(check at least one box)

.

- a. Except as may be indicated below in (b), all of the patents, publications or other information are in the English language or were cited in an English language Search Report, a copy of which is attached hereto (concise explanation not required).
- b. A concise explanation of the relevance of all patents, publications or other information listed that is not in the English language is as follows:
- c. \square The following additional information is provided for the Examiner's consideration:

The listed reference was cited by the Chinese Patent Office in an Office Action dated December 12, 2003, in a corresponding application.

IV. <u>THIS IDS IS BEING FILED UNDER 37 C.F.R. § 1.97(b)</u> (check one box)

- a. within three months of the filing date of a national application (37 C.F.R. § 1.97(b) (1)). No fee or certification is required.
- b. within three months of the date of entry of the national stage as set forth in §1.491 in an international application (37 C.F.R. § 1.97(b) (2)). No fee or certification is required.
- c. before the mailing date of a first Action on the merits (37 C.F.R. § 1.97(b) (3)). No fee or certification is required. In the event that a first Office Action on the merits has been issued, please consider this IDS under 37 C.F.R. § 1.97(c) and see the certification under 37 C.F.R. § 1.97(e) below, or, if no certification has been made, charge our deposit account a fee in the amount of \$240.00 as required by 37 C.F.R. § 1.17(p).

V. <u>THIS IDS IS BEING FILED UNDER 37 C.F.R. § 1.97(c):</u> (check one box)

before the mailing date of a Final Office Action under 37 C.F.R. § 1.113 (See 37 C.F.R. § 1.97(c) (1)) or before the mailing date of a Notice of Allowance under 37 C.F.R. § 1.311 (See 37 C.F.R. § 1.97(c) (2)).

a. No certification; therefore, a fee in the amount of \$180.00 is required by 37 C.F.R. § 1.17(p).

or

b. See the certification below. No fee is required.

VI. <u>THIS IDS IS BEING FILED UNDER 37 C.F.R. § 1.97(d)</u>: (check both boxes if applicable)

before payment of the Issue Fee (See 37 C.F.R. § 1.97(d).

- a. \boxtimes See the certification below; and
- b. A fee in the amount of \$180.00 is enclosed as required by 37 C.F.R. § 1.17(p).

VII. <u>CERTIFICATION UNDER 37 C.F.R. § 1.97(e)</u> (check <u>only</u> one box)

The undersigned hereby certifies that

c. 🗌

 \square

- a. each item of information contained in the IDS was cited in a communication from a foreign Patent Office in a counterpart foreign application not more than three months prior to the filing of this IDS; or
- b. no item of information contained in the IDS was cited in a communication from a foreign Patent Office in a counterpart foreign application or, to the best of my knowledge after making reasonable inquiry, was known to any individual designated in 37 C.F.R. § 1.56(c) more than three months prior to the filing of this statement.

Some of the items of information were cited in a communication from a foreign Patent Office. As to this information, the undersigned certifies that each item of information contained in the IDS was cited in a communication from a foreign Patent Office in a counterpart foreign application not more than three months prior to the filing of this IDS. As to the remaining information, the undersigned hereby certifies that no item of this remaining information contained in the IDS was cited in a communication from a foreign Patent Office in a counterpart foreign application or, to the best of my knowledge after making reasonable inquiry, was known to any individual designated in 37 C.F.R. § 1.56(c) more than three months prior to the filing of this statement.

Please charge Deposit Account No. 04-1105 in the amount of ______ for the aboveindicated fee. A triplicate copy of this paper is attached.

No fee is required.

If the Examiner has any questions concerning this IDS, he/she is requested to contact the undersigned. If it is determined that this IDS has been filed under the wrong rule, the PTO is requested to consider this IDS under the proper rule (with a petition, if necessary) and charge the appropriate fee to Deposit Account No. 04-1105.

Respectfully submitted,

Date: 2-12-04

his c.h

Christine C. O'Day (Reg. No. 38,256) EDWARDS & ANGELL, LLP P.O. Box 55874 Boston, MA 02205 Tel: (617) 439-4444

Customer No. 21874

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FORM PTO-1	449	DEMERSON		ATTY DOO	CKET NO.		Sh SEF	eet 1 RIAL N	of 1 	
INFORMATIC	N DISC	LOSURE STATEMEN	58827(45	5107)		10/130,214				
			APPLICANT(S): Claus Meese							
			FILING I May 14,	DATE: 2002		ART UNIT: 1624				
		UNITED :	STATES PATI	ENT DOCUM	ENTS					
EXAM. INITIAL		DOCUMENT NUMBER	DATE	NAME		CLA	ss	SUB CLASS	FILE DATE IF APPR	
•	AA	5,686,464	11/11/97	R.A. Jo et	514		315	· · · · · · · · · · · · · · · · · · ·		
		FOR	EIGN PATENT	DOCUMENTS						
		DOCUMENT NUMBER	PUBLISHED DATE	COUNTRY	CLASS		SUB CLASS		AN YES/NO	

OTHER DOCUMENTS (INCLUDING AUTHOR, TITLE, DATE, PERTINENT PAGES, ETC.)

Examiner:

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Date:

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UNITED STATES PATENT AND TRADEMARK OFFICE



UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

NOTICE OF ALLOWANCE AND FEE(S) DUE

7590

01/28/2004

Peter F. Corless P O Box 9169 Boston, MA 02209

EXAMINER										
TUCKER, Z	ACHARY C									
ART UNIT	PAPER NUMBER									

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1624 DATE MAILED: 01/28/2004

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/130,214	05/14/2002	Claus Meese	41946/32854	9833

TITLE OF INVENTION: STABLE SALTS OF NOVEL DERIVATIVES OF 3,3-DIPHENYLPROPYLAMINES

APPLN, TYPE	SMALL ENTITY	ISSUE FEE	PUBLICATION FEE	TOTAL FEE(S) DUE	DATE DUE
nonprovisional	NO	\$1330	\$0	\$1330	04/28/2004

THE APPLICATION IDENTIFIED ABOVE HAS BEEN EXAMINED AND IS ALLOWED FOR ISSUANCE AS A PATENT. PROSECUTION ON THE MERITS IS CLOSED. THIS NOTICE OF ALLOWANCE IS NOT A GRANT OF PATENT RIGHTS. THIS APPLICATION IS SUBJECT TO WITHDRAWAL FROM ISSUE AT THE INITIATIVE OF THE OFFICE OR UPON PETITION BY THE APPLICANT. SEE 37 CFR 1.313 AND MPEP 1308.

THE ISSUE FEE AND PUBLICATION FEE (IF REQUIRED): MUST BE PAID WITHIN THREE MONTHS FROM THE MAILING DATE OF THIS NOTICE OR THIS APPLICATION SHALL BE REGARDED AS ABANDONED. THIS STATUTORY PERIOD CANNOT BE EXTENDED. SEE 35 U.S.C. 151. THE ISSUE FEE DUE INDICATED ABOVE REFLECTS A CREDIT FOR ANY PREVIOUSLY PAID ISSUE FEE APPLIED IN THIS APPLICATION: THE PTOL-85B (OR AN EQUIVALENT) MUST BE RETURNED WITHIN THIS PERIOD EVEN IF NO FEE IS DUE OR THE APPLICATION WILL **BE REGARDED AS ABANDONED.**

HOW TO REPLY TO THIS NOTICE:

I. Review the SMALL ENTITY status shown above.

If the SMALL ENTITY is shown as YES, verify your current SMALL ENTITY status: A. If the status is the same, pay the TOTAL FEE(S) DUE shown above.	If the SMALL ENTITY is shown as NO: A. Pay TOTAL FEE(S) DUE shown above, or
B. If the status is changed, pay the PUBLICATION FEE (if required) and twice the amount of the ISSUE FEE shown above and notify the United States Patent and Trademark Office of the change in status, or	 B. If applicant claimed SMALL ENTITY status before, or is now claiming SMALL ENTITY status, check the box below and enclose the PUBLICATION FEE and 1/2 the ISSUE FEE shown above. Applicant claims SMALL ENTITY status. See 37 CFR 1.27.

II. PART B - FEE(S) TRANSMITTAL should be completed and returned to the United States Patent and Trademark Office (USPTO) with your ISSUE FEE and PUBLICATION FEE (if required). Even if the fee(s) have already been paid, Part B - Fee(s) Transmittal should be completed and returned. If you are charging the fee(s) to your deposit account, section "4b" of Part B - Fee(s) Transmittal should be completed and an extra copy of the form should be submitted.

III. All communications regarding this application must give the application number. Please direct all communications prior to issuance to Mail Stop ISSUE FEE unless advised to the contrary.

IMPORTANT REMINDER: Utility patents issuing on applications filed on or after Dec. 12, 1980 may require payment of maintenance fees. It is patentee's responsibility to ensure timely payment of maintenance fees when due.

Page 1 of 3

PTOL-85 (Rev. 11/03) Approved for use through 04/30/2004.

PART B - FEE(S) TRANSMITTAL

Complete and send	l this form,	together	with applicable	fee(s), to:	<u>Mail</u>
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Mail Stop ISSUE FEE Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450

			or Fax	Alexandria, Virginia 22313-1450 (703) 746-4000							
INSTRUCTIONS: This for appropriate. All further corr indicated unless corrected b maintenance fee notification	m should be used for trans respondence including the P elow or directed otherwise s	smitting the ISSUE F Patent, advance orders in Block 1, by (a) spe	EE and PUBLIC and notification ecifying a new c	ATION FEE (if req of maintenance fees prrespondence addres	uired). Blocks 1 through 4 s will be mailed to the current s; and/or (b) indicating a separate	hould be completed where correspondence address as arate "FEE ADDRESS" for					
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Peter F. Corless P O Box 9169 Boston, MA 02209				C I hereby certify that States Postal Service addressed to the M	ertificate of Mailing or Tran: this Fee(s) Transmittal is bein with sufficient postage for fir ail Stop ISSUE FEE address	smission g deposited with the United st class mail in an envelope above, or being facsimile					
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	FILING DATE	FIRS	T NAMED INVEN	TOR	ATTORNEY DOCKET NO	CONFIRMATION NO					
	05/14/2002		Claus Massa	Tok	41046/32854	0922					
TITLE OF INVENTION: ST	FABLE SALTS OF NOVEL	DERIVATIVES OF 3	3,3-DIPHENYLP	ROPYLAMINES							
APPLN. TYPE	SMALL ENTITY	ISSUE FEE	. PI	BLICATION FEE	TOTAL FEE(S) DUE	DATE DUE					
nonprovisional	NO	\$1330		\$0	\$1330	04/28/2004					
EXAM	INER	ART UNIT	CI	ASS-SUBCLASS							
TUCKER', ZA	ACHARY C	1624	•	514-530000							
 Change of corresponde Address form PTO/SB/12 "Fee Address" indicatio PTO/SB/47; Rev-03-02 o Number is required. ASSIGNEE NAME:AND PLEASE NOTE: Unless been previously submittee (A) NAME OF ASSIGNE 	nce address (or Change of C (2) attached. on (or "Fee Address" Indicat or more recent) attached: Use RESIDENCE DATA TO Bl an assignee is identified bel d to the USPTO or is being s EE	orrespondence ion form of a Customer E PRINTED ON THE ow, no assignee data ubmitted under separat (B) RE	gent) and the na ttorneys or agen vill be printed. PATENT, (printed will appear on the te cover, Complet ESIDENCE: (CIT	member a registered mes of up to 2 regi s. If no name is list or type) patent. Inclusion of ion of this form is NG Y and STATE OR CO	assignee data is only appropri T a substitute for filing an ass OUNTRY)	ate when an assignment has ignment.					
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This collection of informat obtain or retain a benefit l application. Confidentiality estimated to take 12 minut completed application forr case. Any comments on suggestions for reducing til Patent and Trademark (22313-1450. DO NOT S SEND TO: Commissioner	tion is required by 37 CFR by the public which is to fi is governed by 35 U.S.C. I es to complete, including ge n to the USPTO. Time will the amount of time you or his burden, should be sent t Office, U.S. Department END FEES OR COMPLE for Patents, Alexandria, Virg	1.311. The informatic le (and by the USPTC 22 and 37 CFR 1.14. T thering, preparing, and l vary depending upo equire to complete tt o the Chief Informatic of Commerce, Alexa TED FORMS TO TF ginia 22313-1450.	on is required to D to process) an This collection is d submitting the n the individual his form and/or on Officer, U.S. andria, Virginia HS ADDRESS.								

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

TRANSMIT THIS FORM WITH FEE(S)

PTOL-85 (Rev. 11/03) Approved for use through 04/30/2004.

OMB 0651-0033 U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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	TED STATES PATENT AN	ND TRADEMARK OFFICE	UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov					
APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO				
10/130,214	05/14/2002	Claus Meese	41946/32854	9833				
759	0 01/28/2004		EXAMINER					
Peter F. Corless			TUCKER, Z	ACHARY C				
Boston, MA 02209			ART UNIT	PAPER NUMBER				
			1624					
			DATE MAILED: 01/28/200	4				

Determination of Patent Term Adjustment under 35 U.S.C. 154 (b) (application filed on or after May 29, 2000)

The Patent Term Adjustment to date is 20 day(s). If the issue fee is paid on the date that is three months after the mailing date of this notice and the patent issues on the Tuesday before the date that is 28 weeks (six and a half months) after the mailing date of this notice, the Patent Term Adjustment will be 20 day(s).

If a Continued Prosecution Application (CPA) was filed in the above-identified application, the filing date that determines Patent Term Adjustment is the filing date of the most recent CPA.

Applicant will be able to obtain more detailed information by accessing the Patent Application Information Retrieval (PAIR) system (http://pair.uspto.gov).

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (703) 305-1383. Questions relating to issue and publication fee payments should be directed to the Customer Service Center of the Office of Patent Publication at (703) 305-8283.

Page 3 of 3

· · · · · · · · · · · · · · · · · · ·	Application No	Applicant(s)								
Notice of Allowability	10/130,214 Examiner	MEESE, CLAUS								
		1624								
The MAILING DATE of this communication apper All claims being allowable, PROSECUTION ON THE MERITS IS herewith (or previously mailed), a Notice of Allowance (PTOL-85) NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATENT R of the Office or upon petition by the applicant. See 37 CFR 1.313 1. This communication is responsive to <u>10 November 2003</u> . 2. The allowed claim(s) is/are <u>1-6,8-17,26 and 28-34</u> . 3. The drawings filed on HMM ¹ / ₂ are accepted by the Examine	ears on the cover sheet with the o (OR REMAINS) CLOSED in this ap or other appropriate communicatio IGHTS. This application is subject to and MPEP 1308.	correspondence address oplication. If not included n will be mailed in due course. THIS to withdrawal from issue at the initiative								
 4. Acknowledgment is made of a claim for foreign priority ur a) All b) Some* c) None of the: 1. Certified copies of the priority documents have 2. Certified copies of the priority documents have 3. Copies of the certified copies of the priority do 	nder 35 U.S.C. § 119(a)-(d) or (f). e been received. e been received in Application No cuments have been received in this	national stage application from the								
 International Bureau (PCT Rule 17.2(a)). * Certified copies not received: 5. Acknowledgment is made of a claim for domestic priority un reference was included in the first sentence of the specifica (a) The translation of the foreign language provisional a 6. Acknowledgment is made of a claim for domestic priority un in the first sentence of the specification or in an Application 	nder 35 U.S.C. § 119(e) (to a provis ation or in an Application Data Shee application has been received. nder 35 U.S.C. §§ 120 and/or 121 s Data Sheet. 37 CFR 1.78.	ional application) since a specific t. 37 CFR 1.78. ince a specific reference was included								
 Applicant has THREE MONTHS FROM THE "MAILING DATE" of below. Failure to timely comply will result in ABANDONMENT of the submerse of	this communication to file a reply c this application. THIS THREE-MO itted. Note the attached EXAMINER es reason(s) why the oath or declara to be submitted. on's Patent Drawing Review (PTO-	omplying with the requirements noted NTH PERIOD IS NOT EXTENDABLE I'S AMENDMENT or NOTICE OF ation is deficient. 948) attached								
(b) including changes required by the proposed drawing c (c) including changes required by the attached Examiner's Identifying indicia such as the application number (see 37 CFR 1. each sheet. Replacement sheet(s) should be labeled as such in t	orrection filed, which has be s Amendment / Comment or in the C .84(c)) should be written on the drawi he margin according to 37 CFR 1.121(een approved by the Examiner. Office action of Paper No ngs in the front (not the back) of d).								
9. DEPOSIT OF and/or INFORMATION about the deposit attached Examiner's comment regarding REQUIREMENT FOR T	sit of BIOLOGICAL MATERIAL r HE DEPOSIT OF BIOLOGICAL MA	nust be submitted. Note the TERIAL.								
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1□ Notice of References Cited (PTO-892) 5□ Notice of Informal Patent Application (PTO-152) 2□ Notice of Draftperson's Patent Drawing Review (PTO-948) 6□ Interview Summary (PTO-413), Paper No 3⊠ Information Disclosure Statements (PTO-1449 or PTO/SB/08), Paper No. 14May02 7⊠ Examiner's Amendment/Comment 4□ Examiner's Comment Regarding Requirement for Deposit of Biological Material 8⊠ Examiner's Statement of Reasons for Allowance										
U.S. Patent and Trademark Office PTOL-37 (Rev. 11-03)	tice of Allowability	Part of Paper No. 1601000								

Application/Control Number: 10/130,214 Art Unit: 1624

EXAMINER'S AMENDMENT

An examiner's amendment to the record appears below. Should the changes and/or additions be unacceptable to applicant, an amendment may be filed as provided by 37 CFR 1.312. To ensure consideration of such an amendment, it MUST be submitted no later than the payment of the issue fee.

Authorization for this examiner's amendment was given in a telephone voicemail message from Christine C. O'Day 18 January 2004, which was in response to a message left by the examiner for Ms. O'Day on 16 January 2004.

IN THE SPECIFICATION -

Insert the heading "**Brief Description of the Drawing**" before the second to last paragraph on page 14 of the specification.

Response to Amendment

As requested in the correspondence from applicants dated 5 November 2003, which is in reply to the Office action of 4 August 2003, claims 3, 5, 6, 8 and 12-15 have been amended, claims 7, 18-25 and 27 have been cancelled and new claims 31-34 added.

Status of Claim Rejections - 35 USC § 112

In the previous Office action, dated 4 August 2003, claims 7, 18-25 and 27-30 were rejected under 35 U.S.C. 112, second paragraph, for indefiniteness.

Claims 7, 18-25 and 27 have been cancelled, mooting the rejection of that claim under this statute.

Claims 28-30 have been amended so as to define proper methods of manufacture under this statute, and therefore the rejection of claims 28-30 under 35 U.S.C. 112, second paragraph, is hereby withdrawn.

Status of Claim Rejections - 35 USC § 101

In the previous Office action, dated 4 August 2003 claims 21, 22 and 27-30 were rejected under 35 U.S.C. 101 for specifying non-statutory subject matter.

Claims 21, 22 and 27 have been cancelled, mooting the rejection of those claims under this statute.

Claims 28-30 have been amended so as to define proper methods of manufacture under this statute, and therefore the rejection of claims 28-30 under 35 U.S.C. 101 is hereby withdrawn.

Status of Obviousness-type Double Patenting

In the previous Office action, dated 4 August 2003, claims 1-7 were provisionally rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over allowed claims 3 and 4 of U.S. application no. 09/700,094.

The rejection is withdrawn in view of the Terminal Disclaimer over the commonlyowned cited application.

Status of Claim Rejections - 35 USC § 102

In the previous Office action, dated 4 August 2003, claims 18 and 23 were rejected as being anticipated by WO 94/11337 (Johansson et al).

Both claims 18 and 23 have been cancelled, mooting the rejection.

Status of Claim Rejections - 35 USC § 103

In the previous Office action, dated 4 August 2003, claims 20 and 25 were

rejected under 35 U.S.C. 103(a) as being unpatentable over Johansson et al.

Both claims 20 and 25 have been cancelled, mooting the rejection.

In the previous Office action, dated 4 August 2003, claims 19 and 24 were

rejected as being unpatentable over WO 98/43242 (Johansson et al '942).

Both claims 19 and 24 have been cancelled, mooting the rejection.

Allowable Subject Matter

Claims 1-6, 8-17, 26 and 28-34 are allowed.

The following is an examiner's statement of reasons for allowance:

Application/Control Number: 10/130,214 Art Unit: 1624

All of the previously stated rejections have been obviated by cancellation of the rejected claim, or by amendment.

An updated search afforded no additional applicable prior art.

New claims 31-34 are patentable under 35 U.S.C. 112, first paragraph, as they comply with the written description and enablement requirement.

If not for the Terminal Disclaimer over 09/700,094, claims 31-34 would be the subject of an obviousness-type double patenting rejection over that application.

Any comments considered necessary by applicant must be submitted no later than the payment of the issue fee and, to avoid processing delays, should preferably accompany the issue fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allowance."

Information Disclosure Statement

A PTO-1449 form, with a received dated of 14 May 2002, is enclosed herewith,

initialed and signed.

to:

Conclusion

All Post-Allowance Correspondence concerning this application must be mailed

Mail Stop Issue Fee Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Or you can fax them to the Office of Patent Publications at 703-308-5083, in order to expedite the handling of such correspondence as amendments under 37 CFR 1.312; information disclosure statements, and formal drawings. Sending Post-Allowance papers to Technology Center 1600 will only cause delays in matching papers with the case.

For information concerning status of correspondence sent after receipt of the Notice of Allowance, please contact the Correspondence Branch at (703) 305-8027.

Application/Control Number: 10/130,214 Art Unit: 1624

The Notice of Allowance also has an insert containing contact information on other items, including Issue Fees, receipt of formal drawings and the status of the application. zt

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Richard L. Raymond

Primary Examiner Art Unit 1624

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Burden Hour Statement: This form is estimated to take 2 hours to complete. Time will vary depending upon the needs of the individual case. Any comments on the amount of time you are required to complete this form should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, Washington, DC 20231. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Assistant Commissioner for Patents, Washington, DC 20231.

1894918



U.S. Patent and Trademark Office

Part of Paper No. 16012004

	Application No.	Applicant(s)	
Issue Classification	10/130,214	MEESE, CLAUS	
	Examiner	Art Unit	
	Zachary C. Tucker	1624	

ISSUE CLASSIFICATION																			
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U.S. Patent and Trademark Office



Application No.	Applicant(s)	
10/130,214	MEESE, CLAUS	
Examiner	Art Unit	
Zachary C. Tucker	1624	

SEARCHED				
Class	Subclass	Date	Examiner	
514	530,531, 534,548, 551	<pre>\$1/16/04</pre>	Zŗ	
560	61,122, 123,124,1 250	38,142,	1/16/64 -Zr	
564	319,	1/16/04	27	
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INTERFERENCE SEARCHED					
Class	Subclass	Date	Examiner		
514	530,531,	2			
	534,548, 551				
560	61, 122, 123, 124, 13	38, (9 27		
564	142' ₁ 250' 319	$\left(\right)$			

U.S. Patent and Trademark Office

SEARCH NOTES (INCLUDING SEARCH STRATEGY)				
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AND "SALTS" <u>AND</u> "ESTERS")			

Part of Paper No. 16012004



SPECIFICATION

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THIS APP-ICATION WAS FILED UNDER 35 U.S.C. 371 AND IS THE U.S. NATIONAL STACE OF PCT/EPO0/11309, FILED 'S NOVEMBER 2000. Stable salts of novel derivatives of

3,3-diphenylpropylamines

The present invention concerns highly pure, crystalline, stable compounds of novel derivatives of 3,3diphenylpropylamines in the form of their salts, a method for manufacturing these and highly pure, stable, intermediate products.

From document PCT/EP99/03212 novel derivatives of 3,3diphenylproprylamines are known.

These are valuable prodrugs for the treatment of urinary incontinence and other spasmodic complaints, which overcome the disadvantage of the active substances available to date, namely inadequate absorption of the active substance by biological membranes or the unfavourable metabolism of these.

Furthermore these novel prodrugs have improved pharmacokinetic characteristics compared with Oxybutynin and Tolterodin.

Preferred compounds from the group of these novel derivatives of 3,3-diphenylpropylamines are esters of aliphatic or aromatic carboxylic acids with the general formula A referred to below

HO

Formula A



-63-1/1

Figure 1

Reaction diagram 1

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(i), (ii), (iii), (iv), (v) stand for: (i), LiAlH₄, (ii),
Raney nickel/H₂, (iii), Me₂CH-CoCl, Et₃N, (iv), fumaric acid,
(v), hydrochloric acids; R stands for isopropyl (iPr)





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TECH CENTER 1600/20	PATENT CENTER 1600/2900

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of:	C. Meese		
Application No.:	10/130,214	GROUP:	1624
Filed:	May 14, 2002	EXAMINER:	Z. Tucker
For:	STABLE SALTS OF NOVEL DERIVA 3.3-DIPHENYLPROPYLAMINES	TIVES OF	

Mail Stop: _____ Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

AMENDMENT TRANSMITTAL

1. Transmitted herewith is an amendment for this application.

STATUS

2. Applicant is

[] a small entity.

[X] other than a small entity.

CERTIFICATE OF MAILING/TRANSMISSION (37 C.F.R. 1.8(a))

I hereby certify that, on the date shown below, this correspondence is being:

MAILING

[X] deposited with the United States Postal Service with sufficient postage as First Class Mail in an envelope addressed to the Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

FACSIMILE

[] transmitted by facsimile to the Patent and Trademark Office.

urkle Signature

Lee Dunkle (type or print name of person certifying)

(Amendment Transmittal-page 1 of 4)

3,3-DIPHENYLPROPYI

Date: ____11/3/03_____

NOTE: "Extension of Time in Patent Cases (Supplement Amendments) — If a timely and complete response has been filed after a Non-Final Office Action, an extension of time is not required to permit filing and/or entry of an additional amendment after expiration of the shortened statutory period.

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If a timely response has been filed after a Final Office Action, an extension of time is required to permit filing and/or entry of a Notice of Appeal or filing and/or entry of an additional amendment after expiration of the shortened statutory period unless the timely-filed response placed the application in condition for allowance. Of course, if a Notice of Appeal has been filed within the shortened statutory period, the period has ceased to run." Notice of December 10, 1985 (1061 O.G. 34-35).

- NOTE: See 37 C.F.R. 1.645 for extensions of time in interference proceedings, and 37 C.F.R. 1.550(c) for extensions of time in reexamination proceedings.
- 3. The proceedings herein are for a patent application and the provisions of 37 C.F.R. 1.136 apply.

(complete (a) or (b), as applicable)

(a) [] Applicant petitions for an extension of time under 37 C.F.R. 1.136 (fees: 37 C.F.R. 1.17(a)(1)-(4)) for the total number of months checked below:

	Extension	Fee for other than	Fee for
	(months)	small entity	small entity
[]	one month	\$110.00	\$55.00
[]	two months	\$420.00	\$210.00
[]	three months	\$950.00	\$475.00
[]	four months	\$1,480.00	\$740.00
[]	five months	\$2,010.00	\$1,005.00

Fee: \$ _____

If an additional extension of time is required, please consider this a petition therefor.

(check and complete the next item, if applicable)

[] An extension for _____ months has already been secured. The fee paid therefor of \$_____ is deducted from the total fee due for the total months of extension now requested.

Extension fee due with this request \$_____

OR

(b) [X] Applicant believes that no extension of term is required. However, this conditional petition is being made to provide for the possibility that applicant has inadvertently overlooked the need for a petition for extension of time.

(Amendment Transmittal-page 2 of 4)

FEE FOR CLAIMS

	(Col.1)	(Col.	. 2) (Col. 3) SM	IALL EN	ΓΙΤΥ	OT SM	HER 7 ALL I	THAN A ENTITY	
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Indep.	*	Minus	***	=	x \$42 =	\$0		x \$84 =	\$0
[] Firs	st Presentati	on of Mul	tiple Depender	nt Claim	+ \$140 =	\$0		+ \$280 =	\$0
			<u>.</u>		Total		OR	Total	
					Addit. Fee	\$		Addit. Fee	\$

4. The fee for claims (37 C.F.R. 1.16(b)-(d)) has been calculated as shown below:

* If the entry in Col. 1 is less than the entry in Col. 2, write "0" in Col. 3,

** If the "Highest No. Previously Paid For" IN THIS SPACE is less than 20, enter "20".

*** If the "Highest No. Previously Paid For" IN THIS SPACE is less than 3, enter "3".

The "Highest No. Previously Paid For" (Total or Indep.) is the highest number found in the appropriate box in Col. 1 of a prior amendment or the number of claims originally filed.

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"After final rejection or action (§ 1.113) amendments may be made canceling claims or complying with any requirement of form which has been made." 37 C.F.R. 1.116(a) (emphasis added).

(complete (c) or (d), as applicable)

(c)	[X]	No additional fee for claims is required.	
		OD	

(d) [] Total additional fee for claims required \$ _____

FEE PAYMENT

 5. [X] Attached is a check in the sum of \$ <u>110.00 (for the terminal disclaimer filed herewith)</u>.
 [] Charge Account No. ______ the sum of \$ ______. A duplicate of this transmittal is attached.

FEE DEFICIENCY

NOTE: If there is a fee deficiency and there is no authorization to charge an account, additional fees are necessary to cover the additional time consumed in making up the original deficiency. If the maximum, six-month period has expired before the deficiency is noted and corrected, the application is held abandoned. In those instances where authorization to charge is included, processing delays are encountered in returning the papers to the PTO Finance Branch in order to apply these charges prior to action on the cases. Authorization to charge the deposit account for any fee deficiency should be checked. See the Notice of April 7, 1986, (1065 O.G. 31-33).

6. [X] If any additional extension and/or fee is required, charge Account No. <u>04-1105.</u>

(Amendment Transmittal-page 3 of 4)

WARNING:

AND/OR

[X] If any additional fee for claims is required, charge Account No. _____04-1105.____

Chine C.n

SIGNATURE OF PRACTITIONER

Christine C. O'Day (type or print name of practitioner)

EDWARDS & ANGELL, LLP P.O. Box 9169 P.O. Address

Customer No. 21874

Tel. No. (617) 439-4444

Reg. No. 38,256

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Boston, Massachusetts 02209

(Amendment Transmittal—page 4 of 4)

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

APPLICANT:	C. Meese	
SERIAL NO.:	10/130,214	ART UNIT: 1624
FILED:	May 14, 2002	EXAMINER: Z. Tucker
FOR:	STABLE SALTS OF NOV DIPHENYLPROPYLAMI	YEL DERIVATIVES OF 3,3- NES

Mail Stop: _ Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

AMENDMENT

Applicants are in receipt of the Office Action dated August 4, 2003. Kindly amend the above-identified application as follows:

Amendments to the claims are reflected in the listing of claims which begins on page 2 of this paper.

Remarks begin on page 14 of this paper.

C. Meese, et al. U.S.S.N. 10/130,214 Page -2-

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of claims:

Claim 1 (original): Compounds of general formula I



Formula I

in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X^- is the acid residue of a physiologically compatible inorganic or organic acid.

Claim 2 (original): Compounds in accordance with claim 1, characterised in that X⁻ in each case is an acid ester of hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L- (-) -malic acid, D- (+) -malic acid, DL-tartaric acid, L- (+) -tartaric acid, D- (-) -tartaric acid, citric acid, L-aspartic acid, L- (+) -ascorbic acid, D- (+) glucuronic acid, 2 - oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4 - hydroxybenzoic acid, salicyclic acid, vanillic acid, 4 - hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3- (4 - hydroxyphenyl) - propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

Claim 3 (currently amended): Compounds in accordance with claims 1 and 2, characterised in that they have general formula 2.

C. Meese, et al. U.S.S.N. 10/130,214 Page -3-



Formula 2

in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid.

Claim 4 (original): Compounds in accordance with claim 3, characterised in that X in each case is an acid ester of hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L- (-)-malic acid, D- (+) -malic acid, DL-tartaric acid, L- (+) tartaric acid, D- (-) -tartaric acid, citric acid, L-aspartic acid, L- (+) -ascorbic acid, D- (+) glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3 - (4 - hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

Claim 5 (currently amended): Compounds in accordance with claims 3-and 4, characterised in that they are R- (+) – 2 - (3 - (diisopropylamino- 1 -phenylpropyl) – 4 – hydroxymethyl – phenylisobutyrate ester hydrogen fumarate, R- (+) – 2 - (3 - (diisopropylamino-1-phenylpropyl) – 4 - hydroxymethylphenylisobutyrate ester-hydrochloride hydrate.

Claim 6 (currently amended): Compounds in accordance with claims 3 and 4, characterised in that R stands for cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, 4 - (1-cyclopropyl-

C. Meese, et al. U.S.S.N. 10/130,214 Page -4-

methanoyloxy)-phenyl, 4 - (1 – cyclobutyl - methanoyloxy) - phenyl, 4 - (1 - cyclohexylmethanoyloxy) -phenyl or 4 - (2, 2 – dimethyl -propanoyloxy) -phenyl and X⁻ denotes chloride.

Claim 7 (cancelled).

Claim 8 (currently amended): Method for manufacturing compounds of general formula I



in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid, characterised in that

a) a compound of formula III



Formula III

is split with a hydrogenation agent to form a compound of Formula V

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7



whereupon

b) the compound of formula V so obtained is converted with a reducing agent, in order to give a compound of formula VI



Formula VI

which

c) is converted with an acylation agent, in order to obtain a compound of formula A

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Formula A

in which R has the significance stated above, which

d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula I



Formula I

in which R denotes C_1 - C_6 -alkyl, C_3 - $Cl\theta C_{10}$ -cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

Claim 9 (original): Method in accordance with claim 8, characterised in that for the manufacture of the compounds of general formula I hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L- (-) - malic acid, D- (+) - malic acid, DL-tartaric acid, L- (+) -tartaric acid, D- (-) -tartaric acid, citric acid, L-aspartic acid,

C. Meese, et al. U.S.S.N. 10/130,214 Page -7-

L- (+) -ascorbic acid, D- (+) -glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (Naectylglycine), phloretinic acid (3- (4 - hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

Claim 10 (original): Method for manufacturing compounds of general formula 2



Formula 2

in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid, characterised in that

a) a compound of the formula 3



s split with a hydrogenation agent to form a compound of formula 5

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C. Meese, et al. U.S.S.N. 10/130,214 Page -8-



Formula 5

whereupon

b) the compound of formula 5 so obtained is converted with a reducing agent, in order to give a compound of formula 6



Formula 6

which

c) is converted with an acylation agent, in order to obtain a compound of formula 1

C. Meese, et al. U.S.S.N. 10/130,214 Page -9-



Formula 1

in which R has the significance stated above, which

d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula 2



in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

Claim 11 (original): Method in accordance with claim 10, characterised in that for the manufacture of the compounds of general formula 2 hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L- (-) -malic acid, D- (+) - malic acid, DL-tartaric acid, L- (+) -tartaric acid, D- (-) -tartaric acid, citric acid, L-aspartic acid,

C. Meese, et al. U.S.S.N. 10/130,214 Page -10-

L- (+) -ascorbic acid, D- (+) -glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (Naectylglycine), phloretinic acid (3- (4 - hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

Claim 12 (currently amended): Method in accordance with claims 8 to 11, characterised in that as the hydrogenation agent, Raney nickel/H₂ in methanol is preferably used as the solvent.

Claim 13 (currently amended): Method in accordance with claims 8 to 11, characterised in that for the reducing agent NaBH₄/EtOH, preferably LiAlH₄/THF, is used.

Claim 14 (currently amended): Method in accordance with claims 8 to 11, characterised in that for the acylation agent isobutyrylchloride and for the base triethylamine are used.

Claim 15 (currently amended): Method in accordance with claims 10 to 14, characterised in that a compound of general formula 6 is converted with an equivalent isobutyryl chloride in the presence of triethylamine using one of the respective solvents ethylacetate, dichloromethane, tetrahydrofurane, acetonitrile or toluene regio- and chemoselectively into R- (+) -2-(3 diisopropylamino-l-phenylpropyl) -4-hydroxymethylphenylisobutyrate.

Claim 16 (currently amended): Method in accordance with claims 10 to 15, characterised in that *R*- (+) -2-(3 - diisopropylamino-1-phenylpropyl) -4-hydroxymethylphenylisobutyrate ester and fumaric acid or hydrochloric acid are converted with the formation of the respective salt.

Claim 17 (currently amended): Method in accordance with claims 10 to 13 for the manufacture of R-(+)-2-(3-diisopropylamino-1-phenyl-propyl)-4-hydroxy-methylphenylisobutyrate ester hydrochloride hydrate, characterised in that the phenolic esterification of R-(+)-2-(3-

C. Meese, et al. U.S.S.N. 10/130,214 Page -11-

diisopropylamino-1-phenylpropyl)-4-hydroxy-methylphenol (6) is carried out without the addition of an external base, in that solutions of (6) are dropped into solutions of isobutyrate isobutyryl_chloride, that contain at least 1 mole equivalent of water, in order to directly obtain a corresponding stable, hydrate-containing hydrochloride.

Claims 18-25 (cancelled).

Claim 26 (original): Compound of formula 7



Claim 27 (cancelled).

Claim 28 (currently amended): <u>A method of Use of a compound in accordance with claims 23 to 26 as an</u> intermediate product in the manufacture of phenolic monoesters of general formula 1 C. Meese, et al. U.S.S.N. 10/130,214 Page -12-



Formula 1

wherein the method comprises the steps of:

providing a compound of claim 26;

deprotecting the hydroxyl residues of the 4-hydroxy-benzyl alcohol residue; and

acylating the phenol residue.

Claim 29 (currently amended): <u>A method of Use of a compound in accordance with claims 23 to 26 as</u> an intermediate product in the manufacture of salts of phenolic monoesters of general formula 2:



Formula 2

in which R has the same meaning as given in claim 3 denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X is the acid residue of a physiologically compatible inorganic or organic acid, wherein the method comprises the steps of:
C. Meese, et al. U.S.S.N. 10/130,214 Page -13-

> providing a compound of claim 26; deprotecting the hydroxyl residues of the 4-hydroxy-benzyl alcohol residue; and acylating the phenol residue.

Claim 30 (currently amended): A method of Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of R- (+) -2- (3-diisopropylamino-l-phenylpropyl) -4hydroxymethylphenylisobutyrate ester hydrogen fumarate and or R- (+) -2- (3diisopropylamino-l-phenylpropyl) -4- hydroxymethylphenylisobutyrate ester hydrochloride hydrate, the method comprising the steps of:

providing a compound of claim 26;

deprotecting the hydroxyl residues of the 4-hydroxy-benzyl alcohol residue; and acylating the phenol residue.

Claim 31 (new): A method of treating a patient suffering from urinary incontinence, which method comprises the step of administering to said patient an effective amount of a compound according to claim 1.

Claim 32 (new): A method of treating a patient suffering from urinary incontinence, which method comprises the step of administering to said patient an effective amount of a compound according to claim 3.

Claim 33 (new): A method of treating a patient suffering from urinary incontinence, which method comprises the step of administering to said patient an effective amount of a compound according to claim 5.

Claim 34 (new): The method of any one of claims 31-33, wherein the urinary incontinence disorder is urge incontinence.

C. Meese, et al. U.S.S.N. 10/130,214 Page -14-

REMARKS

Claims 3, 5, 6, 8, 12-17, and 28-30 have been amended; claims 7, 18-25, and 27 stand cancelled; and new claims 31-34 are added. No new matter has been added by virtue of the within amendment. Support for the amended and newly presented claims appears throughout the specification and in the original claims of the application.

As an initial matter, Applicants appreciate the indication of allowable subject matter, i.e., that claims 8-11 and 26 stand allowed, and that claims 12-17 would be allowable if amended to correct improper multiple dependencies and other minor informalities.

Referring now to the Office Action, claims 7, 18-25, and 27-30 stand rejected under 35 USC §112, 2nd paragraph.

Applicants believe that the subject matter of the noted claims is indeed clear and definite. However, in an effort to expedite allowance of the application, each of the noted claims has been cancelled or amended to further define the present invention. For instance, claims 28-30 have been amended to provide method of preparation claims. Withdrawal of the rejection is therefore proper.

Claims 21, 22 and 27-30 stand rejected under 35 USC §101.

Again, in an effort to expedite allowance of the application, claims 21, 22, and 27 have been cancelled and claims 28-30 have been amended to provide method of preparation claims. Withdrawal of the rejection is therefore proper.

Claims 1-7 stand rejected under the judicially created doctrine of obviousness-type double patenting over copending application USSN 09/700,094. Applicants confirm that the cited application and the present application are commonly owned. Additionally, in order to

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obviate the rejection, Applicants enclose herewith a terminal disclaimer. Withdrawal of the rejection is therefore proper.

The remaining rejections relate to the prior art and are summarized as follows.

Claims 18 and 23 stand rejected under 35 USC §102(b) over WO 94/11337 (Johansson et al.).

Claims 20 and 25 stand rejected under 35 USC §103(a) over WO 94/11337 (Johansson et al.).

Claims 19 and 24 stand rejected under 35 USC §103(a) over WO 98/43942 (Johansson et al. '942).

The rejections are traversed. Applicants submit that the cited references do not teach or suggest the present invention in a manner sufficient to sustain the instant rejection. For example, see *In re Marshall*, 198 USPQ 344, 346 (CCPA 1978) ("[r]ejections under 35 U.S.C. §102 are proper only when the claimed subject matter is identically disclosed or described in the prior art.") Additionally, it is well-known that to establish a *prima facie* case of obviousness, three basic criteria must be met: (1) there must be some suggestion or motivation, either in the references themselves or in the knowledge generally available to one of ordinary skill in the art, to modify the reference or to combine reference teachings; (2) there must be a reasonable expectation of success; and (3) the prior art reference(s) must teach or suggest all the claim limitations. The teaching or suggestion to make the claimed combination and the reasonable expectation of success must both be found in the prior art, and not based on applicant's disclosure. *In re Vaeck*, 947 F.2d 488, 20 USPQ2d 1438 (Fed. Cir. 1991). See MPEP § 2143.

Nonetheless, in an effort to expedite allowance of the application, the rejected claims

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have been cancelled. Withdrawal of the rejection is therefore proper.

It also is noted that claims 3 and 5-6 have been amended to remove improper multiple dependencies. Withdrawal of the objections related to those claims is requested.

Lastly, claim 8 has been amended merely to correct a typographical error.

It is believed the application is in condition for immediate allowance, which action is earnestly solicited.

Respectfully submitted,

Christ C. h

Christine C. O'Day (Reg. No.: 38,256) John B. Alexander, Ph.D. (Reg. No. 48,399) EDWARDS & ANGELL, LLP P.O. Box 9169 Boston, MA 02209 Tel. (617) 439-4444



PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of: Application No.: Filed: For: C. Meese 10/130,214 GROUP: 1624 May 14, 2002 EXAMINER: Z. Tucker STABLE SALTS OF NOVEL DERIVATIVES OF 3,3-DIPHENYLPROPYLAMINES

[] *Patent No.:

Issue Date: Reexamination Date:

*NOTE: Preferably also insert inventor's name and invention title.

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

TERMINAL DISCLAIMER TO OBVIATE A DOUBLE PATENTING REJECTION (37 C.F.R. SECTION 1.321(c))

Identification of Person(s) Making This Disclaimer

I, Christine C. O'Day

CERTIFICATE OF MAILING/TRANSMISSION (37 C.F.R. SECTION 1.8(a))

[]

I hereby certify that, on the date shown below, this correspondence is being:

MAILING

[X] deposited with the United States Postal Service with sufficient postage as first class mail in an envelope addressed to the Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

11/06/2003 SDENBOB1 00000030 10130214

01 FC:1814

110.00 OP

Date: <u>November 3, 2003</u>

Signature

Trademark Office, (703) ____-

FACSIMILE

transmitted by facsimile to the Patent and

(type or print name of person certifying)

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 1 of 7)

(type or print names of all inventors or assigns or name of attorney signing disclaimer)

- (a) represent that I am
 - [] an inventor (applicant) of this invention.
 - [] an assignee of this invention.

WARNING: "If the patent or patent application is assigned to an organization, such as a corporation, partnership, university, [g] overnment agency or similar entity, and the disclaimer is signed by the assignee, the assignee must comply with Section 3.73(b)." Notice of Oct. 15, 1993, 1156 O.G. 54-61 at 56, Section 1490, M.P.E.P., 7th Edition.

- [] a representative authorized to sign on behalf of the assignee identified below.
- [] A statement under 37 C.F.R. Section 3.73(b) is attached.
- **WARNING:** See the above "WARNING".
 - [X] the attorney of record for this invention.
- NOTE: The rules "permit an attorney or agent of record to sign a terminal disclaimer without the need to comply with Section 3.73(b)." Notice of Oct. 15, 1993, 1156 O.G. 54-61, at 56. See also Section 1490, M.P.E.P., 7th Edition.

IDENTITY OF ASSIGNEE AND TITLE OF DISCLAIMANT (*if applicable*)

The assignee is

Name of assignee _____ Schwarz Pharma AG

Address of assignee ______ Alfred-Nobel-Strasse 10, 40789 Monheim, Germany

Title of disclaimant authorized to sign on behalf of assignee _____n/a____

EXTENT OF DISCLAIMANT'S INTEREST

The extent of the interest in this invention that the disclaimant owns is in:

- [X] the whole of this invention.
- [] a sectional interest in this invention, as follows:

(state the exact interest of the disclaimant)

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 2 of 7)

RECORDAL OF ASSIGNMENT IN PTO (if applicable)

[X] The assignment was recorded on: May 14, 2002

Reel 013122 Frame 0883

[] Authorization for recordal of the assignment is separately attached.

[] A separate [] "ASSIGNMENT (DOCUMENT) COVER SHEET" or [] FORM PTO 1595 is also attached.

ESTABLISHING RIGHT OF ASSIGNEE TO TAKE ACTION (if applicable)

[] Attached is a STATEMENT UNDER 37 C.F.R. Section 3.73(b) establishing the right of the assignee to take action in this case.

DISCLAIMER

(select one of the following)

(Provisional Obviousness-Type Double Patenting Rejection Over A Pending Application)

Petitioner hereby disclaims, except as provided below, the terminal part of any patent granted on the instant application, which would extend beyond the expiration date of any patent granted on Application No. 09/700,094, filed on January 2, 2001, as shortened by any terminal disclaimer. Petitioner hereby agrees that any patent so granted on the instant application shall be enforceable only for and during such period that it and any patent granted on the above-listed application are commonly owned. This agreement runs with any patent granted on the instant application and is binding upon the grantee, its successors, or assigns.

In making the above disclaimer, disclaimant does not disclaim the terminal part of any patent granted on the instant application that would extend to the expiration date of the full statutory term as defined in 35 U.S.C. Sections 154 to 156 and 173 of any patent granted on the application forming the basis of the double patenting rejection, namely, any patent granted on Application No.: 09/700,094, in the event that it later: expires for failure to pay a maintenance fee, is held unenforceable, is found invalid by a court of

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 3 of 7)

competent jurisdiction, is statutorily disclaimed in whole or terminally disclaimed under 37 C.F.R. Section 1.321, has all claims cancelled by a reexamination certificate, is reissued, or is in any manner terminated prior to expiration of its full statutory term as presently shortened by any terminal disclaimer, except for the separation of legal title stated above.

DISCLAIMER FEE (37 C.F.R. Section 1.20(d))

[X] Other than a small entity--fee \$110.00

[] Small entity--fee \$55.00

- [] Small entity statement attached
- [] Small entity statement already filed
 - [] in patent application __/ ____ on

(date)

OR

(Obviousness-Type Double Patenting Rejection Over A Prior Patent)

Petitioner hereby disclaims, except as provided below, the terminal part of any patent granted on the instant application, which would extend beyond the expiration date of Patent No. ______ as presently shortened by any terminal disclaimer. Petitioner hereby agrees that any patent so granted on the instant application shall be enforceable only for and during such period that it and the above-listed patent are commonly owned. This agreement runs with any patent granted on the instant application and is binding upon the grantee, its successors, or assigns.

In making the above disclaimer, disclaimant does not disclaim the terminal part of any patent granted on the instant application that would extend to the expiration date of the full statutory term as defined in 35 U.S.C. Sections 154 to 156 and 173 of the patent forming the basis of the double patenting rejection, namely, Patent No.: ______, as presently shortened by any terminal disclaimer, in the event that it later: expires for failure to pay a maintenance fee, is held unenforceable, is found invalid by a court of competent jurisdiction, is statutorily disclaimed in whole or terminally disclaimed under 37 C.F.R. Section 1.321, has all claims cancelled by a reexamination certificate, is reissued, or is in any manner terminated prior to expiration of its full statutory term as presently shortened by any terminal disclaimer, except for the separation of legal title stated above.

DISCLAIMER FEE (37 C.F.R. Section 1.20(d))

- [] Other than a small entity--fee \$110.00
- [] Small entity--fee \$55.00
 - [] Small entity statement attached
 - [] Small entity statement already filed

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 4 of 7)

[] in patent application __/____ on (date)

OR

(Provisional Obviousness-Type Double Patenting Rejection Over A Pending Application-Reexamination Proceeding)

Petitioner hereby disclaims, except as provided below, the terminal part of any patent being reexamined, which would extend beyond the expiration date of any patent granted on Application No. _______, filed on _______, as shortened by any terminal disclaimer. Petitioner hereby agrees that any reexamination certificate issued on the instant patent being reexamined shall be enforceable only for and during such period that it and any patent granted on the a bove-listed application are commonly owned. This agreement runs with any reexamination certificate issued on the instant patent granted and is binding upon the grantee, its successors, or assigns.

DISCLAIMER FEE (37 C.F.R. Section 1.20(d))

[] Other than a small entity--fee \$110.00

[] Small entity--fee \$55.00

[] Small entity statement attached

[] Small entity statement already filed

[] in patent application __/ ____

(date)

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 5 of 7)

__ on

(Provisional Obviousness-Type Double Patenting Rejection Over A Prior Patent---Reexamination Proceeding)

Petitioner hereby disclaims, except as provided below, the terminal part of the patent being reexamined, which would extend beyond the expiration date of Patent No. _______ as presently shortened by any terminal disclaimer. Petitioner hereby agrees that the patent for which a reexamination certificate is issued as a result of this proceeding shall be enforceable only for and during such period that it and the above listed patent granted are commonly owned. This agreement runs with any reexamination certificate issued on the instant patent and is binding upon the grantee, its successors, or assigns.

In making the above disclaimer, disclaimant does not disclaim the terminal part of any reexamination certificate granted on the instant patent that would extend to the expiration date of the full statutory term as defined in 35 U.S.C. Sections 154 to 156 and 173 of the patent forming the basis of the double patenting rejection, namely, Patent No.: ______, as presently shortened by any terminal disclaimer, in the event that it later: expires for failure to pay a maintenance fee, is held unenforceable, is found invalid by a court of competent jurisdiction, is statutorily disclaimed in whole or terminally disclaimed under 37 C.F.R. Section 1.321, has all claims cancelled by a reexamination certificate, is reissued, or is in any manner terminated prior to expiration of its full statutory term as presently shortened by any terminal disclaimer, except for the separation of legal title stated above.

DISCLAIMER FEE (37 C.F.R. Section 1.20(d))

- [] Other than a small entity--fee \$110.00
- [] Small entity--fee \$55.00

[]

- [] Small entity statement attached
 - Small entity statement already filed
 [] in patent application ___/____ on

(date)

FEE PAYMENT

- [X] Attached is a check in the sum of \$ 110.00.
 - [X] Charge Account 04-1105 for any fee deficiency.
- [] Charge Deposit Account ______ the sum of \$_____. A duplicate of this disclaimer is attached.

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 6 of 7)

Signature of disclaimant

Date: //- 3 - 0 3

Reg. No.: 38,256

China C.M

SIGNATURE OF PRACTITIONER

Christine C. O'Day (Reg. No.: 38,256) EDWARDS & ANGELL, LLP P.O. Box 9169 Boston, MA 02209 Tel. (617) 439-4444

Customer No.: 21874

(Terminal Disclaimer to Obviate a Double Patenting Rejection--page 7 of 7)



Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)
	10/130,214	MEESE, CLAUS
Office Action Summary	Examiner	Art Unit
	Zachary C. Tucker	1624
The MAILING DATE of this communication a	appears on the cover sheet	with the correspondence address
 THE MAILING DATE OF THIS COMMUNICATION Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication. If the period for reply specified above is less than thirty (30) days, a If NO period for reply is specified above, the maximum statutory per Failure to reply within the set or extended period for reply will, by sta Any reply received by the Office later than three months after the ma earned patent term adjustment. See 37 CFR 1.704(b). 	N. R 1.136(a). In no event, however, may reply within the statutory minimum of th iod will apply and will expire SIX (6) Mo atute, cause the application to become ailing date of this communication, even	a reply be timely filed hirty (30) days will be considered timely. ONTHS from the mailing date of this communication. ABANDONED (35 U.S.C. § 133). if timely filed, may reduce any
1) Responsive to communication(s) filed on		
2a) This action is FINAL . 2b)⊠	This action is non-final.	
3) Since this application is in condition for allo closed in accordance with the practice und Disposition of Claims	owance except for formal m ler <i>Ex parte Quayl</i> e, 1935 (atters, prosecution as to the merits is C.D. 11, 453 O.G. 213.
4) Claim(s) <u>1-30</u> is/are pending in the applicat	tion.	
4a) Of the above claim(s) is/are witho	Irawn from consideration.	
5)⊠ Claim(s) <u>8-11 and 26</u> is/are allowed.		
6) Claim(s) <u>1-7, 18-25 and 27-30</u> is/are rejecte	ed.	
7)⊠ Claim(s) <u>12-17</u> is/are objected to.		
8) Claim(s) are subject to restriction and	d/or election requirement.	
Application Papers		
9) The specification is objected to by the Exam	iner.	
10) The drawing(s) filed on is/are: a) □ ac	ccepted or b) objected to by	the Examiner.
Applicant may not request that any objection to	the drawing(s) be held in abe	eyance. See 37 CFR 1.85(a).
11) The proposed drawing correction filed on	is: a) approved b)	disapproved by the Examiner.
If approved, corrected drawings are required in	reply to this Office action.	
12) The oath or declaration is objected to by the	Examiner.	· · ·
Priority under 35 U.S.C. §§ 119 and 120		
13) Acknowledgment is made of a claim for fore	eign priority under 35 U.S.C	:. § 119(a)-(d) or (f).
a)⊠ All b)[_ Some * c)[_ None of:		
1. Certified copies of the priority docume	ents have been received.	
2. Certified copies of the priority docume	ents have been received in	Application No
3. Copies of the certified copies of the p application from the International * See the attached detailed Office action for a	riority documents have bee Bureau (PCT Rule 17.2(a)) list of the certified copies no	n received in this National Stage). ot received
14) Acknowledgment is made of a claim for dome	estic priority under 35 U.S.C	C. § 119(e) (to a provisional application).
a) The translation of the foreign language 15) Acknowledgment is made of a claim for dom	provisional application has estic priority under 35 U.S.	been received. C. §§ 120 and/or 121.
Attachment(s)	• •	
 1) X Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) X Information Disclosure Statement(s) (PTO-1449) Paper No(statement) 	4) ☐ Intervie 5) ☐ Notice o s) <u>6 and 7</u> . 6) ☐ Other:	w Summary (PTO-413) Paper No(s) of Informal Patent Application (PTO-152)
U.S. Patent and Trademark Office PTO-326 (Rev. 04-01) Petitin	Action Summary oner Mylan Pharmace	Part of Paper No. 10 uticals Inc Exhibit 1002 - Page 302

DETAILED ACTION

Claim Rejections - 35 USC § 112

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

Claims 7, 18-25 and 27-30 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

The term "bulk material" in claim 7 is indefinite. Though clearly, a compound specified in one of claims 1 to 6 would be a "bulk material" if it were present in an amount on the order of 1 metric tonne, it is not clear where the delineation between "bulk" and "not bulk" lies. Perhaps most importantly, the patentability of a chemical compound cannot be based on the amount of that compound, thus claim does not provide for any patentable distinction over the claims from which it depends. Claim 7 is not viewed as being further limiting.

The term "highly pure" in claims 18-20 and 23-25 is a relative term which renders the claim indefinite. The term "highly pure" is not defined by the claim, the specification does not provide a standard for ascertaining the requisite degree, and one of ordinary skill in the art would not be reasonably apprised of the scope of the invention. The characterization of a compound as being "highly pure" may signify different purities to different persons of ordinary skill in the art in different settings, such as to the analytical chemist, forensic chemist, medicinal chemist or pharmacist.

Page 2

Claims in which "highly pure" is recited as a limitation have been examined from the point of view that if there is a disclosure of the specified compound in the absence of any characterization explicitly stating said compound is impure, the limitation is met.

Claim 21, 22 and 27-30 provide for the use of the specified chemical compounds, but since the claims do not set forth any steps involved in the method/process, it is unclear what method/process applicant is intending to encompass. A claim is indefinite where it merely recites a use without any active, positive steps delimiting how this use is actually practiced.

Claim Rejections - 35 USC § 101

35 U.S.C. 101 reads as follows:

Whoever invents or discovers any new and useful process, machine, manufacture, or composition of matter, or any new and useful improvement thereof, may obtain a patent therefor, subject to the conditions and requirements of this title.

Claims 21, 22 and 27-30 are rejected under 35 U.S.C. 101 because the claimed invention is directed to non-statutory subject matter.

The claimed recitation of a use, without setting forth any steps involved in the

process, results in an improper definition of a process, i.e., results in a claim which is

not a proper process claim under 35 U.S.C. 101. See for example Ex parte Dunki, 153

USPQ 678 (Bd.App. 1967) and Clinical Products, Ltd. v. Brenner, 255 F. Supp. 131,

149 USPQ 475 (D.D.C. 1966).

Claims 21, 22 and 27-30 have not been further examined on the merits.

Obviousness-type Double Patenting

The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent

and to prevent possible harassment by multiple assignees. See *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970);and, *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent is shown to be commonly owned with this application. See 37 CFR 1.130(b).

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

Claims 1-7 are provisionally rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over allowed claims 3 and 4

of copending Application No. 09/700,094 in view of the specification of 09/700,094.

Though claims 3 and 4 of U.S. serial no. 09/700,094 do not recite salts of the

compounds specified in those claims, the specification of U.S. serial no. 09/700,094

makes it clear that salts of the compounds of claims 3 and 4 of that application are

within the scope of the invention disclosed in that application. Example "bb"

demonstrates hydrochloride salt formation of several species, while the specification

(page no. 6 in the PCT publication on which 09/700,094 is based - WO 99/58478)

teaches that the compounds disclosed in that application comprise salts thereof with

physiologically acceptable organic and inorganic acids. Page no. 35 in the PCT

publication again contemplates salts of the compounds disclosed therein with

"physiologically acceptable acids." One of ordinary skill in the art, given the disclosure

of U.S. serial no. 09/700,094, and claims 3 and 4 of U.S. serial no. 09/700,094 would at

once envisage the compounds of claims 3 and 4 of U.S. serial no. 09/700,094 in the

form of salts commonly employed in formulating pharmaceutical dosage forms of drugs

Page 4

hydrochloride, hydrobromide, sulfate, phosphate, tartrate, fumarate, maleate, citrate
 and aspartate.

This is a provisional obviousness-type double patenting rejection.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 18 and 23 are rejected under 35 U.S.C. 102(b) as being anticipated by

WO 94/11337 (Johansson et al).

Compounds of instant claims 18 and 23 are disclosed on page 12, lines 15-28.

The compound was recovered in the form of white crystals after recrystallization from

diisopropyl ether. The identity of the compound was verified by N.M.R. and melting

point. The compound is stable, as no decomposition was reported.

The compound in the aforementioned Example is pure (+) isomer, which in this

case has the (R) configuration (tolterodine is the (+) isomer and has the (R)

configuration). The (-) isomer is crystallized in lines 29-34 of page 12.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all

obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 20 and 25 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 94/11337 (Johansson et al).

A compound having the molecular structure depicted in claim 20 or 25, in pure, crystalline form would have been obvious to one of ordinary skill in the art at the time the invention was made.

Compounds having the molecular structures depicted in instant claims 20 and 25 are known from the prior art. The compounds are stable, that is, they do not spontaneously decompose. Compounds having the molecular structures depicted in instant claims 20 and 25 have an established utility as antimuscarinic drugs in treatment of, for example, urinary incontinence.

Page 13, lines 16-30 discloses synthesis of a compound having the molecular structure depicted in claims 20 and 25 from the 2-benzyloxy substituted precursor. The compound is the (+) isomer, which corresponds to the (R) configuration for the particular compound (tolterodine is the (+) isomer and has the (R) configuration). The corresponding (-) isomer is synthesized in a similar manner on page 14, lines 6-11.

The Johansson et al publication teaches that free bases of the 3,3diphenylproprylamines disclosed therein are suitable for pharmaceutical application. Page 6, lines 36-38 – page 7, lines 1-3 clearly state that free bases of any of the compounds of formula (I) disclosed in that publication are contemplated for pharmaceutical application. Thus, though the compound on page 13, lines 16-30 is converted to the mandelic acid salt, Johansson et al teaches that the free base, the form in which the compound was present initially, is suited for incorporation into

Page 6

compositions for oral use, injection or nasal spray, and may be combined with materials so that it may be delivered enterally, percutaneously or parenterally.

A pharmacist or medicinal chemist of ordinary skill in the art, knows that free amines are more lipophilic than their corresponding salts, and therefore are suited for incorporation into dosage forms requiring a drug with lipophilic character such as a patch for percutaneous administration of the drug. Therefore, a crystallized form of the free base of a compound having the molecular structure depicted in instant claims 20 and 25 would be obvious to make. Doing so would be well within the skill of the average practitioner, and Johannson et al demonstrates crystallization of other free bases in the examples of WO 94/11337 (see above in *Claim Rejections – 35 USC* § 102). A crystalline form of the free base would be necessary for conveniently weighing and compounding the substance into a pharmaceutically elegant dosage form, such as a dosage form for percutaneous administration of the substance.

Claims 19 and 24 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 98/43242 (Johansson et al '942).

A compound having the molecular structure depicted in claim 19 and 24, in pure, crystalline form would have been obvious to one of ordinary skill in the art at the time the invention was made.

Johansson et al '942 discloses compounds having the molecular structures depicted in claims 19 and 24 on page 26 in Example 9 (the (R) isomer). The hydrochloride of this compound is prepared, however. Therefore the deficiency of

Johansson et al '942 with respect to claims 19 and 24 is that no example is present in Johansson et al '942 demonstrating crystallization of a free base of a compound having the molecular structure depicted in claims 19 and 24.

Johansson et al '942 expressly suggests that free bases, as well as the salts of compounds of formula I in that publication, are effective treatments for a disorder relating to urinary incontinence and may be incorporated into a variety of different types of dosage forms, including dosage forms for percutaneous administration of the compound (page 3, lines 16-21 and page 13, lines 6-25).

Page 7, lines 4 and 5, disclose N,N-diisopropyl-3-(2-hydroxy-5methyloxycarbonylphenyl)-3-phenylpropanamine, and its (R) isomer. This compound is not a salt. The compound is apparently stable, as Johannson et al does not describe any difficulties in its synthesis (Example 9, referred to above), and teaches that the compound is preferred and suitable for incorporation into pharmaceutical compositions.

A pharmacist or medicinal chemist of ordinary skill in the art, knows that free amines are more lipophilic than their corresponding salts, and therefore are suited for incorporation into dosage forms requiring a drug with lipophilic character such as a patch for percutaneous administration of the drug. Therefore, a crystallized form of the free base of a compound having the molecular structure depicted in instant claims 19 and 24 would be obvious to make.

A crystalline form of the free base would be necessary for conveniently weighing and compounding the substance into a pharmaceutically elegant dosage form, such as a dosage form for percutaneous administration of the substance, but equally necessary

for any dosage form in which a free base of the active substance was to be compounded.

Cited of Interest

Pertinent to the compounds of claims 20 and 25 is Nilvebrant et al, "Antimuscarinic Potency and Bladder Selectivity of PNU-200577, a Major Metabolite of Tolterodine" Pharmacology and Toxicology. vol. 81, pages 169-172 (1997).

The Nilvebrant et al reference demonstrates the antimuscarinic potency of "PNU-200557," which has the molecular structure depicted in claims 20 and 25. The compound is employed as the mandelate salt in the experiments of that reference (page 170 – "Drugs and Chemicals").

Claim Objections/Allowable Subject Matter

Claims 3, 5-7, 21, 22 and 27-30, in addition to those claims being rejected for reasons set forth *supra*, are objected to for improper multiple dependency. A multiply dependent claim cannot simultaneously depend from more than one claim at a time. Claims 3, 5-7 have been examined on the merits in this Office action as though they were in proper multiply dependent form.

Claims 8-11 are allowed. The only disclosure of the compounds of formulae I or 2, as specified in claims 8 and 10, respectively, is in allowed U.S. patent application 09/700,094, over which the compounds of claims 1-7 were rejected for Obviousness-type Double Patenting, in WO 99/58478, which is the PCT application on which the U.S. application is based, and in the German patent application DE 98108608.5, which is the basis for WO 99/58478. The compounds of formulae I and 2 are made by a different

process in WO 99/58478 and US serial no. 09/700,094. In these disclosures, the methyl ester group is reduced first, followed by cleavage of the benzyloxy group. There is no suggestion to reverse these steps in WO 99/58478 and U.S. serial no. 09/700,094.

Claims 12-17 are objected to for being improper multiple dependent claims. A multiply dependent claim cannot simultaneously depend from more than one claim at a time. Claims 12-17 have been examined on the merits in this Office action as though they were in proper multiply dependent form. Claim 17 erroneously refers to isobutyryl chloride as "isobutyrate chloride." Isobutyryl chloride is recited in claim 15. Correction would be appreciated.

If amended so as to place them in proper form, claims 12-17 would be allowable.

Claim 26 is allowed. There is no disclosure of the compound of claim 26 in the

prior art.

Conclusion

Any inquiry concerning this communication should be directed to Zachary Tucker whose telephone number is (703) 305-2050. The examiner can normally be reached Monday-Friday from 7:00am to 3:30pm. If Attempts to reach the examiner are unsuccessful, the examiner's supervisor, Mukund Shah, can be reached at (703) 308-4716. The fax number for the organization where this application or proceeding is assigned is (703) 308-4556 for regular communications and (703) 308-4242 for after-final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703) 308-1235.

zt Zach Tu-

Mukund Shah Supervisory Patent Examiner Art Unit 1624

Notice of References Cited	Application/Control No. 10/130,214	Applicant(s)/Pa Reexamination MEESE, CLAU	atent Under 1 JS
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U.S. PATENT DOCUMENTS

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NON-PATENT DOCUMENTS

*		Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)
	U	Nilvebrant et al, "Antimuscarinic Potency and Bladder Selectivity of PNU-200577, a Major Metabolite of Tolterodine" Pharmacology and Toxicology. vol. 81, pages 169-172 (1997).
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*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

Notice of References Cited

Part of Paper No. 10



PTO/-A820 (08-00)

Approved for use through 10/31/2002. OMB 0651-0031 U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number. Information Disclosure Statement - PTO 1449 (Modified) Sheet 1 of 1

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This file contains CAS Registry Numbers for easy and accurate substance identification.

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ANSWER 1 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN L9 ACCESSION NUMBER: 2003:335062 CAPLUS DOCUMENT NUMBER: 138:353732 TITLE: Quarternary ammonium compounds and their use as antimuscarinic agents INVENTOR(S): Richards, Ivan; Cammarata, Sue K.; Wegner, Craig D.; Hawley, Michael; Warchol, Mark P.; Kontny, Mark; Morozowich, Walter; Kolbasa, Karen P.; Moon, Malcolm W.; Bonafoux, Dominique; Wolfson, Sergey G.; Lennon, Patrick J. PATENT ASSIGNEE(S): Pharmacia & Upjohn Company, USA PCT Int. Appl., 69 pp. SOURCE: CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE --------------WO 2002-US34529 20021025 WO 2003035599 A1 20030501 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, ΤM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,

Petitioner Mylan Pharmaceuticals Inc. - Exhibit 1002 - Page 318

NE, SN, TD, TG PRIORITY APPLN. INFO.:

US	2001-348930P	Р	20011026
US	2002-361979P	Р	20020306
US	2002-391521P	Ρ	20020625

OTHER SOURCE(S): MARPAT 138:353732 GI



AB Novel quaternary ammonium compds. I [R1-R3 = (un)substituted alkyl; NR1R2, NR2R3, NR1R3 = heterocyclic; R4 = H, Me,acyl, alkoxycarbonyl, (un)substituted NH2; R5-R7 = H, OMe, OH, CONH2, SO2NH2, F, Cl, Br, I, CF3, (un)substituted alkyl; X = anion of a pharmaceutically acceptable acid] were prepd. for use as antimuscarinic agents. Thus, tolterodine tartrate was converted to the free base and quaternized with MeI to give (R)-5,2-Me(OH)C6H3CHPhCH2CH2N+(CHMe2)2Me I- which has high affinity, but little selectivity for M1-M5 muscarinic receptors.

IT 518360-93-5P
RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological
study); PREP (Preparation); USES (Uses)

(prepn.of diarylpropylammonium salts as antimuscarinic agents)

RN 518360-93-5 CAPLUS

CN Benzenepropanaminium, 5-(hydroxymethyl)-N-methyl-N,N-bis(1-methylethyl)-2-(2-methyl-1-oxopropoxy)-.gamma.-phenyl-, bromide, (.gamma.R)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.



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REFERENCE COUNT:

THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT





CAPLUS COPYRIGHT 2003 ACS on STN Ь9 ANSWER 2 OF 14 ACCESSION NUMBER: 2002:818306 CAPLUS DOCUMENT NUMBER: 138:407019 TITLE: Sonic spray ionization interface for liquid chromatography-mass spectrometry AUTHOR(S): Bjorkman, Helena T.; Edlund, Per-Olof; Jacobsson, Sven Ρ. CORPORATE SOURCE: Department of Analytical Chemistry, Stockholm University, Stockholm, SE-10691, Swed. SOURCE : Analytica Chimica Acta (2002), 468(2), 263-274 CODEN: ACACAM; ISSN: 0003-2670 PUBLISHER: Elsevier Science B.V. DOCUMENT TYPE: Journal LANGUAGE: English AR A sonic spray ionization (SSI) interface for liq. chromatog.-mass spectrometry (LC-MS) anal. was optimized for anal. of 2-[(1R)-3-[bis(1methylethyl)amino]-1-phenylpropyl]-4-methyl-phenol (tolterodine), used as a model drug substance, and the influence of different parameter settings was evaluated using factorial design. A comparison between SSI and electrospray ionization (ESI) was made for tolterodine, tolterodine metabolites, and a set of steroids. SSI was found to give slightly poorer repeatability and broader peaks for tolterodine compared to ESI. However, there was no significant difference in chromatog. peak shape, and the repeatability using SSI was similar to that obtained using ESI if a ratio (area of tolterodine/area of metabolite) was used. In this study, the sensitivity was higher using SSI. For the anal. of pregnanolone, less water loss was obtained using SSI, probably due to less energy being transferred to the analyte upon ionization. IT 194482-44-5 RL: ANT (Analyte); ANST (Analytical study) (sonic spray ionization interface for lig. chromatog.-mass spectrometry) RN 194482-44-5 CAPLUS CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME) Absolute stereochemistry. Rotation (-). OH Ph N(Pr-i)2 CO₂H **REFERENCE COUNT:** 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L9 ANSWER 3 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2002:51413 CAPLUS DOCUMENT NUMBER: 136:102178 TITLE: Preparation of 3,3-diarylpropylamines via hydroformylation-amination of diarylethenes in presence of a transition metal catalyst INVENTOR(S): Donsbach, Martin; Eilbracht, Peter; Buss, Christian; Schmidt, Andreas PATENT ASSIGNEE(S): Schwarz Pharma A.-G., Germany SOURCE : PCT Int. Appl., 72 pp. CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE : German FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

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AB	The invention relates to a novel method for producing 3,3- diarylpropylamines I [Ar = (un)substituted aryl; X = (un)substituted OH; Y = Cl, Br, I, CN, CH2OR, CHO, CO2H, CO2R; R = alkyl, aryl; R1, R2 = alkyl, cycloalkyl; NR1R2 = heterocyclic] by hydroformylation/hydrocarbonylation and subsequent reductive amination using a transition metal catalyst. Thus, 5,2-Me(HO)C6H3COPh was methylated and methylenated with MeP+Ph3 Br- to give 5,2-Me(MeO)C6H3CPh:CH2 which was treated with (Me2CH)2NH, CO, and H in presence of Ph(acac)(CO)2 and Bu3P to give 85% 5.2-
	Me (MeO) C6H3CHPhCH2CH2N (CHMe2) 2.
IT	286930-05-0P 389068-25-1P
	RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP (Preparation)
	(prepn. of 3,3-diarylpropylamines via hydroformylation-amination of diarylethenes in presence of a transition metal catalyst)
RN	286930-05-0 CAPLUS

286930-05-0 CAPLUS CNBenzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(phenylmethoxy) -, methyl ester (9CI) (CA INDEX NAME)



RN 389068-25-1 CAPLUS CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(phenylmethoxy)-, ethyl ester (9CI) (CA INDEX NAME)

Ph-CH₂-O Ph
CH-CH₂-CH₂-N(Pr-i)₂
EtO-C

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REFERENCE COUNT: THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS 4 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT Гð ANSWER 4 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2001:923742 CAPLUS DOCUMENT NUMBER: 136:37403 TITLE: Shortened synthesis of 3,3-diarylpropylamine derivatives INVENTOR(S): Meese, Claus PATENT ASSIGNEE(S): Schwarz Pharma A.-G., Germany SOURCE: PCT Int. Appl., 37 pp.

CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: German FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

> PATENT NO. KIND APPLICATION NO. DATE DATE -----_ _ _ _ ---------WO 2001096279 WO 2001-EP6577 20010611 A1 20011220 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG 20020529 DE 10028443 DE 2000-10028443 20000614 C1 EP 1289929 EP 2001-947355 20010611 A1 20030312 AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, R : IE, SI, LT, LV, FI, RO, MK, CY, AL, TR

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	WO	20	01-EF	P65	77	W	2001	0611
CASREACT	136:37403	3;	MARPA	ΥA	136	:3740	3	

OTHER SOURCE(S): GI





- AB 3,3-Diarylpropylamines I [R = H, alkyl; R1, R2 = alkyl] are prepd. by reaction of R02CC6H4OH-4 with PhCH:CHC02H to give a 2-oxo-4-phenyl-3,4dihydrobenzopyran-6-carboxylate which is resolved via its cinchonidine salt, the (R)-isomer hydrolyzed to the acid which is reesterified, reduced to the benzopyranol, and subjected to aminolysis to give I. I [R = Me, R1, R2 = CHMe2], thus obtained, was then reduced to the benzyl alc. II.

(shortened synthesis of 3,3-diarylpropylamine derivs.)

- RN 214601-16-8 CAPLUS
- CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-hydroxy-, methyl ester (9CI) (CA INDEX NAME)

RN 380636-45-3 CAPLUS
CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-hydroxy-,
methyl ester, hydrochloride (9CI) (CA INDEX NAME)



IT 214601-17-9P 380636-47-5P
RL: SPN (Synthetic preparation); PREP (Preparation)
 (shortened synthesis of 3,3-diarylpropylamine derivs.)

RN 214601-17-9 CAPLUS

CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, methyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 380636-47-5 CAPLUS
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, hydrochloride (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

4



HCl

REFERENCE COUNT:

THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 ANSWER 5 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2001:449738 CAPLUS DOCUMENT NUMBER: 135:61141 TITLE: Preparation of stable salts of 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl esters. INVENTOR(S): Meese, Claus PATENT ASSIGNEE(S): Schwarz Pharma A.-G., Germany SOURCE: Ger. Offen., 22 pp. CODEN: GWXXBX PREORDER KPP M DOCUMENT TYPE: Patent LANGUAGE: German FAMILY ACC. NUM. COUNT: 2 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE _ _ _ _ ----------_ _ _ _ _ _ _ _ _ DE 19955190 A1 20010621 DE 1999-19955190 19991116 DE 29923134 U1 20000803 DE 1999-29923134 19991116 WO 2001035957 A1 20010525 WO 2000-EP11309 20001115 WO 2001035957 A3 20011227 AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, ₩: CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TMRW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG BR 2000015610 Α 20020730 BR 2000-15610 20001115 EP 1230209 A2 20020814 EP 2000-989857 20001115 AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, R: IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2003514018 т2 20030415 JP 2001-537950 20001115 NO 2002002314 А 20020515 NO 2002-2314 20020515 PRIORITY APPLN. INFO.: DE 1999-19955190 IA 19991116 WO 2000-EP11309 W 20001115 OTHER SOURCE(S): MARPAT 135:61141

GI

HO O R Ph O R I

AB Title compds. [I; R = alkyl, cycloalkyl, (substituted) Ph; X- = residue of a physiol. acceptable (in)org. acid], were prepd. Thus, (R)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl isobutyrate (II) (prepn. given) in 2-butanone was treated with fumaric acid under warming to give 83.1% II.hydrogen fumarate.
IT 156755-33-8 286930-05-0 RL: RCT (Reactant); RACT (Reactant or reagent)




(prepn. of stable salts of 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl esters)

RN 156755-33-8 CAPLUS

```
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-
(phenylmethoxy)-, hydrochloride (9CI) (CA INDEX NAME)
```

Absolute stereochemistry. Rotation (-).



HC1

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RN 286930-05-0 CAPLUS
CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-
(phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)
```



```
IT 156755-35-0P 214601-16-8P 214601-17-9P
286930-02-7P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
        (prepn. of stable salts of 2-(3-diisopropylamino-1-phenylpropyl)-4-
        hydroxymethylphenyl esters)
RN 156755-35-0 CAPLUS
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-
        (phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)
```

Absolute stereochemistry. Rotation (-).



RN 214601-16-8 CAPLUS
CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-hydroxy-,
methyl ester (9CI) (CA INDEX NAME)



RN 214601-17-9 CAPLUS CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, methyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



- RN 286930-02-7 CAPLUS
- CN Propanoic acid, 2-methyl-, 2-[(1R)-3-[bis(1-methylethyl)amino]-1phenylpropyl]-4-(hydroxymethyl)phenyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IT 286930-03-8P 345663-07-2P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of stable salts of 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl esters)

- RN 286930-03-8 CAPLUS
 CN Propanoic acid, 2-methyl-, 2-[(1R)-3-[bis(1-methylethyl)amino]-1-
- phenylpropyl]-4-(hydroxymethyl)phenyl ester, (2E)-2-butenedioate (1:1) (salt) (9CI) (CA INDEX NAME)

CM 1

CRN 286930-02-7 CMF C26 H37 N O3

Absolute stereochemistry. Rotation (+).



CM 2 CRN 11

CRN 110-17-8 CMF C4 H4 O4

Double bond geometry as shown.

но2С Е СО2Н

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RN 345663-07-2 CAPLUS
CN Propanoic acid, 2-methyl-, 2-[(1R)-3-[bis(1-methylethyl)amino]-1-
phenylpropyl]-4-(hydroxymethyl)phenyl ester, hydrochloride (9CI) (CA
```

INDEX NAME)

Absolute stereochemistry. Rotation (+).



HC1

Ь9 CAPLUS COPYRIGHT 2003 ACS on STN ANSWER 6 OF 14 ACCESSION NUMBER: 2000:533448 CAPLUS DOCUMENT NUMBER: 133:155419 TITLE: Stable salts of novel derivatives of 3,3-diphenylpropylamines PATENT ASSIGNEE(S): Schwarz Pharma A.-G., Germany SOURCE: Ger. Gebrauchsmusterschrift, 37 pp. CODEN: GGXXFR DOCUMENT TYPE: Patent LANGUAGE: German FAMILY ACC. NUM. COUNT: 2 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE ------ - - -_ _ _ _ _ _ _ _ _ ----------DE 29923134 U1 20000803 DE 1999-29923134 19991116 DE 19955190 A1 20010621 DE 1999-19955190 IA 19991116 - APPLECANTS DE 1999-19955190 19991116 PRIORITY APPLN. INFO.: F.P. App¹N OTHER SOURCE(S): MARPAT 133:155419



GI

AB 3,3-Diphenylpropylamine salts I [R1 = RCO2; R = C1-6 alkyl, C3-10

Ι

Petitioner Mylan Pharmaceuticals Inc. - Exhibit 1002 - Page 329

cycloalkyl, (substituted) Ph; R2 = CH2OH; X = inorg. or org. acid] are prepd. for use as prodrugs of agents for treatment of urinary incontinence and other spasmogenic disorders. I show improved absorption through biol. membranes and improved metabolic patterns and are easily crystd. I are prepd. from I free base (R1 = PhCH2O, R2 = CO2Me) by debenzylation, redn., acylation, and combination with HX. Thus, R-(-)-I-HCl (R1 = PhCH2O, R2 = CO2H) was esterified by refluxing in acidic MeOH, the ester was reduced with LiAlH4, the resulting carbinol was reduced with Raney Ni/H2, and the product [R-(+)-I free base, R = CHMe2] was converted to its H fumarate salt by heating with equimolar fumaric acid in 2-butanone; the salt was crystd. by addn. of cyclohexanone and cooling to 0.degree..

```
286930-03-8P 286930-04-9P
```

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (stable salts of novel derivs. of diphenylpropylamines)

RN 286930-03-8 CAPLUS
CN Propanoic acid, 2-methyl-, 2-[(1R)-3-[bis(1-methylethyl)amino]-1phenylpropyl]-4-(hydroxymethyl)phenyl ester, (2E)-2-butenedioate (1:1)
(salt) (9CI) (CA INDEX NAME)

CM 1

IT

CRN 286930-02-7 CMF C26 H37 N O3

Absolute stereochemistry. Rotation (+).



CM 2

CRN 110-17-8 CMF C4 H4 O4

Double bond geometry as shown.

HO₂C

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RN 286930-04-9 CAPLUS
CN Propanoic acid, 2-methyl-, 2-[(1R)-3-[bis(1-methylethyl)amino]-1-
phenylpropyl]-4-(hydroxymethyl)phenyl ester, hydrochloride, monohydrate
(9CI) (CA INDEX NAME)
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Absolute stereochemistry. Rotation (+).



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IT 156755-33-8
RL: RCT (Reactant); RACT (Reactant or reagent)
      (stable salts of novel derivs. of diphenylpropylamines)
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RN 156755-33-8 CAPLUS
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```
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-
(phenylmethoxy)-, hydrochloride (9CI) (CA INDEX NAME)
```

Absolute stereochemistry. Rotation (-).



• HCl

IT 156755-35-0P 214601-16-8P 214601-17-9P
286930-05-0P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
 (stable salts of novel derivs. of diphenylpropylamines)
RN 156755-35-0 CAPLUS
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-

(phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 214601-16-8 CAPLUS
CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-hydroxy-,
methyl ester (9CI) (CA INDEX NAME)



RN 214601-17-9 CAPLUS CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, methyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 286930-05-0 CAPLUS CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)



L9 ANSWER 7 OF 14 ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

AUTHOR(S): CORPORATE SOURCE:

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

Journal English A capillary solid-phase extn. (SPE) system has been coupled directly to AB electrospray tandem mass spectrometry for quantification of free tolterodine and metabolite concns. in plasma. The unbound fraction of these compds. was obtained by ultrafiltration of plasma. The ultrafiltrate was directly injected onto the SPE capillary (4 mm.times.200 .mu.m, 5 .mu.m C18). After desalting and clean-up of the sample, the analytes were eluted in backflush mode with methanol-1 mM triethylamine (70:30, vol./vol.), providing considerable solute focusing. Elution from the SPE capillary was improved by inserting a short trapping capillary between the SPE capillary and the MS interface, by which analyte focusing was increased. The unresolved compds. eluted simultaneously with the remaining matrix compds. and were detected in a multiple-reaction monitoring (MRM) mode. No interference of the sample matrix on detection was obsd., allowing aq. stds. to be used for calibration. Linear calibration curves were obtained between 0.05 and 1000 ng/mL (corresponding to 150 pM-3 .mu.M) and the limit of detection was 50 pg/mL injecting 10 .mu.l. Equilibration of the SPE capillary, sample loading, elution and detection took less then 6 min per sample.

CAPLUS COPYRIGHT 2003 ACS on STN

ultrafiltered plasma samples

CODEN: JCBBEP; ISSN: 0378-4347

Elsevier Science B.V.

Capillary solid-phase extraction-tandem mass spectrometry for fast quantification of free

Swart, R.; Koivisto, P.; Markides, K. E.

Applications (1999), 736(1 + 2), 247-253

concentrations of tolterodine and two metabolites in

Journal of Chromatography, B: Biomedical Sciences and

Institute of Chemistry, Department of Analytical Chemistry, Uppsala University, Uppsala, 751 21, Swed.

1999:779638 CAPLUS

132:202559

IT **194482-44-5**

RL: ANT (Analyte); ANST (Analytical study)

(capillary solid-phase extn.-tandem mass spectrometry for fast quantification of free concns. of tolterodine and two metabolites in ultrafiltered plasma samples)

RN 194482-44-5 CAPLUS

CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



REFERENCE COUNT:

PATENT INFORMATION:

THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 8 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN Гð ACCESSION NUMBER: 1999:736261 CAPLUS DOCUMENT NUMBER: 131:336818 TITLE: Preparation of 3,3-diphenylpropylamines as antimuscarinic agents. INVENTOR (S) : Sparf, Bengt; Meese, Claus O. PATENT ASSIGNEE(S): Schwarz Pharma AG, Germany SOURCE: Eur. Pat. Appl., 27 pp. CODEN: EPXXDW DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

6

PATENT NO. KIND DATE APPLICATION NO. DATE - - - -_._ _ _ _ _ _ _ ------_____ EP 957073 A1 19991117 EP 1998-108608 19980512 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO CA 1999-2328920 CA 2328920 AA 19991118 19990511 WO 9958478 A1 19991118 WO 1999-EP3212 19990511 W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG AU 9941412 A1 19991129 AU 1999-41412 19990511 AU 748057 В2 20020530 BR 9910406 Α 20010109 BR 1999-10406 19990511 EP 1077912 A1 20010228 EP 1999-924929 19990511 EP 1077912 Β1 20020703 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO AT 220056 AT 1999-924929 Е 20020715 19990511 EP 1254890 A1 20021106 EP 2002-13481 19990511 AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, R : IE, SI, LT, LV, FI, RO, MK, CY, AL NZ 507487 Α 20021126 NZ 1999-507487 19990511 ES 2181443 . Т3 20030216 ES 1999-924929 19990511 RU 2199525 C2 20030227 RU 2000-125813 19990511 JP 2003519079 Т2 20030617 JP 2000-548284 19990511 NO 2000005669 А 20010111 NO 2000-5669 20001110 PRIORITY APPLN. INFO.: EP 1998-108608 A 19980512 EP 1999-924929 A3 19990511 WO 1999-EP3212 W 19990511



- AB Title compds. (I; R = H, Me, Et, Pr, Me2CH, Bu, iso-Bu, pentyl, hexyl, PhCH2, alkyl, CHO, Ac, propionyl, isobutyryl, aminocarbonyl, aminosulfonyl, MeO2C, etc.; R1 = H, Me, Et, Pr, Me2CH, Bu, iso-Bu, pentyl, hexyl, PhCH2, alkyl, phenylalkyl; Z = NR8R9; R8, R9 = hydrocarbyl; NR8R9 = atoms to form a ring; with a proviso), were prepd. as antimuscarinic agents (no data). Thus, 4-bromophenol, cinnamoyl chloride, and Et3N were stirred 18 h in CH2Cl2 to give 99.8% 3-phenylacrylic acid 4-bromophenyl ester. This was refluxed 2 h with HOAc/H2SO4 to give 43.8% 6-bromo-4-phenylchroman-2-one. The latter was refluxed with benzyl bromide, K2CO3, and NaI in acetone/MeOH to give 102.1% crude Me 3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionate, which was stirred with LiAlH4 in THF to give 96.3% 3-(2-benzyloxy-5-bromophenyl)-3-phenylpropan-1ol. This was stirred with tosyl chloride and pyridine in CH2Cl2 for 18 h to give 93.6% tosylate ester, which was refluxed 97 h with diisopropylamine in MeCN to give 77.9% [3-(2-benzyloxy-5-bromophenyl)-3phenylpropyl]diisopropylamine. The latter was converted in several steps to 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol, which was acylated to give I. IT 250214-41-6P 250214-42-7P 250214-43-8P
- 250214-44-9P 250214-45-0P 250214-46-1P 250214-47-2P 250214-48-3P 250214-49-4P 250214-50-7P 250214-88-1P 250214-89-2P 250214-91-6P 250214-92-7P 250214-94-9P 250214-96-1P 250215-02-2P RL: BAC (Biological activity or effector, except adverse); BSU (Biological activity or effector, except adverse); BSU (Biological activity or effector, except adverse); BSU (Biological

study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

- (prepn. of 3,3-diphenylpropylamines as antimuscarinic agents) RN 250214-41-6 CAPLUS
- CN Benzenemethanol, 4-(acetyloxy)-3-[3-[bis(1-methylethyl)amino]-1phenylpropyl]- (9CI) (CA INDEX NAME)

Ph OAc $CH-CH_2-CH_2-N(Pr-i)_2$ HO--CH2

RN 250214-42-7 CAPLUS
CN Benzenemethanol, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(1oxopropoxy)- (9CI) (CA INDEX NAME)





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RN 250214-44-9 CAPLUS
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CN Propanoic acid, 2-methyl-, 2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl ester (9CI) (CA INDEX NAME)



HO-CH2

RN 250214-45-0 CAPLUS
CN Propanoic acid, 2,2-dimethyl-, 2-[3-[bis(1-methylethyl)amino]-1phenylpropyl]-4-(hydroxymethyl)phenyl ester (9CI) (CA INDEX NAME)



RN 250214-46-1 CAPLUS .CN Benzenemethanol, 4-(benzoyloxy)-3-[3-[bis(1-methylethyl)amino]-1phenylpropyl]- (9CI) (CA INDEX NAME)



- RN 250214-47-2 CAPLUS
- CN Propanedioic acid, bis[2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl] ester (9CI) (CA INDEX NAME)





RN 250214-49-4 CAPLUS

CN Pentanedioic acid, bis[2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl] ester (9CI) (CA INDEX NAME)





RN 250214-88-1 CAPLUS CN Carbamic acid, ethyl-, 2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl ester (9CI) (CA INDEX NAME)



RN 250214-89-2 CAPLUS
CN Benzenemethanol, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4[[(phenylamino)carbonyl]oxy]- (9CI) (CA INDEX NAME)





- RN 250214-91-6 CAPLUS
- CN Carbamic acid, 1,4-butanediylbis-, bis[2-[3-[bis(1-methylethyl)amino]-1phenylpropyl]-4-(hydroxymethyl)phenyl] ester (9CI) (CA INDEX NAME)



RN 250214-92-7 CAPLUS CN Carbonic acid, 2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl ethyl ester (9CI) (CA INDEX NAME)



RN 250214-94-9 CAPLUS
CN Benzenemethanol, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4(formyloxy)- (9CI) (CA INDEX NAME)



CN Carbonic acid, 2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl phenyl ester (9CI) (CA INDEX NAME)



RN 250215-02-2 CAPLUS CN Carbamic acid, ethyl-, 2-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(hydroxymethyl)phenyl ester, monohydrochloride (9CI) (CA INDEX NAME)



• HCl

(prepn. of 3,3-diphenylpropylamines as antimuscarinic agents) 250214-38-1 CAPLUS

CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(phenylmethoxy)-, hydrochloride (9CI) (CA INDEX NAME)



HCl

3

REFERENCE COUNT:

RN

THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 ANSWER 9 OF 14 (ACCESSION NUMBER: DOCUMENT NUMBER:	CAPLUS COPYRIGHT 2003 ACS on STN 1999:692703 CAPLUS 132:87770
TITLE:	Ketoconazole inhibits the metabolism of tolterodine in subjects with deficient CYP2D6 activity
AUTHOR (S) :	Brynne, N.; Forslund, C.; Hallen, B.; Gustafsson, L. L.; Bertilsson, L.
CORPORATE SOURCE:	Department of Clinical Pharmacology, Pharmacia and Upjohn AB, Stockholm, SE-112 87, Swed.
SOURCE :	British Journal of Clinical Pharmacology (1999), 48(4), 564-572 CODEN: BCDERM, ISSN: 0206-5251
PUBLISHER:	Blackwell Science Ltd.
DOCUMENT TYPE:	Journal
LANGUAGE :	English
AB The pharmacokinet	cics and safety of tolterodine and tolterodine metabolites

was studied after single-and multiple-dose administration in the absence and presence of ketoconazole, an inhibitor of cytochrome P 450 (CYP) 3A4, in healthy volunteers with deficient CYP2D6 activity, i.e. poor metabolizers of debrisoquine. Eight healthy volunteers received single oral doses (2 mg) of tolterodine L-tartrate. Following a wash-out period of about 3 mo, six of the subjects participated in a multiple-dose (1 mg twice daily) phase of the study. Ketoconazole 200 mg was given once daily for 4-4.5 days during both the single and multiple dose tolterodine administration phases. Blood samples were drawn and the pharmacokinetics of tolterodine and its metabolites were detd. A decrease (P < 0.01) in apparent oral clearance of tolterodine, from 10-12 l h-1 to 4.3-4.7 l h-1, was obtained during concomitant administration of ketoconazole, yielding at least a two-fold increase in the area under the serum concn.-time curve after single as well as after multiple doses following single dose administration of tolterodine. The mean (.+-.s.d.) terminal half-life increased by 50% from 9.7.+-.2.7 h to 15.+-.5.4 h in the presence of ketoconazole. CYP3A4 is the major enzyme involved in the elimination of tolterodine in individuals with deficient CYP2D6 activity (poor metabolizers), since oral clearance of tolterodine decreased by 60% during ketoconazole coadministration. This inhibition resulted in 2.1-fold increase in AUC.

IT 194482-44-5

RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(as tolterodine metabolite, ketoconazole inhibits the metab. of tolterodine in human subjects with deficient CYP2D6 activity)

- RN 194482-44-5 CAPLUS
- CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



REFERENCE COUNT:

27

THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT





L9 ANSWER 10 OF 14 CA ACCESSION NUMBER:	PLUS COPYRIGHT 2003 ACS on STN 1999:692702 CAPLUS
TTTLE.	IJ2:07/09 Fluovetine inhibits the metabolism of
	toltoroding pharmagekingtig impligations and property
	aliginal welcowned
	cillical felevance
AUTHOR (S):	Brynne, N.; Svanstrom, C.; Aberg-Wistedt, A.; Hallen,
	B.; Bertilsson, L.
CORPORATE SOURCE:	Departments of Clinical Pharmacology, Pharmacia and
	Upjohn AB, Stockholm, SE-112 87, Swed.
SOURCE :	British Journal of Clinical Pharmacology (1999),
	48(4), 553-563
	CODEN: BCPHBM; ISSN: 0306-5251
PUBLISHER:	Blackwell Science Ltd.
DOCUMENT TYPE:	Journal
LANGUAGE :	English
AB The change in dispo	sition of tolterodine during coadministration of the
potent cytochrome P	450 2D6 (CYP2D6) inhibitor fluoxetine was studied
Thirteen natients r	are used to the roding I start and a maturing doily for a r
THITTCEEN PACTENES I	everyed correrourne preatrigle 2 mg twice daily for 2.5

days, followed by fluoxetine 20 mg once daily for 3 wk and then concomitant administration for an addnl. 2.5 days. They were characterized as extensive metabolizers (EM1 with one functional CYP2D6 gene, EM2 with two functional genes) or poor metabolizers (PM). Nine patients, three EM2 and four EM1 and two PM, completed the trial. Following tolterodine administration, the area under the serum concn.-time curve (AUC) of tolterodine was 4.4-times and 30-times higher among EM1 and PM, resp., compared with EM2. The AUC of the 5-hydroxymethyl metabolite (5-HM) was not quantifiable in PM. Fluoxetine significantly decreased (P < 0.002) the oral clearance of tolterodine by 93% in EM2 and by 80% in EM1. The AUC of 5-HM increased in EM2 and decreased in EM1. However, the exposure to the active moiety (unbound tolterodine +5-HM) was not significantly increased in the two phenotypes. The subdivision of the EM group showed a 2.1-fold increase in active moiety in EM2 but the exposure was still similar to EM1 compared with before the interaction. The study suggests a difference in the pharmacokinetics of tolterodine and its 5-hydroxymethyl metabolite depending on the no. of functional CYP2D6 genes. Fluoxetine significantly inhibited the hydroxylation of tolterodine. Despite the effect on the pharmacokinetics of tolterodine in extensive metabolizers, the clin. effect is expected to be within normal variation.

IT 194482-44-5

RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(fluoxetine inhibits the metab. of tolterodine-pharmacokinetics) RN 194482-44-5 CAPLUS

CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

30



REFERENCE COUNT:

THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS



RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 A	NSWER	11 0	F 14	CA	PLUS	CO	PYRI	GHT	2003	ACS	on	STN						
ACCESS	ION NU	MBER	:		199	8:68	2217	CA	PLUS									
DOCUME	NT NUM	IBER:			129	:316	029											
TITLE:					Nov	el 3	-ary	1-3-	phen	ylpr	opan	amin	es w	ith a	anti	chol	inergi	.c
					act	ivit	y, tì	heir	use	in	the	trea	tmen	t of	uri	nary	-	
					inc	onti	nence	e, a	nd tì	heir	pre	para	tion			-		
INVENT	OR (S) :				Joh	anss	on, 1	Rolf	; Ha	rald	sson	, Ma	rtin	; Ri	ngbe	rg, i	Erik;	
					Vag	berg	, Jai	n; B	eier	lein	, Ka	tari	na;	Emon	d, R	ikar	d;	
					Sjo	berg	, Bi:	rger										
PATENT	ASSIG	NEE (S):		Pha	rmac	ia a	nd U	pjoh	n AB	, Sw	ed.					-	
SOURCE	:				PCT	Int	. Apj	pl.,	88	pp.								
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DOCUME	NT TYF	E:			Pat	ent												
LANGUA	GE:				Eng	lish												
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		KP,	KR,	KΖ,	LС,	LК,	LR,	LS,	LT,	LU,	LV,	MD,	MG,	MK,	MN,	MW,	MX,	
		NO,	NZ,	PL,	PΤ,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ΤJ,	ΤM,	TR,	ΤT,	
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	RW:	GH,	GM,	KE,	LS,	MW,	SD,	SZ,	UG,	ZW,	ΑT,	BE,	CH,	DE,	DK,	ES,	FI,	
		FR,	GB,	GR,	IE,	IT,	LU,	MC,	NL,	PΤ,	SE,	BF,	BJ,	CF,	CG,	CI,	CM,	
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19981008 ZA 1998-2478 19980324 19981022 AU 1998-67552 19980326 20011004 20000308

BR 1998-8069 19980326 EP 1998-912864 19980326

AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO JP 2001522355 20011113 JP 1998-541548 19980326 т2

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19	9991	126		NO	1999	-4438		1999	0913
20	0000	228		MX	1999	-8862		1999	0927
. 20	011	106		US	1999	-3818	68	1999	0927
			SE	199	7-114	44	А	1997	0327
			WO	199	8-SE	556	W	1998	0326
MARPA	AT 1	29:33	6029						

OTHER SOURCE(S): GI

ZA 9802478

AU 9867552

BR 9808069

EP 1019358

EP 1019358

R:

AT 239693

NO 9904438

MX 9908862

US 6313132

PRIORITY APPLN. INFO.:

AU 739186



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AB The invention relates to novel compds. I [wherein R1 = H, OH, alkyl, alkoxy, CF3, amino, alkanoylamino, alkanoyloxy, halo, hydroxyalkyl; R2, R3 = H, OH, alkyl, alkoxy, hydroxyalkyl, halo, carbamoyl, etc.; R4 =





(un)substituted alkyl or amino, CHO, CO2H, NO2, cyano, N3, alkoxy, and may also be H, Me, OMe, etc. under some circumstances; R5 = H, halo, alkyl; Ar = (un)substituted (hetero)aryl; R6, R7 = hydrocarbyl with optional OH groups or O bridge(s), and may form a ring; with several provisos], their salts with physiol. acceptable acids, their racemic mixts., and the individual enantiomers. The compds. have anticholinergic activity, and in particular are of use in the treatment of urinary incontinence. Sixty synthetic examples are given, and approx. 90 compds. (including free bases and salts) were prepd. and/or claimed. For instance, Wittig-type reaction of (EtO)2P(O)CH2CON(Pr-iso)2 with 2-fluorobenzophenone, followed by hydrogenation of the formed olefin and redn. of the amide with LiAlH4, gave after acidification, title compd. II.HCl. In a test for inhibition of carbachol-induced contraction of isolated guinea pig bladder strips, II had a KB value of 10 nM, and other compds. had values ranging from 1.18 nM to 3315 nM.

IT 214600-45-0P 214601-16-8P 214601-17-9P

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

- (prepn. of arylphenylpropanamines as anticholinergic agents) RN 214600-45-0 CAPLUS
- CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, methyl ester, hydrochloride (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



O HCl

RN 214601-16-8 CAPLUS
CN Benzoic acid, 3-[3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-hydroxy-,
methyl ester (9CI) (CA INDEX NAME)

Ph OH $CH^-CH_2^-CH_2^-N(Pr-i)_2$ MeO О

RN 214601-17-9 CAPLUS CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy-, methyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



IT 156755-34-9
RL: RCT (Reactant); RACT (Reactant or reagent)
 (starting material; prepn. of arylphenylpropanamines as anticholinergic
 agents)
RN 156755-34-9 CAPLUS

CN Benzoic acid, 3-[(1S)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-(phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)

5

Absolute stereochemistry. Rotation (+).



REFERENCE COUNT:

THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

1.9 ANSWER 12 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN 1998:393013 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 129:156415 TITLE: Biotransformation of tolterodine, a new muscarinic receptor antagonist, in mice, rats, and dogs AUTHOR(S): Andersson, Stig H. G.; Lindgren, Anders; Postlind, Hans CORPORATE SOURCE: Department of Drug Metabolism, Pharmacia & Upjohn AB, Uppsala, S-751 82, Swed. SOURCE : Drug Metabolism and Disposition (1998), 26(6), 528-535 CODEN: DMDSAI; ISSN: 0090-9556 PUBLISHER: Williams & Wilkins. DOCUMENT TYPE: Journal LANGUAGE: English AB Tolterodine is intended for the treatment of urinary urge incontinence and other symptoms assocd. with an overactive bladder. The in vivo metab. of 14C-labeled tolterodine was investigated in rats, mice, and dogs by anal. of blood and urine samples, whereas in vitro metab. studies were performed by incubation of [14C]tolterodine with mouse, rat, dog, and human liver

microsomes in the presence of NADPH. Tolterodine was extensively

metabolized in vivo. Mice and dogs showed similar metabolite patterns,

which correlated well with that obsd. in humans. In these species, tolterodine was metabolized along 2 different pathways, with the more important being the stepwise oxidn. of the 5-Me group to yield the 5-hydroxymethyl metabolite of tolterodine and then, via the aldehyde, the 5-carboxylic acid metabolite. The other pathway involved dealkylation of the nitrogen. In the subsequent phase II metab., tolterodine and the metabolites were conjugated with glucuronic acid to various degrees. Rats had a more extensive metab. and a markedly different metabolite pattern, with metabolites also being formed by hydroxylation of the nonsubstituted benzene ring. Gender differences were also obsd., with male rats showing more extensive metab. than females. Incubation of [14C]tolterodine yielded 5 metabolites with rat microsomes and 3 metabolites with mouse, dog, and human microsomes. The 5-hydroxymethyl metabolite of tolterodine and N-dealkylated tolterodine were major metabolites in all incubations, representing 83-99% of total metab. Although the extent of metab. varied among the species, the metabolic profiles were similar. Rat liver microsomes also formed metabolites hydroxylated in the nonsubstituted benzene ring. Thus, the metab. of tolterodine in mice and dogs corresponds to that obsd. in humans, whereas rats have a different metabolite pattern.

194482-44-5 CAPLUS

CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN

RN 210573-53-8 CAPLUS CN .beta.-D-Glucopyranosiduronic acid, 2-[(1R)-3-[bis(1-methylethyl)amino]-1phenylpropyl]-4-carboxyphenyl (9CI) (CA INDEX NAME)

Absolute stereochemistry.



15

REFERENCE COUNT:

THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT





L9 ANSWER 13 OF 14 CAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1997:478930 CAPLUS DOCUMENT NUMBER: 127:199591 TITLE: Pharmacokinetics and pharmacodynamics of tolterodine in man. A new drug for the treatment of urinary bladder overactivity AUTHOR (S): Brynne, N.; Stahl, M. M. S.; Hallen, B.; Edlund, P. O.; Palmer, L.; Hoglund, P.; Gabrielsson, J. CORPORATE SOURCE: Department Clinical Pharmacology, Pharmacia and Upjohn AB, Uppsala, S-75182, Swed. SOURCE: International Journal of Clinical Pharmacology and Therapeutics (1997), 35(7), 287-295 CODEN: ICTHEK; ISSN: 0946-1965 PUBLISHER: Dustri-Verlag Dr. Karl Feistle DOCUMENT TYPE: Journal LANGUAGE : English

AB The pharmacokinetics, pharmacodynamics, and safety of tolterodine was detd. following single oral and i.v. doses in healthy volunteers. Major urinary metabolites were identified and mass balance was detd. Single oral doses of 0.2, 0.4, 0.8, 1.6, 3.2, 6.4, and 12.8 mg of tolterodine (as the tartrate salt) were given to 17 healthy male volunteers. Two i.v. doses (0.64, 1.28 mg) were administered to 8 of the volunteers and mass balance was studied after a single oral dose of 5 mg (14C)-tolterodine in 6 subjects. Tolterodine was rapidly absorbed following oral administration. The abs. bioavailability was highly variable, ranging from 10-70%. The vol. of distribution at steady-state ranged from 0.9-1.6 L/kg and systemic clearance ranged from 0.23-0.52 L/h/kg, which resulted in a terminal half-life of 2-3 h. Tolterodine exhibited high first-pass metab. and 2 hepatic metabolic pathways were identified: oxidn. and dealkylation. Independent of route of administration, < 1% of the parent compd. was excreted unchanged in urine. Five metabolites were structurally identified in urine. Following oral administration of (14C)-tolterodine, the excretion of radioactivity into urine and feces was 77 and 17%, resp. Tolterodine decreased stimulated salivation after 3.2, increased heart rate after 6.4, and near point of vision after 12.8 mg. Six of 8 subjects reported micturition difficulties after a dose of 12.8 The lack of a direct relationship between tolterodine blood serum mg. concns. and effects on stimulated salivation suggested the presence of pharmacol. active metabolite(s).

IT 194482-44-5

RN

RL: BPR (Biological process); BSU (Biological study, unclassified); MFM (Metabolic formation); BIOL (Biological study); FORM (Formation, nonpreparative); PROC (Process)

(pharmacokinetics and pharmacodynamics of tolterodine in man) 194482-44-5 CAPLUS

CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4hydroxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).







ACCESSION NUMBER: 1994:508197 CAPLUS DOCUMENT NUMBER: 121:108197 TITLE: Preparation of 3,3-diphenylpropylamines and their use INVENTOR (S) : Johansson, Rolf Arne; Moses, Pinchas; Nilverbant, Lisbeth; Sparf, Bengt Aake PATENT ASSIGNEE(S): Kabi Pharmacia AB, Swed. SOURCE : PCT Int. Appl., 30 pp. CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PA'	LENL V	10.		KII	ND	DATE				API	5PIC	CAT.	ION	NC).	DATI	3			
WO	94113	37		A	 1	1994	0526			WO	199	93-8	SE9	27		1993	31105	i		
	W :	AU,	CA,	FI,	HU,	JP,	NO,	US												
	RW:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GE	i, (GR,	IE,	, I'	т,	LU,	MC,	, NL,	PΤ,	SE	
CA	21488	327		A	A	1994	0526			CA	199	93-2	214	882	:7	1993	31105	i		
AU	94543	80		A	1	1994	0608			AU	199	94 - 5	543	80		1993	31105	i		
AU	67245	58		B	2	1996	1003													
EP	66785	52		A	1	1995	0823			EΡ	199	93 - 9	924	876	5	1993	31105	i		
EP	66785	52		B	1	1998	0408													
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JP	08503	208		T	2	1996	0409			JP	199	94 - 5	511	977	,	1993	31105			
JP	33432	256		B	2	2002	1111													
HU	72742	2		A	2	1996	0528			HU	199	95-2	132	9		1993	31105	i		
AT	16482	8		Ε		1998	0415			AT	199	93 - 9	924	876	;	1993	31105	i		
ES	21171	.55		T3	3	1998	0801			ES	199	93 - 9	924	876	;	1993	31105	i		
FI	95021	.79		Α		1995	0505			FI	199	95-2	217	9		1995	50505			
NO	95017	75		Α		1995	0505			NO	199	95-2	177	5		1995	50505			
US	55592	69		Α		1996	0924			US	199	95-4	432	113		1995	50505			
US	56864	64		Α		1997	1111			US	199	96-6	584	638	1	1996	50722	1		
PRIORITY	Y APPL	N. 1	INFO.	:					SE	199	92-3	318	3		А	1992	21106	;		
									WO	199	93-5	SE92	27		W	1993	31105			
									US	199	95-4	1321	113		A3	1995	50505			
OTHER SC	OURCE ((S):			MAR	PAT	121:3	1081	97											





Title compds. I (R1 =H, Me; R2, R3 = H, Me, MeO, HO, H2NCO, H2NSO2, halo; AB X = R4R5N wherein R4, R5 = non-arom. hydrocarbyl and which together contain at least three carbon atoms, or R4R5N = heterocyclyl), salts, optical isomers, racemic mixt. and individual enantiomers are useful as anticholinergics. P-Br-C6H4OH, PhCH:CHCO2H, AcOH and H2SO4 were refluxed to give 6-bromo-4-phenyl-3,4-dihydrocoumarin which was converted in 4 steps to N,N-diisopropyl3-(2-benzyloxy-5-brophenyl)-3-phenylpropylamine (II). II was resolved to the (-)-isomer and converted in 4 steps to (-)-I



Absolute stereochemistry. Rotation (+).



O HC1

RN 156755-33-8 CAPLUS
CN Benzoic acid, 3-[(1R)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4 (phenylmethoxy)-, hydrochloride (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



🕑 HCl

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RN 156755-34-9 CAPLUS
CN Benzoic acid, 3-[(1S)-3-[bis(1-methylethyl)amino]-1-phenylpropyl]-4-
(phenylmethoxy)-, methyl ester (9CI) (CA INDEX NAME)
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Absolute stereochemistry. Rotation (+).



Absolute stereochemistry. Rotation (-).



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FULL ESTIMATED COST	72.26	522.12
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Under the Paderwork Reduction Act or 1995	Application Number	10/130,214
TRANSMITTAL	Filing Date	May 14, 2002
FORM	First Named Inventor	C. Meese
(to be used for all correspondence after initial	filing) Art Unit	1614
	Examiner Name	Not Yet Assigned
Total Number of Pages in This Submission	7 Attorney Docket Number	58827 (45107)
	ENCLOSURES (Check all tha	nt apply)
Fee Transmittal Form Fee Attached Amendment/Reply After Final After Final Stress Attached Extension of Time Request Express Abandonment Request Information Disclosure Statement Certified Copy of Priority Document(s) Response to Missing Parts/ Incomplete Application Response to Missing Parts under 37 CFR 1.52 or 1.53	□ Drawing(s) □ Licensing-related Papers □ Petition □ Petition to Convert to a Provisional Application ▶ Power of Attorney, Revocation Change of Correspondence Addr □ Terminal Disclaimer □ Request for Refund □ CD, Number of CD(s)	ress
Firm Edwards & Angell 11 P	TURE OF AFFEICANT, ATTORNE	ET, OR AGENT
or Christine C. O'Day		·
Signature /	r.m	
Date 5-19-02	<u>ر</u> ،	
I hereby certify that this correspondence is being fa first class mail in an envelope addressed to: Comm	csimile transmitted to the USPTO or deposited with issioner for Patents, Washington, DC 20231 on this	n the United States Posta I Service with sufficient postage as
Typed or printed Susan M. Dillon	- 	
Signature	m Dillon	Date 5/19/03

In sollection of information is required by 37 CFR 1.5. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 1 2 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Infor mation Officer, U.S. Patent and . Trademark Office, U.S. Department of Commerce, Washington, DC 202 31. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, Washington, DC 20231.

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

\mathbf{v}	
$ \Lambda $	

In re application of:C. MeeseSerial No.:10/130,214Group No.:1614Filed:May 14, 2002Examiner:Not Yet AssignedFor:STABLE SALTS OF NOVEL DERIVATIVES OF 3,3-DIPHENYLPROPYLAMINES

*NOTE: Insert name(s) of all inventor(s) and title also for patent.

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

POWER OF ATTORNEY BY ASSIGNEE OF ENTIRE INTEREST (REVOCATION OF PRIOR POWERS)

As assignee of record of the entire interest of the above identified

[X] application,

[] patent,

REVOCATION OF PRIOR POWERS OF ATTORNEY

all powers of attorney previously given are hereby revoked and

NEW POWER OF ATTORNEY

the following attorney(s) and/or agent(s) are hereby appointed to prosecute and transact all business in the Patent and Trademark Office connected therewith.

(list name and registration number)

(O	,	
Reg. No. 33,860	David A. Tucker	Reg. No. 27,840
Reg. No. 38,256	John J. Penny, Jr.	Reg. No. 36,984
Reg. No. 27,026	John B. Alexander	Reg. No. 48,399
Reg. No. 26,964	Steven M. Jensen	Reg. No. 42,693
Reg. No. 31,003	Kathryn A. Piffat	Reg. No. 34,901
Reg. No. 35,487	Richard J. Roos	Reg. No. 45,053
Reg. No. 38,227	Dianne M. Rees	Reg. No. 45,281
Reg. No. 40,927	George W. Hartnell, III	Reg. No. 42,639
Reg. No. 44,368	Howard M. Gitten	Reg. No. 32,138
Reg. No. 47,715	Jennifer K. Rosenfield	Reg. No. 53,531
Reg. No. 34,558	Richard J. Roos	Reg. No. 45,053
	Reg. No. 33,860 Reg. No. 38,256 Reg. No. 27,026 Reg. No. 26,964 Reg. No. 31,003 Reg. No. 35,487 Reg. No. 38,227 Reg. No. 40,927 Reg. No. 44,368 Reg. No. 44,368 Reg. No. 47,715 Reg. No. 34,558	Reg. No. 33,860David A. TuckerReg. No. 38,256John J. Penny, Jr.Reg. No. 27,026John B. AlexanderReg. No. 26,964Steven M. JensenReg. No. 31,003Kathryn A. PiffatReg. No. 35,487Richard J. RoosReg. No. 38,227Dianne M. ReesReg. No. 40,927George W. Hartnell, IIIReg. No. 44,368Howard M. GittenReg. No. 34,558Richard J. Roos

(Power of Attorney by Assignee of Entire Interest-page 1 of 2)

(check the following item, if applicable)

[] Attached, as part of this power of attorney, is the authorization of the above-named attorney(s) to accept and follow instructions from my representative(s).

SEND CORRESPONDENCE TO:

DIRECT TELEPHONE CALLS TO:

į

Peter F. Corless P.O. Box 9169 Boston, MA 02209 USA Peter F. Corless (617) 439-4444

Customer No.:

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	Schwarz Pharma AG			
(type or	print identity of assignee of	entire interest)		
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	40789 Monheim, Ger	rmany		
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[]	Recorded herewith	<u> </u>		
	A	ASSIGNEE STATEMENT		
Attached to the	is power is a "STATEM	IENT UNDER 37 C.F.R. 3.73(b))."	
	Mumm)	aliant	Date:	25 April 2003
Signatu	ire /			
КС). Hommerich	D.W. Schacht		
(type or	print name of person author	ized to sign on behalf of assignee)		
Auth	norized Officer	Assistant Manager		
Title		·····		

NOTE: The assignee of the entire interest may revoke previous powers and be represented by an attorney of his or her selection. 37 C.F.R. 1.36.

(check the following item, if it forms a part of this power of attorney)

[] Added page—Authorization of attorney(s) to accept and follow instructions from representative.

(Power of Attorney by Assignee of Entire Interest-page 2 of 2)

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Not Yet Assigned

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

[X]

In re application of: C. Meese Serial No.: 10/130,214 Group No.: Filed: May 14, 2002 Examiner: STABLE SALTS OF NOVEL DERIVATIVES OF 3,3-DIPHENYLPROPYLAMINES

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

For:

STATEMENT UNDER 37 C.F.R. 3.73(b) ESTABLISHING RIGHT OF ASSIGNEE TO TAKE ACTION

	· · · · -	CERTIEICATION UND		
		(When using Express Mail the l	PER 37 C.F.R. 3 Express Mail Ial	99 1.8(a) and 1.10* and number is mandatory
		Express Mail c	ertification is of	ptional.)
I hereb	y certify th	at, on the date shown below, this correspon	ndence is being:	
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Г	deposite 1450, A	ed with the United States Postal Service in lexandria, VA 22313-1450.	an envelope add	Iressed to the Commissioner for Patents, P.O. Box
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*WAR	NING:	Each paper or fee filed by AExpress Ma placed thereon prior to mailing. 37 C.F. ASince the filing of correspondence und oversight that can be avoided by the exe not be granted on petition. ≈ Notice of O	il≅ must have th R. ∋ I.10(b). er ∋ I.10 withou rcise of reasona ct. 24, 1996, 60	te number of the AExpress Mail≅mailing label at the Express Mail mailing label thereon is an ble care, requests for waiver of this requirement will Fed. Reg. 56.439. at 56.442.

(Statement under 37 C.F.R. 3 3.73(b) Establishing Right of Assignee to Take Action page 1 of 4)

patent application, ..., patent, registration, or reexamination proceeding, the assignee must establish its ownership of the property to the satisfaction of the Commissioner. Ownership is established by submitting to the Office, in the Office file related to the matter in which action is sought to be taken, documentary evidence of a chain of title from the original owner to the assignee (e.g., copy of an executed assignment submitted for recording) or by specifying (e.g., reel and frame number) where such evidence is recorded in the Office. The submission establishing ownership must be signed by a party authorized to act on behalf of the assignee. Documents submitted to establish ownership may be required to be recorded as a condition to permitting the assignee to take action in a matter pending before the Office. \simeq

- NOTE: ASection 3.73(b) is amended to remove the sentence requiring an assignee to specifically state that the evidentiary documents have been reviewed and to certify that title is in the assignee seeking to take action. The sentence is deemed to be unnecessary in view of the amendment to 331.4(d) and 10.18. Notice of Oct. 10, 1997, 62 Fed. Reg. 53,131, at 53,174.
- 1. The assignee(s) of the entire right, title and interest hereby seek(s) to take action in the PTO in this matter.

IDENTIFICATION OF ASSIGNEE

2. <u>Schwarz Pharma AG</u> Name of assignee

Corporation

Type of assignee, e.g., corporation, partnership, university, government agency, etc.

PERSON AUTHORIZED TO SIGN

K.-D. Hommerich

3.

D.W. Schacht

(type name of person authorized to sign on behalf of assignee) Authorized Officer Assistant Manager

Title of person authorized to sign

NOTE: The Notice of April 30, 1993 (1150 O.G. 62-64) points out:

AThe statement under 37 CFR 3.73(b) may be signed on behalf of the assignee in the following two manners if the assignee is an organization (e.g., corporation, partnership, university, government agency, etc.).

A(1) The statement may be signed by a person in the organization having apparent authority to sign on behalf of the organization. An officer (president, vice-president, secretary, or treasurer) is presumed to have authority to sign on behalf of the organization. The signature of the chairman of the board of directors is acceptable, but not the signature of an individual director. A person having a title (manager, director, administrator, general counsel) that does not clearly set forth that person as an officer of the assignee is not presumed to be an officer of the assignee or to have authority to sign the statement on behalf of the assignee. A power of attorney from the inventors in an organization to a practitioner to prosecute a patent application does not make the practitioner an official of an assignee or empower the practitioner to sign the statement on behalf of the assignee.

A(2) The statement may be signed by any person, if the statement includes an averment that the person is empowered to sign the statement on behalf of the assignee and, if not signed by a registered practitioner, the statement must be in oath or declaration form. Where a statement does not include such an averment, and the person signing does not hold a position in the organization that would give rise to a presumption that the person is empowered to sign the statement on behalf of the assignee, evidence of the person₌s authority to sign will be

(Statement under 37 C.F.R. > 3.73(b) Establishing Right of Assignee to Take Action page 2 of 4)

 $\textit{required}.\underline{\simeq}$

(complete the following, if applicable)

[X] I, the person signing below, state that I am empowered to sign this statement on behalf of the assignee.

BASIS OF ASSIGNEE'S INTEREST

Ownership by the assignee is established as follows:

A.

1.	[X]	An assignment	from the	inventor(s)	of the	matter	identified	above,	which	was
	recorde	d in the PTO at _	05/14/02	2						
		D1	012122	Γ	A	002				

Reel 013122 , Frame 0883

2. [] Other:

AND/OR

B. [] A chain of title from the inventor(s) to the current assignee as shown below:

1.	From:								
	Name of inventor(s)								
	То:								
	Recorded in PTO: Reel, Frame								
2.	From:								
	Name of inventor(s) or assignee								
	То:								
	Recorded in PTO: Reel, Frame								
3.	From:								
	Name of inventor(s) or assignee								
	То:								
	Recorded in PTO: Reel Frame								

(check item below, and add details, if applicable)

[] Additional documents in the chain of title are listed in the attached Supplemental Sheet.

(Statement under 37 C.F.R. > 3.73(b) Establishing Right of Assignee to Take Action page 3 of 4)

COPIES OF DOCUMENTS IN CHAIN OF TITLE

(complete this item, if copies are being sent)

[X] Copies of the assignment(s) or other document(s) in the chain of title are attached as follows:

[X] []	A B	[X] []	1 1	[] []	2 2	[]	3			
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					V -	Signat	ure of author	ized per	son	
				КD.	Homme	eri <u>ch</u>		D.W.	Schacht	
				(type or print name of authorized person)						

Authorized Officer Assistant Manager

Title of authorized person

TCM

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SIGNATURE OF PRACTITIONER

Christine C. O'Day (type or print name of practitioner)

P.O. Box 9169 P.O. Address

Customer No.: 21874

Tel. No.: (617) 439-4444

Reg. No.: 38,256

Boston, MA 02209

(Statement under 37 C.F.R. 3 3.73(b) Establishing Right of Assignee to Take Action page 4 of 4)

Petitioner Mylan Pharmaceuticals Inc. - Exhibit 1002 - Page 357



Peter F. Corless P.O Box 9169 Boston, MA 02209

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Date Mailed: 05/30/2003

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OC00000010151521

NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 05/22/2003.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

NORMA M VILLARIVERA 1641 (703) 308-0377

OFFICE COPY



UNITED STATES	<u>5 Patent and Traden</u>	MARK OFFICE United States DEPARTMENT OF COMMERCE United States Patent and Trademark Office Addres: COMMISSIONER OF PATENTS AND TRADEMARKS PO. Dox 1450 Alexandria, Virginia 22313-1450 www.usplo.gov				
APPLICATION NUMBER	FILING DATE	FIRST NAMED APPLICANT	ATTY. DOCKET NO./TITLE			
10/130,214	05/14/2002	Claus Meese	41946/32854			
21888 THOMPSON COBURN, LLP ONE FIRSTAR PLAZA		*OC000000	CONFIRMATION NO. 983			

THOMPSON COBURN, LL ONE FIRSTAR PLAZA SUITE 3500 ST LOUIS, MO 63101

Date Mailed: 05/30/2003

NOTICE REGARDING CHANGE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 05/22/2003.

• The Power of Attorney to you in this application has been revoked by the assignee who has intervened as provided by 37 CFR 3.71. Future correspondence will be mailed to the new address of record(37 CFR 1.33).

NORMA M VILLARIVERA 1641 (703) 308-0377

OFFICE COPY



21888 THOMPSON COBURN, LLP ONE FIRSTAR PLAZA SUITE 3500 ST LOUIS, MO 63101

11/16/1999

11/15/2000

Date Mailed: 08/05/2002

NOTICE OF ACCEPTANCE OF APPLICATION UNDER 35 U.S.C 371 AND 37 CFR 1.494 OR 1.495

The applicant is hereby advised that the United States Patent and Trademark Office in its capacity as an Elected Office (37 CFR 1.495), has determined that the above identified international application has met the requirements of 35 U.S.C. 371, and is ACCEPTED for national patentability examination in the United States Patent and Trademark Office.

The United States Application Number assigned to the application is shown above and the relevant dates are:

<u>05/14/2002</u> DATE OF RECEIPT OF 35 U.S.C. 371(c)(1), (c)(2) and (c)(4) REQUIREMENTS

05/14/2002 DATE OF RECEIPT OF ALL 35 U.S.C. REQUIREMENTS

A Filing Receipt (PTO-103X) will be issued for the present application in due course. THE DATE APPEARING ON THE FILING RECEIPT AS THE "FILING DATE" IS THE DATE ON WHICH THE LAST OF THE 35 U.S.C. 371 REQUIREMENTS HAS BEEN RECEIVED IN THE OFFICE. THIS DATE IS SHOWN ABOVE. The filing date of the above identified application is the international filing date of the international application (Article 11(3) and 35 U.S.C. 363). Once the Filing Receipt has been received, send all correspondence to the Group Art Unit designated thereon.

The following items have been received:

- U.S. Basic National Fee
- Assignee Statement
- Copy of IPE Report
- Copy of references cited in ISR
- Copy of the International Application
- Copy of the International Search Report
- Information Disclosure Statements
- Oath or Declaration
- Preliminary Amendments





BARBARA A CAMPBELL Telephone: (703) 305-3631

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FORM PCT/DO/EO/903 (371 Acceptance Notice)

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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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C07C 1/00, 217/62, 217/48, 219/28, 219/22, C07D 207/06, 295/06, C07C 271/08, C07F 7/18, C07C 307/02, A61K 31/135, 31/325, 31/40, 31/435	A1	(43) International Publication Date: 18 November 1999 (18.11.99)
(21) International Application Number:PCT/EP9(22) International Filing Date:11 May 1999 (1)	99/032 11.05.9	 (81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK
(30) Priority Data: 98108608.5 12 May 1998 (12.05.98)	I	 MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK,
(71) Applicant (for all designated States except US): SCI PHARMA AG [DE/DE]; Alfred-Nobel-Strat D-40789 Monheim (DE).	HWAF sse 1	KZ ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAP. 0, patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR NE, SN, TD, TG). NE, SN, TD, TG).
 (72) Inventors; and (75) Inventors/Applicants (for US only): MEESE, Claus [Kreuzberger Strasse 50, D-40789 Monheim (DE). Bengt [SE/SE]; Drottningstigen 6, S-142 65 Tr (SE). 	DE/DI SPAR rångsu	3]; Published F, With international search report. nd.
(74) Agent: ALBRECHT, Thomas; Kraus & Thomas-Wimmer-Ring 15, D-80539 Munich (DE	Weise !).	ort,

(54) Title: NOVEL DERIVATIVES OF 3,3-DIPHENYLPROPYLAMINES

(57) Abstract

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The invention concerns novel derivatives of 3,3-diphenylpropylamines, methods for their preparation, pharmaceutical compositions containing the novel compounds, and the use of the compounds for preparing drugs. More particularly, the invention relates to novel prodrugs of antimuscarinic agents with superior pharmacokinetic properties compared to existing drugs such as oxybutynin and tolterodine, methods for their preparation, pharmaceutical compositions containing them, a method of using said compouds and compositions for the treatment of urinary incontinence, gastrointestinal hyperactivity (irritable bowel syndrome) and other smooth muscle contractile conditions.

NOT PRIDE ART. SAME AS SPEC. FROM 09/700094.

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Description

Novel derivatives of 3,3-diphenylpropylamines

The present invention relates to novel derivatives of 3,3diphenylpropylamines, methods for their preparation, pharmaceutical compositions containing the novel compounds, and the use of the compounds for preparing drugs.

In man, normal urinary bladder contractions are mediated mainly through cholinergic muscarinic receptor stimulation. There is reason to believe that muscarinic receptors mediate not only normal bladder contractions, but also the main part of the contractions in the overactive bladder resulting in symptoms such as urinary frequency, urgency and urge incontinence. For this reason, antimuscarinic drugs have been proposed for the treatment of bladder overactivity.

Among the antimuscarinic drugs available on the market, oxybutynin is currently regarded as the gold standard for pharmacological treatment of urge incontinence and other symptoms related to bladder overactivity. The effectiveness of oxybutynin has been demonstrated in several clinical studies, but the clinical usefulness of oxybutynin is limited due to antimuscarinic side effects. Dryness of the mouth is the most common experienced side effect which may be severe enough to

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result in poor compliance or discontinuation of treatment (Andersson, K.-E., 1988, Current concepts in the treatment of disorders of micturition, Drugs 35, 477-494; Kelleher et al. 1994).

Tolterodine is a new, potent and competitive, muscarinic receptor antagonist intended for the treatment of urinary urge incontinence and detrusor hyperactivity. Preclinical pharmacological data show that tolterodine exhibits a favourable tissue selectivity in vivo for the urinary bladder over the effect on the salivation (Nilvebrant et al., 1997, Tolterodine - a new bladder-selective antimuscarinic agent, Eur. J. Pharmacol. 327 (1997), 195-207), whereas oxybutynin exhibits the reversed selectivity. Tolterodine is equipotent to oxybutynin at urinary bladder muscarinic receptors and the favourable tissue selectivity of tolterodine demonstrated in the preclinical studies has been confirmed in clinical studies. Thus a good clinical efficacy has been combined with a very low number of incidences of dry mouth and antimuscarinic side effects.

A major metabolite of tolterodine, the 5-hydroxymethyl derivative is also a potent muscarinic receptor antagonist and the pharmacological in vitro and in vivo profiles of this metabolite are almost identical to those of tolterodine (Nilvebrant et al., 1997, Eur. J. Pharmacol. 327 (1997), 195-207). Combined pharmacological and pharmacokinetic data indicate that it is most likely that the metabolite gives a major contribution to the clinical effect in most patients.

WO 94/11337 proposes the active metabolite of tolterodine as a new drug for urge incontinence. Administration of the active metabolite directly to patients has the advantage com-

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pared to tolterodine that only one active principle (compound) has to be handled by the patient which normally should result in a lower variation in efficacy and side effects between patients and lower risk of interaction with other drugs.

However, the introduction of an additional hydroxy group in the tolterodine results in an increased hydrophilic property of the new compounds (3,3-diphenylpropylamines) compared to the parent compounds which normally results in a lower absorption/bioavailability, leading to pre-systemic side effects or interactions due to non-absorbed antimuscarinic drug. In a method to circumvent this disadvantage, different prodrugs of the metabolite have been synthesized and tested for their antimuscarinic activity, potential absorption through biological membranes and enzymatic cleavage.

It is an object of the present invention to provide novel derivatives of 3,3-diphenylpropylamines. It is a further object of the present invention to provide new derivatives of 3,3-diphenylpropylamines which will be more useful as prodrugs for treatment of urinary incontinence and other spasmogenic conditions that are caused by muscarinic mechanisms while avoiding the disadvantage of a too low absorption through biological membranes of the drugs or an unfavourable metabolism.

A further object of the invention is to provide novel prodrugs of antimuscarinic agents with superior pharmacokinetic properties compared to present drugs as oxybutynin and tolterodine, methods for preparing thereof, pharmaceutical compositions containing them, a method of using said compounds

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and compositions for the treatment of urinary incontinence, gastrointestinal hyperactivity (irritable bowel syndrome) and other smooth muscle contractile conditions.

According to the present invention, novel 3,3-diphenylpropylamines are provided, which are represented by the general formulae I and VII'



wherein R and R' are independently selected from

a) hydrogen, C_1-C_6 alkyl, C_3-C_{10} cycloalkyl, substituted or unsubstituted benzyl, allyl or carbohydrate; or

b) formyl, C_1-C_6 alkylcarbonyl, cycloalkylcarbonyl, substituted or unsubstituted arylcarbonyl, preferably benzoyl; or

c) C_1-C_6 alkoxycarbonyl, substituted or unsubstituted aryloxycarbonyl, benzoylacyl, benzoylglycyl, a substituted or unsubstituted amino acid residue; or

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wherein R^4 and R^5 independently

represent hydrogen, C_1 - C_6 alkyl, substituted or unsubstituted aryl, preferably substituted or unsubstituted phenyl, benzyl or phenoxyalkyl wherein the alkyl residue has 1 to 4 carbon atoms and wherein R^4 and R^5 may form a ring together with the amine nitrogen; or

e)

d)



N-SOr wherein R^6 and R^7 independently

represent C1-C6 alkyl, substituted or unstubstituted aryl, preferably substituted or unsubstituted phenyl, benzyl or phenoxyalkyl wherein the alkyl residue has 1 to 6 carbon atoms; or

f) an ester moiety of inorganic acids,

 $-\text{SiR}_a\text{R}_b\text{R}_c,$ wherein $\text{R}_a,$ $\text{R}_b,$ R_c are independently selected g) from C₁-C₄ alkyl or aryl, preferably phenyl,

with the proviso that R' is not hydrogen, methyl or benzyl if R is hydrogen,

X represents a tertiary amino group of formula Ia



Formula la

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wherein R^8 and R^9 represent non-aromatic hydrocarbyl groups, which may be the same or different and which together contain at least three carbon atoms, and wherein R^8 and R^9 may form a ring together with the amine nitrogen,

Y and Z independently represent a single bond between the $(CH_2)_n$ group and the carbonyl group, O, S or NH,

A represents hydrogen (¹H) or deuterium (²H),

n is 0 to 12

and

their salts with physiologically acceptable acids, their free bases and, when the compounds can be in the form of optical isomers, the racemic mixture and the individual enantiomers.

The aforementioned compounds can form salts with physiologically acceptable organic and inorganic acids. Furthermore, the aforementioned compounds comprise the free bases as well as the salts thereof. Examples of such acid addition salts include the hydrochloride, hydrobromide and the like.

When the novel compounds are in the form of optical isomers, the invention comprises the racemic mixture as well as the individual isomers as such.

Preferably each of R^8 and R^9 independently signifies a saturated hydrocarbyl group, especially saturated aliphatic hydrocarbyl groups such as C_{1-8} -alkyl, especially C_{1-6} -alkyl, or adamantyl, R^8 and R^9 together comprising at least three, preferably at least four carbon atoms.

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According to another embodiment of the invention, at least one of R^8 and R^9 comprises a branched carbon chain.

Presently preferred tertiary amino groups X in formula I include the following groups a) to h):

a) CH(CH₃)₂

C(CH3)3

b)

c) $-N < CH_3 \\ C(CH_3)_2 CH_2 CH_3$











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Group a) is particularly preferred.

The aforementioned tertiary amino groups X are described in WO 94/11337 and the compounds according to the present invention can be obtained by using the corresponding starting compounds.

In the compounds according to the present invention, the term "alkyl" preferably represents a straight-chain or branchedchain hydrocarbon group having 1 to 6 carbon atoms. Such hydrocarbon groups may be selected from methyl, ethyl, propyl, isopropyl, butyl, isobutyl, pentyl and hexyl. The term "cycloalkyl" denotes a cyclic hydrocarbon group having 3 to 10 carbon atoms which may be substituted conveniently.

The term "substituted or unsubstituted benzyl" denotes a benyl group $-CH_2-C_6H_5$ which is optionally substituted by one or more substituents on the phenyl ring. Suitable substituents are selected from alkyl, alkoxy, halogen, nitro and the like. Suitable halogen atoms are fluorine, chlorine and iodine atoms. Preferred substituted benzyl groups are 4-methylbenzyl, 2-methylbenzyl, 4-methoxybenzyl, 2-methoxybenzyl, 4nitrobenzyl, 2-nitrobenzyl, 4-chlorobenzyl and 2-chlorobenzyl.

In the compounds according to the present invention the term $"C_1-C_6$ alkylcarbonyl" denotes a group R-C(=O) - wherein R is an alkyl group as defined hereinbefore. Preferred C_1-C_6 alkylcarbonyl groups are selected from acetyl, propionyl, isobutyryl, butyryl, valeroyl and pivaloyl. The term "cyclo-alkylcarbonyl" denotes a group R-C(=O) - wherein R is a cyclic hydrocarbon group as defined hereinbefore. The same counts to the selected carbonyl groups.

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The term "aryl" denotes an aromatic hydrocarbon group such as phenyl- (C_6H_5-) , naphthyl- $(C_{10}H_7-)$, anthryl- $(C_{14}H_9-)$, etc. Preferred aryl groups according to the present invention are phenyl and naphthyl with phenyl being particularly preferred.

The term "benzoyl" denotes an acyl group of the formula $-CO-C_6H_5$ wherein the phenyl ring may have one or more substituents.

Preferred substituents of the aryl group and in particular of the phenyl group are selected from alkyl, alkoxy, halogen and *nitro*. As substituted benzoyl groups 4-methylbenzoyl, 2methylbenzoyl, 4-methoxybenzoyl, 2-methoxybenzoyl, 4-chlorobenzoyl, 2-chlorobenzoyl, 4-nitrobenzoyl and 2-nitrobenzoyl may be mentioned.

The term $"C_1-C_6$ alkoxycarbonyl" refers to a group ROC(=0) wherein R is an alkyl group as defined hereinbefore. Preferred C_1-C_6 alkoxycarbonyl groups are selected from $CH_3OC(=0) -$, $C_2H_5-OC(=0) -$, $C_3H_7OC(=0) -$ and $(CH_3)_3COC(=0)$ and alicyclic alkyloxycarbonyl.

The term "amino acid residue" denotes the residue of a naturally occurring or synthetic amino acid. Particularly preferred amino acid residues are selected from the group consisting of glycyl, valyl, leucyl, isoleucyl, phenylalanyl, prolyl, seryl, threonyl, methionyl, hydroxyprolyl.

The amino acid residue may be substituted by a suitable group and as substituted amino acid residues, benzoylglycyl and Nacetylglycyl may be mentioned.

The term "carbohydrate" denotes the residue of a polyhydroxy aldehyde or polyhydroxy ketone of the formula $C_nH_{2n}O_n$ or $C_n(H_2O)_n$ and correponding carbohydrate groups are, for example, described in Aspinal, The Polysaccharides, New York: Academic Press 1982, 1983. A preferred carbohydrate group in the compounds according to the present invention is a glucuronosyl group, in particular a 1 β -D-glucuronosyl group.

The term "LG" as used herein denotes a leaving group selected from halogenides, carboxylates, imidazolides and the like.

The term "Bn" as used herein denotes a benzyl group.

Suitable ester moieties of inorganic acids may be derived from inorganic acids such as sulfuric acid and phosphoric acid.

Preferred compounds according to the present invention are:

A)

Phenolic monoesters represented by the general formulae II and II'









Particularly preferred phenolic monoesters are listed below:

(±)-formic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-propionic acid 2-(3-diisopropylamino-1-phenyl-

propyl)-4-hydroxymethylphenyl ester,

(±)-n-butyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

R-(+)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2,2-dimethylpropionic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-acetamidoacetic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-cyclopentanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-cyclohexanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-benzoic acid 2-(3-diisopropylamino+1-phenylpropyl)--4-hydroxymethylphenyl ester,

R-(+)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-4-methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-acetoxybenzoic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-1-naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-2-naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-4-chlorobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-4-methoxybenzoic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester, (±)-2-methoxybenzoic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester, (±)-4-nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-2-nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-malonic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester, (±)-succinic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester, (±)-pentanedioic acid bis-[2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethyl-phenyl]ester, (±)-hexanedioic acid bis-[2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethyl-phenyl]ester.

B) Identical diesters represented by the general formula III



wherein R^1 is as defined above.

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Particularly preferred identical diesters are listed below:

(±)-formic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester, (±)-acetic acid 4-acetoxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester, (±)-propionic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-propionyloxymethylphenyl ester, (±)-n-butyric acid 4-n-butyryloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-isobutyryloxymethylphenyl ester, (±)-2,2-dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-(2,2-dimethyl-propionyloxy)-benzyl ester, (±)-benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, R-(+)-benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-pent-4-enoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-(pent-4-encyloxymethyl)-phenyl ester, cyclic oct-4-ene-1,8-dioate of Intermediate B, cyclic octane-1,8-dioate of Intermediate B, poly-co-DL-lactides of Intermediate B.

C)

Mixed diesters represented by the general formula IV



Formula IV

wherein R¹ is as defined above

and

 R^2 represents hydrogen, C_1-C_6 alkyl or phenyl

with the proviso that R^1 and R^2 are not identical.

Particularly preferred mixed diesters are listed below:

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester, (±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-formyloxymethylphenyl ester, (±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-acetoxymethylphenyl ester, R-(+)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-acetoxymethylphenyl ester, (±)-isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, R-(+)-isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-2,2-dimethylpropionic acid 4-acetoxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, (±)-2,2-dimethylpropionic acid 4-acetoxymethyl-2-(3diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-benzoic acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester.

D) Benzylic monoesters represented by the general formula V



wherein R^1 is as defined above.

Particularly preferred benzylic monoesters are listed below:

(±)-formic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-acetic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-propionic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-hydroxybenzyl ester,

(±)-butyric acid 3-(3-diisopropylamino-1-phenylpropyl)-4-hydroxybenzyl ester,

(±)-isobutyric acid 3-(3-diisopropylamino-l-phenylpropyl)-4-hydroxybenzyl ester,

(±)-2,2-dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-hydroxybenzyl ester,

(±)-benzoic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-hydroxybenzyl ester.

E) Ethers and silyl ethers represented by the general formula VI .`

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wherein at least one of R^{10} and R^{11} is selected from C_1-C_6 alkyl, benzyl or $-SiR_aR_bR_c$ as defined above and the other one of R^{10} and R^{11} may additionally represent hydrogen, C_1-C_6 alkylcarbonyl or benzoyl.

Particularly preferred ethers and silyl ethers are listed below:

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-methoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-ethoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-propoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-isopropoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-butoxymethylphenol,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4methoxymethylphenyl ester,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4ethoxymethylphenyl ester,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-trimethylsilanyloxymethylphenol,

.*

(±)-diisopropyl-[3-phenyl-3-(2-trimethylsilanyloxy-5trimethylsilanyloxymethylphenyl)-propyl]-amine,

(±) - [3-(3-diisopropylamino-1-phenylpropyl) - 4-trimethylsilanyloxyphenyl] - methanol,

(±)-diisopropyl-[3-(5-methoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropylamine,

(±)-diisopropyl-[3-(5-ethoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropylamine,

(±)-[4-(tert.-butyl-dimethylsilanyloxy)-3-(3-diiso-

propylamino-1-phenylpropyl)-phenyl]-methanol,

(±)-acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-3-(3diisopropylamino-1-phenylpropyl)-benzyl ester,

(±)-4-(tert.-butyl-dimethylsilanyloxy)-3-(3-diisopropylamino-1-phenylpropyl)-phenol,

(±)-acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-2-(3diisopropylamino-1-phenylpropyl)-phenyl ester,

(±)-{3-[2-(tert.-butyl-dimethylsilanyloxy)-5-(tert.-

butyl-dimethylsilanyloxymethyl)-phenyl]-3-phenylpropyl}diisopropylamine,

(±)-[4-(tert.-butyl-diphenylsilanyloxy)-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-methanol,

(±)-acetic acid 4-(tert.-butyl-diphenylsilanyloxymethyl)-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester,

(±)-4-(tert.-butyl-diphenylsilanyloxymethyl)-2-(3-diisopropylamino-1-phenylpropyl)-phenol,

(±)-{3-[2-(tert.-butyl-diphenylsilanyloxy)-5-(tert.butyl-diphenylsilanyloxymethyl)-phenyl]-2-phenylpropyl}diisopropylamine,

(±)-acetic acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester,

(±)-benzoic acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester, . '

(±)-isobutyric acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-(1 β -D-glucuronosyloxymethyl)-phenol.

F) Carbonates and carbamates represented by the general formulae VII and VIII



wherein Y, Z and n are as defined above and wherein R^{12} and R^{13} represent a C_1-C_6 alkoxycarbonyl group or



wherein R^4 and R^5 are as defined above.

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Particularly preferred carbonates and carbamates are listed below:

(±)-N-ethylcarbamic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-N,N-dimethylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-N,N-diethylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±) -N-phenylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±) - [2-(3-Diisopropylamino-1-phenylpropyl) - 4-hydroxymethyl-phenoxycarbonylamino]acetic acid ethyl ester hydrochloride,

(±)-N-ethylcarbamic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-N-ethylcarbamoyloxybenzyl ester,

(±) -N,N-dimethylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N,N-dimethylcarbamoyloxybenzyl ester, (±)-N,N-diethylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N,N-diethylcarbamoyloxybenzyl ester, (±)-N-phenylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N-phenylcarbamoyloxybenzyl ester, (±)-{4-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenoxycarbonylamino]-butyl}-carbamic acid 2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester ethyl ester,

(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester phenyl ester,

(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-ethoxycarbonyloxymethylphenyl ester ethyl ester,

(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-

4-phenoxycarbonyloxymethylphenyl ester phenyl ester.

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G) 3,3-Diphenylpropylamines selected from

(i) compounds of the formulae IX and IX'



wherein o and p are the same or different and represent the number of methylene units $\{ CH_2 \}$ and may range from 0 to 6,

(ii) (±)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-sulphooxymethyl-phenyl ester

(iii) Poly-co-DL-lactides of 2-(3-diisopropylaminophenylpropyl)-4-hydroxymethyl-phenol

(iv) $(\pm)-2-(3-Diisopropylamino-1-phenylpropyl)-4-(1\beta-D-glucuronosyloxymethyl)-phenol having the formula$



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and

their salts with physiologically acceptable acids, their free bases and, when the compounds can be in the form of optical isomers, the racemic mixture and the individual enantiomers.

The present invention, moreover, relates to processes for the preparation of the aforementioned compounds. In particular, according to the present invention, the following processes are provided:

A process for the production of phenolic monoesters represented by the general formula II



Formula II

as defined above, which comprises treatment of a compound of the formula



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with an equivalent of an acylating agent selected from

wherein LG represents a leaving group selected from halogenide, carboxylate and imidazolide and R^1 is as defined above, in an inert solvent in the presence of a condensating agent.

Preferably, the acylating agent is selected from

$$\begin{array}{ccc} O & O \\ \parallel & \parallel & \parallel \\ R^{1}-C-Hal & or & R^{1}-C-O-C-R^{1}, \end{array}$$

wherein Hal represents a halogen atom, preferably a chlorine atom, and R^1 is as defined above.

A process for the production of phenolic monoesters represented by the general formula II'



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as defined above, which comprises treatment of two equivalents of a compound of the formula



with an acylating agent selected from

$$\begin{array}{cccc} 0 & 0 & 0 & 0 \\ \parallel & \parallel & \parallel & \parallel \\ \text{Hal-C-} (CH_2)_n - \text{C-Hal} & \text{or} & \text{C-} (CH_2)_n - \text{C} \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ \end{array}$$

wherein Hal represents a halogen atom, preferably a chlorine atom.

Hence, in these processes, an Intermediate B having the formula



is treated with an equivalent of an acylating agent (e.g. an acyl halogenite or acyl anhydride) in an inert solvent and in the presence of a condensating agent (e.g. amine) to provide phenolic monoesters of formula II or formula II' (wherein n

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is 0-12), respectively, if polyfunctional acylating agents (e.g. acid halides, preferably acid chlorides of dicarboxylic acids) are used.

The Intermediate B as used in the processes for the production of the 3,3-diphenylpropylamines according to the present invention can be in the form of a racemic mixture or of optically active compounds in accordance with the formulae shown below:







Intermediate RS Intermediate R-(+) Intermediate S-(-)

Alternatively, structures of formula II or II' may be obtained by regioselective deprotection of a protected benzylic hydroxy group (chemically or enzymatically: T. W. Greene, P. G. M. Wuts, "Protective Groups in Organic Chemistry", 2nd Ed., J. Wily & Sons, New York 1991).

The identical diesters represented by the general formula III



Formula III

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as defined above can be prepared by a process which comprises treatment of a compound of the formula



with at least two equivalents of the acylating agent R^1 -C(=0)-LG as defined above.

Thus, the aforementioned di-acyl compounds are readily accessible if an at least two-molar excess of an acylating agent is used in the above-mentioned conversion of Intermediate B or, more general, on treatment of compounds of formula I with acylating agents in the presence of suitable catalysts. In the above process, the following Intermediate A



wherein R' denotes a benzyl group can be used instead of Intermediate B. The Intermediate A can be used in the form of a racemic mixture or of optically active compounds (similar to Intermediate B).

Benzylic monoestes represented by the general formula V

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wherein R^1 is as defined above can be prepared by a process which comprises treatment of a compound of the formula



at room temperature and under anhydrous conditions with activated esters in the presence of enzymes selected from lipases or esterases.

Hence, this process relates to the preparation of phenols with para acyloxymethyl substituents (cf. formula V). These compounds can be prepared in several chemical steps from intermediates such as formula I, where R represents hydrogen and R' is hydrogen or any suitable protective group which can be removed by known methods (T. W. Greene, P.G.M. Wuts, "Protective Groups in Organic Chemistry", 2nd Ed., J. Wily & Sons, New York 1991) in the presence of the newly introduced substituent R¹CO. It was found, however, that the benzylic substituent R¹CO can be introduced more conveniently and in only one step if Intermediate B is treated at room tempera-

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ture and under anhydrous conditions with activated esters (e.g. vinyl acylates, isopropenyl acylates) in the presence of enzymes such as lipases or esterases.

The mixed diesters represented by the general formula IV



Formula IV

wherein R^1 and R^2 are as defined above can be prepared by a process which comprises acylation of the above-mentioned benzylic monoester represented by the general formula V



wherein R^1 is as defined above or of a phenolic monoester represented by the general formula II



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as defined hereinbefore.

In general, mixed diesters of formula IV can be obtained by acylation of compounds of the general formula I wherein R and R' are different substituents selected from the group consisting of hydrogen, acyl residues or protecting groups that are cleavable under the acylation reaction conditions.

Ethers represented by the general formula VI



as defined hereinbefore wherein R^{11} is hydrogen can be prepared by a process which comprises reacting a compound of the formula



with an alcohol R^{10} -OH in the presence of an esterification catalyst.

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A further process for the preparation of ethers represented by the general formula VI



wherein R^{10} and R^{11} are as defined hereinbefore, comprises acid or base treatment of free benzylic alcohols selected from



and



and



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or



wherein R^{10} is hydrogen and R^{11} is as defined above or



Formula VII

wherein R^{12} is hydrogen and R^{13} represents a $C_1\text{-}C_6$ alkoxycarbonyl group or



wherein R^4 and R^5 are as defined above

or of benzylic acylates selected from

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wherein R^1 and R^2 are as defined hereinbefore in the presence of suitable hydroxy reagents.

Finally, ethers of formula VI can be prepared by a process which comprises treating a compound of the formula



wherein R¹⁰ is as defined above with an alkylating agent selected from alkyl halogenides, alkyl sulphates and alkyl triflates, said alkyl group having 1 to 6 carbon atoms.
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In summary, regioselective modification of the benzylic hydroxy groups is achieved either by acid or base treatment of benzylic acylates in the presence of suitable hydroxy reagents (e.g. alcohols) or by catalytic ether formation as described in the literature for other benzylic substrates (J.M. Saa, A. Llobera, A. Garcia-Raso, A. Costa, P.M. Deya; J. Org. Chem. 53: 4263-4273 [1988]). Both free benzylic alcohols such as Intermediates A and B or compounds of formulas II or VI (in which R^{10} is hydrogen) or formula VII (in which R^{12} is hydrogen) as well as benzylic acylates such as formulae III, IV, V may serve as starting materials for the preparation of benzylic ethers (B. Loubinoux, J. Miazimbakana, F. Gerardin; Tetrahedron Lett. 30: 1939-1942 [1989]).

Likewise the phenolic hydroxy groups are readily transformed into phenyl ethers (R¹¹ = alkyl) using alkylating agents such as e.g. alkyl halogenides, alkyl sulphates, alkyl triflates or employing Mitsunobu type reaction conditions (Synthesis **1981**, 1-28). Similarly, both phenolic and alcoholic monosilyl ethers are obtained by regioselective silylation or by desilylation of bis-silyl ethers of Intermediate B as described for other compounds in the literature (J. Paladino, C. Guyard, C. Thurieau, J.-L. Fauchere, Helv. Chim. Acta **76**: 2465-2472 [1993]; Y. Kawazoe, M. Nomura, Y. Kondo, K. Kohda, Tetrahedron Lett. **26**: 4307-4310 [1987]).

Carbonates and carbamates represented by the general formulae VII and VIII





as defined hereinbefore can be prepared by a process which comprises reacting a compound selected from the group consisting of



Formula (i



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wherein R^1 is defined as above, n is 0 to 12, Bn is benzyl, R^{10} or R^{11} is hydrogen with activated carbonyl compounds or carbonyl precursor reagents selected from haloformates, ketenes, activated esters, mixed anhydrides of organic or inorganic acids, isocyanates and isothiocyanates.

The coupling reactions can be carried out in inert solvents over periods of several hours at temperatures from -10° C to the refluxing temperature of the solvent or reagent used to provide compounds of the general formula VII where R¹² represents hydrogen, alkyl, aliphatic or aromatic acyl, or carbamoyl, and R¹³ represents $-C(=0)-Y-R^3$, wherein Y and R³ represent O, S, NH and alkyl or aryl, respectively. Polyfunctional reagents give the corresponding derivatives. For example, diisocyanates or di-carbonylchlorides provide compounds of formula VIII where X, Y have the meaning of O, S, or NH and n is zero to twelve.

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The invention, moreover, relates to pharmaceutical compositions comprising one or more of the aforementioned 3,3-diphenylpropylamines. In other words, the compounds according to the present invention can be used as pharmaceutically active substances, especially as antimuscarinic agents.

They can be used for preparing pharmaceutical formulations containing at least one of said compounds.

The compounds according to the present invention in the form of free bases or salts with physiologically acceptable acids, can be brought into suitable galenic forms, such as compositions for oral use, for injection, for nasal spray administration or the like, in accordance with accepted pharmaceutical procedures. Such pharmaceutical compositions according to the invention comprise an effective amount of the compounds of claims 1 to 15 in association with compatible pharmaceutically acceptable carrier materials, or diluents, as is well known in the art. The carriers may be any inert material, organic or inorganic, suitable for enteral, percutaneous or parenteral administration, such as water, gelatine, gum arabicum, lactose, microcrystalline cellulose starch, sodium starch glycolate, calcium hydrogen phosphate, magnesium stearate, talcum, colloidal silicon dioxide, and the like. Such compositions may also contain other pharmaceutically active agents, and conventional additives, such as stabilizers, wetting agents, emulsifiers, flavouring agents, buffers, and the like.

The composition according to the invention can e.g. be made up in solid or liquid form for oral administration, such as tablets, capsules, powders, syrups, elixirs and the like, in WO 99/58478

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the form of sterile solutions, suspensions or emulsions for parenteral administration, and the like.

The compounds according to the invention may be used in a patch formulation. The compounds can be administered transdermally with a reduced incidence of side effects and improved individual compliance.

The compounds and compositions can, as mentioned above, be used for the treatment of urinary incontinence and other spasmogenic conditions that are caused by muscarinic mechanisms. The dosage of the specific compound will vary depending on its potency, the mode of administration, the age and weight of the patient and the severity of the condition to be treated. The daily dosage may, for example, range from about 0.01 mg to about 5 mg, administered singly or multiply in doses e.g. from about 0.05 mg to about 50 g each.

The invention will be further illustrated by the following non-limiting examples and pharmacological tests.

I. Experimental

1. General

All compounds were fully characterized by ¹H and ¹³C NMR spectroscopy (Bruker DPX 200). The chemical shifts reported for ¹³C NMR spectra (50 MHz, ppm values given) refer to the solvents CDCl₃ (77.10 ppm), dideuterio dichloromethane $(CD_2Cl_2, 53.8 \text{ ppm})$, CD_3OD (49.00 ppm) or hexadeuterio dimethylsulphoxide (DMSO-d₆, 39.70 ppm), respectively. ¹H NMR data (200 MHz, ppm) refer to internal tetramethylsilane).

Thin-layer chromatography (tlc, R_f values reported) was conducted on precoated 5x10 cm E. Merck silica gel plates (60F254), spots were visualized by fluorescence quenching or spaying with alkaline potassium permanganate solution. Solvent systems: (1), ethyl acetate/n-hexane (30/70, v/v-%);(2), toluene/acetone/methanol/acetic acid (70/5/20/5, v/v-%); (3), n-hexane/acetone/diethylamine (70/20/10, v/v-%); (4), n-hexane/acetone/triethylamine (70/20/10, v/v-\$); (5), ethyl acetate/n-hexane/2-propanol/triethylamine (60/40/20/1, v/v-%); (6), ethyl acetate/triethylamine (90/10, v/v-%); (7), cyclohexane/acetone/acetic acid (80/20/0.5, v/v-%). Optical rotations were measured at 589.3 nm and room temperature on a Perkin Elmer Polarimeter Type 241. Melting points (mp) reported are uncorrected and were determined on a Mettler FP 1 instrument. IR spectra were taken from a Perkin-Elmer FTIR spectrometer Series 1610, resolution 4 cm⁻¹. Gas chromatography-mass spectrometry (GC-MS): spectra (m/z values and relative abundance (%) reported) were recorded on a Finnigan TSQ 700 triple mass spectrometer in the positive (P-CI) or negative (N-CI) chemical ionization mode using methane or ammonia as reactant gas. Hydroxylic compounds were analyzed as their trimethylsilyl ether derivatives. Combined liquid chromatography-mass spectrometry (LC-MS):

Waters Integrety System, Thermabeam Mass Detector (EI, 70 eV), m/z values and relative abundance reported.

2. Synthesis of Intermediates A and B

3-Phenylacrylic acid 4-bromophenyl ester

An ice-cooled solution of 4-bromophenol (69.2 g) and cinnamoyl chloride (66.8 g) in dichloromethane (150 ml) was treated with triethylamine (40.6 g). After stirring for 18 hrs at



room temperature the mixture was washed with water (250 ml), 1 M aqueous HCl, and dried over anhydrous sodium sulphate. Evaporation in vacuum left solid 3-phenylacrylic acid 4-bromophenyl ester (121.0 g, 99.8% yield), m.p. 113.3°C, tlc: (1) 0.83. NMR(CDCl₃): 116.85, 118.87, 123.49, 128.38, 129.06, 130.90, 132.49, 134.02, 147.07, 149.84, 165.06.

(±)-6-Bromo-4-phenylchroman-2-one

A portion of the ester (60.0 g) was dissolved in a mixture of acetic acid (60 ml) and concentrated sulphuric acid (18 ml) and refluxed for 2 hrs. After cooling, the reaction mixture was poured into ice water and the product was isolated by extraction with ethylacetate. Evaporation of the solvent and recrystallization of the residue from boiling ethanol (150 ml) yielded 26.3 g (43.8% yield) of pure, crystalline (\pm) -6-bromo-4-phenylchroman-2-one, m.p. 117.8°C, tlc: (1) 0.67. NMR (CDCl₃): 36.56, 40.51, 117.29, 118.87, 127.47, 127.89, 128.33, 129.32, 131.07, 131.79, 139.42, 150.76, 166.84.

(\pm) -3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid methyl ester

A suspension consisting of (\pm) -6-bromo-4-phenylchroman-2-one (85.0 g), anhydrous potassium carbonate (46.7 g), sodium iodide (20.5 g) and benzyl chloride (40.6 g) in methanol (350 ml) and acetone (350 ml) was refluxed for 3 hrs. After evaporation of the solvents the residue was extracted with diethyl ether (2 x 300 ml) and the extract was washed with water (2 x 200 ml) and aqueous sodium carbonate. Drying (Na₂SO₄) and rotoevaporation left 121.8 g (102.1% crude yield) of (\pm)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid methyl ester as a light yellow oil, tlc: (1) 0.77; NMR (CDCl₃): 39.22, 40.53, 51.63, 70.16, 113.10, 113.77, 126.46, WO 99/58478

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126.92, 127.88, 128.08, 128.34, 128.45, 130.31, 130.55, 134.41, 136.44, 142.37, 154.94, 172.08.

(±)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid A solution of (\pm) -3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid methyl ester (0,391 g, 0,92 mmol) in ethanol (5 ml) was treated at 50°C with excess aqueous sodium hydroxide solution until the milky emulsion became clear. The reaction mixture was then acidified (pH 3), evaporated and extracted with dichloromethane. The organic extract was evaporated and the remaining oil was redissolved in a minimum of boiling ethanol. The precipitation formed after 18 hrs at 4°C was filtered off and dried in vacuo to yield 0,27 g (71.4%) of (\pm) -3-(2-Benzyloxy)-5-bromophenyl)-3-phenylpropionic acid, colourless crystals, m.p. 124.9°C; tlc: (1) 0.15 (starting material methyl ester 0.75); NMR (CDCl₃): 39.15, 40.26, 70.25, 113.21, 113.90, 126.62, 127.27, 127.98, 128.17, 128.47, 128.54, 130.46, 130.68, 134.34, 136.45, 142.16, 154.95, 177.65. LC-MS: 412/410 (14/11%, M^{+.}), 394/392 (15/13%), 321/319 (17/22%), 304/302 (17/21%), 259 (24%), 194 (22%), 178 (21%), 167 (65%), 152 (49%), 92 (100%). IR (KBr): 3434, 3030, 1708, 1485, 1452, 1403, 1289, 1243, 1126, 1018, 804, 735, 698, 649. Calculated for $C_{22}H_{19}BrO_3$ (mol-wgt. 411.30): C 64.25%, H 4.66%, Br 19.43%, O 11.67%; found: C 63.72%, H 4.70%, Br 19.75%, O 11.80%.

Alternatively, the crude reaction mixture from the above described synthesis of (\pm) -3-(2-benzyloxy-5-bromophenyl)-3phenylpropionic acid methyl ester was evaporated, redissolved in warm ethanol, and treated with excess aqueous potassium hydroxide solution. Acidification to pH 3 (conc. hydrochloric acid) and cooling to 4°C resulted in the formation of a solid, which was filtered off after 18 hrs, washed repeatedly WO 99/58478

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with water and dried to yield (t)-3-(2-benzyloxy-5-bromo-phenyl)-3-phenylpropionic acid in 82% yield.

a) Resolution of 3-(2-benzyloxy-5-bromophenyl)-3phenylpropionic acid

R-(-)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid Warm solutions of (\pm) -3-(2-benzyloxy-5-bromophenyl) -3-phenylpropionic acid (815.6 g, 1.85 mol) and 1S, 2R-(+)-ephedrine hemihydrate (232.1 g, 1.85 mol) in 2000 ml and 700 ml, respectively, of absolute ethanol were combined and then allowed to cool to 0°C. The precipitate formed was collected, washed with cold ethanol and dried in vacuum to give 553.2 g of the ephedrinium salt of the title compound (m.p. 153°C, e.e. 65% as determined by NMR and HPLC). The salt was recrystallized twice from boiling ethanol to give R-(-)-3-(2-)benzyloxy-5-bromophenyl)-3-phenylpropionic acid 15,2R-(+)ephedrinium salt in 75% yield, colourless crystalls, m.p. 158.6°C, e.e. 97.6% (HPLC). NMR (CDCl₃): 9.53, 30.90, 41.54, 42.83, 61.45, 70.15, 70.42, 113.05, 113.68, 125.89, 126.03, 127.33, 127.85, 128.19, 128.28, 128.45, 129.86, 130.70, 135.91, 136.65, 140.40, 144.09, 155.20, 178.94.

1.2 g (2.0 mmol) of the ephedrinium salt were dissolved in a mixture of acetone (5 ml) and ethanol (10 ml). After treatment with water (0.4 ml) and conc. (37%) aqueous hydrochloric acid (0.34 ml), the solution was evaporated in vacuum, and the residue was redissolved in 1M aqueous hydrochloric acid (2 ml) and dichloromethane (10 ml). The organic phase was separated, washed twice with water (2 ml), and evaporated to dryness to give R-(-)-3-(2-Benzyloxy-5-bromophenyl)-3-phenyl-propionic acid as a colourless oil which slowly solidified (0.4 g, 98% yield), m.p. 105.6°C (from ethyl acetate/n-

heptane); tlc: (7) 0.21; $[\alpha]_D^{20} = -21.1$ (c = 1.0, ethanol), e.e. 99.9% (HPLC). NMR: identical with the racemic acid.

S-(+)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid The combined mother liquids from the above resolution and recrystallizations were treated under stirring and cooling (18°C) with excess conc. aqueous hydrochloric acid. The precipitate (ephedrinium hydrochloride) was filtered off, and the filtrate was evaporated to dryness. The residue was redissolved in dichloromethane (1.5 litre) and then washed with several portions of 1 M aqueous hydrochloric acid followed by water. After drying (Na₂SO₄), filtration, and evaporation 479 g of crude S-(+)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid were obtained as a yellow viscous oil. The pure S-(+) enantiomeric acid was converted into the 1R,2S-(-)-ephedrine salt as described above for the R-(-) acid. Two recrystallizations from boiling ethanol provided colourless crystals of S-(+)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid 1R,2S-(-)-ephedrinium salt in 83% yield, m.p. 158.7°C, e.e. 97.8% (HPLC). NMR (CDCl₃): 9.47, 30.85, 41.54, 42.92, 61.48, 70.13, 70.30, 113.04, 113.66, 125.89, 126.01, 127.32, 127.84, 128.18, 128.44, 129.83, 130.68, 135.94, 136.63, 140.44, 144.13, 155.19, 178.94.

S-(+)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid was obtained in quantitative yield from this ephedrinium salt by the method described above for the R-(-) acid, tlc: (7) 0.20, e.e. (NMR) > 99%, mp 105.5°C; $[\alpha]_{p}^{20} = +22.6$ (c = 1.0, ethanol); NMR: identical with the racemic acid. b) Enantioselective Synthesis of R-(-) - and S-(+)-3-(2benzyloxy-5-bromophenyl)-3-phenylpropionic acid



2-Benzyloxy-5-bromobenzaldehyde

To a solution of 0.1 mol of 5-bromo-2-benzaldehyde in THF (150 ml) was added 0.1 mol of K_2CO_3 and 0.11 mol of benzyl bromide. The mixture was refluxed for 2 hrs and water (500 ml) was added. After addition of ethyl acetate (400 ml) and stirring the organic layer was washed with water, dried (sodium sulphate) and evaporated to dryness. The resulting slightly yellow solid of pure (tlc) 2-benzyloxy-5-bromobenzaldehyde was used as such in the next step.

3-(2-Benzyloxy-5-bromophenyl)-acrylic acid

A mixture of 2-benzyloxy-5-bromobenzaldehyde (0.10 mol), malonic acid (15.0 g), and piperidine (2.0 ml) in 150 ml of pyridine was first heated at 90°C for 90 min and subsequently refluxed for 0.5 hrs. After cooling to room temperature, the reaction was poured on a mixture of ice (1 kg) and concentrated aqueous hydrochloric acid (250 ml). The solid

material that precipitated after stirring for 2 hrs. was collected by suction and recrystallized from a minimum of boiling methanol.

3-[3-(2-Benzyloxy-5-bromophenyl)-acryloyl]-(4R)-4-phenyloxazolidin-2-one

Pivaloylchloride (7 g) was added dropwise at -30°C to a stirred solution of 3-(2-benzyloxy-5-bromophenyl)-acrylic acid (50.0 mmol) and triethylamine (15.0 ml) in 200 ml of tetrahydrofuran. After an additional hour the temperature was lowered to -50°C and (R)-2-phenyloxazolidin-2-one (9.0 g) and lithium chloride (2.5 g) were added in one portion. The cooling bath was then removed and stirring was continued over 18 hrs. The reaction was diluted with water and 3-[3-(2-benzyloxy-5-bromophenyl)-acryloyl]-(4R)-4-phenyloxazolidin-2-one was isolated by extraction with ethyl acetate.

3-[3-(2-Benzyloxy-5-bromophenyl)-(3S)-3-phenylpropionyl]-(4R)-4-phenyloxazolidin-2-one

To a precooled (-30°C) mixture of copper-(I) chloride (21.0 g) and dimethylsulfide (45 ml) in dry tetrahydrofuran (150 ml) was added dropwise an ethereal solution of phenylmagnesiumbromide (0.3 mol). The mixture was stirred 20 min at the same temperature and then cooled to -40°C. A solution of 3-[3-(2-Benzyloxy-5-bromophenyl)-acryloyl]-(4R)-4-phenyloxazolidin-2-one (50.0 mmol) in dry tetrahydrofuran (150 ml) was added during 10 min. The cooling bath was removed and stirring was continued for 18 hrs. The mixture was quenched with half-saturated aqueous ammonium chloride solution and the product was isolated by extraction with ethyl acetate. WO 99/58478

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S-(+)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid A solution of the above described 3-[3-(2-benzyloxy-5-bromophenyl)-(3S)-3-phenylpropionyl]-(4R)-4-phenyloxazolidin-2-one in tetrahydrofuran (300 ml) and water (100 ml) was cooled to 0°C and then treated with 30% aqueous hydrogen peroxide (20 ml) followed by solid lithium hydroxide (4.3 g). Water was added after 2 hrs and the chiral auxiliary was removed by extraction with ethyl acetate. The aqueous phase was acidified with aqueous hydrochloric acid (10%) and crude S-(+)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid was extracted with tert.-butyl-methylether.

HPLC analysis (Chiralpak AD, mobile phase hexane/2-propanol/trifluoro acetic acid [92:8:0.1, vol/vol-%); flow 1.0 ml/min, detection 285 nm) indicated an enantiomeric ratio 93:7 (retention times 14.8 min and 11.5 min, respectively). The e.e. of 86% of the S-(+) enantiomer can be improved to >98.5% by recrystallization of the diastereomeric salts using "nitromix" (Angew. Chem. Int. Ed. Engl. 1998, Vol. 37, p. 2349) or (1R,2S)-(-)-ephedrine hemihydrate as described above. The S-(+)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid was isolated after acidification of aqueous solutions of the diastereomeric salts. It forms colourless crystals which gave an optical rotation of $[\alpha]_p^{22} = +21.6$ (c = 0.5, MeOH).

R-(-)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionic acid Conjugate organocuprate addition of phenylmagnesiumbromide to 3-[3-(2-benzyloxy-5-bromophenyl)-acryloyl]-(4S)-4-phenoyloxazolidin-2-one as described above for the S-(+)enantiomer gave crystalline R-(-)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid in an e.e. of 99.6% after two recrystalliza-

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tions, $[\alpha]_{p}^{22} = -21.7$ (c = 0.5, MeOH).

c) Synthesis of the R- and S- Enantiomers of Intermediate B

(i) Phenylpropanol Route



(±)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropan-1-ol

A solution of the methyl(±)-propionate (121.0 g) in 350 ml of dry tetrahydrofuran was slowly added under an atmosphere of nitrogen to a suspension of lithium aluminiumhydride (7.9 g) in tetrahydrofuran (350 ml). After stirring at room temperature for 18 hrs, 20% aqueous HCl was added dropwise and the product was isolated by repeated extraction with diethyl ether. The combined extracts were gradually washed with hydrochloric acid, sodium hydroxide solution, distilled water, and then dried (Na₂SO₄) to give a light yellow viscous oil (108.8 g, 96.3% yield) after evaporation which gradually crystallized, m.p. 73.8°C, tlc: (1) 0.47, (±)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropan-1-ol. NMR (CDCl₃): 37.52, 39.52, 60.84, 70.54, 113.54, 113.83, 126.29, 127.30, 127.51, 129.99, 128.24, 128.38, 129.99, 130.88, 135.69, 136.40, 143.53, 155.12.

The same product was obtained after reduction of (\pm) -3-(2benzyloxy-5-bromophenyl)-3-phenylpropionic acid with lithium aluminium hydride in tetrahydrofuran (30 min, 25°C), 31% yield. (±)-Toluene-4-sulph nic acid 3-(2-benzyloxy-5-bromophenyl)-3phenylpropyl ester

A cooled (5°C) solution of (\pm) -3-(2-benzyloxy-5-bromophenyl)-3-phenylpropan-1-ol (108.0 g) in dichloromethane (300 ml) was treated with pyridine (79.4 ml) and then p-toluenesulphonyl chloride (60.6 g) in dichloromethane (200 ml). After 18 hrs. at room temperature the solvent was removed in vacuum and the residue was extracted with diethyl ether. The extract was washed with hydrochloric acid, water, and dried over anhydrous sodium sulphate to give (\pm)-toluene-4-sulphonic acid 3-(2-benzyloxy-5-bromophenyl)-3-phenylpropyl ester as a light yellow oil after concentration under reduced pressure (140.3 g, 93.6% yield), tlc: (1) 0.66. NMR (CDCl₃): 21.67, 33.67, 39.69, 68.58, 70.28, 113.21, 113.76, 126.47, 127.84, 128.10, 128.25, 128.41, 128.51, 129.81, 130.26, 130.42, 132.91, 134.39, 136.41, 142.16, 155.07.

(±)-[3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropyl]-diisopropylamine

A solution of the (\pm) -toluenesulphonate $((\pm)$ -toluene-4-sulphonic acid 3-(2-benzyloxy-5-bromophenyl)-3-phenylpropyl ester, 139.3 g) in acetonitrile (230 ml) and N,N-diiso-propylamine (256 g) was refluxed for 97 hrs. The reaction mixture was then evaporated to dryness and the residue thus formed was partitioned between diethyl ether (500 ml) and aqueous sodium hydroxide (2 M, 240 ml). The organic phase was washed twice with water (250 ml) and then extracted with 1 M sulphuric acid. The aqueous phase was adjusted to about pH 12-13 and reextracted with ether (500 ml). The organic phase was washed with water, dried (Na₂SO₄) and evaporated to provide (\pm) -[3-(2-benzyloxy-5-bromophenyl)-3-phenylpropyl]-diisopropylamine as a brown and viscous syrup (94.5 g, 77.9%

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yield), tlc: (2) 0.49. NMR (CDCl₃): 20.65, 20.70, 36.70, 41.58, 43.78, 48.77, 70.24, 113.52, 126.02, 127.96, 128.20, 128.36, 129.82, 130.69, 136.34, 136.76, 144.20, 155.15.

(ii) Phenylpropionamide Route



S-(+)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropionyl chloride

Thionylchloride (4.5 g, 2.8 ml, 37.8 mmol) and some drops of dimethylformamide were added to a solution of $S - (+) - 3 - (2 - benzyloxy-5-bromophenyl) - 3-phenylpropionic acid (10.3 g, 25 mmol) in ethyl acetate (60 ml). The mixture was refluxed until tlc control indicated complete consumption of the starting material (2 hrs). Evaporation in vacuum gave the acid chloride as a light yellow liquid in almost quantitative yield (10.7 g). Conversion of an aliquot to the methyl ester showed a single spot in tlc (<math>R_f$ 0.54, solvent system (7)).

S-(+)-N,N-Diisopropyl-3-(2-benzyloxy-5-bromophenyl)-3phenylpropionamide

A solution of S-(+)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionyl chloride (9.6 g, 22.3 mmol) in ethyl acetate (40 ml) was added dropwise to a stirred and cooled (3°C) solution of diisopropylamine (6.4 g, 49.0 mmol) in 60 ml of ethyl acetate. The reaction was stirred for 18 hrs at room temper-

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ature and then washed with water, aqueous hydrochloric acid (1 M) and half saturated brine. The organic phase was dried (sodium sulphate) and evaporated to dryness. The colourless oily residue (10.7 g, 97% yield) of S-(+)-N,N-diisopropyl-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionamide showed a single spot on tlc: (R_f 0.70 (4)). NMR (CDCl₃): 18.42, 20.46, 20.63, 20.98, 39.51, 41.44, 45.76, 48.63, 70.00, 112.84, 113.64, 126.10, 126.45, 127.34, 127.78, 128.20, 128.36. 129.93, 130.59, 135.18, 136.52, 143.52, 155.17, 169.61.

(±)-N,N-Diisopropyl-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionamide

The amide was prepared from diisopropylamine and the racemic acid chloride as described above for the S-(+) enantiomer. The viscous colourless oil was dissolved in ethanol and the solution stored at -30°C. From this solution colourless crystals were obtained, m.p. 101.8°C.

(±) - [3-(2-Benzyloxy-5-bromophenyl) -3-phenylpropyl] -diisopropylamine

To a stirred solution of (\pm) -N,N-diisopropyl-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionamide (11.8 g) in 40 ml of dry tetrahydrofuran was added 1 M lithium aluminium hydride/tetrahydrofuran (36 ml). The reaction was refluxed for 4 hrs and then quenched with the dropwise addition of water. After removal of the precipitate the solvent was evaporated and the oily residue dissolved in diluted sulphuric acid. The aqueous phase was washed several times with diethyl ether, adjusted to pH 10-12 (aqueous NaOH), and extracted with diethyl ether. The extract was dried (sodium sulphate), filtered and evaporated to dryness in vacuum to leave 8.1 g (76.7%) of the title compound as a viscous colourless oil, tlc:(4) 0.86. The NMR spectrum corresponds to the product, obtained from the tosylate precursor (see above).

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S-(+)-[3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropyl]-diiso-
propylamine
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Repetition of the reaction sequence by using S-(+)-3-(2benzyloxy-5-bromophenyl)-3-phenylpropionic acid as the starting material gave S-(+)-[3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropyl]-diisopropylamine as a viscous colourless oil, $[\alpha]_{D}^{22} = +18.5$ (c = 10.0, ethanol), e.e. of a representative batch 99.4%

R-(-)-[3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropyl]-diisopropylamine

Repetition of the reaction sequence by using R-(-)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid as the starting material gave <math>R-(-)-[3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropyl]-diisopropylamine as a viscous colourless oil, $<math>[\alpha]_p^{22} = -17.3$ (c = 10.0, ethanol), e.e. of a representative batch 98.3%.

The optical purities were determined by chiral HPLC using Chiralpak OD columns.

(±)-4-Benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzoic acid hydrochloride

An ethereal Grignard solution, prepared from the above (\pm) amine (22.8 g), ethyl bromide (17.4 g) and magnesium (6.1 g) under an atmosphere of nitrogen was diluted with dry tetrahydrofuran (200 ml) and then cooled to -60°C. Powdered solid carbon dioxide (ca. 50 g) was then added in small portions and the green reaction mixture was warmed to room temperature. After the addition of an aqueous solution of ammonium chloride (200 ml, 10%) and adjustment of the aqueous phase to

pH 0.95, a white solid was recovered by filtration to provide (±)-4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzoic acid hydrochloride (14.7 g, 64.3% yield), m.p. 140°C (dec.), tlc: (2) 0.33. NMR (CD₃OD): 17.07, 18.77, 33.55, 43.27, 56.50, 71.50, 112.89, 124.10, 127.94, 129.07, 129.25, 129.34, 129.59, 129.66, 130.18, 131.60, 132.78, 137.60, 143.30, 161.11, 169.70.

(±)-[4-Benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)phenyl]-methanol

Intermediate A (n = 1)

The (±)-hydrochloride was converted into its methyl ester (MeOH, trace sulphuric acid, 6h reflux) and the free oily base thus obtained (28 g; tlc (2): R_f 0.46) was dissolved in dry diethyl ether (230 ml). This solution was slowly (2h) dropped under a nitrogen atmosphere to a suspension of lithium aluminium hydride (1.8 g) in ether (140 ml). After stirring for 18 hrs, the reaction was quenched by the addition of water (4.7 ml). The organic phase was dried over anhydrous sodium sulphate, filtered and evaporated to dryness to provide (±)-[4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-methanol (26 g, 98.9% yield), as an oil which gradually crystallized, m.p. 86.4°C, tlc: (2) 0.32. NMR (CDCl₃): 20.53, 20.61, 36.87, 41.65, 44.14, 48.82, 65.12, 70.09, 111.80, 125.77, 125.97, 126.94, 127.55, 128.08, 128.37, 128.44, 133.27, 134.05, 134.27, 137.21, 144.84.



Intermediate A

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(±) - [4-Benzyloxy-3-(3-diisopropylamino-1-phenylpropyl) - phenyl] - [C²H]methanol

Intermediate d_2 -A (n = 2)

Repetition of the above described reduction of the methylester of (\pm) -4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzoic acid by the use of lithium aluminium deuteride gave (\pm) -[4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-[C²H] methanol, colourless amorphous solid in 77% yield; tlc: (2) 0.33. NMR (CDCl₃): 20.46, 20.55, 36.77, 41.62, 44.09, 48.77, multiplett centred at 64.96, 70.05, 111.76, 125.72, 127.34, 128.03, 128.32, 128.38, 133.15, 133.99, 137.17, 144.80, 155.52.

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol

Intermediate B (n = 1)

A solution of Intermediate A (9.1 g) in methanol (100 ml) was hydrogenated over Raneynickel (4.5 g) under ambient conditions. After 5 hrs thin layer chromatography indicated complete hydrogenolysis. The catalyst was filtered off and the solution evaporated to dryness to leave an oil (6.95 g, 96.5% yield) which gradually solidified, (±)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol, m.p. 50°C, tlc: (2) 0.15. NMR (CDCl₃): 19.42, 19.83, 33.22, 39.62, 42.27, 48.27, 65.19, 118.32, 126.23, 126.55, 127.47, 128.33, 132.50, 144.47, 155.38. Hydrochloride: colourless crystalls, m.p. 187-190°C (with

decomposition)

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Intermediate B

S-(-)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol

Hydrogenolysis of $S - (-) - [4 - benzyloxy - 3 - (3 - diisopropylamino - 1 - phenylpropyl) - phenyl] - methanol (prepared from <math>S - (+) - 3 - (2 - benzyloxy - 5 - bromophenyl) - 3 - phenylpropionic acid as described for the racemic series) gave the title compound in 85% yield, colourless solid; m.p. <math>\geq 50 \,^{\circ}$ C, $[\alpha]_{p}^{22} = -19.8$ (c = 1.0, ethanol); NMR (CDCl₃): 19.58, 19.96, 33.30, 39.52, 42.10, 48.00, 65.40, 118.58, 126.31, 126.57, 127.16, 127.54, 128.57, 132.63, 132.83, 144.55, 155.52. S - (+) hydrochloride: colourless, non-hygroscopic solid, m.p. 186.4°C (dec.); $[\alpha]_{p}^{22} = +6.6$ (c = 0.5, water). NMR (DMSO-d₆): 16.58, 18.17, 31.62, 41.37, 45.90, 54.02, 63.07, 115.18, 126.05, 126.37, 128.03, 128.45, 129.04, 133.12, 143.88, 153.77.

R-(+)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol

Hydrogenolysis of R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-methanol (prepared from <math>R-(-)-3-(2-benzyloxy-5-bromophenyl)-3-phenylpropionic acid as described for the racemic series) gave the title compound in 87% yield,

colourless solid; m.p. $\geq 50 \,^{\circ}$ C, $[\alpha]_{D}^{22} = +21.3$ (c = 1.0, ethanol). R-(-) hydrochloride: colourless, non-hygroscopic solid, m.p. 179.8°C (dec.); $[\alpha]_{D}^{22} = -7.2$ (c = 0.5, water); NMR (DMSO-d₆): 16.59, 18.19, 31.64, 41.38, 45.92, 54.07, 63.08, 115.19, 126.07, 126.39, 128.04, 128.46, 129.05, 133.13, 143.89, 153.79. S-(+)-mandelate: m.p. 139.7°C, $[\alpha]_{D}^{21} = +38.3$ (c = 1.0, ethanol)

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxy-[²H₂]methyl-phenol

Intermediate d_2 -B (n = 2)

A stirred suspension of lithium aluminium deuteride (0.1 g, 2.38 mmol) in 5 ml of dry diethyl ether was treated during 30 min at room temperature under an atmosphere of dry nitrogen with a solution of (\pm) -4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzoic acid methyl ester (1.0 g, 2.17 mmol) in dry diethyl ether (5 ml). After an additional stirring at room temperature for 18 hrs the reaction was quenched by the dropwise addition of 0.17 ml of ${}^{2}\text{H}_{2}\text{O}$. The resultant precipitation was filtered off, washed with small portions of ether, and the combined organic phases were evaporated to dryness in vacuum to leave

(±) - [4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl) phenyl] - [²H₂]methanol

as slightly yellow, viscous oil which gradually crystallized, m.p. 84.1°C; tlc: (2) 0.33 (starting material 0.46), 0.725 g, 77.2% yield. NMR (CDCl₃): 20.46, 20.55, 36.77, 41.62, 44.09, 48.77, multiplett centred at 64.30, 70.05, 111.76, 125.72, 125.94, 126.92, 127.34, 127.71, 128.03, 128.32, 128.38, 133.15, 133.99, 137.17, 144.80, 155.52. A solution of the above (±)-[4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-[²H₂]methanol (0.129 g, 0.29 mmol) in a suspension of methanol (5 ml) and wet Raney-Nickel (0.1-0.2 g) was stirred at room temperature under an atmosphere of deuterium gas $(^{2}H_{2})$. After 1 hr tlc indicated complete disappearance of the starting material. The mixture was filtered, evaporated and the residue was redissolved in diethyl ether (5 ml). The solution was washed with water (2 x 5 ml), dried over sodium sulphate, filtered and evaporated to dryness to leave a pale yellow oil, 76.3 mg, in 74.6% yield, which gradually solidified to give a colourless solid of a m.p. range of 46-49°C. Tlc:(4) 0.57 (starting material 0.77). NMR (CDCl₃): 19.57, 19,94, 33.33, 39.56, 42.18, 48.07, 48.43, multiplett centred at 64.61, 118.47, 126.29, 126.58, 127.55, 127.94, 128.38, 132.53, 144.53, 155.37. GC-MS (P-CI, ammonia, TMS derivative): 488.43 (100%), 489.56 (70%), 490.56 (31%), 491.57 (8%).



Intermediate d,-B

n = 2, deuterium

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxy-[²H₂]methyl-phenol Intermediate d₂-B - 55 -

(iii) Heck-Cuprate-Route to Intermediate B



Intermediate B

N, N-Diisopropyl-acrylamide

A solution of acroyl chloride (42.2 g, 40.6 ml, 0.467 mol) in 125 ml of dichloromethane was slowly added to a cooled (0-5°C) solution of N,N-diisopropylamine in dichloromethane (500 ml). After 2 hrs the precipitated ammonium salt was filtered off and the filtrate was washed with 1M hydrochloric acid (3 x 100 ml), dried (sodium sulphate), and evaporated to dryness. N,N-diisopropyl-acrylamide was obtained as a slight yellow liquid in 48% yield and ca. 99% purity. NMR (CDCl₃): 20.54, 21.25, 45.66, 48.10, 125.62, 130.70, 166.17.

(E)-N,N-Diisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl)acrylamide

((E)-3-(2-Diisopropylcarbamoyl-vinyl)-4-methoxybenzoic acid methyl ester)

The reaction was carried out under an atmosphere of dry and oxygen-free nitrogen gas. All solvents and reagents were

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dried before use.

A stirred suspension consisting of N,N-dimethylglycine (6.0 mmol), anhydrous sodium acetate (40 mmol), methyl 3-bromo-4methoxybenzoate (20 mmol, 4.90 g), N,N-diisopropylacrylamide (24 mmol, 3.72 g), bis-(benzonitrile)-palladium-II chloride (1.5 mol%), and 20 ml of N-methyl-2-pyrrolidinone was heated at 130°C until no starting material could be detected by tlc (starting material methyl 3-bromo-4-methoxybenzoate: Rf 0.73; N, N-diisopropylacrylamide: Rf 0.46; solvent system (1)). After cooling to room temperature 50 ml of an aqueous 2N HCl solution was added. The reaction was diluted with dichloromethane (50 ml) and the precipitated grey palladium metal was filtered off. The organic phase was washed with five portions (50 ml each) of 2N aqueous hydrochloric acid, dried (MgSO_4) and evaporated to dryness. The remaining off-white solid was recrystallized from ethyl acetate/n-hexane to give 4.40 g (E) -N, N-diisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl) acrylamide in 69% yield, m.p. 139-140°C, tlc: (1) R_f 0.40. NMR (CD₂Cl₂): 21.22, 22.10, 46.39, 48.87, 52.59, 56.61, 111.42, 123.39, 123.78, 125.54, 130.32, 132.53, 135.07. MS (EI, DI, 105°C): 319 (M^{*} , 22), 304 (6%), 276 (8%), 219 (100%), 187 (18%), 160 (7%).

(±)-N,N-Diisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl)-3phenylpropionamide

 $((\pm)-3-(2-Diisopropylcarbamoyl-1-phenylethyl)-4-methoxy$ benzoic acid methyl ester)

The reaction was carried out under an atmosphere of dry and oxygen-free nitrogen gas. All solvents and reagents were dried before use.

A dark green solution of lithium diphenylcuprate was prepared by addition of phenyllithium solution (12 ml, 24 mmol, cyclo-

hexane/diethyl ether) to a cooled (0°C) and stirred suspension of copper-I bromide dimethylsulphide adduct (2.71 g, 13 mmol) in diethyl ether (40 ml). This solution was cooled to -78°C and then subsequently solutions were added of trimethylchlorosilane (1.5 ml, 12 mmol) in diethyl ether (5 ml) followed by the above cinnamide (3.19 g, 10.0 mmol, (E)-N,Ndiisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl)-acrylamide) in 10 ml of tetrahydrofuran. The reaction was stirred for one hour at -78°C, warmed to room temperature and then quenched by the addition of 150 ml of a saturated aqueous solution of ammonium chloride. After 90 min the organic phase was washed with two portions (100 ml) of half saturated aqueous sodium chloride, dried (MgSO₄) and evaporated to dryness. The yellow oily residue was dissolved in a minimum of ethyl acetate and purified by column chromatography on silica gel (mobile phase (1)). Evaporation of the combined fractions of the title compound gave

(±)-N,N-diisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl)-3phenylpropionamide

as a viscous slightly yellow syrup (1.8 g, 44% yield). NMR (CD₂Cl₂): 19.45, 19.56, 19.74, 38.86, 44.87, 47.92, 50.80, 54.76, 109.41, 121.32, 125.53, 128.10, 128.43, 128.78, 132.03, 143.20, 159.95, 165.95, 168.87. MS (EI, DI, 105°C): 397 (M^{+.}, 41%), 366 (5%), 322 (2%), 269 (3%), 255 (14%), 237 (7%), 165 (5%), 128 (12%), 91 (43%), 58 (100%).

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol

A solution of (±)-N,N-diisopropyl-3-(2-methoxy-5-methoxycarbonylphenyl)-3-phenylpropionamide (0.79 g, 2.0 mmol) in 20 ml of tetrahydrofuran was cooled to 5°C and then treated with 2.5 ml of 1M LiAlH₄/THF. After stirring at room tem-

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perature for 18 hrs. finely powdered aluminium chloride (0.3 g) was added and stirring was continued for additional 4 hrs. The reaction was quenched at 5°C by the dropwise addition of water followed by aqueous sodium hydroxide solution. The mixture was diluted with diethyl ether (150 ml) and the organic phase was washed with half saturated brine, dried (sodium sulphate), and evaporated to dryness to give the title compound as a solid off-white foam. Tlc (2) 0.16, m.p. 48-51°C. A portion of the material was converted into the hydrochloride (ethereal hydrochloric acid), m.p. 186-189°C (dec.).

Hydrogenolytic Deoxygenation of S-(-)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol

A mixture of S-(-)-2-(3-Diisopropylamino-1-phenylpropyl)-4hydroxymethylphenol (683 mg, 2.0 mmol, $[\alpha]_{p}^{22} = -19.8$ (c = 1.0, ethanol)), platinium-on-carbon catalyst (120 mg) and acetic acid (1.0 ml) was diluted with ethyl acetate (50 ml) and then hydrogenated at room temperature under a pressure of 4 bar hydrogen gas for 5 hrs. The catalyst was filtered off and the filtrate was evaporated to leave an oil. The residue was redissolved in dichloromethane (25 ml) and the solution was washed with aqueous sodium hydrogencarbonate solution. The organic phase was concentrated to dryness and the oily residue taken up in ethanol (7 ml). Addition of D-(-)-tartaric acid (300 mg) and storage of the clear solution at -25°C gave colourless crystals (310 mg) of S-(-)-2-(3-diisopropylamino-1-phenylpropyl)-4-methylphenol D-(-) hydrogentartrate in 33% yield, tlc: (4): 0.66 (starting material 0.31), $\left[\alpha\right]_{D}^{22}$

 $= -26.7 (c = 1.0, methanol). NMR (CD_3OD): 17.98, 18.37, 20.69, 33.68, 43.12, 56.33, 74.17, 116.31, 127.51, 129.11, 129.50, 129.70, 129.89, 130.41, 144.57, 153.67, 176.88.$

A portion of the tartrate was treated with aqueous sodium hydrogencarbonate solution and the free base was isolated in quantitative yield as a colourless oil by extraction with ethyl acetate and evaporation of the extract. $[\alpha]_{D}^{22} = -26.3$ (c = 1.0, methanol).

Preferred intermediates in the processes for the preparation of the 3,3-diphenylpropylamines according to the present invention are:

(±)-3-(2-Benzyloxy-5-bromophenyl)-3-phenylpropanoic acid and its salts,

R-(-)-(2-Benzyloxy-5-bromophenyl)-3-phenylpropanoic acid and its salts,

S-(+)-(2-Benzyloxy-5-bromophenyl)-3-phenylpropanoic acid and its salts,

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxy-

 $[C^{2}H_{2}]$ methyl-phenol,

S-(-)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxy-[C²H₂]methyl-phenol,

R-(+)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxy- $[C^{2}H_{2}]$ methyl-phenol and their salts.

3. Examples

a) Phenolic monoesters

aa) General procedure

Esters of Carboxylic Acids

A stirred solution of (±)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol (Intermediate B, 1.71 g, 5.01 mmol) and acid chloride (5.00 mmol carboxylic acid monochloride for compounds of formula II, 2.50 mmol for compounds

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of formula II') in 60 ml of dichloromethane was cooled to 0°C and then triethylamine (0.502 g, 4.96 mmol for compounds of formula II, 1.05 g, 9.92 mmol for compounds of formula II'), dissolved in 10 ml of dichloromethane, was added dropwise during 5-10 min. Stirring was continued for 18 hrs at room temperature, and then the mixture was washed successively with water (25 ml), aqueous sodium hydrogen carbonate (5%, 25 ml), and water (25 ml). The organic phase was then dried (sodium sulphate) and evaporated under reduced pressure and at low temperature. The oily residues thus formed were finally exposed to high vacuum (2-4 hrs.) to remove traces of residual solvents.

The esters of formula II or II' were obtained as colourless to light yellow solids or viscous syrups in purities between 90% and 99% (tlc, HPLC, NMR).

Esters of N-Acylamino Acids

Phenolic Monoesters

To a solution of the respective amino acid (2.0 mmol) in 0.7 ml to 5 ml of N,N-dimethylformamide and 0.5 ml of triethylamine was added at 5°C in one portion methyl chloroformate (2.0 mmol, 288 mg). After stirring for 2 hrs. at the same temperature the cooling bath was removed and a solution of Intermediate B (2.0 mmol, 682 mg) in 5 ml of dichloromethane and triethylamine (0.5 ml) was added. The reaction was allowed to stir for 2-8 hrs and then diluted with diethyl ether (70 ml). Solid precipitates were filtered off and the mixture was washed with aqueous sodium hydrogen sulphate solution (5%) and water. After drying (sodium sulphate), filtration and evaporation in vacuum the residue was purified by flash chromatography on silica gel (eluent: solvent system (4)). Nacylamino acid esters were obtained as viscous oils or waxy solids in yields between 24% and 73%.

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bb) Salt formation (Example hydrochloride)

A cooled (0°C) solution of 4.94 mmol amino base in 30 ml of dry diethyl ether was treated under an atmosphere of nitrogen with 4.70 mmol (monoamines of formula II) or 9.4 mmol (diamines of formula II') ethereal (1 M) hydrochloric acid. The oily precipitation was washed repeatedly with dry ether and then evaporated in high vacuum. The residual product solidificated in most cases as an amorphous foam. The highly hygroscopic solids show a wide melting range above 100°C (with decomposition).

The following compounds were prepared according to the method described above and their analytical data are listed below:

(±)-Acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester, tlc: R_f 0.47 (4), NMR (CDCl₃): 20.36, 20.68, 20.97, 36.59, 42.35, 43.83, 48.76, 64.58, 122.69, 125.61, 126.22, 126.71, 127.96, 128.34, 136.82, 138.97, 143.73, 147.77, 169.24; GC-MS/P-CI (ammonia, trimethylsilyl derivative): 456.8 (100%), 398.4 (4%)

(±) -Propionic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester, tlc: R_f 0.52 (4); NMR (CDCl₃): 20.44, 20.64, 27.67, 36.67, 42.21, 43.87, 48.78, 64.70, 122.71, 125.62, 126.52, 126.78, 127.97, 128.53, 136.86, 138.82, 143.82, 147.86, 172.68; GC-MS/P-CI (ammonia, trimethylsilyl derivative): 470.38 (100%), 398.4 (4%)

(±)-n-Butyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester, tlc: R_f 0.43 (4); NMR (CDCl₃): 13.77, 18.40, 20.43, 20.51, 20.59, 36.15, 36.82, 42.16,

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43.90, 48.83, 49.20, 64.58, 122.66, 125.98, 126.17, 126.74, 127.33, 127.94, 128.33, 136.79, 138.91, 143.82, 171.88; GC-MS/N-Cl (methane, trimethylsilyl derivative): 482.3 (20%), 412.3 (100%), 340.1 (33%), 298.1 (89%), 234.7 (15%); GC-MS/P-Cl (methane, trimethylsilyl derivative): 484.5 (100%), 468.4 (62%), 394.3 (22%); GC-MS/P-CI (ammonia, trimethylsilyl derivative): 484.4 (100%), 398.4 (3%)

(±)-Isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester, tlc: R_f 0.43 (4); NMR (CDCl₃): 18.99, 19.11, 20.54, 34.21, 36.88, 41.84, 43.91, 48.78, 64.61, 122.54, 125.57, 126.14, 126.81, 127.94, 128.34, 136.84, 138.84, 143.89, 147.85, 175.36

R-(+)-Isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester

Tlc: R_f 0.38 (4), starting material: 0.26; colourless oil (yield 95%); NMR (CDCl₃): 19.02, 19.14, 19.96, 20.61, 34.26, 36.92, 41.87, 43.90, 48.80, 64.84, 122.63, 122.63, 125.64, 126.19, 126.92, 127.98, 128.39, 136.96, 138,76, 143.93, 147.97, 175.39.

Hydrochloride: colourless hygroscopic solid; $[\alpha]_{D}^{20} = +5.5$ (c = 1.0, chloroform); NMR (CDCl₃): 17.03, 17.53, 18.30, 18.52, 18.95, 19.12, 31.23, 34.10, 41.69, 45.40, 54.22, 54.47, 64.00, 122.32, 126.62, 126.81, 127.40, 128.06, 128.70, 133.88, 140.64, 142.25, 147.81, 175.89.

(±)-2,2-Dimethylpropionic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester, tlc: R_f 0.49 (1); NMR (CDCl₃): 20.46, 20.66, 26.53, 27.34, 37.12, 39.21, 41.46, 43.98, 48.81, 64.65, 122.42, 125.58, 126.16, 126.92, 128.37, 134.27, 136.92, 138.82, 143.97, 148.02, 176.97; GC-MS/P-CI

(ammonia, trimethylsilyl derivative): 498.8 (100%), 482.5 (10%), 398.4 (4%)

(±)-2-Acetamidoacetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester ((±)-2-[Diisopropylamino)-1-phenylpropyl]-4-(hydroxymethyl)phenyl 2-(acetylamino)acetate) NMR (CD₃OD): 20.33, 20.61, 22.17, 30.54, 42.39, 48.62, 51.04, 64.88, 117.99, 124.73, 125.51, 127.01, 127.75, 129.31, 131.63, 137.33, 146.67, 147.43, 171.47, 173,82

(±)-Cyclopentanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.66 (4), starting material Intermediate B (0.50), colourless oil, yield: 82%. NMR (CDCl₃): 20.42, 25.87, 30.25, 36.57, 41.89, 43.97, 47.15, 49.02, 64.63, 122.56, 125.60, 126.16, 126.81, 127.60, 127.94, 128.35, 128.77, 136.74, 138.88, 143.85, 147.92, 175.05.

(±)-Cyclohexanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.67 (4), starting material Intermediate B (0.50), colourless oil, yield: 93%. NMR (CDCl₃): 20.27, 25.40, 25.74, 29.03, 29.16, 36.29, 41.82, 43.31, 44.08, 49.36, 64.62, 122.56, 125.68, 126.22, 126.92, 127.92, 128.38, 136.65, 139.00, 143.72, 147.86, 174.40.

(±)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester

Tlc: R_f 0.31 (4); colourless syrup (99% yield, purity >
95%);gradually crystallized upon refrigeration; NMR (CDCl₃):
20.41, 20.51, 36.65, 42.42, 43.85, 48.79, 64.70, 122.79,
125.74, 126.17, 126.83, 128.13, 128.28, 128.58, 129.48,
130.25, 133.62, 137.21, 139.10, 143.67, 148.00, 164.99.

R-(+)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester tlc R_f 0.30 (4); colourless syrup Hydrochloride: colourless amorphous solid; [α]_D²⁰ = +14.9 (c = 1.0, chloroform); NMR (CDCl₃): 17.06, 17.53, 18.25, 18.61, 31.23, 42.19, 45.49, 54.26, 54.53, 64.09, 122.55, 126.77, 127.13, 127.58, 128.10, 128.50, 128.72, 128.78, 129.02, 130.17, 133.96, 134.27, 140.81, 142.13, 147.91, 165.40.

(±)-4-Methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.30 (4), starting material Intermediate B: 0.24; yield: quantitative, viscous light yellow oil; NMR (CDCl₃): 20.32, 20.50, 21.78, 36.13, 42.35, 43.98, 49.29, 64.66, 122.79, 125.81, 126.19, 126.70, 127.04, 128.30, 129.32, 129.76, 130.29, 136.94, 139.20, 143.61, 144.46, 148.04, 165.07. LC-MS: 459 (M^{**}, 3.5%), 444 (17%), 223 (2.5%), 195 (2%), 119 (48%), 114 (100%).

(±)-2-Methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester viscous colourless oil, tlc: (4) 0.64 (starting material R_f 0.51), yield 84%. NMR (CDCl₃): 20.44, 20.53, 21.86, 22.01, 36.74, 42.36, 43.87, 48.81, 64.76, 122.93, 123.11, 125.71, 126.12, 126.88, 128.10, 128.48, 130.76, 131.26, 131.70, 132.03, 132.79, 137.28, 139.00, 141,73, 143.72, 148.04, 165.25. LC-MS: 459 (M⁺, 21%), 444 (100%), 326 (1%), 223 (10%), 213 (6%), 195 (9%), 165 (14%), 115 (94%), 91 (99%).

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(±)-2-Acetoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester colourless syrup, tlc: (4) 0.47 (starting material R_f 0.51), yield 82%. NMR (CDCl₃): 20.39, 20.57, 20.96, 36.92, 42.29, 43.88, 48.87, 64.64, 122.39, 122.64, 124.05, 125.80, 126.11, 126.75, 128.09, 128.32, 132.23, 134.66, 137.27, 139.32, 143.64, 147.63, 151.37, 162.72, 169.73. LC-MS: 503 (M⁺⁻, 7%), 488 (59%), 446 (6%), 326 (22%), 223 (9%), 213 (9%), 195 (9%), 163 (14%), 121 (100%), 114 (88%).

 (\pm) -1-Naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester

colourless viscous oil, tlc: (4) 0.57 (starting material R_f 0.51), yield 82%. NMR (CDCl₃): 20.46, 20.58, 36.82, 42.46, 43.89, 48.76, 64.81, 122.98, 124.51, 125.64, 125.79, 125.98, 126.15, 126.44, 126.94, 128.12, 128.36, 128.65, 131.37, 131.82, 133.98, 134.45, 137.44, 139.08, 143.73, 148.13, 165.49. LC-MS: 495 (M⁺, 8%), 480 (100%), 213 (7%), 165 (8%), 155 (95%), 127 (100%), 114 (90%).

(±)-2-Naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester

colourless slightly yellow viscous oil, tlc: (4) 0.57 (starting material R_f 0.51), yield 71%. NMR (CDCl₃): 20.47, 20.59, 36.71, 42.59, 43.85, 48.81, 64.82, 122.89, 126.89, 127.89, 128.19, 128.41, 128.68, 129.50, 132.03, 132.55, 135.87, 137.22, 139.08, 143.83, 148.20, 165.14. LC-MS: 495 (M^{+.}, 7%), 480 (98%), 223 (8%), 213 (6%), 195 (6%), 165 (8%), 155 (96%), 127 (100%), 114 (81%).

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(±)-4-Chlorobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.54 (4), starting material Intermediate B: 0.44; yield: quantitative, viscous light yellow oil; NMR (CDCl₃): 20.34, 20.50, 36.41, 42.51, 43.84, 48.93, 64.66, 122.72, 125.82, 126.88, 127.27, 128.06, 128.56, 128.96, 131.60, 133.80, 136.95, 139.30, 140.16, 143.60, 147.87, 164.10. LC-MS: 479 (M⁺⁺, 1.5%), 464 (10%), 223 (2%), 195 (2%), 165 (1.5%), 139 (25%), 114 (100%).

(±)-4-Methoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.47 (4), starting material Intermediate B: 0.42; yield: 89%, viscous light yellow oil; NMR (CDCl₃): 20.31, 20.47, 36.43, 42.39, 43.90, 48.97, 55.53, 64.71, 121.79, 122.86, 125.72, 126.14, 126.79, 128.11, 128.27, 131.27, 131.77, 132.36, 132.84, 137.15, 139.01, 143.74, 148.08, 163.92, 164.71. LC-MS: 475 (M^{**}, 3.5%), 460 (20%), 223 (2%), 195 (2%), 135 (48%), 114 (100%).

(±)-2-Methoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.40 (4), starting material Intermediate B: 0.42; yield: 98%, viscous light yellow oil; NMR (CDCl₃): 20.29, 20.42, 36.50, 41.92, 44.02, 49.09, 55.95, 64.72, 119.10, 120.20, 122.86, 125.64, 126.10, 126.82, 128.06, 128.30, 132.38, 134.32, 137.11, 139.01, 143.87, 148.00, 159.82, 164.40. LC-MS: 475 (M⁺⁻, 3.5%), 460 (18%), 223 (1%), 195 (1%), 135 (49%), 114 (100%).



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(±)-4-Nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.44 (4), starting material Intermediate B: 0.42; yield: 78%, viscous yellow oil which slowly solidified; m.p. 123.6°C; NMR (CDCl₃): 20.47, 20.62, 36.52, 42.66, 43.70, 48.75, 64.69, 122.61, 123.72, 125.91, 126.33, 127.04, 128.02, 128.37, 131.32, 134.86, 136.83, 139.55, 143.56, 147.75, 150.93, 163.04. LC-MS: 490 (M⁺, 1.5%), 475 (15%), 327 (0.8%), 223 (3%), 195 (3%), 150 (15%), 114 (100%).

(±)-2-Nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester Tlc: R_f 0.32 (4), starting material Intermediate B: 0.42; yield: 92%, viscous yellow oil which slowly solidified; NMR (CDCl₃): 20.39, 20.50, 36.74, 42.14, 43.89, 48.71, 48.92, 64.59, 122.15, 123.95, 124.18, 125.89, 126.25, 127.23, 127.99, 128.39, 129.95, 132.95, 133.08, 136.72, 139.62, 143.64, 147.63, 148.15, 163.90. LC-MS: 490 (M⁺⁺, 1%), 475 (11%), 327 (2.5%), 223 (2.5%), 195 (3%), 165 (3%), 150 (7%), 114 (100%).

(±)-N-Acetylglycine 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester/(±)-2-Acetamidoacetic acid 2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester ((±)-2-[Diisopropylamino-1-phenylpropyl]-4-(hydroxymethyl)phenyl 2-(acetylamino)acetate) NMR (CD₃OD): 20.33, 20.61, 22.17, 30.54, 42.39, 48.62, 51.04, 64.88, 117.99, 124.73, 125.51, 127.01, 127.75, 129.31, 131.63, 137.33, 146.67, 147.43, 171,47, 173.82.

(±)-Malonic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl]ester, tlc: R_f 0.38 (4); NMR (CDCl₃): 20.52, 20.62, 20.69, 36.95, 41.84, 42.82, 43.89, 48.23,


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64.83, 123.37, 127.36, 127.97, 128.42, 128.38, 129.06, 131.55, 137.50, 138.90, 148.23, 148.32, 160.54

(±)-Succinic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl]ester, tlc: R_f 0.40 (4); NMR (CDCl₃): 20.54, 20.63, 20.73, 30.69, 36.91, 41.80, 43.92, 48.20, 64.81, 122.60, 127.41, 127.93, 128.39, 129.31, 131.80, 136.73, 138.92, 143.82, 148.17, 168.01

(±)-Pentanedioic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl]ester, tlc: R_f 0.43; NMR (CDCl₃): 20.47, 20.60, 32.87, 36.93, 41.82, 43.90, 48.22, 64.81, 64.83, 122.85, 127.39, 127.99, 128.35, 129.31, 131.84, 136.98, 138.94, 143.80, 147.40, 169.05

(±)-Hexanedioic acid bis-[2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl]ester, tlc: R_f 0.43; NMR (CDCl₃): 20.64, 23.40, 34.37, 36.95, 41.84, 43.88, 48.25, 64.87, 122.88, 127.34, 127.97, 128.39, 129.33, 131.80, 136.99, 138.94, 143.82, 147.65, 168.72

b) Identical diesters

(\pm)-Identical diesters (formula III) were prepared and worked up as described above with the exception that 2.4 mmol of both triethylamine and acyl chloride (R¹-COCl) were used. The physical properties were similar to the bases and salts described above.

Diesters of N-acylaminoacids were prepared as described for phenolic monoesters with the exception that an additional molar equivalent of acylating agent (mixed acid anhydride) was used. WO 99/58478

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In particular, the following compounds were prepared and their analytical data are given below:

(±)-Formic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester, tlc: R_f 0.65 (4). This diester was prepared from mixed formic acetic anhydride and Intermediate B as described for other substrates previously (F. Reber, A. Lardon, T. Reichstein, Helv. Chim. Acta 37: 45-58 [1954])

(±)-Acetic acid 4-acetoxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, tlc: R_f 0.76 (4); GC-MS/P-CI (ammonia): 426.3 (100%), 368.3 (22%); GC-MS/P-CI (methane, trimethylsilyl derivative): 426.4 (64%), 410.3 (16%), 366.3 (100%); hydrochloride, NMR (DMSOd₆)- 16.50, 16.76, 18.05, 20.94, 21.04, 27.02, 31.39, 41.28, 45.26, 53.80, 65.21, 123.39, 126.84, 127.61, 127.85, 128.70, 134.41, 135.49, 142.68, 148.20, 169.32, 170.42

(±) -Propionic acid 2-(3-diisopropylamino-1-phenylpropyl)-4propionyloxymethylphenyl ester, tlc: R_f 0.82 (4); NMR (CDCl₃): 20.53, 20.73, 21.14, 27.66, 36.73, 42.10, 43.68, 48.65, 65.75, 122.65, 126.10, 127.01, 127.70, 128.34, 128.78, 133.73, 136.81, 143.76, 148.45, 172.45, 174.21; GC-MS/P-CI (ammonia): 454.8 (100%), 438.5 (9%), 382.4 (27%)

(±)-n-Butyric acid 4-n-butyryloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, tlc: R_f 0.86 (4); NMR (CDCl₃): 13.70, 13.76, 18.44, 20.53, 20.69, 21.13, 36.14, 36.76, 37.09, 42.08, 43.73, 48.71, 65.64, 122.81, 125.97, 126.97, 127.92, 128.35, 128.77, 133.78, 136.99, 143.76, - 70 -

148.41, 171.68, 173.40; GC-MS/P-CI (ammonia): 482.8 (100%), 396.4 (67%)

(±)-Isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4isobutyryloxymethylphenyl ester, tlc: R_f 0.83 (4), NMR (CDCl₃): 18.97, 19.10, 20.64, 20.67, 34.01, 34.23, 36.98, 41.72, 43.70, 48.65, 65.61, 122.50, 126.18, 126.73, 127.92, 128.13, 128.36, 133.90, 137.01, 143.85, 148.41, 175.17, 176.81; GC-MS/N-CI (methane): 480.3 (15%); GC-MS/P-CI (methane): 482.5 (63%), 466.4 (18%), 394.3 (100%)

(±)-2,2-Dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-(2,2-dimethylpropionyloxy)-benzyl ester, Tlc: R_f 0.96 (4); NMR (CDCl₃): 20.44, 20.75, 27.09, 27.24, 37.18, 38.68, 39.15, 41.25, 43.66, 48.20, 65.50, 122.36, 126.32, 127.22, 127.48, 127.83, 128.29, 133.99, 136.98, 143.87, 148.37, 176.70, 178.10; GC-MS/P-CI (methane): 510.5 (76%), 494.5 (21%), 408.4 (100%)

(±)-Benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester, tlc: R_f 0.80 (4); NMR (CDCl₃): 20.62, 36.95, 41.72, 43.89, 48.23, 66.76, 122.22, 125.33, 127.36, 127.62, 127.89, 127.89, 127.97, 128.38, 129.49, 130.52, 130.64, 131.15, 131.22, 131.98, 136.38, 137.66, 143.82, 148.95, 164.77, 166.60

(+)-Benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester

Hydrochloride: colourless solid; tlc: (4) 0.70, $[\alpha]_{D}^{20} =$ +24.2 (c = 1.0, chloroform). NMR (DMSO-d₆): 16.52, 17.99, 18.06, 26.99, 31.32, 53.94, 65.98, 123.58, 127.65, 127.98, 128.62, 128.90, 129.02, 129.45, 129.71, 130.10, 133.64, 134.32, 134.55, 135.60, 142.52, 148.37, 164.53, 165.76.

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c) Mixed diesters

Mixed diesters (formula IV) were prepared by acylation of the respective benzylic or phenolic monoesters. Working up and physical properties corresponded to the bases and salts described above.

In particular, the following compounds were prepared and their analytical data are given below:

(±) -Acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester, tlc: R_f 0.76 (4); NMR (CDCl₃): 20.62, 20.91, 33.25, 42.20, 42.28, 48.23, 70.70, 122.96, 127.36, 127.97, 128.38, 128.73, 132.02, 135.41, 137.11, 143.81, 149.35, 161.34, 168.95

(±)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester, tlc: R_f 0.74 (4); NMR (CDCl₃): 20.60, 36.93, 41.72, 43.89, 48.23, 70.71, 122.50, 125.33, 127.30, 127.89, 127.97, 128.36, 129.57, 130.65, 131.13, 132.05, 135.41, 136.66, 143.80, 149.15, 161.35, 164.78

(±)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4acetoxymethylphenyl ester Viscous colourless oil, tlc: R_f 0.70 (4); NMR (CDCl₃): identical with R-(+) enantiomer, see below.

 $\begin{array}{l} R_{-}(+) - Benzoic \ acid \ 2 - (3 - diisopropylamino - 1 - phenylpropyl) - 4 - \\ acetoxymethylphenyl \ ester \\ tlc: \ R_f \ 0.70 \ (4) \\ Hydrochloride: \ colourless \ non-hygroscopic \ solid \ \left[\alpha\right]_{D}^{20} = \\ +27.1 \ (c \ = \ 1.0, \ chloroform) . \ NMR \ (CDCl_3): \ 17.14, \ 18.53, \end{array}$

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21.04, 31.51, 42.25, 46.27, 54.74, 65.58, 123.18, 127.07, 127.55, 127.61, 127.99, 128.80, 130.22, 134.14, 134.81, 135.27, 141.44, 148.54, 165.19, 170.81.

 (\pm) -Isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester, tlc: $R_f 0.77$ (4); NMR (CDCl₃): 18.99, 19.12, 20.65, 21.05, 34.24, 37.02, 41.79, 43.79, 48.72, 65.98, 122.75, 126.76, 127.14, 127.94, 128.39, 128.84, 133.55, 137.04, 143.84, 148.56, 170.84, 175.18

(+)-Isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester colourless oil Hydrochloride: colourless hygroscopic solid; [α]_D²⁰ = +14.6 (c = 1.0, chloroform); NMR (CDCl₃): 16.89, 17.04, 18.31, 18.54, 18.92, 19.06, 20.95, 31.49, 34.07, 41.64, 46.17, 54.55, 65.49, 122.91, 126.93, 127.48, 127.83, 128.74, 134.50, 134.88, 141.61, 148.44, 170.67, 175.63.

(±)-2,2-Dimethylpropionic acid 4-acetoxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, tlc: R_f 0.80 (4); NMR (CDCl₃): 20.63, 20.93, 27.19, 33.25, 37.49, 42.21, 42.25, 48.22, 67.37, 123.18, 127.36, 127.84, 128.39, 131.16, 137.34, 143.84, 148.29, 168.93, 178.40

(±)-2,2-Dimethylpropionic acid 4-acetoxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, tlc: R_f 0.81 (4); NMR (CDCl₃): 20.60, 20.79, 27.09, 36.93, 37.35, 41.85, 42.29, 48.25, 65.91, 122.36, 127.37, 127.99, 128.39, 129.38, 132.69, 136.00, 136.85, 143.80, 170.45, 176.60 - 73 -

d) Benzylic monoesters

A mixture consisting of Intermediate B (80 mg, 0.23 mmol), vinyl ester (0.4 ml), tert.-butyl methylether (18 ml), and lipase enzyme (1.0 g) was gently shaken at room temperature. Benzylic formate, acetate, and n-butyrate were prepared from the corresponding vinyl ester donors using SAM I lipase (Amano Pharmaceutical Co.). Benzoylation was achieved with vinyl benzoate in the presence of Lipozym IM 20 (Novo Nordisk), whereas pivalates and isobutyrates were obtained from the corresponding vinyl esters under catalysis of Novozym SP 435 (Novo Nordisk). Tlc analysis indicated after 2 - 24 hrs complete disappearence of the starting material ($R_f = 0.45$ (3)). The mixture was filtered and then evaporated under high vacuum (< 40°C) to give the carboxylic acid (R^1 -CO₂H) salts of the respective benzylic monoesters as colourless to light yellow oils.

In particular, the following compounds were prepared and their analytical data are given below:

(±)-Formic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.25 (2); NMR (CDCl₃): 19.43, 33.24, 39.61, 42.25, 48.21, 68.44, 118.09, 127.34, 127.66, 128.31, 128.39, 133.97, 144.47, 156.63, 161.32

(±)-Acetic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.26 (2); NMR (CDCl₃): 19.45, 20.96, 33.26, 39.63, 42.27, 48.23, 63.59, 118.00, 127.36, 128.33, 128.48, 128.53, 129.13, 131.59, 133.88, 144.49, 155.74, 170.44 - 74 -

(±)-Propionic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.45 (2); NMR (CDCl₃): 19.02, 19.43, 27.58, 33.20, 39.61, 42.25, 48.21, 64.08, 118.30, 125.30, 127.03, 127.39, 128.31, 130.12, 134.22, 144.51, 155.64, 173.22

(±)-Butyric acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.54 (2); NMR (CDCl₃): 13.58, 18.40, 19.45, 33.29, 35.88, 39.65, 42.23, 48.25, 63.96, 118.32, 124.55, 126.20, 127.35, 128.32, 129.91, 134.22, 144.50, 155.60, 169.05

(±)-Isobutyric acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.56 (4); NMR (CDCl₃): 19.09, 19.45, 33.28, 33.59, 39.65, 42.29, 48.25, 64.63, 118.35, 125.35, 127.03, 127.38, 128.35, 128.49, 129.79, 134.22, 144.52, 155.65, 175.48

(±)-2,2-Dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-hydroxybenzyl ester, tlc: R_f 0.61 (4); NMR (CDCl₃): 19.41, 27.15, 33.24, 37.46, 39.61, 42.25, 48.21, 65.10, 118.30, 125.32, 127.00, 127.34, 128.31, 129.42, 134.18, 144.47, 155.61, 178.39

(±) -Benzoic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester, tlc: R_f 0.77 (4); NMR (CDCl₃): 18.01, 19.40, 33.24, 39.60, 42.40, 48.20, 66.93, 117.13, 127.18, 127.81, 128.33, 129.98, 130.17, 132.96, 133.58, 142.33, 156.95, 166.60

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e) Ethers and silyl ethers

A mixture of Intermediate B (3.4 g, 10 mmol), methanesulphonic acid (2 ml, 31 mmol), and alcohol R^{10} -OH (50-150 ml) was stirred at room temperature until no starting material was detectable (2-24 hrs). After evaporation to dryness (< 35°C) the residue was redissolved in aqueous sodium hydrogen carbonate solution (100-200 ml, 5%, w/v) and the solution was extracted with ethyl acetate (75 ml). The organic phase was separated, dried (Na₂SO₄), filtered and evaporated to give bases of formula VI ($R^{11} = H$) as colourless to light yellow oils.

Mixed ester ether derivatives, e.g. of Intermediate A, were prepared by benzylic acylation of phenolic ethers, such as Intermediate A, according to the procedure described for examples of the structure of formula IV.

Hydrochlorides:

Molar equivalents of bases of formula VI $(R^{11} = H)$, dissolved in tert.-butyl methylether, and ethereal hydrochloric acid were combined at room temperature. Oily precipitates were separated and dried in vacuum, crystalline hydrochlorides were isolated and recrystallized from acetonitrile or acetone to give colourless crystalline material.

In particular, the following compounds were prepared and their analytical data are given below:

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-methoxymethylphenol, tlc: R_f 0.61 (4); GC-MS/P-CI (methane, trimethylsilyl derivative): 428.4 (100%), 412.3 (49%), 396.3 (52%); hydrochloride: amorphous hygroscopic colourless solid; m.p. 161°C; NMR (CD₃OD): 17.39/18.75 (broad signals), 33.79, 43.13, 56.47, 58.00, 75.59, 116.19, 120.79, 127.62, 129.04, 129.14, 129.42, 129.55, 130.43, 144.32, 155.85

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-ethoxymethylphenol, tlc: R_f 0.72 (4); GC-MS/P-CI (ammonia, trimethylsilyl derivative): 444.8 (100%), 398.4 (6%); hydrochloride: colourless non-hygroscopic crystals, m.p. 158-161°C, NMR (CD₃OD): 15.43, 17.12, 18.82, 33.80, 56.49, 66.49, 73.62, 116.19, 127.63, 128.99, 129.13, 129.36, 129.55, 130.58, 130.75, 144.32, 155.77

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-propoxymethylphenol, NMR (CDCl₃): 18.62, 19.44, 23.10, 33.24, 39.61, 42.26, 48.22, 71.87, 73.94, 117.78, 124.95, 127.35, 127.57, 128.32, 128.47, 133.66, 134.23, 144.48, 155.25

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-isopropoxymethylphenol, NMR (CDCl₃): 19.44, 22.32, 33.27, 39.65, 42.29, 48.25, 69.28, 72.10, 117.90, 127.38, 128.03, 128.41, 131.10, 133.76, 134.37, 144.51, 154.65. Hydrochloride: colourless crystals, m.p. 140.4°C, tlc (4) 0.61. LC-MS: 383 (6%, [M-HCl]^{+.}), 368 (11%), 324 (1%), 223 (6%), 195 (3%), 165 (2%), 155 (5%), 114 (100%). NMR (DMSOd₆): 16.57, 18.09, 18.19, 22.29, 31.58, 41.25, 45.87, 53.97, 69.26, 69.92, 115.28, 126.34, 127.08, 127.25, 127.96, 128.45, 129.07, 129.70, 132.31, 143.88, 154.22.

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-butoxymethylphenol, NMR (CDCl₃): 13.75, 19.44, 19.75, 32.24, 33.28, WO 99/58478

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39.60, 42.20, 48.20, 72.45, 117.87, 125.50, 127.29, 128.39, 133.70, 134.30, 144.47, 155.36

(±)-Acetic acid 2-(3-Diisopropylamino-1-phenylpropyl)-4methoxymethylphenyl ester, NMR (CDCl₃): 19.99, 20.62, 20.90, 33.33, 42.30, 48.21, 58.41, 75.94, 122.92, 127.37, 127.95, 128,35 131.85, 136.99, 138.81, 143.88, 147.88, 168.95

(±) -Acetic acid 2-(3-Diisopropylamino-1-phenylpropyl)-4ethoxymethylphenyl ester, NMR (CDCl₃): 15.49, 19.94, 20.95,
33.23, 42.25, 48.25, 65.70, 73.73, 122.63, 127.46, 127.95,
128.36, 131.65, 136.79, 139.71, 143.80, 147.66, 168.99

(±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-trimethylsilanyloxymethylphenol, NMR (CDCl₃): 0.10, 19.40, 19.43, 33.25, 39.65, 42.25, 48.20, 64.93, 117.90, 124.90, 126.60, 127.35, 128.35, 128.48, 133.80, 137.15, 144.49, 155.28

(±)-Diisopropyl-[3-phenyl-3-(2-trimethylsilanyloxy-5trimethylsilanyloxymethylphenyl)-propyl]amine, NMR (CDCl₃): 0.10, 0.29, 19.40, 19.53, 33.28, 41.19, 42.27, 48.25, 66.40, 121.37, 127.36, 128.25, 128.50, 136.42, 144.10, 154.98

(±)-[3-(3-Diisopropylamino-1-phenylpropyl)-4-trimethylsilanyloxyphenyl]methanol, NMR (CDCl₃): 0.29, 0.33, 19.40, 19.53, 33.27, 41.16, 42.27, 48.23, 65.22, 118.04, 124.99, 126.52, 127.30, 128.25, 134.16, 136.80, 144.14, 155.06

(±)-Diisopropyl-[3-(5-methoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropyl]amine, NMR (CDCl₃): 0.28, 0.32, 19.39, 19.43, 33.28, 41.22, 42.33, 48.19, 58.40, 75.95, 117.68, 124.92, 126.60, 127.35, 128.25, 128.55, 134.00, 136.47, 144.16, 155.09 - 78 -

(±) -Diisopropyl-[3-(5-ethoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropyl]amine, NMR (CDCl₃): 0.28, 0.31, 15.50, 19.42, 19.58, 33.29, 41.17, 42.25, 48.20, 65.70, 72.48, 117.50, 124.75, 126.39, 127.39, 128.25, 128.50, 134.99, 136.28, 144.19, 154.28

(±) - [4-(tert.-Butyl-dimethylsilanyloxy) - 3-(3-diisopropylamino-1-phenylpropyl) - phenyl]methanol, R_f 0.65 (3)

(±) -Acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-3-(3diisopropylamino-1-phenylpropyl)-benzyl ester, NMR (CDCl₃): -4.92, -5.00, 19.40, 19.49, 20.40, 20.83, 23.49, 33.25, 41.22, 42.25, 48.25, 72.55, 81.55, 121.24, 124.88, 127.40, 128.26, 128.44, 128.48, 133.37, 135.74, 144.11, 155.20

(±)-4-(tert.-Butyl-dimethylsilanyloxymethyl)-2-(3-diisopropylamino-1-phenylpropyl)-phenol, tlc: R_f 0.70 (3); GC-MS/N-CI (methane, trimethylsilyl derivative): 526.5 (59%), 454.3 (100%), 412.2 (14%), 340.1 (42%); GC-MS/P-CI (methane, trimethylsilyl derivative): 528.6 (100%), 512.5 (85%), 470.43 (10%), 396.3 (31%)

(±) -Acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-2-(3diisopropylamino-1-phenylpropyl)-phenyl ester, NMR (CDCl₃): -4.77, -4.88, 19.15, 20.65, 20.93, 24.77, 33.25, 42.20, 48.20, 67.90, 122.79, 125.15, 127.44, 127.90, 128.41, 136.99, 140.55, 143.85, 147.86, 168.95

(±)-{3-[2-(tert.-Butyl-dimethylsilanyloxy)-5-(tert.-butyl-dimethylsilanyloxymethyl)-phenyl]-3-phenylpropyl}-diisopropylamine, tlc: R_f 0.94 (3); GC-MS/N-CI (methane): 568.6 (62%), 454.3 (100%), 438.2 (10%), 340.2 (58%), 324.8 (16%), 234.7

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(78%); GC-MS/P-CI (methane): 570.6 (70%), 554.5 (52%), 512.5 (18%), 438.4 (24%)

(±)-Acetic acid 4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, tlc: R_f 0.56 (5); GC-MS/P-CI (ammonia): 474.4 (100%), 416.4 (54%); NMR (CDCl₃): 20.44, 20.56, 21.07, 36.73, 41.53, 44.01, 48.79, 66.43, 70.00, 111.61, 125.75, 127.34, 127.55, 127.76, 127.90, 128.03, 128.27, 128.39, 133.98, 136.98, 144.63, 156.05, 170.94

(±)-Benzoic acid 4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, tlc: R_f 0.87 (4); NMR (CDCl₃): 20.54, 20.60, 36.80, 41.51, 43.95, 48.67, 66.83, 70.04, 111.66, 125.76, 127.35, 127.45, 127.78, 128.06, 128.27, 128.30, 128.42, 128.85, 129.66, 130.55, 132.86, 134.05, 137.03, 144.75, 156.08, 166.46; GC-MS/P-CI (ammonia): 536.5 (100%), 416.4 (42%)

(±)-Isobutyric acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester, tlc: R_f 0.77 (4); NMR (CDCl₃): 19.01, 20.62, 20.65, 34.04, 36.85, 41.54, 43.97, 48.71, 66.15, 70.06, 111.62, 125.79, 125.96, 126.97, 127.24, 127.55, 127.81, 128.08, 128.34, 128.45, 134.05, 137.10, 144.79, 156.00, 177.01; GC-MS/P-CI (ammonia): 502.4 (100%), 416.4 (49%)

f) Carbamates and carbonates

Mono N-substituted carbamates

A solution of 4.0 mmol of Intermediate B, benzylic ether (formula VI, $R^{11} = H$) or monoester of formula II in dichloromethane (20 ml) was treated at room temperature for 16 hrs with isocyanate (4.8 mmol) or diisocyanate (2.2 mmol). After

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washing with 10 ml aqueous sodium hydrogen carbonate (5%, w/v), drying (Na₂SO₄) and evaporation oily residues or colourless solids of the free bases were obtained.

N-disubstituted carbamates

N,N-dialkyl-carbamoylchloride (4.4 mmol) was dissolved in dichloromethane and dropped into a cooled (0°C) and stirred mixture consisting of Intermediate B (4.0 mmol), dichloromethane (30 ml) and triethylamine (7.0 mmol, 0.71 mg, 1 ml). Stirring was continued for 6 hrs. The mixture was then washed with 5 portions (10 ml) of aqueous sodium hydrogen carbonate, dried (sodium sulphate), filtered and evaporated to give the carbamates as colourless oils or solids.

Bis-carbamates were prepared in like manner using Intermediate B and excess isocyanate (4.8 mmol) and toluene as solvent at 65°C over 18 hrs. Carbonates were prepared and worked-up according to the methods described for the preparation of compounds of formulae II to IV. Alkyl chloroformates were used as acylation re-

Hydrochlorides:

agents.

The oils or solids were redissolved in tetrahydrofuran (10 ml). Addition of ethereal hydrochloric acid and evaporation to dryness in high vacuum gave crystalline or amorphous carbamate hydrochlorides.

In particular, the following compounds were prepared and their analytical data are given below:

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 (\pm) -N-Ethylcarbamic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, tlc: R_f 0.38 (4); GC-MS/P-CI (ammonia, trimethylsilyl derivative): 486.8 (100%), 413.4 (5%), 398.4 (6%); hydrochloride: m.p. 64°C (with decomposition); NMR (DMSO-d₆): 15.16, 16.68, 18.05, 18.13, 25.33, 31.26, 35.46, 53.94, 62.65, 67.22, 123.04, 125.70, 126.72, 127.86, 128.67, 135.42, 136.02, 140.07, 142.98, 147.53, 154.52

(±)-N,N-Dimethylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester NMR (CDCl₃): 20.34, 20.66, 30.51, 36.33, 36.77, 42.00, 48.28, 50.21, 65.65, 119.83, 123.44, 125.19, 126.60, 127.38, 127.54, 129.31, 136.62, 143.33, 150.99, 155.67.

(±)-N,N-Diethylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester NMR (CDCl₃): 20.54, 20.66, 30.49, 35.61, 42.42, 48.31, 50.20, 65.56, 119.43, 123.40, 125.33, 126.66, 126.99, 127.05, 136.30, 143.27, 149.13, 154.97

(±) -N-Phenylcarbamic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester; NMR (CDCl₃): 20.52, 20.61, 36.91, 39.44, 42.25, 48.22, 62.66, 118.36, 119.46, 123.50, 125.32, 127.11, 127.99, 130.15, 132.63, 139.65, 141.33, 145.16, 152.21, 156.00

(±)-[2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenoxycarbonylamino]acetic acid ethyl ester hydrochloride Tlc: R_f 0.14 (4); m.p. colourless crystals (from acetone, 21% yield); NMR (CDCl₃): 16.76, 16.86, 18.45, 20.96, 31.37, 42.20, 46.13, 54.56, 65.50, 123.10, 126.98, 127.66, 128.72,

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130.14, 134.05, 134.72, 135.22, 141.37, 148.47, 165.12, 170.71

(±)-N-Ethylcarbamic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-N-ethylcarbamoyloxybenzyl ester, tlc: R_f 0.36 (3); NMR (CDCl₃): 15.00, 19.23, 19.40, 33.26, 36.00, 39.62, 42.35, 48.12, 65.95, 118.30, 125.45, 127.08, 128.33, 130.37, 134.24, 144.44, 155.44, 157.74

(±)-N,N-Dimethylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N,N-dimethylcarbamoyloxybenzyl ester NMR (CDCl₃): 20.59, 20.66, 30.59, 35.96, 36.40, 36.74, 36.98, 42.03, 48.26, 50.09, 67.09, 119.04, 123.23, 123.49, 125.01, 126.67, 127.72, 129.33, 133.65, 143.43, 150.99, 155.63.

(±)-N,N-Diethylcarbamic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-N,N-diethylcarbamoyloxybenzyl ester NMR (CDCl₃): 13.31, 13.64, 13.89, 20.33, 20.71, 31.57, 37.97, 41.55, 42.37, 48.46, 51.00, 67.23, 120.00, 123.39, 124.82, 126.31, 126.95, 127.33, 150.36, 157.18, 158.97.

(±) - {4-[2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenoxycarbonylamino]-butyl}-carbamic acid 2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester (formula VII', X = Y = NH, n = 4) tlc: R_f 0.60 (6); dihydrochloride m.p. 142.5-145.6°C

(±)-Carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester ethyl ester, $R_t = 0.67$ (4)

(\pm)-Carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4ethoxycarbonyloxymethylphenyl ester ethyl ester, R_f 0.87 (4) - 83 -

g) Intramolecular cyclic diesters via Ring Closing Metathesis (RCM)



Example:

(±)-Pent-4-enoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-(pent-4-encyloxymethyl)-phenyl ester (x = y = 2) A cooled (4°C) mixture of pent-4-enoic acid, isobutyl chloroformate, and triethylamine (each 5.84 mmol) in 10 ml of dichloromethane was stirred 5 hrs under an atmosphere of dry nitrogen gas. The cooling bath was then removed and both triethylamine (1.46 mmol) and 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol (1.46 mmol) were added in one portion. After 18 hrs the mixture was diluted with dichloromethane (30 ml), washed several times with water and finally aqueous 5% sodium hydrogen carbonate solution. After drying (sodium sulphate), filtration and evaporation the oily residue was re-dissolved in a small volume of a solvent mixture consisting of ethyl acetate/heptane/triethylamine (65/30/5, vol.-%) and applied on a silica gel flash chromatography column. Elution of the column with the same solvent mixture, collection of the appropriate fractions, and evaporation of the combined fractions gave (\pm) -pent-4-enoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-(pent-4-enoyloxy-

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methyl)-phenyl ester as a pale yellow syrupy oil (50% yield), tlc: (4) 0.75. NMR (CDCl₃): 18.95, 20.77, 27.75, 28.87, 33.58, 36.83, 42.13, 43.72, 48.71, 65.85, 70.55, 115.47, 115.99, 122.45, 126.26, 127.08, 127.96, 128.11, 128.83, 133.73, 136.38, 136.79, 137.04, 143.77, 148.46, 171.11, 172.78.

Intramolecular cyclic diesters of $1, \omega$ -dioic acids and Intermediate B

Example

Intramolecular cyclic diester of octane-1,8-dioic acid and 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenol Grubbs catalyst (benzylidene-bis-(tricyclohexylphosphine)dichlororuthenium, 16 mg, 0.002 mmol, 2 mol-%) was added to a solution of (\pm) -pent-4-enoic acid 2-(3-diisopropylamino-1phenylpropyl)-4-(pent-4-enoyloxymethyl)-phenyl ester (483 mg, 0.96 mmol) in dichloromethane (150 ml) and the mixture was refluxed for 96 hrs. under an atmosphere of nitrogen gas, after which all of the starting material was consumed as indicated by tlc. The mixture was filtered through a short pad of basic alumina, and the solvent was removed in vacuum. Flash chromatography (solvent system (4)) afforded the intermediate intramolecular cyclic diester of oct-4-ene-1,8dioic acid and 2-(3-diisopropylamino)-1-(phenylpropyl)-4hydroxymethyl-phenol (324 mg) as a colourless syrup (tlc: (4) R_f 0.68) in 71% yield, mixture of two geometrical isomers. NMR (CDCl₃, major isomer): 19.24, 20.61, 23.11, 25.62, 30.55, 33.53, 35.02, 42.41, 48.29, 50.20, 65.30, 114.46, 124.33, 125.58, 127.15, 128.70, 129.29, 131.10, 132.46, 139.54, 146.76, 147.98, 173.76, 174.39.

A portion of this material (140 mg) was dissolved in ethyl acetate (10 ml) and hydrogenated at room temperature in the

presence of palladium-on carbon catalyst to afford the intramolecular cyclic diester of octane-1,8-dioic acid and 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenol in essentially quantitative yield, 139 mg, colourless oil, tlc: (4) 0.71.

NMR (CDCl₃): 19.36, 20.73, 24.84, 25.28, 28.90, 29.70, 30.57, 33.72, 34.37, 42.39, 48.26, 50.20, 65.26, 114.45, 124.37, 127.11, 128.67, 129.29, 131.18, 132.45, 139.52, 146.77, 147.69, 173.90, 174.15.

Poly-co-DL-Lactides of Intermediate B

All reagents were dried over P_2O_5 in vacuum (< 1 mbar) and at room temperature. The reactions were carried out at room temperature in an atmosphere of dry, oxygen-free nitrogen.

Low Molecular Weight Copolymer

A 15% solution of n-butyllithium (0.36 ml) was injected through a rubber septum into a stirred solution of 2-(3diisopropylamino-phenylpropyl)-4-hydroxymethyl-phenol (100 mg, Intermediate B) and DL-dilactide (1.5 g) in 15 ml of dry toluene. The polymerization was allowed to proceed for 4 days at room temperature. Distilled water (10 ml) was then added in order to terminate the polymerization. The organic phase was separated and slowly dropped into 200 ml of methanol. The precipitated colourless oil was treated with water (100 ml) and then dried in high vacuum for 48 hrs. The copolymer was obtained in 72.7% yield. NMR analysis (see below) indicated an average molecular weight range of ${\tt M}_n$ 2000-4000 and a weight content of Intermediate B of about 8.4% (NMR). Tlc analysis showed the absence of monomeric Intermediate B. Gel permeation chromatography (GPC) analysis showed a Mw of 1108 and a Mn of 702.

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High Molecular Weight Copolymer

The high molecular weight copolymer was prepared as described above with the exception that 3.0 g of DL-dilactide was used. Precipitation by methanol gave a fluffy white solid which was carefully washed with water and then dried as desribed to give the copolymer in 81% yield. NMR analysis (see below) indicated an average molecular weight range of M_n 4000-8000 and a weight content of Intermediate B of about 2.0%. The analysis showed the absence of monomeric Intermediate B. Gel permeation chromatography (GPC) showed a Mw of 9347 and a Mn of 6981. Differential scanning calorimetry (DSC) provided a Tg of 42.5°C.

NMR Analysis

The ¹H NMR resonance signals of the poly-lactyl chain were clearly separated from the copolymeric part of Intermediate B (solvent CDCl₃):

CH₃ resonances of the poly-lactyl chain: 1.30-1.60 ppm CH resonances of the poly-lactyl chain: 5.10-5.30 ppm CH resonances of the connecting lactyl units with the two hydroxy groups of Intermediate B: 4.8-5.0 ppm and 5.5-5.7 ppm.

Polymer bound Intermediate B: 1.06-1.11 (CH₃), 2.20-2.30 (CH₂CH₂), 2.40-2.80 (NCH₂), 3.30-3.50 (NCH), 4.45-4.55 (CHCH₂), 4.70-4.80 (CH₂-OCO-lactyl), 6.70-7.30 (aryl CH).



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h) Inorganic ester

Example:

(±) -Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4sulphooxymethyl-phenyl ester Hydrochloride

To a stirred solution of chlorosulphonic acid (116 mg, 1.0 mmol) in 5 ml of dry diethyl ether was slowly added at 0°C a solution of (±)-benzoic acid 2-(3-diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl ester (445.6 mg, 1.0 mmol) in 3 ml of dry diethyl ether. The gel formed immediately during the addition was stirred at room temperature until it became a crystalline consistency (ca. 1 hr). The precipitate was washed several times with diethyl ether and then dried in vacuum to give 0.52 g (46% yield) colourless crystals, m.p. 63-65°C. NMR (CDCl₃): 16.85, 17.03, 18.32, 18.49, 32.01, 42.29, 46.23, 55.23, 55.50, 69.24, 122.52, 126.94, 127.15, 129.04, 129.76, 130.25, 133.89, 134.93, 136.85, 141.87, 147.80, 165.19.

Benzylic 1-O-β-D-glucuronide of 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol
 ((±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-(1β-D-glucuronosyloxymethyl)-phenol)



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A solution of methyl 2,3,4-triacetyl-1- α -D-glucuronosylbromide (2.07 g, 4.64 mmol) in 24 ml of dry toluene was cooled to -25°C under an atmosphere of nitrogen and then treated with a solution of (±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester in 7 ml of toluene. To this mixture was added dropwise with stirring and under protection from light a solution of silver triflate in 14 ml of toluene (immediate formation of a white precipitate). The cooling bath was removed after 15 min and pyridine (0.38 ml) was added. The mixture was diluted with ethyl acetate (200 ml), filtered and the clear yellow filtrate was washed sequentially with aqueous solutions of sodium thiosulphate (5%), sodium hydrogen carbonate (5%), and sodium chloride (20%). The solution was dried with solid sodium sulphate, treated with charcoal, filtered and evaporated to dryness. The waxy residue was re-dissolved in a small volume of a solvent mixture consisting of ethyl acetate/heptane/triethylamine (65/30/5, vol.-%) and applied on a silica gel flash chromatography column. Elution of the column with the same solvent mixture, collection of the appropriate fractions, and evaporation of the combined fractions gave (±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4- $(2, 3, 4-triacetyl-1\beta$ -D-glucuronosyloxymethyl)-phenyl ester, colourless syrup, tlc (4) 0.70 (starting amine: 0.31, bromo glycoside: 0.23), yield 14%. NMR (CDCl₃, mixture of diastereomers): 20.41, 20.50, 20.60, 20.65, 20.84, 36.49, 42.44, 43.65, 48.73, 52.91, 69.46, 70.43, 71.12, 72.11, 72.60, 73.99, 99.19, 122.91, 126.23, 126.38, 126.54, 127.60, 127.92, 128.06, 128.09, 128.31, 128.59, 129.38, 130.22, 133.67, 134.31, 137.41, 143.52, 148.46, 164.82, 167.26, 169.21, 169.39, 170.07.

A portion (350 mg) of the above described material was dissolved and hydrolyzed in a solvent mixture consisting of tetrahydrofuran/methanol/aqueous potassium hydroxide (excess, 12 hrs, 22°C). The mixture was evaporated, re-dissolved in 5 ml of water and the pH was adjusted to 8.3. This solution was applied to a chromatography column charged with prewashed XAD 2 resin (50 g). The column was washed with water (ca. 250 ml) and then eluted with methanol. Collection of the appropriate methanol fractions, and evaporation of the combined fractions in vacuum gave 111 mg of $(\pm) -2 - (3 - diisopropylamino - 1 - phenylpropyl) - 4 - (1\beta - D - glucurono$ syloxymethyl) - phenol, sodium salt, $amorphous colourless solid, m.p. <math>\cong$ 110-124°C (dec.), tlc (4)

0.12. NMR (CD₃OD, major isomer): 19.43, 19.67, 33.26, 39.63, 42.27, 48.23, 69.76, 73.55, 74.70, 75.95, 78.03, 107.64, 117.95, 125.51, 127.36, 128.33, 133.83, 134.77, 144.49, 155.36, 176.76.

II. Incubations of different compounds of the invention with human liver S 9-fraction

a) Incubation of unlabelled substrates

A pooled human liver S 9-preparation was used to show the invitro metabolism of different compounds of the invention and to prove the generation of the active metabolite by enzymatic process.

The pooled human liver S 9-preparation was delivered by Gentest, Woburn, MA, USA.

In a routine assay, 25 μ L of pooled human liver S9 (20 mg protein/mL, H961, Gentest, Woburn, MA, USA) was incubated

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for 2 hrs at 37°C with 40 μ M substrate in a 0.01 M potassium phosphate buffer in the presence of NADPH (1 mM). The reaction was quenched by the addition of concentrated perchloric acid and precipitating protein was removed by centrifugation. The supernatant was adjusted to pH 3 with concentrated potassium phosphate solution, centrifuged, and injected into the HPLC for analysis of the respective products.

The analysis of the non-deuterated compounds was performed by a routine High Pressure Liquid Chromatography (HPLC) method with UV-detection.

The incubation results expressed in (%) of theoretical turnover are presented in Fig. 1.

They ranged from 96 to 63.2%. The formation of the active metabolite is dependent on the substituents both at the benzylic and phenolic side of the respective compounds.

Explanation:

The prodrugs introduced in the assay show the following chemical structure:



chemical structure	X-/-Y	
AcO-/-OAc	means	acetate
HO-/-OBut	means	hydroxy and <u>n</u> -butyrate
HO-/-OiBut	means	hydroxy and iso-butyrate

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iBut0-/-OiBut	means	iso-butyrate
ButO-/-OBut	means	<u>n</u> -butyrate
Prop0-/-OProp	means	proprionate
HO-/-OProp	means	hydroxy and proprionate
HO-/-OAc	means	hydroxy and acetate
BzO-/-OBz	means	benzoate and benzoate
AcO-/-OiBut	means	acetate and isobutyrate
AcO-/-OBz	means	acetate and benzoate

b) Incubation of labelled substrates

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The metabolic degradation of the unlabelled hydroxy metabolite (i.e. Intermediate B) and the deuteriated hydroxymetabolite (Intermediate d_2B) were compared in vitro. Used were the respective enantiomers and the racemates.

The hydroxy metabolite and the deuteriated hydroxy-metabolite expressed significant differences in the rate to produce the corresponding carboxylic acid.

The measurement was performed with an incubation time of 3 hrs at 37.0°C in a concentration of 40 μ M. The formation of the carboxylic acid from the deuteriated hydroxy-metabolite showed a significantly decreased velocity of 10%.

These in-vitro experiments indicate a reduced metabolic turnover of the deuteriated compound in vitro, which may result in higher plasma levels.

c) Receptor binding study

WO 94/11337 discloses that the active metabolite has high affinity to muscarinic receptors in the guinea-pig bladder. Different compounds of the present invention were tested in - 92 -

a well established standardized assay, measuring the binding of $[{}^{3}H]$ -methylscopolamine to recombinant human M3 receptors. BSR-M3H cells transfected with a plasmid encoding the human muscarinic M3 receptor were used to prepare membranes in modified Tris-HCl pH 7.4 buffer using standard techniques. An aliquot of the membrane preparation was incubated with $[{}^{3}H]$ methylscopolamine in the presence or absence of different concentrations of several compounds of the invention for 60 minutes at 25°C. Nonspecific binding was estimated in the presence of 1 μ M atropine. Membranes were filtered and washed three times and the filters were counted to determine the amount of $[{}^{3}H]$ -methylscopolamine specifically bound. The following table shows the IC₅₀ values of several compounds of the invention in the M3 receptor binding assay.

Prodrug	IC ₅₀ [nM]		
(+) HO-/-OH	8.7		
(-) HO-/-OH	1300		
(+)HO-/-OiBut	159		
(+)HO-/-OBz	172		
BzO-/-OBz	2400		
AcO-/-OiBut	3600		
AcO-/-OBz	5400		

Interaction with human M3 receptors in vitro

These data clearly showed that derivatization at the phenolic hydroxyl moiety results in an about 20 times less potent binding. If both functionalities are derivatized, the binding is even more dramatically reduced. Furthermore, it is demonstrated that the enantiomers of the active metabolite exhibit a marked difference in the binding characteristics to human M3 receptors. - 93 -

The compounds were tested for their anticholinergic activity in a standard tissue assay, the guinea-pig ileum. A segment of ileum was obtained from Duncan Hartley guinea-pigs which were sacrified by cervical dislocation. The tissue was placed under 1 g tension in a 10 ml bath containing Krebs' solution (pH 7.4, 32°C) and the concentration-dependent ability of different compounds to reduce the methacholine-induced (0.6 μ M) contractile response was recorded. The IC₅₀ values for the different substances were calculated and examples are presented in the following table.

Prodrug	IC ₅₀ [nM]
(+) HO-/-OH	20
(-) HO-/-OH	680
(+)HO-/-OiBut	57
(+)HO-/-OBz	180
(+)BzO-/-OBz	220
(+)AcO-/-OiBut	240

Anticholinergic activity in guinea-pig ileum in vitro

These data confirm the results obtained in the receptor binding assays and demonstrate that the anticholinergic activity of the compounds decreases with increased derivatization.

d) Biological membranes

Different compounds of the invention were tested for their ability to penetrate the human skin (200 μ m thick) in the "Flow through cell" at 32°C according to Tiemessen et al. (Acta Pharm. Technol. 1998; 34:99-101). Phosphate buffer (pH 6.2) was used as the acceptor medium. Samples were drawn at different time points and analysed by RP-HPLC with UV de- 94 -

tection (220 nm). Permeation profiles were plotted and mean flux rates of different substances were calculated by linear regression analysis. The data obtained for different compounds of the invention are summarized in the following table.

Penetration	through	human	skin

Prodrug	Flux rate
	$[\mu g/cm^2/24hrs]$
но-/-он	3
HO-/-OiBut	150
iButO-/-OiBut	60
Prop0-/-OProp	70

Disubstitution of the hydroxy group of HO-/-OH leads to a ≥ 20-fold increase in skin permeation in relation to the parent HO-/-OH. Suprisingly monosubstitution of the penolic hydroxy group resulted in even higher 50-fold penetration rate through human skin.

Taken together, these biological data clearly demonstrate that the compounds of the invention have a reduced affinity to bind to human muscarinic M3 receptors. They exhibit an increased penetration through biological membranes, e.g. the human skin, and they are rapidly transformed to the active metabolite, once they have entered the systemic circulation as shown by the in vitro metabolism by the human liver S9 preparation.

Thus, the antimuscarinic prodrugs according to this invention showed a profile that defines excellent prodrugs.

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<u>Claims</u>

1. 3,3-Diphenylpropylamines of the general formulae I and
VII':



wherein R and R' are independently selected from

a) hydrogen, C_1-C_6 alkyl, C_3-C_{10} cycloalkyl, substituted or unsubstituted benzyl, allyl or carbohydrate; or

 b) formyl, C₁-C₆ alkylcarbonyl, cycloalkylcarbonyl, substituted or unsubstituted arylcarbonyl, preferably benzoyl;
 or

c) C_1-C_6 alkoxycarbonyl, substituted or unsubstituted aryloxycarbonyl, benzoylacyl, benzoylglycyl, a substituted or unsubstituted amino acid residue; or

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d)

e)

 R^4 N-CO- wherein R^4 and R^5 independently R^5

represent hydrogen, C_1-C_6 alkyl, substituted or unsubstituted aryl, preferably substituted or unsubstituted phenyl, benzyl or phenoxyalkyl wherein the alkyl residue has 1 to 4 carbon atoms and wherein R^4 and R^5 may form a ring together with the amine nitrogen; or

 R° N-SO₂ wherein R° and R^{7} independently

represent C_1 - C_6 alkyl, substituted or unsubstituted aryl, preferably substituted or unsubstituted phenyl, benzyl or phenoxyalkyl wherein the alkyl residue has 1 to 6 carbon atoms; or

f) an ester moiety of inorganic acids,

g) $-SiR_aR_bR_c$, wherein R_a , R_b , R_c are independently selected from C_1-C_4 alkyl or aryl, preferably phenyl,

with the proviso that R' is not hydrogen, methyl or benzyl if R is hydrogen, R is not ethyl if R' is hydrogen,

X represents a tertiary amino group of formula Ia



Formula la

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wherein R^8 and R^9 represent non-aromatic hydrocarbyl groups, which may be the same or different and which together contain at least three carbon atoms, and wherein R^8 and R^9 may form a ring together with the amine nitrogen,

Y and Z independently represent a single bond between the $(CH_2)_n$ group and the carbonyl group, O, S or NH,

A represents hydrogen (^{1}H) or deuterium (^{2}H) ,

n is 0 to 12

and

their salts with physiologically acceptable acids, their free bases and, when the compounds can be in the form of optical isomers, the racemic mixture and the individual enantiomers.

2. 3,3-Diphenylpropylamines as claimed in claim 1, wherein X is



3. 3,3-Diphenylpropylamines as claimed in claim 2 selected from phenolic monoesters represented by the general formulae II and II'



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wherein R^{1} represents hydrogen, $C_{1}-C_{6}$ alkyl or phenyl.

4. 3,3-Diphenylpropylamines as claimed in claim 3 selected from:

(±)-formic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-propionic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-n-butyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

R-(+)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2,2-dimethylpropionic acid 2-(3-diisopropylamino-1-

phenylpropyl)-4-hydroxymethylphenyl ester,

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(±)-2-acetamidoacetic acid 2-(3-diisopropylamino-1-
phenylpropyl)-4-hydroxymethylphenyl ester,
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(±)-cyclopentanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-cyclohexanecarboxylic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

R-(+)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-4-methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-methylbenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-acetoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-1-naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-2-naphthoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester,

(±)-4-chlorobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-4-methoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-methoxybenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-4-nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-2-nitrobenzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-malonic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester,

(±)-succinic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester,

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(±)-pentanedioic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester,

(±)-hexanedioic acid bis-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethyl-phenyl]ester.

3,3-Diphenylpropylamines as claimed in claim 2 selected 5. from identical diesters represented by the general formula III



Formula ill

wherein R^2 is defined as in claim 3.

3,3-Diphenylpropylamines as claimed in claim 5 selected 6. from:

(±)-formic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester,

(±)-acetic acid 4-acetoxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester,

(±)-propionic acid 2-(3-diisopropylamino-1-phenylpropyl)-4propionylcxymethylphenyl ester,

(±)-n-butyric acid 4-n-butyryloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester,

(±)-isobutyric acid 2-(3-diisopropylamino-1-phenylpropyl)-4isobutyryloxymethylphenyl ester,

(±)-2,2-dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-(2,2-dimethyl-propionyloxy)-benzyl ester, (±)-benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester,



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R-(+)-benzoic acid 4-benzoyloxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-pent-4-enoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-(pent-4-enoyloxymethyl)-phenyl ester, cyclic oct-4-ene-1,8-dioate of Intermediate B, cyclic octane-1,8-dioate of Intermediate B, poly-co-DL-lactides of Intermediate B.

7. 3,3-Diphenylpropylamines as claimed in claim 2 selected from mixed diesters represented by the general formula IV



Formula IV

wherein R^1 is defined as in claim 3

and

 R^2 represents hydrogen, C_1-C_6 alkyl or phenyl

with the proviso that R^1 and R^2 are not identical.

8. 3,3-Diphenylpropylamines as claimed in claim 7 selected from:

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester,

(±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4formyloxymethylphenyl ester,

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(±)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4acetoxymethylphenyl ester, R-(+)-benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4acetoxymethylphenyl ester, (±)-isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1phenylpropyl)-phenyl ester, R-(+)-isobutyric acid 4-acetoxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-2,2-dimethylpropionic acid 4-acetoxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester, (±)-2,2-dimethylpropionic acid 4-acetoxymethyl-2-(3-diisopropylamino-1-phenylpropyl)-phenyl ester, (±)-benzoic acid 4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester.

9. 3,3-Diphenylpropylamines as claimed in claim 2 selected from benzylic monoesters represented by the general formula $\rm V$



Formula V

wherein R^1 is defined as in claim 3.

10. 3,3-Diphenylpropylamines as claimed in claim 9 selected from:

(±)-formic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-acetic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

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(±)-propionic acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-butyric acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-isobutyric acid 3-(3-diisopropylamino-1-phenylpropyl)-4hydroxybenzyl ester,

(±)-2,2-dimethylpropionic acid 3-(3-diisopropylamino-1phenylpropyl)-4-hydroxybenzyl ester,

(±)-benzoic acid 3-(3-diisopropylamino-1-phenylpropyl)-4-hydroxybenzyl ester.

11. 3,3-Diphenylpropylamines as claimed in claim 2 selected from ethers and silyl ethers represented by the general formula VI



wherein at least one of R^{10} and R^{11} is selected from C_1-C_6 alkyl, benzyl or $-SiR_aR_bR_c$ as defined in claim 1 and the other one of R^{10} and R^{11} may additionally represent hydrogen, C_1-C_6 alkylcarbonyl or benzoyl.

12. 3,3-Diphenylpropylamines as claimed in claim 11 selected from:

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-methoxymethylphenol,
(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-ethoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-propoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-isopropoxymethylphenol,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-butoxymethylphenol,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4methoxymethylphenyl ester,

(±)-acetic acid 2-(3-diisopropylamino-1-phenylpropyl)-4ethoxymethylphenyl ester,

(±)-2-(3-diisopropylamino-1-phenylpropyl)-4-trimethylsilanyloxymethylphenol,

(±)-diisopropyl-[3-phenyl-3-(2-trimethylsilanyloxy-5-trimethylsilanyloxymethylphenyl)-propyl]-amine,

(±) - [3-(3-diisopropylamino-1-phenylpropyl) -4-trimethylsilanyloxyphenyl] -methanol,

(±) -diisopropyl-[3-(5-methoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropylamine,

(±)-diisopropyl-[3-(5-ethoxymethyl-2-trimethylsilanyloxyphenyl)-3-phenylpropylamine,

(±) - [4 - (tert. -butyl-dimethylsilanyloxy) - 3 - (3 - diisopropyl amino-1-phenylpropyl) - phenyl] - methanol,

(±)-acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-3-(3diisopropylamino-1-phenylpropyl)-benzyl ester,

(±)-4-(tert.-butyl-dimethylsilanyloxy)-3-(3-diisopropylamino-1-phenylpropyl)-phenol,

(±)-acetic acid 4-(tert.-butyl-dimethylsilanyloxy)-2-(3-

diisopropylamino-1-phenylpropyl)-phenyl ester,

(±) -{3-{2-(tert.-butyl-dimethylsilanyloxy)-5-(tert.-butyl-

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dimethylsilanyloxymethyl)-phenyl]-3-phenylpropyl}-
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diisopropylamine,

(±) - [4-(tert.-butyl-diphenylsilanyloxy) - 3-(3-diisopropylamino-1-phenylpropyl) - phenyl] - methanol,

(±)-acetic acid 4-(tert.-butyl-diphenylsilanyloxymethyl)-2-

(3-diisopropylamino-1-phenylpropyl)-phenyl ester,

(±)-4-(tert.-butyl-diphenylsilanyloxymethyl)-2-(3-diisopropylamino-1-phenylpropyl)-phenol,

(±) - {3- {2- (tert.-butyl-diphenylsilanyloxy) - 5- (tert.-butyl-diphenylsilanyloxymethyl) - phenyl] - 2-phenylpropyl }-diisopropylamine,

(±)-acetic acid 4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester,

(±)-benzoic acid 4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzyl ester,

(±)-isobutyric acid 4-benzyloxy-3-(3-diisopropylamino-1phenylpropyl)-benzyl ester,

(±) -2-(3-diisopropylamino-1-phenylpropyl) -4-(1 β -D-glucurono-syloxymethyl) -phenol.

13. 3,3-Diphenylpropylamines as claimed in claim 2 selected from carbonates and carbamates represented by the general formulae VII and VIII



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wherein Y, Z and n are as defined in claim 1 and wherein R^{12} and R^{13} represent a C_1-C_6 alkoxycarbonyl group or



wherein R^4 and R^5 are as defined in claim 1.

14. 3,3-Diphenylpropylamines as claimed in claim 13 selected from:

(±)-N-ethylcarbamic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester,

(±)-N,N-dimethylcarbamic acid 2-(3-diisopropylamino-1-

phenylpropyl)-4-hydroxymethylphenyl ester

(±)-N,N-diethylcarbamic acid 2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenyl ester

(±)-N-phenylcarbamic acid 2-(3-diisopropylamino-l-phenylpropyl)-4-hydroxymethylphenyl ester,

(±) - [2-(3-Diisopropylamino-1-phenylpropyl) - 4-hydroxymethyl-phenoxycarbonylamino]acetic acid ethyl ester hydrochloride,
(±) -N-ethylcarbamic acid 3-(3-diisopropylamino-1-phenyl-propyl) - 4-N-ethylcarbamoyloxybenzyl ester,

(±)-N,N-dimethylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N,N-dimethylcarbamoyloxybenzyl ester, (±)-N,N-diethylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N,N-diethylcarbamoyloxybenzyl ester, (±)-N-phenylcarbamic acid 3-(3-diisopropylamino-1phenylpropyl)-4-N-phenylcarbamoyloxybenzyl ester, (±)-{4-[2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenoxycarbonylamino]-butyl}-carbamic acid 2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl ester, (±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester ethyl ester, - 107 -

(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl ester phenyl ester,
(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4ethoxycarbonyloxymethylphenyl ester ethyl ester,
(±)-carbonic acid 2-(3-diisopropylamino-1-phenylpropyl)-4phenoxycarbonyloxymethylphenyl ester phenyl ester.

15. 3,3-Diphenylpropylamines selected from

(i) compounds of the formulae IX and IX'





Formula IX

Formula IX¹

wherein o and p are the same or different and represent the number of methylene units $\frac{1}{2}$ CH₂ $\frac{1}{2}$ and may range from 0 to 6,

- (ii) (±)-Benzoic acid 2-(3-diisopropylamino-1-phenylpropyl)-4-sulphooxymethyl-phenyl ester
- (iii) Poly-co-DL-lactides of 2-(3-diisopropylamino-phenylpropyl)-4-hydroxymethyl-phenol
- (iv) (±)-2-(3-Diisopropylamino-1-phenylpropyl)-4-(1β-Dglucuronosyloxymethyl)-phenol having the formula



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and

their salts with physiologically acceptable acids, their free bases and, when the compounds can be in the form of optical isomers, the racemic mixture and the individual enantiomers.

16. A process for the production of phenolic monoesters represented by the general formula II



as defined in claim 3, which comprises treatment of a compound of the formula





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with an equivalent of an acylating agent selected from

O II R¹-C-LG

wherein LG represents a leaving group selected from halogenide, carboxylate and imidazolide and R^1 is as defined in claim 3, in an inert solvent in the presence of a condensating agent.

17. A process for the production of phenolic moncesters represented by the general formula II'



as defined in claim 3, which comprises treatment of two equivalents of a compound of the formula



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with an acylating agent selected from

wherein Hal represents a halogen atom.

18. A process for the production of identical diesters represented by the general formula III



as defined in claim 5, which comprises treatment of a compound of the formula



with at least two equivalents of the acylating agent as defined in claim 16. - 111 -

19. A process for the preparation of benzylic moncesters represented by the general formula V



as defined in claim 9, which comprises treatment of a compound of the formula



at room temperature and under anhydrous conditions with activated esters in the presence of enzymes selected from lipases or esterases.

20. A process for the preparation of mixed diesters represented by the general formula IV



Formula IV

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as defined in claim 7, which comprises acylation of a benzylic monoester represented by the general formula V



as defined in claim 9 or of a phenolic monoester represented by the formula II as defined in claim 3.

21. A process for the production of ethers represented by the general formula VI



as defined in claim 11 wherein R^{11} is hydrogen which comprises reacting a compound of the formula



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with an alcohol R^{10} -OH in the presence of an esterification catalyst.

22. A process for the preparation of ethers represented by the general formula VI



wherein R^{10} and R^{11} are as defined in claim 11, which comprises acid or base treatment of free benzylic alcohols selected from



and



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and









Formula VI

wherein R^{10} is hydrogen or



Formula VII

wherein R^{12} is hydrogen and R^{13} represents a $C_1\text{-}C_6$ alkoxycarbonyl group or



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wherein R^4 and R^5 are as defined in claim 1 or of benzylic acylates selected from





Formula IV



Formula V

wherein R^1 and R^2 are as defined in claim 7 in the presence of suitable hydroxy reagents.

23. A process for the preparation of ethers of formula VI as defined in claim 11, which comprises treating a compound of the formula



with an alkylating agent selected from alkyl halcgenides, alkyl sulphates and alkyl triflates, said alkyl group having 1 to 6 carbon atoms.

24. A process for the preparation of carbonates and carbamates represented by the general formulae VII and VIII



as defined in claim 13, which comprises reacting a compound selected from the group consisting of



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wherein R^{1} is defined as in claim 3, n is 0 to 12, Bn is benzyl, one of R^{10} or R^{11} is hydrogen and the other one is as defined in claim 11 with activated carbonyl compounds or carbonyl precursor reagents selected from haloformates, ketenes, activated esters, mixed anhydrides of organic or inorganic acids, isocyanates and isothiocyanates.

25. 3,3-Diphenylpropylamines as claimed in claims 1 to 15 for use as pharmaceutically active substances, especially as antimuscarinic agents.

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26. A pharmaceutical composition comprising a 3,3-diphenylpropylamine as claimed in claim 1 to 15 and a compatible pharmaceutical carrier.

27. Use of a 3,3-diphenylpropylamine as claimed in claims 1 to 15 for preparing an antimuscarinic drug.







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A. CLASSIF IPC 6	219/22 307/02				
B. FIELDS	SEARCHED				
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A	LISBETH NILVEBRANT ET AL.: "Tolt a new bladder-selective antimusca agent" EUROPEAN JOURNAL OF PHARMACOLOGY, vol. 327, 1997, pages 195-207, XP cited in the application the whole document	erodine - rinic 002079629		1,25-27	
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* Special categories of cited documents : "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "A" document defining the general state of the art which is not considered to be of particular relevance "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "E" earlier document but published on or after the international filing date to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered to a nother or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "Y" document is combined with one or more other such document is combined with one or more other such document is combined with one or more other such document is the art. "P" document published prior to the international filing date but tater than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 19 July 1999 26/07/1999					
Name and	mailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized office Rufet ,	J		

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Patent Abandoned Unintentionally Under 37 CFR 1.137(b)) SIGNATURE OF APPLICANT, ATTORNEY, OR AGENT Firm or Individual Name Paul A. Lesko Signature Date August 14, 2002 CERTIFICATE OF EXPRESS MAILING Express Mail No. EL474165379US I hereby certify that this correspondence is being deposited on August 14, 2002 with the United States Postal Service as Express Mail under 37 C.F.R. 1.10 and addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231. Typed or printed name Paul A. Lesko Signature Date August 14, 2002 Typed or printed name Paul A. Lesko Signature Date August 14, 2002	 Charge Deposit Accoun Fee Attached Amendment / Reply Affidavits/declarations(s Extension of Time Requ Express Abandonment f Information Disclosure S Certified Copy of Priority Document(s) Response to Missing Pa Incomplete Application Response to Missing Pa under 37 CFR 1.52 or 1. Petition For Revival of a) est Request statement rts 53 n Application for	Assignment (for an Appli Drawing(s) Licensing-re Petition Petition to C Provisional / Power of Att Change of C Address Terminal Dis Request for CD, Number	Papers ication) elated Papers Convert to a Application formey, Revocation Correspondence sclaimer Refund r of CD(s) Commission application overpayme enclosed a	on oner is here and any fe ent, to Depo	 After Allowance Communication to Group Appeal Communication to Board of Appeals and Interferences Appeal Communication to Group (Appeal Notice, Brief, Repty Brief) Proprietary Information Status Letter Request To Rescind Previous Nonpublication Request Response to Notice of Allowability Other Enclosure(s) (please identify below): aby authorized to charge fees in this sees which may be required, or any cosit Account 20-0823. I have copy of this sheet 		
Firm or Individual Name Paul A. Lesko Signature Paul A. Lesko Date August 14, 2002 CERTIFICATE OF EXPRESS MAILING Express Mail No. EL474165379US I hereby certify that this correspondence is being deposited on August 14, 2002 with the United States Postal Service as Express Mail under 37 C.F.R. 1.10 and addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231. date: August 14, 2002 Typed or printed name Paul A. Lesko Date Signature Date August 14, 2002	37 CFR 1.137(b))			. Amount				
Date August 14, 2002 CERTIFICATE OF EXPRESS MAILING Express Mail No. EL474165379US I hereby certify that this correspondence is being deposited on August 14, 2002 with the United States Postal Service as Express Mail under 37 C.F.R. 1.10 and addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231. date: August 14, 2002 Typed or printed name Paul A. Lesko Signature Date August 14, 2002	Firm or Pau Individual Name Signature	JI A. Lesko	TURE OF APPLICA	NI, ALIORNEY,	, UR AGENT			
CERTIFICATE OF EXPRESS MAILING Express Mail No. EL474165379US I hereby certify that this correspondence is being deposited on August 14, 2002 with the United States Postal Service as Express Mail under 37 C.F.R. 1.10 and addressed to: Assistant Commissioner for Patents, Washington, D.C. 20231. date: August 14, 2002 Typed or printed name Paul A. Lesko Signature Data August 14, 2002	Date Au	just 14, 2002						
Z002 Typed or printed name Paul A. Lesko Signature Data	I hereby certify that this corres 37 C.F.R. 1.10 and addressed	C pondence is being d to: Assistant Comm	ERTIFICATE OF Express Mail No deposited on August hissioner for Patents	EXPRESS MA . EL474165375 14, 2002 with the , Washington, D.	ILING 9US e United Stat C. 20231.	tes Postal Service as Express Mail under date: August 14,		
Signature Date August 14 2002	Typed or printed name	Poul A Lest				2002		
	Signature	Faul A. Lesk			Data	August 14, 2002		

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TR	ANSMITTAL (OF INFORMATIC Under 37 CFR 1	ON DISCLOS .97(b) or 1.9	SURE STATE	MENTOIP	Do 419	cket No. 46/32854	
In Re A	pplication Of:			······································	AU8 1 4 2	2002		
Meese,	Clause				PATE			
	Serial No.	Fili	ing Date	Ex	aminer TRADEN	ARI	iroup Art Linit	
10/1302	214	5/14/2002		Not Assigned	- TUCKER	2 - 1 614 -10	524	
Title: STA	BLE SALTS OF	NOVEL DERIVATIV	ES OF 3,3-DIP	HENYLPROPYL	AMINES	TECHCE	RECE	
			A Assistant Com Washing	ddress to: imissioner for Pater iton, D.C. 20231	its	TER TOUR		
			37 C	FR 1.97(b)		ECHO	Er, O	
1. 🛛	The Information Disclosure Statement submitted herewith is being filed within three months of the filing of a national application other than a continued prosecution application under 37 CFR 1.58(8), within three months of the date of entry of the national stage as set forth in 37 CFR 1.491 in an international application; before the mailing of a first Office Action on the merits, or before the mailing of a first Office Action after the filing of a request for continued examination under 37 CFR 1.114.							
			37 C	FR 1.97(c)				
2. 🗌	The Informatio CFR 1.97(b), p Final Action u otherwise close	n Disclosure Statem provided that the Info nder 37 CFR 1.113 es prosecution in the	ent submitted h ormation Disclo , a Notice of A application, an	erewith is being sure Statement i Allowance under d is accompanied	filed after the p s filed before t 37 CFR 1.31 d by one of:	eriod specifie he mailing da 1, or an Acti	ed in 37 ate of a on that	
	the stat	ement specified in 37	7 CFR 1.97(e);			•		
		OR						
	the fee	set forth in 37 CFR 1	.17(p).					

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FORM PT	O-1390 U.S. DEPARTMENT OF CO	MMERCE PATENT AND TRADEMARK OFFICE	ATTORNEY'S DOCKET NUMBER				
(REV. 11-2			41946/32854				
2	TRANSMITTAL LETTER	TO THE UNITED STATES	U.S. APPLICATION NO. (If known see 37 CFR 1.5)				
	CONCERNING A FUN	ED OFFICE (DO/EO/US)	New $90/920291$				
	CONCERNING A FILIN	G UNDER 35 U.S.C. 371	10/150214				
INTERI PCT/E	NATIONAL APPLICATION NO. P00/11309	INTERNATIONAL FILING DATE 15 November 2000	PRIORITY DATE CLAIMED 16 NOVEMBER 1999				
TITLE STABL	OF INVENTION LE SALTS OF NOVEL DERIVAT	TIVES OF 3,3,-DIPHENYLPROPYLAN	MINES				
APPLIC	CANT(S) FOR DO/FO/US						
MEESI	E, Claus						
Applica	int herewith submits to the United St	tates Designated/Elected Office (DO/EO/	IS) the following items and other information				
$1 \qquad \square$	This is a FIRST submission of ite	ms concerning a filing under 35 U.S.C. 37					
	This is a SECOND or SUBSEOU	ENT submission of items concerning a fil	$\frac{1}{1}$				
3	This is an express request to begin	national examination procedures (35 U S	$C_{271}(f_{0})$ The submission must include				
	items (5), (6), (9) and (21) indicate	ed below.	.c. 571(1)). The submission must menude				
4. 🖾	The US has been elected by the ex	piration of 19 months from the priority da	te (Article 31).				
5. 🖂	A copy of the International Applic	ation as filed (35 U.S.C. 371(c)(2))	· · ·				
	a. is attached hereto (require	ed only if not communicated by the Interna	ational Bureau).				
	b. 🛛 has been communicated b	by the International Bureau.					
	c. 🔲 is not required, as the app	lication was filed in the United States Rec	eiving Office (RO/US).				
6. 🖾	An English language translation of	the International Application as filed (35	U.S.C. 371(c)(2)).				
	a. 🔀 is attached hereto.						
	b.• 🔲 has been previously subm	itted under 35 U.S.C. 154 (d)(4).					
7. 🖾	Amendments to the claims of the I	nternational Application under PCT Articl	le 19 (35 U.S.C. 371(c)(3))				
	a. 🔲 are attached hereto (requi	red only if not communicated by the Intern	national Bureau).				
	b. 🗌 have been communicated	by the International Bureau.					
	c. have not been made; how	ever, the time limit for making such amend	dments has NOT expired.				
	d. 🔀 have not been made and w	vill not be made.					
8. 🗌	An English language translation of	the amendments to the claims under PCT	Article 19 (35 U.S.C. 371 (c)(3)).				
9. 🖂	An oath or declaration of the inver	tor(s) (35 U.S.C. 371(c)(4)).					
10. 🗌	An English language translation of Article 36 (35 U.S.C. 371(c)(5)).	the annexes of the International Prelimina	ary Examination Report under PCT				
. Iter	ms 11 to 20 below concern docum	ent(s) or information included:					
11. 🛛	An Information Disclosure Statem	ent under 37 CFR 1.97 and 1.98.					
12. 🖂	An assignment document for recor	ding. A separate cover sheet in compliance	ce with 37 CFR 3.28 and 3.31 is included.				
13. 🖂	A FIRST preliminary amendment.						
14. 🗌	A SECOND or SUBSEQUENT preliminary amendment.						
15. 🗌	A substitute specification.						
16. 🛛	A change of power of attorney and	/or address letter.					
17. 🗌	A computer-readable form of the set	equence listing in accordance with PCT R	ule 13ter.2 and 35 U.S.C. 1.821 – 1.825.				
18. 🔲	A second copy of the published int	ernational application under 35 U.S.C. 15	4(d)(4).				
19. 📋	A second copy of the English lange	uage translation of the international application	ation under 35 U.S.C. 154(d)(4).				
20. 🔀	Other items or information: Certif	icate of Express Mailing;					
ľ	Postca	ard					
	Staten	nent Under 37 CFR 3.73(b)					
page 1 of 2			· · · · · · · · · · · · · · · · · · ·				

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U.S. APPLICATION NO- of known see 37 CFR-1.51 INTERNATIONAL APPLICATION NO ATTORNEY'S DOCKET NUMBER New DD 502 14 PCT/EP00/11309 41946/32854								
21. X The following	g fees are submitted:	CAI	LCULATIONS F	TO USE ONLY				
BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) – (5)):								
Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO\$1000.00								
International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO\$860.00								
International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO\$710.00								
International preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4)\$690.00								
International prelimit and all claims satisfie	hary examination fee (3 ed provisions of PCT A							
ENI	FER APPROPRIATE	BASIC FEE AMOUN	T =	\$	860.00			
Surcharge of \$130.00 months from the earlie	for furnishing the oath est claimed priority date	or declaration later than e (37 CFR 1.492(e)).	20 30	\$	· · · · · · · · · · · · · · · · · · ·			
CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE	\$	· · · · · · · · · · · · · · · · · · ·			
Total claims	30-20 =	10	x \$18.00	\$	180.00			
Independent claims	10- 3 =	7	x \$80.00	\$	560.00			
MULTIPLE DEPEND	DENT CLAIM(S) (if ap	oplicable)	+ \$270.00	\$	270.00			
Applicant claims	TC	DTAL OF ABOVE CA	LCULATIONS =	\$				
are reduced by 1.	/2		+	\$				
Drogogging for of \$120	00 for furnishing the	English translation later	$\frac{\text{SUBTOTAL}}{\text{them}} = \frac{20}{20} + \frac{20}{20}$	\$				
months from the earlie	est claimed priority date	e (37 CFR 1.492(f)).		\$				
		TOTAL NA	ATIONAL FEE =	\$				
Fee for recording the e accompanied by an ap	Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31), \$40.00 per property + \$40.00							
TOTAL FEES ENCLOSED = 1910.00 1910								
Amount to be								
					refunded:	\$		
· · · · · · · · · · · · · · · · · · ·					charged:	5		
 a. A check in the amount of \$_1910.00 to cover the above fees is enclosed. b. Please charge my Deposit Account No in the amount of \$ to cover the above fees. 								
A duplicate copy of this sheet is enclosed.								
c. I The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. <u>20-0823</u> . A duplicate copy of this sheet is enclosed.								
d. Fees are to be charged to a credit card. WARNING: Information on this form may become public. Credit card information should not be included on this form. Provide credit card information and authorization on PTO-2038.								
NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to review (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.								
SEND ALL CORRESPONDENCE TO:								
Paul A Lesko Esg								
Thompson Coburn L	LP		Signature					
One U.S. Bank Plaza			5					
St. Louis, MO 63101		Paul A. Lesko						
Telephone No.: 314.552.6443Name								
Facsimile No.: 314.552.7000								
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page 2 of 2	Registration Number							
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JC13 Rec'd PCT/PTO 1 4 MAY 2002

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

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In re Application of: MEESE, Claus et al.

Filed: Herewith Title: STABLE SALTS OF NOVEL DERIVATIVES OF 3,3-

Application No.: To be assigned

Examiner: To be assigned

Group Art Unit: To be assigned

Docket No.: 41946/32854

Commissioner for Patents Box PCT Washington, DC 20231

DIPHENYLPROPYLAMINES

PRELIMINARY AMENDMENT

Sir:

Prior to calculation of the filing fee and examination on the merits, kindly amend the above-identified patent application per the following instructions.

Kindly amend the specification at page one after the title and before the first line of text, by inserting at that point the following sentence -- This patent application claims the benefit of priority under 35 U.S.C. § 119 of German Patent Application No. 199 55 190.1, filed November 16, 1999. German Patent Application No. 199 55 190.1 is incorporated herein in its entirety by reference.--

The amendments to claims 18-21, 23-25, 27, and 28 are pursuant to an Article 34 amendment made to the PCT application on October 5, 2001.

IN THE CLAIMS

At page 56, amend claims 18-21, 23-25, 27, and 28 as follows:

18. (once amended) Compound of formula III

.



in highly pure, crystalline and stable form.

19. (once amended) Compound of formula V



Formula V

20. (once amended) Compound of formula VI



in highly pure, crystalline and stable form.

- 21. (once amended) Use of a compound in accordance with claims 18 to 20 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds of formula 1 in accordance with claim 1.
- 23. (once amended) Compound of formula 3



24. (once amended) Compound of formula 5



in highly pure, crystalline and stable form.

25. (once amended) Compound of formula 6



- 27. (once amended) Use of a compound in accordance with claims 23 to 26 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds of formula 2 in accordance with claim 3.
- 28. (once amended) Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of phenolic monoesters of general formula 1



in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl or unsubstituted or substituted phenyl.

Kindly consider this preliminary amendment and enter it into the record of this application. Attached is a clean copy of the claims. All correspondence should to be directed to Paul A. Lesko, Thompson Coburn LLP, One U.S. Bank Plaza, St. Louis, MO 63101, Telephone No.: 314.552.6443, Facsimile No.: 314.552.7000.

Respectfully submitted,

Paul/A. Lesko Registration No. 45,364 Thompson Coburn LLP One U.S. Bank Plaza St. Louis, MO 63101 Telephone: 314.552.6443 Facsimile: 314.552.7000

CLEAN COPY OF PARAGRAPH TO BE INSERTED INTO SPECIFICATION

-- This patent application claims the benefit of priority under 35 U.S.C. § 119 of German Patent Application No. 199 55 190.1, filed November 16, 1999. German Patent Application No. 199 55 190.1 is incorporated herein in its entirety by reference.--

Claims

18. Compound of formula III

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in highly pure, crystalline and stable form.

19. Compound of formula V



Formula V

20. Compound of formula VI



in highly pure, crystalline and stable form.

- 21. Use of a compound in accordance with claims 18 to 20 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds of formula 1 in accordance with claim 1.
- 23. Compound of formula 3

.



24. Compound of formula 5



in highly pure, crystalline and stable form.

25. Compound of formula 6



in highly pure, crystalline and stable form.

27. Use of a compound in accordance with claims 23 to 26 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds of formula 2 in accordance with claim 3.

28. Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of phenolic monoesters of general formula 1



in which R denotes C_1 - C_6 -alkyl, C_3 - C_{10} -cycloalkyl or unsubstituted or substituted phenyl.

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Figure 1

Reaction diagram 1

(i), (ii), (iii), (iv), (v) stand for: (i), LiAlH₄, (ii), Raney nickel/H₂, (iii), Me₂CH-CoCl, Et₃N, (iv), fumaric acid, (v), hydrochloric acids; R stands for isopropyl (iPr)





SPECIFICATION

JC13 Rec'd PCT/PTO 1 4 MAY 2002

Stable salts of novel derivatives of 3,3-diphenylpropylamines

The present invention concerns highly pure, crystalline, stable compounds of novel derivatives of 3,3diphenylpropylamines in the form of their salts, a method for manufacturing these and highly pure, stable, intermediate products.

From document PCT/EP99/03212 novel derivatives of 3,3diphenylproprylamines are known.

These are valuable prodrugs for the treatment of urinary incontinence and other spasmodic complaints, which overcome the disadvantage of the active substances available to date, namely inadequate absorption of the active substance by biological membranes or the unfavourable metabolism of these.

Furthermore these novel prodrugs have improved pharmacokinetic characteristics compared with Oxybutynin and Tolterodin.

Preferred compounds from the group of these novel derivatives of 3,3-diphenylpropylamines are esters of aliphatic or aromatic carboxylic acids with the general formula A referred to below

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Formula A

Petitioner Mylan Pharmaceuticals Inc. - Exhibit 1002 - Page 505
-2-

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl or unsubstituted or substituted phenyl. These can occur in their optical isomers form as racemic mixtures and in the form of their individual enantiomers.

Compounds with the structure of formula A do, however, have low solubility in water. This restricts their oral bioavailability.

Finally, monoesters of the structure, as shown in formula A, have a tendency towards intermolecular transesterification.

During long periods of storage, therefore, as the content of the compounds with the structure of general formula A drops an increase in diesters and free diol can be detected.

Basically salts of the compounds of general formula A can be obtained if solutions of the compounds of formula A (base component) are purified with solutions of acids in suitable solvents, but the salts obtained in the form of solid matter can prove to be altogether amorphous and/or hygroscopic and cannot be directly crystallized from the normal solvents either. Such salts have inadequate chemical stability to be galenically processed as valuable pharmaceutically active substances.

Surprisingly, it has now been found that the abovementioned disadvantages can be avoided if compounds with the structure of general formula A, once they have been prepared under a special reaction process, are converted with a physiologically compatible inorganic or organic acid with general formula H-X, in which ⁻X represents the respective acid residue, into their respective salt with general formula I.



Formula I

The problem for the present invention is therefore to provide highly pure, crystalline, stable compounds of novel derivatives of 3,3-diphenylpropylamines in the form of their salts, that avoid the stated disadvantages and are well suited to use in pharmaceutical-technical formulations and can be processed into these.

A further problem for the present invention is to provide a method for manufacturing such highly pure, crystalline, stable compounds in the form of their salts, as well as highly pure, stable intermediate products.

The final problem for the invention is to provide a method for manufacturing the abovementioned compounds with which a high yield of the products of the process and the respective intermediate products can be obtained chemo- or regioselectively. This problem is solved in that highly pure, crystalline, stable compounds of the 3,3-diphenylpropylamines in the form of their salts with general formula I are provided,



Formula I

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid.

X٠

In accordance with a design of the invention the salts of general formula I can contain the respective acid residue X^- of the acids mentioned below:

hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)-tartaric acid, citric acid, L-aspartic acid, L-(+)-ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4hydroxybenzoic acid, salicyclic acid, vanillic acid, 4hydroxycinammic acid, gallic acid, hippuric acid (N-benzoylglycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

In accordance with a further design form of the invention Rconfigured compounds with general formula 2 are provided



Formula 2

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid.

χ.

In accordance with an advantageous design form of the invention the compounds in the form of their salts of general formula 2 can contain the respective acid residue X^- of the acids mentioned below:

hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)-tartaric acid, citric acid, L-aspartic acid, L-(+)-ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4hydroxybenzoic acid, salicyclic acid, vanillic acid, 4hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-6-

glycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

Preferred compounds of the present invention are the salts

R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydro xymethylphenylisobutyrate ester hydrogen fumarate

- and

- R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate.

Furthermore, compounds are preferred in which R stands for cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, 4-(1-cyclo-propyl-methanoyloxy)-phenyl, 4-(1-cyclobutyl-methanoyloxy)-phenyl, 4-(1-cyclohexyl-methanoyloxy)-phenyl or 4-(2,2-dimethyl-propanoyloxy)-phenyl and X⁻ denotes chloride.

Particular preference is for [(R)-3-(2-{1-[4-(1-cyclopropylmethanoyloxy)-phenyl]-methanoyloxy}-5-hydroxymethyl-phenyl)-3-phenylpropyl]-diisopropyl-ammonium chloride, [(R)-3-(2-{1-[4-(1-cyclobutyl-methanoyloxy)-phenyl]-methanoyloxy}-5hydroxymethyl-phenyl)-3-phenyl-propyl]-diisopropyl-ammonium chloride, [(R)-3-(2-{1-[4-(1-cyclohexyl-methanoyloxy)phenyl]-methanoyloxy}-5-hydroxymethyl-phenyl)-3-phenylpropyl]-diisopropyl-ammonium chloride, [(R)-3-(2-{1-[4-(2,2dimethyl-propanoyloxy)-phenyl]-methanoyloxy}-5-hydroxymethylphenyl)-3-phenyl-propyl]-diisopropyl-ammonium chloride, {(R)-3-[2-(1-cyclopropyl-methanoyloxy)-5-hydroxymethyl-phenyl]-3phenyl-propyl}-diisopropyl-ammonium chloride, {(R)-3-[2-(1cyclobutyl-methanoyloxy)-5-hydroxymethyl-phenyl]-3-phenyl-7-

propyl}-diisopropyl-ammonium chloride, {(R)-3-[2-(1cyclopentyl-methanoyloxy)-5-hydroxymethyl-phenyl]-3-phenylpropyl}-diisopropyl-ammonium chloride and {(R)-3-[2-(1cyclohexyl-methanoyloxy)-5-hydroxymethyl-phenyl]-3-phenylpropyl}-diisopropyl-ammonium chloride.

In the compounds of the present invention the expression "alkyl" preferably stands for a straight-chain or branchedchain hydrogen group with between 1 and 6 C-atoms. Special preference is for methyl, ethyl, propyl, isopropyl, butyl, isobutyl, pentyl and hexyl. The expression "cycloalkyl" designates cyclical hydrogen groups, that have between 3 and 10 hydrogen atoms, that may also contain suitable substitutes in place of the hydrogen atoms.

The expression "phenyl" designates a $-C_6H_5$ -group that may be substituted or unsubstituted. Suitable substitutes can be, for example, alkyl, alkoxy, halogen, nitro and amine. The expression "alkoxy" has, with respect to the alkyl component, the same meaning as already given above for "alkyl". Suitable halogens are fluorine, chlorine, bromine and iodine atoms

The present invention also includes methods for manufacturing the compounds in accordance with the invention of general formula I as well as valuable intermediate products.

The method is characterised by chemo- and regioselectivity.

Compounds of general formula I



Formula I

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid, are manufactured in that

χ-

a) a compound of formula III

HO



Formula III

, is split with a hydrogenation agent to form a compound of formula \mathtt{V}



Formula V

whereupon

 b) the compound of formula V so obtained is converted with a reducing agent, in order to give a compound of formula VI



which

c) is converted with an acylation agent, in order to obtain a compound of formula A



Formula A

in which ${\tt R}$ has the significance stated above, which

 d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula I



Formula I

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

In accordance with the invention, for the manufacture of the compounds of general formula I hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)-tartaric acid, citric acid, L- aspartic acid, L-(+)-ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

In accordance with an advantageous further development of the invention a method for the manufacture of *R*-configured compounds of the general formula 2 is described,

-11-



Formula 2

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid, in that

a) a compound of formula 3



Formula 3

is split with a hydrogenation agent to form a compound of formula 5



-12-

whereupon

 b) the compound of formula 5 so obtained is converted with a reducing agent, in order to give a compound of formula 6



which

c) is converted with an acylation agent, in order to obtain a compound of formula 1



Formula 1

in which R has the significance stated above, which

d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula 2 -13-



Formula 2

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

Advantageously in order to obtain compounds of general formula 2, in accordance with the method hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)-tartaric acid, citric acid, L-aspartic acid, L-(+)-ascorbic acid, D-(+)glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-hydroxyphenyl)propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

Particular advantageously, on the basis of the crystalline R-(-)-4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)benzoic acid methyl ester, the highly pure,

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crystalline intermediate product R-(-)-3-(3-diisopropylaminophenyl-propyl)-4-hydroxy-benzoic acid methyl ester is prepared, which is reduced to R-(+)-2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenol, is finally acylated in a suitable manner and is then converted with a physiologically compatible inorganic or organic acid under spontaneous crystallisation to the respective highly pure, crystalline, stable salt.

Depending on the acid chloride used, compounds of general formula 1 are obtained,



Formula 1

in which R denotes C_1-C_6 -alkyl, in particular isopropyl, C_3-C_{10} -cycloalkyl or unsubstituted or substituted phenyl.

In order to obtain the compounds in accordance with the invention in the form of their salts the special reaction process via particular intermediate stages and individually identifiable intermediate products is crucial.

This is explained using reaction diagram 1 (see Figure 1), in which the conversions with *R*-configured compounds are described, but without this being restrictive.

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In this:

- 3 = R-(-)-4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-benzoic acid- methyl ester
- 4 = R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1-phenylpropyl)-phenyl]-methanol
- 5 = R-(-)-3-(3-diisopropylamino-phenyl-propyl)-4-hydroxybenzoic acid methyl ester
- 6 = R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenol
- 1 = R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenyl-isobutyrate ester
- 2a = R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenyl-isobutyrate ester hydrogen fumarate
- 2b = R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydro
 xymethylphenyl-isobutyrate ester hydrochloride
 hydrate

In accordance with the reaction process explained in the embodiment the preliminary stage 3 (R-(-)-4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)-benzoic acid-methylester) is prepared in crystalline, pure form.

Using normal methods - such as BBr_3 , $AlCl_3$ - but preferably by means of hydrogen gas via Raney nickel in methanol as the solvent at room temperature (RT), preliminary stage 3 is split into 5 (R-(-)-3-(3-diisopropylamino-phenyl-propyl)-4hydroxy-benzoic acid methylester. This develops in highly pure, crystalline form (melting point 143.7 °C).

Finally, using a suitable reducing agent - such as $NaBH_4/EtOH$ - preferably $LiAlH_4$ 5 is reduced into an inert solvent at low

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temperature (-78°C to + 10°C) and the compound 6 (*R*-(+)-2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol)is obtained. The compound 6 is obtained in a highly pure state and can be crystallised from a suitable solvent such as ethyl acetate. The colourless, compact grained material has a melting point of 102.3°C. This is surprising in that the compound 6 in the state of the art is described as an amorphous solid.

Compound 6 is now acylated with very good yield and regioand chemoselectivity, into a phenolic ester. This reaction is performed at RT or low temperatures with an equivalent acid chloride in the presence of a base in a suitable solvent. Suitable solvents are ethyl acetate, dichloromethane, tetrahydrofurane, acetonitrile or toluene.

The reaction is preferably performed with isobutyrylchloride as the acid chloride and triethylamine as the base at the abovementioned temperatures. The 1 (R-(+)-2-(3disopropylamino-1-phenylpropyl)-4-

hydroxymethylphenylisobutyrate ester) then obtained, occurs with such purity that with solutions of the fumaric acid in suitable solvents spontaneous crystallisation starts with the formation of the hydrogen fumarate salt 2a.

This salt has a high melting point of 103°C, is stable at RT, is non-hygroscopic and does not contain crystallose agents. It can be recrystallised as often as desired.

If instead of fumaric acid anhydrous hydrochloric acid is used - for example as an etheric solution - salt formation also takes place with the crystalline product 2b (R-(+)-2-(3-

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diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate being obtained.

Following a further recrystallisation the product 2b has a melting point range of 97 - 106°C.

Finally the product 2b can particularly advantageously be obtained by the following variants of the inverse reaction process, starting with the compound 6 of reaction diagram 1. The product 2b can thus be obtained without the addition of an external acid-intercepting base, as explained in the following.

Solutions of 6 (*R*-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol) are dripped into solutions of isobutyrate chloride, so that under suitable polarity conditions the anhydrous product 2b rapidly crystallises out. 2b is very hygroscopic.

If the abovementioned reaction is carried out in a humid solvent, that contains at least one mole equivalent of water, a stable and crystalline, hydrate-containing product 2b is obtained, that has the abovementioned melting characteristics.

The compounds in accordance with the invention of general formulae 1 and 2 are suited to bulk material.

Of particular advantage are the highly pure compounds of general formulas III, V, VI, 3, 5, 6 and 7 which can be obtained.

Compound of formula III



Formula III

Compound of formula V



Formula V

Compound of formula VI



Formula VI

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Compound of formula 3



Formula 3

Compound of formula 5



Formula 5

Compound of formula 6



Formula 6

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Compound of formula 7



[(R)-3-(2-{1-[4-(2,2-dimethyl-propanoyloxy)-phenyl]-methaneoyloxy}-5-{1-[4-(2,2-dimethyl-propanoyloxy)-phenyl]-methaneoyloxymethyl}-phenyl)-3-phenyl-propyl]-diisopropyl-ammoniumchloride

The abovementioned compounds III, V, VI, 3, 5, 6 and 7 are particularly suited to use in each case as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds.

Of particular advantage are compounds for use as an intermediate product in the manufacture of R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-

hydroxymethylphenylisobutyrate ester hydrogen fumarate and R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-

hydroxymethylphenylisobutyrate ester hydrochloride hydrate.

Finally, the method can be carried out in a particularly advantageous way by converting a compound of general formula 6 (see reaction diagram 1) with an equivalent isobutyryl -21-

chloride in the presence of triethylamine using one of the respective solvents ethylacetate, dichloromethane, tetrahydrofurane, acetonitrile or toluene regio- and chemoselectively into R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester.

In accordance with the invention R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester is particularly suited to conversion with fumaric acid or hydrochloric acid with the formation of the respective salt.

The following embodiments explain the invention.

Experimental

I. General

All compounds have been fully characterised by ¹H and ¹³C NMRspectroscopy (Bruker DPX 200). The stated chemical displacements in the ¹³C-NMR-spectra (50 MHz, ppm values stated) refer to the solvent resonances of CDCl₃ (77.10 ppm). ¹H NMR data (CDCl₃; 200 MHz, ppm) refer to internal tetramethylsilane).

Thin layer chromatography (DC, R_f given) was carried out on 5x10 cm E. Merck silica gel films (60F254), and the stains were revealed by fluorescence erasure or by spraying with alkaline potassium permanganate solution.

Absorbent systems were: (1), n-hexane / acetone / triethylamine (70/20/10, v/v-%); (2), toluene / acetone / methanol / acetic acid (70/5/20/5, v/v-%). -22-

The optical rotations were measured at a wavelength of 589.3 nm (sodium D-line), at room temperature using ethanol as a solvent (apparatus: Perkin Elmer Polarimeter Type 241), melting points (in °C) are uncorrected and were determined on the Mettler FP apparatus, or by differential thermoanalysis (DSC) on the Perkin Elmer Model DSC7, using "Pyris" evaluation software.

UV/VIS measurements were carried out on the spectrophotometer model Lambda 7 (Perkin-Elmer) with a layer thickness of 1 cm. The specific absorption stated is for a 1% solution $(A^{1} *_{1 cm})$.

IR spectra were recorded on a Perkin-Elmer FTIR spectrometer Series 1610 (resolution 4 cm^{-1}).

Gas chromatography mass spectrometry (GC-MS, m/z values and relative intensity with reference to the base ion (%)) was carried out with a Finnigan TSQ 700 Triple Mass Spectrometer in positive (P-CI) or negative (N-CI) chemical ionisation measurement mode with methane or ammonium as a reactant gas or via electron impact ionisation. Hydroxy compounds were measured as trimethylsilylether-derivatives.

Coupled liquid chromatography-mass spectrometry (LC-MS): Waters Integrity System, Thermabeam Mass Detector (EI, 70 eV), m/z-values and relative intensity (%) are given over a quantity range of 50-500 a.m.u. -23-

II. Embodiments

The Arabic numerals in brackets (3), (4), (5), (6) refer to the identical designations in reaction diagram 1.

1. Preparation of
R-(-)-4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)benzoic acid methylester (3)



A solution of R-(-)-4-benzyloxy-3-(3-diisopropylamino-1phenyl-propyl)-benzoic acid hydrochloride (2.30 kg, 4.77 Mol) in 26.4 litres of methanol and 0.25 litre of concentrated sulphuric acid is heated for 16 hours with recycling. Then a third of the solvent is distilled off, cooled and under agitation mixed with 5 kg ice and 2.5 litres 25% aqueous sodium carbonate solution. The deposit is first extracted with 15 litres and then again with 5 litres of dichloromethane. The organic phases are purified and concentrated on the rotary evaporator until dry. 1.99 kg (90.7% of theoretical) dark yellow oil with a purity of approximately 90% (DC, NMR) are obtained. DC (1): 0.58 -24-

¹³C-NMR (CDCl₃): 20.55, 20.65, 36.83, 41.84, 43.83, 51.82, 70.12, 111.09, 122.46, 125.28, 127.49, 128.02, 128.35, 128.50, 129.22,129.49, 133.20, 136.39, 144.51, 159.87, 167.09.

Recrystallisation

69.0 oily raw material is dissolved in 150 ml boiling methanol. Following the addition of 15 ml distilled water it is left at 0°C, whereupon colourless crystals precipitate. These are filtered off, washed with a little cold methanol and vacuum-dried. Yield: 41.8 g (60.6 % of theoretical) colourless crystals, melting point 89.8 °C; $[I]_D^{20} = -30.7$ (c = 1.0, ethanol).

2. Preparation of
R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)phenyl]-methanol (4)



Raw product (3) (28 g) is dissolved in 230 ml pure diethylether and under agitation is dripped into a suspension of 1.8 g lithium-aluminium hydride in diethylether (140 ml). After 18 hours of agitation at room temperature, 4.7 ml of water are added in drop form. The organic phase is separated -25-

off, dried with anhydrous sodium sulphate, filtered and concentrated on the rotary evaporator until dry. 26 g (98.9% of theoretical) R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1phenyl-propyl)-phenyl]-methanol (4) are obtained as a colourless oil.

DC (2): 0.32; [I]_D²⁰ = + 6.3 (c = 1.0, ethanol). ¹³C-NMR (CDCl₃): 20.53, 20.61, 36.87, 41.65, 44.14, 48.82, 65.12, 70.09, 111.80, 125.77, 125.97, 126.94, 127.55, 128.08, 128.37, 128.44, 133.27, 134.05, 134.27, 137.21, 144.84.

3. Preparation of

R-(-)-3-(3-diisopropylamino-phenyl-propyl)-4-hydroxy-benzoic
acid methyl ester (5)



To an agitated suspension of 5g Raney nickel (washed with water, then with methanol) in 200 ml methanol, 10 g (21.8 mmol) R-(-)-4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)-benzoic acid methyl ester (3) are added. Following brief heating, in order to dissolve all (3) completely, the apparatus is placed under a hydrogen gas atmosphere. After three hours of agitation at normal pressure and room temperature, the thin layer chromatography demonstrates

complete conversion. The deposit is rinsed with nitrogen gas and following addition of some active charcoal is filtered. Following concentration of the methanolic solution on the rotary evaporator 6.0 g (75% of theoretical) R-(-)-3-(3diisopropylaminophenyl-propyl)-4-hydroxy-benzoic acid methyl ester (5) remains in the form of colourless crystals with a purity of 99.6 % (HPLC).

Melting point 143.7 °C; DSC 144.7°C
[I]_D²⁰ = -26.6 (c = 0.93, ethanol).
¹³C-NMR (CDCl₃): 18.74, 19.21, 19.62, 33.12, 39.68, 42.36,
48.64, 51.42, 117.99, 120.32, 126.23, 127.81, 128.85, 129.39,
130.26, 132.21, 144.06, 162.43, 167.35.

4. Preparation of *R*-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenol (6)



a) Starting from the intermediate stage (4), R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)-phenyl] methanol

R-(+)-[4-benzyloxy-3-(3-diisopropylamino-1-phenyl-propyl)phenyl]-methanol (19.7 g, 45.7 mmol) are dissolved in 220 ml methanol and Raney nickel (5 g). The apparatus is rinsed with hydrogen gas and the deposit is agitated for two days at room -27-

temperature. Following the addition of a further 5 g Raney nickel, agitation for a further two days at room temperature takes place under a hydrogen gas atmosphere, followed by filtration off from the catalyser and concentration until dry on the rotary evaporator. The oily, pale yellow residue is dissolved in 100 ml diethylether, washed twice with 100 ml water each time, dried via sodium sulphate, filtered and concentrated until dry. 14.1 g (90.4% of theoretical) R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol are obtained in the form of a cream-coloured, amorphous solid. For recrystallisation see under c).

 b) Starting from the intermediate stage (5); R-(-)-3-(3diisopropylamino-phenyl-propyl)-4-hydroxy-benzoic acid methyl ester

A solution of 370 mg (1.0 mmol) R-(-)-3-(3-diisopropylaminophenyl-propyl)-4-hydroxy-benzoic acid methyl ester in 20 ml anhydrous tetrahydrofurane is slowly and at room temperature dropped into an agitated mixture of dried tetrahydrofurane (10 ml) and a 1M solution of lithium-aluminium hydride in tetrahydrofurane (3 ml) (under a nitrogen protective gas atmosphere). Excess hydride is decomposed by the dropped addition of a saturated sodium carbonate solution. Following separation of the organic phase this is concentrated on the rotary evaporator and then dried in the high-vacuum. 274 mg (74% of theoretical) pale yellow oil is obtained, that slowly solidifies into an amorphous mass.

c) Recrystallisation:

Raw product 6 (1.0 g) is dissolved in ethyl acetate and again concentrated on the rotary evaporator. The diol released in

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this way from foreign solvents (diethyl ether or tetrahydrofurane, see above) has 1.5 ml ethyl acetate added with slight heating. Agitation takes place until a clear solution results, followed by cooling at room temperature and addition of a few seed crystals. These are obtained by purifying raw 6 via HPLC, collecting the main fraction, concentrating this and drying the residue for a number of hours in the high-vacuum. Once clear crystallisation has definitely started, it is left at - 10°C. The crystals are sucked off in the cold and dried in the vacuum. Colourless crystals with a yield of 84% are obtained.

Melting point 102.3 °C

DC (1): 0.57

 $[I]_{D}^{20} = +21.3$ (c = 1.0, ethanol).

¹³C-NMR (CDCl₃): 19.58, 19.96, 33.30, 39.52, 42.10, 48.00, 65.40, 118.58, 126.31, 126.57, 127.16, 127.54, 128.57, 132.63, 132.83, 144.55, 155.52.

5. Preparation of R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenolisobutyrate ester (1)



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A solution of R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenol (6) (65.0 g, 190.3 mmol) and triethylamine (20.4 g, 201.7 mmol) in 750 ml dichloromethane has a solution of isobutyrate chloride (23.4 g, 201.7 mmol) in 250 ml dichloromethane added under agitation and cooling. Following addition agitation takes place for a further 15 minutes at 0°C, then for 30 minutes at room temperature and then one after another washing with water (250 ml) and 5% aqueous sodium hydrogen carbonate solution. The organic phase is separated and concentrated on the rotary evaporator until dry. The ester R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester is obtained as a colourless, viscous oil; yield: 77.1 g (98.4 % of theoretical).

DC (1): 0.26; $[I]_{D}^{22} = + 2.7$ (c = 1.0, ethanol).

¹³C-NMR (CDCl₃):19.01, 19.95, 20.59, 21.12, 34.28, 36.89, 41.88, 42.32, 43.90, 48.78, 64.68, 122.57, 125.59, 126.16, 126.86, 127.96, 128.54, 136.88, 138.82, 143.92, 147.90, 175.96.

6. Preparation of

R-(+)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrogen fumarate.



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A solution of 41.87 g (102 mmol) R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester in 90 ml 2-butanone has fumaric acid (11.81 g, 102 mmol) added while heating. Following dissolution of the acid, cyclohexane (20-30 ml) is slowly added under agitation until the onset of turbidity. The colourless, homogenous deposit is initially left for 18 hours at room temperature, and then for several hours at 0°C. The colourless crystals that have precipitated are sucked off, washed with a little cyclohexane/2-butanone (90:10, vol.-%) and dried in the vacuum at 30°C. 44.6 g (83.1 % of theoretical) hydrogen furate salt of R-(+)-2-(3diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester in the form of colourless flakes are obtained.

Melting point 98.8°C, a second crystallisation from the same solvent mixture provides a product with a melting point of 103°C.

 $[I]_{D}^{20} = +6.0$ (c = 1.0, ethanol).

Elementary analysis: Calculated for $C_{30}H_{41}NO_7$ (molecular weight 527.66) C 68.29 %, H 7.83 %, N 2.65 %, O 21.2 %; found C 68.29 %, H 7.90 %, N 2.72 %, O 21.0 %.

UV/VIS at Σ in nm (A ^{1 *}_{1 cm}): 191 (1306), 193 (1305), 200 (1143), 220 (456).

IR: 3380, 2978, 2939, 2878, 2692, 2514, 1756, 1702, 1680, 1618, 1496, 1468, 1226, 1040, 1019, 806,

¹H-NMR (CDCl₃): 1.198, 1.285, 1.287 (CH₃); 2.541 (CHC=O); 3.589 (NCH); 4.585 (<u>C</u>H₂OH); 6.832 (=CH, fumarate); 6.84-7.62 (aryl, = CH).

¹³C-NMR (CDCl₃): 17.79, 18.95, 19.16 (CH₃); 31.63 (CH<u>C</u>H₂); 34.09 (<u>C</u>H-C=O); 41.87 (CH<u>C</u>H₂); 45.83 (NCH₂); 54.29 (NCH); 63.78 (OCH₂); 122.23, 126.48, 126.77, 127.56, 140.46, 140.52, 142.35, 147.54 (Aryl CH); 135.54 (=CH, fumarate); 170.48 (C=O, fumarate); 175.62 (i-Pr-C=O).

MS in the direct inlet, m/z (%): 411 (1), 396 (9), 380 (1), 223 (2), 165 (2), 114 (100), 98 (4), 91 (3), 84 (3), 72 (10), 56 (7).

7. Preparation of R-(+)-2-(3-Diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate



A solution of R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester (8.54 g, 25.0 mmol) in 50 ml dichloromethane is slowly dropped at 0°C into an agitated solution of isobutyrate chloride (2.66 g, 25.0 mmol) in 100 ml dichloromethane. After an hour the cooling is -32-

removed and re-agitation takes place for an additional hour. Following the drawing off of the volatile components in the vacuum on the rotary evaporator a colourless, amorphous-solid foam remains. This residue is dissolved in acetone (17 ml), with 0.45 to 0.50 g water and diethyl ether is added (approx. 20 - 25 ml) until there is a definite onset of turbidity. Following brief treatment with ultrasound crystallisation starts spontaneously and under agitation a further 80 ml of diethyl ether are slowly added. The precipitated colourless crystals are sucked off and dried overnight in the vacuum via phosphorous pentoxide. 10.5 g (93.7 % of theoretical) colourless crystalline R-(+)-2-(3-diisopropylamino-1phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate with a purity of 97.0% (HPLC) are obtained.

Melting point 97.1 °C.

 $[I]_{D}^{20} = + 4.3$ (c = 1.03, ethanol)

¹³C-NMR (CDCl₃): 16.94, 17.35, 18.24, 18.40, 18.87, 19.05, 31.20, 33.99, 41.64, 45.41, 54.18, 54.42, 63.83, 122.25, 126.50, 126.70, 126.96, 127.34, 128.60, 133.80, 140.55, 142.17, 147.68, 175.79.

8. Phenolic monoester



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General work specification for the manufacture of phenolic monoesters

Into a solution of 120.3 mg $(0.352 \text{ mmol})R-(+)-2-(3-\text{diiso-propylamino-1-phenylpropyl)-4-hydroxyphenol (6) in 5 ml dichloromethane, under agitation at 0°C, a solution of acid chloride (0.352 mmol) in 2 ml dichloromethane is dropped. Then triethylamine-dichloromethane (49.1<math>\mu$ l/0.353 mmol-2 ml) is added. After 18 hours at room temperature the thin layer chromatography shows that conversion is complete. The deposit is washed successively with 5 ml water, aqueous 0.1N-hydrochloric acid, 5 ml 5% aqueous sodium-hydrogen carbonate solution, 5 ml water, dried via sodium sulphate and following filtration concentrated until dry. Then it is dried in the high-vacuum until constant weight.

The following compounds are, by way of example, manufactured using this method:

$R = CH_2CH(CH_3)_2$

<u>R-(+)-3-methylbutyric acid-2-(3-diisopropylamino-1-phenyl-</u> propyl)-4-hydroxymethylphenyl-ester

Colourless oil with 70% yield and >95% purity (NMR). ¹³C-NMR (CDCl₃): 20.45, 20.59, 22.54, 25.70, 36,74, 42.18, 43.27, 43.96, 48.90, 64.67, 122.66, 125.60, 126.20, 126.79, 127.95, 128.37, 136.83, 138.86, 143.83, 147.82, 171.37. DC (1): 0.76. -34-

$R = CH_2C(CH_3)_3$

<u>R-(+)-3.3-dimethylbutyric acid-2-(3-diisopropylamino-1-</u> phenyl-propyl)-4-hydroxymethylphenyl-ester, free base

Colourless oil with 69.7% yield and >95% purity (NMR). ¹³C-NMR (CDCl₃): 20.40, 20.53, 29.73, 30.99, 36.62, 42.17, 44.01, 47.60, 49.01, 64.65, 122.64, 125.60, 126.20, 126.80, 127.96, 128.36, 136.85, 138.90, 143.80, 147.82, 170.55. DC (1): 0.75.

 $R = (CH_3)_3C$

R-(+)-3-pivalic acid-2-(3-diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride.

Colourless crystals, melting point 165-6 °C. ¹³C-NMR (DMSO-d₆ =39.7 ppm): 16.52, 16.68, 17.98, 18.11, 26.87, 31.46, 41.71, 45.33, 53.89, 53.98, 62.65, 122.61, 122.97, 125.94, 126.09, 126.57, 126.75, 127.87, 128.58, 131.80, 134.94, 141.02, 142.69, 147.17, 155.32, 163.92, 176.21.

$\mathbf{R} = \mathbf{C} - \mathbf{C}_3 \mathbf{H}_5$

R-(+)-cyclopropane carboxylic acid-2-(3-diisopropylamino-1phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride.

Colourless, waxy substance.

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¹³C-NMR (DMSO-d₆ =39.7 ppm): 173.02, 172.49, 172.37, 153.10, 147.12, 142.72, 142.03, 140.78, 136.60, 134.79, 134.35, 129.55, 129.13, 128.80, 128.67, 127.87, 126.96, 126.74, 125.94, 125.84, 124.37, 123.71, 122.80, 62.64, 53.92, 45.34, 41.65, 31.44, 18.05, 16.66, 12.84, 9.58, 9.28, 8.49, 7.89.

$R = C - C_4 H_7$

R-(+)-cyclobutane carboxylic acid-2-(3-diisopropylamino-1phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride

Colourless, waxy substance.

¹³C-NMR (DMSO-d₆ =39.7 ppm): 173.53, 147.12, 142.81, 140.74, 134.77, 128.65, 127.81, 126.74, 125.99, 125.87, 122.75, 62.63, 53.92, 45.34, 41.42, 37.38, 31.54, 25.04, 24.92, 18.03, 16.68, 16.61.

$\mathbf{R} = \mathbf{C} - \mathbf{C}_5 \mathbf{H}_9$

R-(+)-cyclopentane carboxylic acid-2-(3-diisopropylamino-1phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride

Colourless, waxy substance.

¹³C-NMR (DMSO-d₆ =39.7 ppm): 174.80, 147.22, 142.86, 140.76, 134.72, 128.66, 127.80, 126.73, 126.04, 125.88, 122.71, 62.62, 53.94, 45.37, 43.24, 41.39, 31.54, 29.78, 29.59, 25.64, 25.59, 18.07, 16.64. $\mathbf{R} = \mathbf{C} - \mathbf{C}_6 \mathbf{H}_{11}$

R-(+)-cyclohexane carboxylic acid-2-(3-diisopropylamino-1phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride

Colourless, waxy substance.

 $^{13}C-NMR$ (DMSO-d₆ = 39.7 ppm):

174.08, 147.15, 142.85, 140.77, 134.78, 128.66, 127.77, 126.74, 126.06, 125.87, 122.69, 62.61, 53.91, 45.36, 42.26, 41.24, 31.53, 28.74, 28.62, 25.48, 25.04, 24.98, 18.05, 16.67, 16.60.

R = 4-(C₂H₅CO₂)-C₆H₄ R-(+)-4-ethylcarbonyloxy-benzoic acid-2-(3-diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride

Colourless crystals, melting point 195-8 °C.

¹H-NMR (DMSO-d₆): 9.87 (s, 1H can be substituted with D₂O, NH), 8.19-8.12 (m, 2H, Phenyl-H), 7.55 (d, J = 1.0 Hz, 1H, Phenyl-H3), 7.41-7.13 (m, 9H, Phenyl-H), 5.28 (br s, 1H can be substituted with D₂O, OH), 4.53 (s, 2H, CH₂), 4.23 (t, J = 7.6 Hz, 1H, CH), 3.61-3.50 (m, 2H, 2 × CH(CH₃)₂), 2.97-2.74 (m, 2H, CH₂), 2.67 (q, J = 7.4 Hz, 2H, CH₂), 2.56-2.43 (m, 2H, CH₂), 1.23-1.13 (m, 15H, 2 × CH(CH₃)₂, CH₃).

$R = 4 - (i - C_3 H_7 CO_2) - C_6 H_4$

<u>R-(+)-4-(isopropylcarbonyloxy)-benzoic_acid-2-(3-</u> <u>diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester</u> hydrochloride

Colourless crystals, melting point 202-4 °C. ¹H-NMR (DMSO-d₆): 9.73 (s, 1H can be substituted with D₂O, NH), 8.19-8.12 (m, 2H, Phenyl-H), 7.55 (d, J = 1.4 Hz, 1H, Phenyl-H3), 7.42-7.14 (m, 9H, Phenyl-H), 5.27 (br s, 1H can be substituted with D₂O, OH), 4.53 (s, 2H, CH₂), 4.23 (t, J = 7.5 Hz, 1H, CH), 3.61-3.50 (m, 2H, $2 \times CH(CH_3)_2$), 2.99-2.78 (m, 3H, CH₂, $CH(CH_3)_2$), 2.54-2.47 (m, 2H, CH₂), 1.29-1.13 (m, 18H, $3 \times CH(CH_3)_2$).

$R = 4 - (t - C_4 H_9 CO_2) - C_6 H_4$

R-(+)-4-(t-butylcarbonyloxy)-benzoic acid-2-(3diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenylester, free base.

Colourless oil. ¹H-NMR (DMSO-d₆): 8.19-8.12

¹H-NMR (DMSO-d₆): 8.19-8.12 (m, 2H, phenyl-H), 7.45-7.33 (m, 3H, phenyl-H), 7.25-7.09 (m, 7H, phenyl-H), 5.20 (t, J = 5.6 Hz, 1H, OH), 4.50 (d, J = 5.6 Hz, 2H, CH₂), 4.20 (t, J = 7.5 Hz, 1H, CH), 2.95-2.80 (m, 2H, $2 \times C\underline{H}(CH_3)_2$), 2.38-2.25 (m, 2H, CH₂), 2.09-2.03 (m, 2H, CH₂), 1.33 (s, 9H, (CH₃)₃), 0.82-0.76 (m, 12H, $2 \times CH(C\underline{H}_3)_2$).

Hydrochloride: colourless crystals, melting point 165-6 °C. ¹H-NMR (CDCl₃): 8.22-8.16 (m, 2H, phenyl-H), 8.02 (d, J = 1.8 Hz, 1H, phenyl-H), 7.27-7.02 (m, 9H, phenyl-H), 4.83-4.60
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('m', 2H, CH₂), 4.01-3.94 (m, 1H, CH), 3.66-3.54 (m, 2H), 3.18-2.80 (m, 3H), 2.53-2.44 (m, 1H) (2 × CH₂, 2 × C<u>H</u>(CH₃)₂), 1.43-1.25 (m, 21H, (CH₃)₃, 2 × CH(CH₃)₂).

$R = 4 - (c - C_3 H_5 CO_2) - C_6 H_4$

<u>R-(+)-4-(cyclopropylcarbonyloxy)-benzoic acid-2-(3-</u> <u>diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester</u> hydrochloride

Colourless crystals, melting point 208-213 °C.

¹H-NMR (DMSO-d₆): 9.04 (s, 1H can be substituted with D₂O, NH), 8.15-8.09 (m, 2H, phenyl-H), 7.53 ('d', 1H, phenyl-H3), 7.42-7.13 (m, 9H, phenyl-H), 5.25 (br s, 1H can be substituted with D₂O, OH), 4.52 (s, 2H, CH2), 4.23 (t, J = 7.5 Hz, 1H, CH), 3.62-3.53 (m, 2H, $2 \times CH(CH3)2$), 3.05-2.70 (m, 2H, CH2), 2.51-2.37 (m, 2H, CH2), 2.01-1.89 (m, 1H, cyclopropyl-CH), 1.20-1.05 (m, 16H, $2 \times CH(CH3)2$, $2 \times cyclopropyl-CH2$).

¹³C-NMR (DMSO-d₆ =39.7 ppm): 172.71, 163.93, 154.92, 147.16, 142.69, 141.03, 134.97, 131.76, 128.60, 127.86, 126.76, 126.56, 126.06, 125.94, 122.95, 122.65, 62.65, 54.00, 53.89, 45.33, 41.63, 31.49, 18.10, 17.98, 16.69, 16.51, 12.86, 9.52.

$R = 4 - (c - C_4 H_7 CO_2) - C_6 H_4$

<u>R-(+)-4-(cyclobutylcarbonyloxy)-benzoic acid-2-(3-</u> <u>diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester</u> hydrochloride

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Colourless crystals, melting point 201-6 °C. ¹H-NMR (DMSO-d₆): 9.50 (s, 1H can be substituted with D₂O, NH), 8.17-8.12 (m, 2H, phenyl-H), 7.54 (d, J = 1.4 Hz, 1H, phenyl-H3), 7.42-7.14 (m, 9H, phenyl-H), 5.25 (br s, 1H can be substituted with D₂O, OH), 4.52 (s, 2H, CH₂), 4.23 (t, J = 7.5 Hz, 1H, CH), 3.62-3.47 (m, 3H, cyclobutyl-CH), 2 × CH(CH₃)₂), 3.00-2.70 (m, 2H, CH₂), 2.51-2.26 (m, 6H, CH₂, 2 × cyclobutyl-CH₂), 2.10-1.85 (m, 2H, cyclobutyl-CH₂), 1.22-1.12 (m, 12H, 2 × CH(CH₃)₂).

 $R = 4 - (c - C_6 H_{11} CO_2) - C_6 H_4$

<u>R-(+)-4-(cyclohexylcarbonyloxy)-benzoic acid-2-(3-</u> diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenyl-ester hydrochloride

Colourless crystals, melting point 212-217 °C. ¹H-NMR (DMSO-d₆): 9.34 (s, 1H, can be substituted with D_2O , NH), 8.16-8.12 (m, 2H, phenyl-H), 7.54 (d, J = 1.4 Hz, 1H, phenyl-H3), 7.39-7.14 (m, 9H, Phenyl-H), 5.26 ('t', 1H, can be substituted with D_2O , OH), 4.53 (d, J = 4.2 Hz, 2H, CH₂), 4.22 (t, J = 7.5 Hz, 1H, CH), 3.62-3.48 (m, 2H, 2 × CH(CH₃)₂), 3.00-2.60 (m, 3H, cyclohexyl-CH), CH₂), 2.51-2.40 (m, 2H, CH₂), 2.07-1.98 (m, 2H, cyclohexyl-CH₂), 1.80-1.11 (m, 20H, 4 × cyclohexyl-CH₂), 2 × CH(CH₃)₂) -40-

9. Identical diesters



General work specification for the manufacture of identical diesters

Into a solution of 7.30 g (21.4 mmol)R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxyphenol (6) in 100 ml dichloromethane, under agitation at 0°C, a solution of acid chloride (49.2 mmol) in 50 ml dichloromethane is dropped. Then triethylamine-dichloromethane (6.86 ml/ 49.2 mmol-50 ml)is added. After 1-3 hours at room temperature the thin layer chromatography shows that conversion is complete. The deposit is washed successively with respectively 100 ml water, aqueous 0.1N-hydrochloric acid, 5 ml 5% aqueous sodium-hydrogen carbonate solution, 5 ml water, dried via sodium sulphate and following filtration concentrated until dry. Then it is dried in the high-vacuum until constant weight. The following compounds are, by way of example, manufactured using this method:

R = Methyl

<u>R-(-)-acetic acid-2-(3-diisopropylamino-1-phenyl-propyl)-4-</u> acetoxymethyl-phenyl-ester, free base

Pale yellow oil, purity (HPLC): 95.2%. ¹³C-NMR (CDCl₃): 20.36, 20.69, 20.94, 20.99, 36.41, 42.27, 43.69, 48.79, 65.89, 122.89, 126.28, 127.17, 127.92, 128.36, 133.69, 136.95, 143.61, 148.46, 168.97, 170.76. LC-MS: 425 (15%, M^{+.}), 410 (97%), 382 (4%), 308 (3%), 266 (7%), 223 (27%), 195 (13%), 165 (8%), 114 (100%). [α]_D²⁰ = -33.1 (c = 1, CH₃CN). DC (1): 0.79.

R = Cyclohexyl

<u>R-(+)-cyclohexane carboxylic acid-2-(3-diisopropylamino-1-</u> phenyl-propyl)-4-cyclohexylcarbonyloxymethyl-phenyl-ester

Pale yellow oil, purity (NMR): >95%. ¹³C-NMR (CDCl₃): 20.30, 25.17, 25.58, 25.73, 28.97, 29.12, 41.70, 43.15, 44.03, 48.64, 65.37, 122.67, 125.88, 126.24, 127.06, 127.31, 127.90, 128.37, 134,03, 136.85, 143.55, 148.33, 174.20, 175.72. DC (1): 0.96. -42-

R = Isopropyl

<u>R-(+)-isobutyrate-2-(3-diisopropylamino-1-phenyl-propyl)-4-</u> isobutyryloxymethyl-phenyl-ester

Free base: pale yellow oil, purity (HPLC): 95.6%.
¹³C-NMR (CDCl₃): 18.96, 19.08, 20.59, 33.98, 34.20, 36.86,
41.72, 43.72, 48.72, 65.58, 122.65, 126.19, 126.73, 127.91,
128.11, 128.36, 133.91, 136.96, 143.81, 148.41, 175.15,
176.77.
DC (1): 0.74.

Hydrogen fumarate salt: colourless syrup, 94.4% HPLC purity.
¹³C-NMR (CDCl₃): 17.89,
18.07, 18.94, 18.97, 19.07, 31.22, 33.93, 34.13, 41.78,
45.62, 53.93, 65.33, 122.93, 126.82, 127.45, 127.53, 127.91,
128.75, 134.74, 135.29, 135.42, 142.04, 148.44, 170.24,
175.71, 176.79.

 $R = 4 - (t - C_4 H_9 CO_2) - C_6 H_4$

R-4-(t-butylcarbonyloxy)-benzoic acid-2-(3-diisopropylamino-1-phenyl-propyl)-4-(4-t-butylcarbonyloxymethyl-benzoic acid)phenyl-ester hydrochloride

Colourless crystals, melting point 105-7 °C. ¹³C-NMR (DMSO-d₆): 16.49, 16.71, 17.97, 18.06, 26.84, 31.36, 38.45, 41.70, 45.24, 53.79, 53.96, 55.09, 66.11, 122.47, 122.62, 123.59, 126.42, 126.83, 127.21, 127.70, 127.88, 128.02, 128.62, 131.17, 131.86, 134.48, 135.64, 142.52, 148.35, 154.86, 155.39, 163.80, 165.09, 176.14, 176.19.

10. Mixed diesters



R` is not equal to R``

General work specification for the manufacture of mixed diesters

Into a solution of 5.30 mmol phenolic monoester of general formula A in 40 ml dichloromethane under agitation at 0°C a solution of acid chloride (5.83 mmol) in 15 ml dichloromethane is dropped. Then triethylaminedichloromethane (0.589g/ 5.82 mmol-15 ml) is added. After 18 hours at room temperature the thin layer chromatography shows that conversion is complete. The deposit is washed successively with respectively 50 ml water, aqueous 0.1Nhydrochloric acid, 5 ml 5% aqueous sodium-hydrogen carbonate solution, 5 ml water, dried via sodium sulphate and following filtration concentrated until dry. Then it is dried in the high-vacuum until constant weight. The following example is manufactured using this method:

R` = CH(CH₃)₂
R`` = CH₃
R-(+)-isobutyrate-2-(3-diisopropylamino-1-phenyl-propyl)-4acetoxymethyl-phenyl-ester

Colourless oil. DC (1): 0.56 ¹³C-NMR (CDCl₃): 19.12, 20.65, 21.05, 34.24, 37.02, 41.79, 43.79, 48.72, 65.98, 122.75, 125.98, 126.22, 127.94, 128.39, 128.84, 133.55, 137.04, 143.84, 148.58, 170.84, 175.18.

Hydrochloride: colourless crystals ¹³C-NMR (CDCl₃): 16.89, 17.04, 18.31, 18.92, 20.95, 31.49, 34.07, 41,64, 46.17, 54.55, 65.49, 122.91, 126.61, 126.93, 127.48, 127.83, 128.74, 134.50, 134.88, 141.61, 148.44, 170.67, 175.63. $[\alpha]_{D}^{20} = +14.6$ (c = 1, CHCl₃). -45-

CLAIMS

1. Compounds of general formula I



Formula I

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid.

2. Compounds in accordance with claim 1, characterised in that X⁻ in each case is an acid ester of hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)tartaric acid, citric acid, L-aspartic acid, L-(+)ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (Naectylglycine), phloretinic acid (3-(4-hydroxyphenyl)propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

 Compounds in accordance with claims 1 and 2, characterised in that they have general formula 2.



Formula 2

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl and X⁻ is the acid residue of a physiologically compatible inorganic or organic acid.

4 Compounds in accordance with claim 3, characterised in that X in each case is an acid ester of hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)- tartaric acid, citric acid, L-aspartic acid, L-(+)ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (Naectylglycine), phloretinic acid (3-(4-hydroxyphenyl)propionic acid), phthalic acid, methanesulfonic acid or orotic acid.

- 5. Compounds in accordance with claims 3 and 4, characterised in that they are R-(+)-2-(3-(diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrogen fumarate, R-(+)-2-(3-(diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester-hydrochloride hydrate
- 6. Compounds in accordance with claims 3 and 4, characterised in that R stands for cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, 4-(1-cyclopropylmethanoyloxy)-phenyl, 4-(1-cyclobutyl-methanoyloxy)phenyl, 4-(1-cyclohexyl-methanoyloxy)-phenyl or 4-(2,2dimethyl-propanoyloxy)-phenyl and X⁻ denotes chloride.
- 7. Compounds in accordance with claims 1 to 6 in the form of a bulk material.
- 8. Method for manufacturing compounds of general formula I



Formula I

in which R denotes $C_1\text{-}C_6\text{-}alkyl\,,\ C_3\text{-}C10\text{-}cycloalkyl\,,$ substituted or unsubstituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid, characterised in that

a compound of formula III a)



Formula III

is split with a hydrogenation agent to form a compound of Formula V



Formula V

whereupon

b) the compound of formula V so obtained is converted with a reducing agent, in order to give a compound of formula VI

HO OH

Formula VI

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which

 c) is converted with an acylation agent, in order to obtain a compound of formula A



Formula A

in which ${\tt R}$ has the significance stated above, which

X-

 d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula I

HO

Formula I

in which R denotes C_1 -C6-alkyl, C_3 -C10-cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

9. Method in accordance with claim 8, characterised in that for the manufacture of the compounds of general formula I hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)tartaric acid, D-(-)-tartaric acid, citric acid, Laspartic acid, L-(+)-ascorbic acid, D-(+)-glucuronic acid, 2-oxopropionic acid (pyruvic acid), furan-2carboxylic acid (mucic acid), benzoic acid, 4hydroxybenzoic acid, salicyclic acid, vanillic acid, 4hydroxycinammic acid, gallic acid, hippuric acid (Nbenzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

10. Method for manufacturing compounds of general formula 2



Formula 2

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in which R denotes C_1-C_6 -alkyl, C_3 -C10-cycloalkyl, substituted or unsubstituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid, characterised in that

a) a compound of the formula 3



is split with a hydrogenation agent to form a compound of formula 5



whereupon

 b) the compound of formula 5 so obtained is converted with a reducing agent, in order to give a compound of formula 6



Formula 6

which

 c) is converted with an acylation agent, in order to obtain a compound of formula 1



Formula 1

in which R has the significance stated above, which

 d) is converted with a physiologically compatible inorganic or organic acid to form a compound of formula 2



Formula 2

in which R denotes C_1 -C6-alkyl, C_3 -C10-cycloalkyl, unsubstituted or substituted phenyl and X- is the acid residue of a physiologically compatible inorganic or organic acid.

11. Method in accordance with claim 10, characterised in that for the manufacture of the compounds of general formula 2 hydrochloric acid, hydrobromic acid, phosphoric acid, sulphuric acid, nitric acid, acetic acid, propionic acid, palmitic acid, stearic acid, maleic acid, fumaric acid, oxalic acid, succinic acid, DL-malic acid, L-(-)-malic acid, D-(+)-malic acid, DL-tartaric acid, L-(+)-tartaric acid, D-(-)-tartaric acid, citric acid, L-aspartic acid, L-(+)-ascorbic acid, D-(+)-glucuronic acid, 2oxopropionic acid (pyruvic acid), furan-2-carboxylic acid (mucic acid), benzoic acid, 4-hydroxybenzoic acid, salicyclic acid, vanillic acid, 4-hydroxycinammic acid, gallic acid, hippuric acid (N-benzoyl-glycine), aceturic acid (N-aectylglycine), phloretinic acid (3-(4-55-

hydroxyphenyl)-propionic acid), phthalic acid, methanesulfonic acid or orotic acid are used.

- 12. Method in accordance with claims 8 to 11, characterised in that as the hydrogenation agent, Raney nickel/ H_2 in methanol is preferably used as the solvent.
- 13. Method in accordance with claims 8 to 11, characterised in that for the reducing agent NaBH₄/EtOH, preferably LiAlH₄/THF, is used.
- 14. Method in accordance with claims 8 to 11, characterised in that for the acylation agent isobutyrylchloride and for the base triethylamine are used.
- 15. Method in accordance with claims 10 to 14, characterised in that a compound of general formula 6 is converted with an equivalent isobutyryl chloride in the presence of triethylamine using one of the respective solvents ethylacetate, dichloromethane, tetrahydrofurane, acetonitrile or toluene regio- and chemoselectively into R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate.
- 16. Method in accordance with claims 10 to 15, characterised in that R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester and fumaric acid or

hydrochloric acid are converted with the formation of the respective salt.

17. Method in accordance with claims 10 to 13 for the manufacture of

R-(+)-2-(3-diisopropylamino-1-phenyl-propyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate, characterised in that the phenolic esterification of R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenol (6) is carried out without the addition of an external base, in that solutions of (6) are dropped into solutions of isobutyrate chloride, that contain at least 1 mole equivalent of water, in order to directly obtain a corresponding stable, hydrate-containing hydrochloride.

18. Compound of formula III



Formula III

19. Compound of formula V

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Formula V

20. Compound of formula VI



Formula VI

- 21. Use of a compound in accordance with claims 18 to 20 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds.
- 22. Use of a compound in accordance with claims 18 to 20 as an intermediate product in the manufacture of phenolic monoesters of general formula A



Formula A

in which R denotes C_1-C_6 -alkyl, C_3-C_{10} -cycloalkyl, substituted or unsubstituted phenyl.

23. Compound of formula 3



Formula 3

24. Compound of formula 5



Formula 5

25. Compound of formula 6



Formula 6

26. Compound of formula 7



- 27. Use of a compound in accordance with claims 23 to 26 as a highly pure, crystalline, stable intermediate product in the manufacture of pharmaceutically useful compounds.
- 28. Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of phenolic monoesters of general formula 1



Formula 1

- 29. Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of salts of phenolic monoesters of general formula 2, in which R has the same meaning as given in claim 3.
- 30. Use of a compound in accordance with claims 23 to 26 as an intermediate product in the manufacture of R-(+)-2-(3diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester hydrogen fumarate and R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4hydroxymethylphenylisobutyrate ester hydrochloride hydrate.

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ABSTRACT

The present invention concerns highly pure, crystalline, stable compounds of novel derivatives of 3,3diphenylpropylamines in the form of their salts, a method for the manufacture and highly pure, stable intermediate products.

The method is in particular characterised by regio- and chemoselectivity and high yield. Salts of phenolic monoesters of 3,3-diphenylpropylamines are provided, that are particularly well-suited for use in pharmaceutical formulations. Preferred compounds are R-(+)-2-(3diisopropylamino-1-phenyl-propyl)-4hydroxymethylphenylisobutyrate ester hydrogen fumarate and R-(+)-2-(3-diisopropylamino-1-phenylpropyl)-4-hydroxymethylphenylisobutyrate ester hydrochloride hydrate. Furthermore, stable, crystalline intermediate products that are essential for obtaining the abovementioned salts are provided. A preferred intermediate product is R-(-)-3-(3diisopropylamino-phenyl-propyl)-4-hydroxy-benzoic acid methyl ester