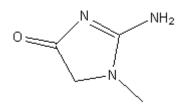
Print Friendly View

Catalog Number: 101423, 101424, 153916 **Creatinine**

Structure (free base):



	Free Base	Hydrochloride	Zinc Chloride
Molecular Formula:	C ₄ H ₇ N ₃ O	C ₄ H ₇ N ₃ O·HCl	$(C_4H_7N_3O)_2 \cdot ZnCl_2$
Molecular Weight:	113.1	149.59	362.5
CAS #	60-27-5	19230-81-0	362.5

Synonyms: 2-Amino-1,5-dihydro-1-methyl-4H-imidazol-4-one; 2-Amino-1-methyl-4-imidazolidinone; 1-Methylhydantoin-2-imide; 1-Methylglycocyamidine; 2-Imino-N-methylhydantoin; 2-Imino-1-methylimidazolidin-4- one

Physical Description: White powder

Solubility:

Free Base: Soluble in 12 parts water; slightly soluble in ethanol; practically insoluble in acetone, ether, chloroform.¹

Hydrochloride: Soluble in water (50 mg/ml - clear, colorless to faint yellow solution) *Zinc Chloride*: Soluble in 1 N Hydrochloric acid (50 mg/ml - clear, colorless solution)

Description: The end product of creatine catabolism.¹ Normal constituent of urine; daily output about 25 mg/kg body weight.¹ Increased amounts in the urine are typically associated with substantially impaired renal function. Also found together with creatine in muscle tissues and blood.¹ Reacts with picric acid under alkaline conditions to form a Janovski complex. The rate of formation of the colored complex, measured at 480-520 nm is proportional to the creatinine concentration.

Availability:

Catalog Number	Description	Size
101423	Creatinine, free base	10 g 25 g 100 g
153916	Creatinine Hydrochloride	

		10 g 25 g 100 g
101424	Creatinine Zinc Chloride	1 g 5 g

References:

- 1. Merck Index, 12th Ed., No. 2638.
- 2. Butler, A.R. and Glidewell, C., "Creatinine: an examination of its structure and some of its reactions by synergistic use of MNDO calculations and nuclear magnetic resonance spectroscopy." *J. Chem. Soc. Perkin Trans.*, v. 2, 1465-1468 (1985).
- 3. Hirate, et al., "The effect of sacrificing method on the tissue concentration of exogeneous creatinine in rats." *Chem. Pharm. Bull.*, v. 33(11), 5105-5107 (1985).
- 4. Kim, et al., "Sarcosine oxidase involved in creatinine degradation in Alcaligenes denitrificans subsp. denitrificans J9 and Arthrobacter spp. J5 and J11." *Agric. Biol. Chem.*, v. 50(11), 2811-2816 (1986).
- 5. Nakamura, et al., "Creatones A and B. Revision of the structure for the product of oxidation of creatinine and creatine." *Bull. Chem. Soc. Jpn.*, v. 63(5), 1540-1542 (1990).
- 6. Valcour, A.A. and Woodworth, R.C., J. Magn. Reson., v. 66, 536 (1986).
- 7. Welch, M.J., et al., "Determination of serum creatinine by isotope dilution mass spectrometry as a candidate definitive method." *Anal. Chem.*, v. 58(8), 1681-1685 (1986)
- 8. Winquist, et al., "Determination of creatinine by an ammonia-sensitive semiconductor structure and immobilized enzymes." *Anal. Chem.*, v. 58(1), 145-148 (1986).
- 9. Yokozawa, et al., "Changes in the urinary constituents in rats with chronic renal failure during oral administration of rhubarb extract." *Chem. Pharm. Bull.*, v. 33(10), 4508-4514 (1985).