Paper No. 10 Entered: March 9, 2016

UNITED STATES PATENT AND TRADEMARK OFFICE

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BEFORE THE PATENT TRIAL AND APPEAL BOARD

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E. I. DU PONT DE NEMOURS AND COMPANY AND ARCHER-DANIELS-MIDLAND COMPANY, Petitioners,

v.

FURANIX TECHNOLOGIES B.V., Patent Owner.

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Case IPR2015-01838 Patent 8,865,921 B2

Before TONI R. SCHEINER, SHERIDAN K. SNEDDEN and CHRISTOPHER G. PAULRAJ, *Administrative Patent Judges*.

PAULRAJ, Administrative Patent Judge.

DECISION Institution of *Inter Partes* Review 37 C.F.R. § 42.108



### I. INTRODUCTION

E. I. du Pont de Nemours and Company and Archer-Daniels-Midland Company (collectively, "Petitioners") filed a Petition (Paper 1, "Pet."), requesting institution of an *inter partes* review of claims 1–10 of U.S. Patent No. 8,865,921 B2 (Ex. 1001, "the '921 patent"). Furanix Technologies B.V. ("Patent Owner") did not file a Preliminary Response. We have jurisdiction under 35 U.S.C. § 314, which provides that an *inter partes* review may not be instituted "unless . . . there is a reasonable likelihood that the petitioner would prevail with respect to at least 1 of the claims challenged in the petition."

Upon consideration of the Petition, and for the reasons explained below, we determine that Petitioners have shown that there is a reasonable likelihood that it would prevail with respect to at least one of the challenged claims. We thus institute an *inter partes* review of claims 1–5 and 7–9 of the '921 patent.

A. Related Proceedings.

The Petition does not identify any separate related matters under 42 C.F.R. § 42.8(b)(2). Pet. 1.

B. The '921 Patent (Ex. 1001)

The '921 patent issued on October 21, 2014, and claims priority to a provisional application filed on October 7, 2009. *See* Ex. 1001, Title Page. It names Cesar Muňoz De Diego, Matheus Adrianus Dam, and Gerardus Johannes Maria Gruter as the inventors. *Id*.

The '921 patent relates generally to methods for preparing 2, 5-furan dicarboxylic acid (FDCA), or a dialkyl ester of FDCA, by contacting 5-hydroxymethylfurfural (HMF), and/or derivatives thereof, with an oxygen-



containing gas in the presence of oxidation catalysts comprising cobalt (Co), manganese (Mn), and bromine (Br), and an acetic acid solvent at elevated temperatures. *Id.*, Abstract, 1:18–26, 2:39–45. The '921 patent states that "FDCA can be produced in particular from esters of HMF, such as for example 5-acetoxymethylfurfural (AMF) or a mixture of one or more of these compounds with HMF, such as for example from a mixture of AMF and HMF." *Id.* at 1:21–24. The '921 patent further discusses the use of FDCA obtained according to the process described therein to prepare a dialkyl ester of 2,5-dicarboxylic acid by the reaction of FDCA with a C<sub>1</sub>–C<sub>5</sub> alkyl alcohol. *Id.* at 5:20–41. The '921 patent acknowledges that the esterification of FDCA was known in the prior art. *Id.* at 5:42–58.

According to the '921 patent, FDCA has been identified as a priority chemical for establishing a "green" chemistry industry, but no commercial process exists for its production. *Id.* at 1:34–38. The specification states that FDCA, a furan derivative, is often synthesized in the laboratory from HMF obtained from carbohydrate containing sources such as glucose, fructose, sucrose, and starch. *Id.* at 1:30–43. The derivatives of HMF are known to be potential and versatile fuel components and precursors for the production of plastics. *Id.* at 1:44–46. The specification identifies prior art processes for the oxidation of HMF to FDCA with Co/Mn/Br catalysts at temperatures ranging from 50 to 125°C, which resulted in low reactivity or yield loss. *Id.* at 1:48–67, 2:1–35. The '921 patent seeks to improve prior art yields by controlling the temperature and/or pressure under which the oxidation reaction occurs. *Id.* at 4:34–61.

In particular, the '921 patent specification explains that "[t]he pressure in a commercial oxidation process may vary within wide ranges," and "is



determined by the solvent (e.g., acetic acid) pressure at a certain temperature." *Id.* at 4:34–39. Moreover, the pressure is preferably selected to maintain the solvent in the liquid phase, which "means that pressures between 5 and 100 bar can be used with a preference for pressures between 10 and 80 bar." *Id.* at 4:39–43. The oxidant can be an oxygen-containing gas, such as air, which "can be continuously fed to and removed from the reactor," in which case "the oxygen partial pressure will suitably be between 1 and 30 bar or more preferably between 1 and 10 bar." Id. at 4:43–46, 51– 55. Conversely, all of the oxygen-containing gas can be supplied at the start of the reaction, but this will require a significantly higher pressure. Id. at 4:45–51. The specification further explains that "[t]he temperature of the reaction mixture is at least 140° C., preferably from 140 and 200° C., most preferably between 160 and 190° C." Id. at 4:56–58. The specification notes that "[g]ood results" were achieved at about 180°C, but cautions that "[t]emperatures higher than 180° C may lead to decarboxylation and to other degradation products." *Id.* at 4:58–61.

The '921 patent includes working examples describing experiments in which the oxidation reaction was carried out with Co/Mn/Br catalysts at an air pressure ranging from 20–60 bars and temperatures ranging from 100 to 220°C. *Id.* at 6:8–11. More particularly, Example 1 describes the oxidation of HMF and/or AMF at 180°C for 1 hour with 20 bar air pressure, which resulted in FDCA yields of up to 76.66%. *Id.* at 6:34–46, Table 1. Example 2 provides a comparative example in which AMF oxidation was conducted at 100°C and 30 bar for 2 hours, showing that FDCA yields under those conditions were lower than the results obtained at higher temperature. *Id.* at 6:50–62, Table 2.



#### C. Illustrative Claims

Petitioners challenge claims 1–10 of the '921 patent. Independent claim 1 is illustrative, and reproduced below:

1. A method for the preparation of 2,5-furan dicarboxylic acid comprising the step of contacting a feed comprising a compound selected from the group consisting of 5-hydroxymethylfurfural ("HMF"), an ester of 5-hydroxymethylfurfural, 5-methylfurfural, 5-methylfurfural, 5-methylfuroic acid, 5-(chloromethyl)furoic acid, 2,5-dimethylfuran and a mixture of two or more of these compounds with an oxygen-containing gas, in the presence of an oxidation catalyst comprising both Co and Mn, and further a source of bromine, at a temperature between 140° C and 200° C at an oxygen partial pressure of 1 to 10 bar, wherein a solvent or solvent mixture comprising acetic acid or acetic acid and water mixtures is present.

Independent claim 7 is directed to the preparation of a dialkyl ester of FDCA, and additionally recites the step of "esterifying the thus obtained product."

## D. The Asserted Grounds of Unpatentability

Petitioners challenge the patentability of the claims of the '921 patent on the following grounds:

References	Basis	Claims challenged
The '732 publication, <sup>1</sup> RU '177, <sup>2</sup> and the '318 application <sup>3</sup>	§ 103(a)	1–5

<sup>&</sup>lt;sup>3</sup> Lilga et al., US 2008/0103318 A1, published May 1, 2008 (Ex. 1008).



<sup>&</sup>lt;sup>1</sup> Grushin et al., WO 01/72732, published Oct. 4, 2001 (Ex. 1002).

<sup>&</sup>lt;sup>2</sup> Slavinskaya et al., USSR Patent RU-448177A1, published Oct. 30, 1974 (Ex. 1007, with certified English translation).

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