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Prediction of Relative Retention Value of the Individual Molecular Species of Diacyl Glycerolipid on High Performance Liquid Chromatography

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Abstract

The relative retention value of the individual molecular species of acetyldiglyceride derived from phosphatidylcholine on reverse phase high performance liquid chromatography was considered to be dependant principally on the addition theorem of chemical potentials of the two fatty acid residues.

It was demonstrated that the chemical potential of each fatty acid residue is equivalent to the relative retention potential of each fatty acid residue; and that the addition of the relative retention potentials of the two fatty acid residues denoted the logarithm of the relative retention value, the relative retention time of each individual molecular species of acetyldiglyceride.

Martin formulated the formulae¹⁾:

$$\begin{aligned}\Delta\mu_B/R \cdot T &= \Delta\mu_A/R \cdot T + \Delta\mu_X/R \cdot T & [1] \\ \log(\alpha_B/\alpha_A) &= \Delta\mu_X/R \cdot T & [2]\end{aligned}$$

where A and B are members of a homologous series differing by the functional group X ; $\Delta\mu_X$ is the difference in chemical potential of the group X in the chromatographic system. R is the gas constant; and T is the absolute temperature. If we consider α_A as the partition coefficient of the standard molecular species and α_B as the partition coefficient of each molecular species, α_B/α_A will correspond to the relative retention value, the relative retention time (RRT) of each molecular species. So from formula [2],

$$\log(\alpha_B/\alpha_A) = \log RRT - \Delta\mu_X/R \cdot T$$

should hold. Under most of the chromatographic system, T is constant. Therefore, $1/R \cdot T$ will also be constant.

$$\therefore \log RRT \propto \Delta\mu_X \quad [3]$$

From the aforementioned aspect, formula [1] can be rewritten as follows:

$$\Delta\mu_B = \Delta\mu_A + \Delta\mu_X \quad [4]$$

This formula [4] can be expanded to cover

$$\Delta\mu_{DG} = \Delta\mu_{FA_1} + \Delta\mu_{FA_2} \quad [5]$$

where DG is the diacylglycerolipid and FA_1 and FA_2 are the fatty acid residues of DG . From formulae [3], [4] and [5],

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Table 1. Relative retention potential of the fatty acid residues of acetyldiglyceride on reverse phase high performance liquid chromatography.*

Fatty acid residue	Relative retention potential	Fatty acid residue	Relative retention potential
14:0	1.023	16:2	0.8670
15:0	1.102	17:2	0.9850
16:0	1.181	18:2 ω 6	0.9757
17:0	1.229	18:3 ω 3	0.8580
18:0	1.347	20:4 ω 6	0.9090
20:0	1.484	22:4	1.027
16:1 ω 9	0.9985	20:5 ω 3	0.7859
17:1	1.081	22:5	0.9261
18:1 ω 9	1.142	22:6 ω 3	0.8247
20:1	1.313		
24:1	1.638		

* Relative retention time of (16:0)(22:6) is regarded as 100. Equipment, Hitachi 638-50; Shodex RI monitor; column, LiChrosorb RP-18 (250×8 mm; tandem); solvent, isopropanol/acetone/methanol/acetonitrile (1:1:3:4, v/v); flow, 1.5 ml/min; column temperature, ambient.

formula [8], the relative retention potential of 16:0 can be calculated as $1.967 - 0.7859 = 1.181$. Incidentally, the relative retention potential of (16:0)(22:6) can be predicted as $1.181 + 0.8247 = 2.006$ and the actual relative retention potential of (16:0)(22:6) was 2.000 (See Table 2). All of the relative retention potentials of the fatty acid residues in Table 1 were calculated in the same manner. From the calculated retention potential of the fatty acid residues shown in Table 1, the relative retention potential of the individual molecular species of DG was theoretically predicted and it was compared with the empirically determined one as shown in Table 2. Though there are small errors between the theoretically determined retention potential and the empirically determined one, the validity of this idea is well demonstrated.

Discussion

As it is obvious from the theory of Martin¹⁾, formulae [1] and [2] should hold for all of the partition chromatographic systems. It then follows that formulae [6], [7], [8] and [9] should also hold for all of the partition chromatographic systems of DG lipid molecular species regardless of the analytical condition employed for the chromatography. By introducing the *RRT* data of the dinitrobenzoyl derivatives from PC presented by Takamura *et al.*⁵⁾ and Kito *et al.*⁶⁾, instead of the acetyl derivatives that have been discussed in this study, it was also demonstrated that the theoretically determined relative retention potential of this derivative coincides well with the empirically determined one as shown in Table 3. This shows that not only

Table 2. Comparison of the predicted and the empirically determined relative retention potential of the individual molecular species of acetyldiglyceride.*

Molecular species of acetyldiglyceride	Predicted relative retention potential	Empirically determined relative retention potential	Relative error (%)
(20:5)(20:5)	1.572	1.572	**
(20:5)(22:6)	1.611	1.613	0.12
(22:6)(22:6)	1.649	1.649	**
(18:3)(22:6)	1.683	1.688	0.30
(20:4)(20:5)	1.695	1.719	1.42
(22:5)(20:5)	1.712	1.712	**
(20:4)(22:6)	1.734	1.722	0.69
(22:5)(22:6)	1.751	1.746	0.29
(18:2)(20:5)	1.762	1.788	1.48
(16:1)(20:5)	1.784	1.784	**
(18:2)(22:6)	1.800	1.825	1.39
(14:0)(20:5)	1.809	1.809	**
(16:1)(22:6)	1.823	1.824	0.01
(18:2)(18:3)	1.834	1.848	0.76
(20:4)(22:5)	1.835	1.869	1.85
(14:0)(22:6)	1.848	1.843	0.27
(22:4)(22:6)	1.852	1.852	**
(22:5)(22:5)	1.852	1.894	2.27
(18:2)(20:4)	1.885	1.917	1.70
(15:0)(20:5)	1.888	1.891	0.16
(17:2)(20:4)	1.894	1.894	**
(18:2)(22:5)	1.906	1.900	0.05
(17:1)(22:6)	1.906	1.906	**
(16:1)(20:4)	1.908	1.918	0.52
(17:2)(22:5)	1.911	1.895	0.84
(16:1)(22:5)	1.925	1.900	1.30
(15:0)(22:6)	1.927	1.927	**
(18:1)(20:5)	1.928	1.945	0.88
(14:0)(22:5)	1.949	1.915	1.74
(18:2)(18:2)	1.951	1.981	1.54
(18:1)(22:6)	1.966	1.966	**
(16:0)(20:5)	1.967	1.967	**
(16:1)(18:2)	1.974	2.000	1.32
(18:1)(18:3)	1.999	1.999	**
(14:0)(18:2)	1.999	2.000	0.05
(16:0)(22:6)	2.006	2.000	0.29
(16:0)(18:3)	2.039	2.030	0.88
(16:0)(16:2)	2.048	2.048	**
(18:1)(20:4)	2.051	2.064	0.29
(17:0)(22:6)	2.053	2.067	0.68

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