oct 27, 1997</u> under 37 CFR 1.312(b) is granted. The paper has been forwarded to the examiner for consideration on the merits. herdus Wan Theodore Morris Director, Patent Examining Group 1100 X The amendment filed on _____Oct 27, 1997 ____ under 37 CFR 1.312 has been considered, and has been: entered. I entered as directed to matters of form not affecting the scope of the invention (Order 3311). disapproved. See explanation below. entered in part. See explanation below. The changes to the specification are minor and the drawing changes have been approved by both the examiner and the Draftsman. MARTIN J. ANGEBRANNDT PRIMARY EXAMINER ART UNIT 1113 U. S. Patent and Trademark Office PTO-271 (Rev. 5-95) Response to Rule 312 Communication Part of Paper No. 16



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PTO-FMD TALBOT-1/97 The file was thoroughly reviewed by our staff. Correspondence Address Change/Power of Attorney dated 04/18/2001 is missing in this file.

This has been brought to your attention so that you will know it has not been overlooked. SAO 120 (Rev. 2/99)

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REPORT ON THE FILING OR DETERMINATION OF AN ACTION REGARDING A PATENT OR TRADEMARK

In Compliance with 35 § 290 and/or 15 U.S.C. § 1116 you are hereby advised that a court action has been

 filed in the U.S. District Court Northern District of California on the following Patents or Trademarks:

 DOCKET NO.
 DATE FILED
 U.S. DISTRICT COURT

 CV 15-01277 JSC
 3/18/15
 450 Golden Gate Avenue, Box 36060, San Francisco, CA 94102

LAM RESEARCH CO	RP	DEFENDANT DANIEL L FLAMM			
PATENT OR TRADEMARK NO.	DATE OF PATENT OR TRADEMARK	HOLDER OF PATENT OR TRADEMARK			
1 see Complaint					
25,711,849					
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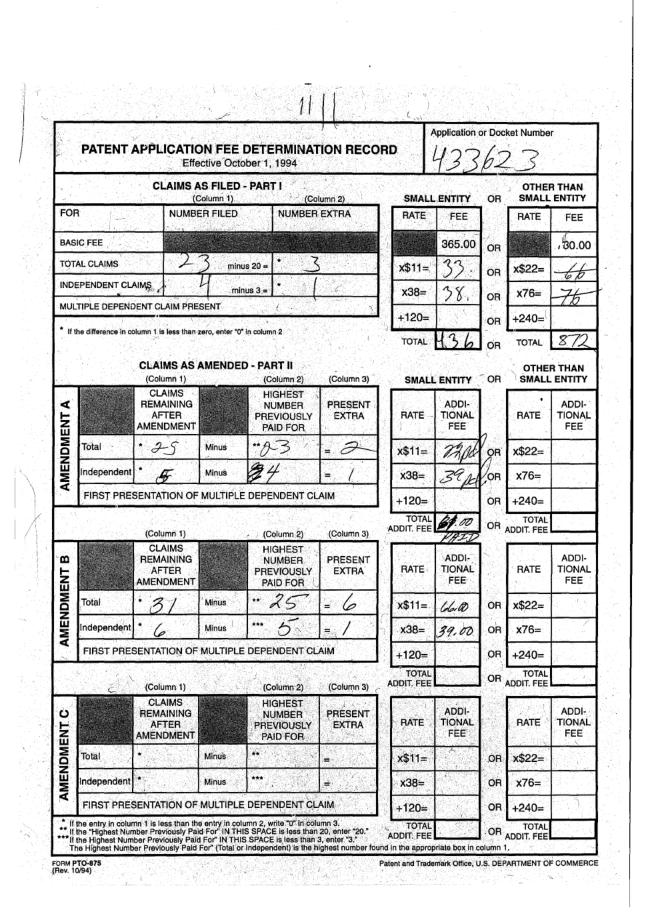
In the above-entitled case, the following patent(s) have been included:

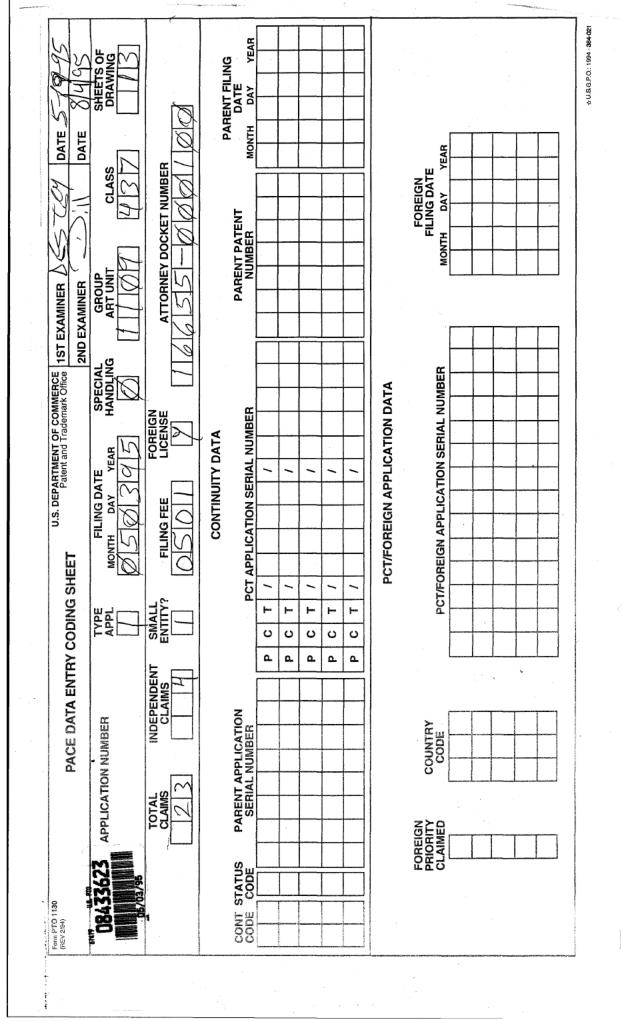
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In the above---entitled case, the following decision has been rendered or judgement issued:

DECISION/JUDGEMENT		
CLERK	(BY) DEPUTY CLERK	DATE
Richard W. W	icking Sheila Rash	March 19, 2015

Copy 1—Upon initiation of action, mail this copy to Commissioner Copy 3—Upon termination of action, mail this copy to Commissioner Copy 2—Upon filing document adding patent(s), mail this copy to Commissioner Copy 4—Case file copy





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SPA	
L5	25 S ((NONUNIFORM? OR UNIFORM?)(8A)(ETCH?))(15A)((POSITION OR
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	(1 1	S (ETCH?) (P) (RATE CONSTANT#)	
L1	18	S (ETCH?) (P) (RAIL CONSTANT#)	
		S' ENTERED AT 16:01:24 ON 17	TAN 97
	FILE JPOAD	S. ENIEKED HI TO.OT.21 ON T.	
L_2	8	S L1	
	0		